

Crystallographic report

1,4-Bis[bis(trimethylsilyl)methyl-dichlorostannylmethyl-dimethylsilyl]benzene,
 p -{[(Me₃Si)₂CH]Sn(Cl)₂CH₂SiMe₂]₂C₆H₄Dainis Dakternieks¹, Bernhard Zobel¹ and Edward R. T. Tiekink^{2*}¹Centre for Chiral and Molecular Technologies, Deakin University, Geelong, Victoria 3217, Australia²Department of Chemistry, The University of Adelaide, South Australia 5005, Australia

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The dinuclear molecule of p -{[(Me₃Si)₂CH]Sn(Cl)₂CH₂SiMe₂]₂C₆H₄ is centrosymmetric and adopts an 'S' conformation that is stabilized by intramolecular C—H... π interactions. The tin atom exists within a distorted tetrahedron defined by a C₂Cl₂ donor set. Copyright © 2002 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; organotin; C—H... π interactions

COMMENT

The title compound, p -{[(Me₃Si)₂CH]Sn(Cl)₂CH₂SiMe₂]₂C₆H₄ (I), is a synthetic precursor for spacer-bridged double ladder molecules.¹ The crystal structure determination shows an 'S' conformation for the centrosymmetric molecule (Fig. 1). This is stabilized by the formation of two intramolecular C—H... π interactions² involving the C1—H and the central aromatic ring. Thus, the distance between the H and the ring centroid is 2.75 Å and the angle subtended at H is 161°. A C₂Cl₂ donor set is found for the tin atom with the widest angle of 124.29(10)° being subtended by the organic groups.

EXPERIMENTAL AND RESULTS

Compound I was prepared from 1,4-bis[bis(trimethylsilyl)methyl-diphenylstannylmethyl-dimethylsilyl]benzene (II) as described. Preparation of II: to a solution of LiCH(SiMe₃)₂³ in 23 ml Et₂O (c = 0.4 mol l⁻¹) was added at room temperature a solution of 1,4-bis(iododiphenylstannylmethyl-dimethylsilyl)benzene¹ (4.43 g, 4.35 mmol) in 40 ml Et₂O over 40 min. The reaction mixture was stirred overnight and hydrolysed with saturated NH₄Cl solution. The organic layer was washed twice with water and dried over Na₂SO₄. After filtration, the organic solvent was evaporated *in vacuo* and the resulting yellow oil was kept 1 h at 130°C and 10⁻³ Torr to

remove all volatile by-products. This gave 4.32 g crude product, which was used without further purification. ¹³C NMR (75.44 MHz, CDCl₃): δ CH -2.31, ¹J(^{117/119}Sn-¹³C) = 246/257, ¹J(²⁹Si-¹³C) = 50; SiMe₂ 0.07; CH₂ 1.61, ¹J(^{117/119}Sn-¹³C) = 170/179, ¹J(²⁹Si-¹³C) = 40; SiMe₃ 3.62 ³J(^{117/119}Sn-¹³C) = 16, ¹J(²⁹Si-¹³C) = 51; C_m 128.10; C_p 128.42; C_{qSi} 132.57; C_o 136.90, ²J(^{117/119}Sn-¹³C) = 36; C_{Si} 141.73; C_i 141.88, ¹J(^{117/119}Sn-¹³C) = 457/478. ²⁹Si NMR (59.6 MHz, CDCl₃): δ -2.01 (s, ²J(^{117/119}Sn-²⁹Si) = 22, SiMe₂); 1.57 (s, ²J(^{117/}

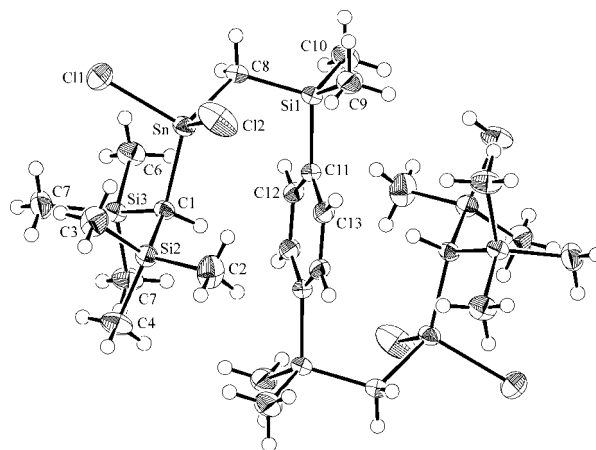


Figure 1. Molecular structure of I. Key geometric parameters: Sn—Cl1 2.3471(9), Sn—Cl2 2.3482(9), Sn1—C1 2.130(2), Sn—C8 2.119(3) Å, Cl1—Sn—Cl2 100.40(4), Cl1—Sn—C1 108.01(7), Cl1—Sn—C8 105.09(8), Cl2—Sn—C1 111.68(7), Cl2—Sn—C8 104.68(8), C1—Sn—C8 124.29(10)°.

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^{119}Sn - ^{29}Si) = 31, $^1J(^{13}\text{C}$ - ^{29}Si) = 51, SiMe_3) ^{119}Sn NMR (111.85 MHz, CDCl_3): δ - 55.2 $^1J(^{13}\text{C}$ - $^{119}\text{Sn})$ = 479, $^1J(^{13}\text{C}$ - $^{119}\text{Sn})$ = 256.

Preparation of **I**: to a magnetically stirred solution (0°C) of **II** (4.32 g, 3.99 mmol) in 50 ml acetone was added dropwise a solution of HgCl_2 (5.50 g, 20.26 mmol) in 50 ml acetone over 30 min. The reaction mixture was stirred for 7 days and the acetone removed *in vacuo*. Hexane (100 ml) was added and the suspension was refluxed for 10 min. The remaining PhHgCl was removed by hot filtration and the organic solvent evaporated *in vacuo* to give 3.74 g yellow oil. This was recrystallized from CHCl_3 (15 ml) to give 1.96 g (51%) of **I**, m.p. 133–136°C. ^1H NMR (299.98 MHz, CDCl_3): δ - 0.11 (s, 1H, $^2J(^{117/119}\text{Sn}$ - $^1\text{H})$ = 101/106, CH); 0.16 (s, 18H, SiMe_3); 0.51 (s, 6H, SiMe_2); 1.00 (s, 2H, $^2J(^{117/119}\text{Sn}$ - $^1\text{H})$ = 84/89, CH_2); 7.59 (s, 2H, C_6H_4). ^{13}C NMR (75.44 MHz, CDCl_3): δ SiMe_2 - 0.67 $^3J(^{117/119}\text{Sn}$ - $^{13}\text{C})$ = 20, $^1J(^{29}\text{Si}$ - $^{13}\text{C})$ = 55; SiMe_3 2.77 $^3J(^{117/119}\text{Sn}$ - $^{13}\text{C})$ = 27, $^1J(^{29}\text{Si}$ - $^{13}\text{C})$ = 52; CH 14.81, $^1J(^{117/119}\text{Sn}$ - $^{13}\text{C})$ = 314/329, $^1J(^{29}\text{Si}$ - $^{13}\text{C})$ = 46; CH_2 18.24, $^1J(^{117/119}\text{Sn}$ - $^{13}\text{C})$ = 204/213, $^1J(^{29}\text{Si}$ - $^{13}\text{C})$ = 35; C_o 133.21; C_i 140.66. ^{119}Sn NMR (111.85 MHz, CDCl_3): δ 123.7 ppm.

Intensity data for **I** were collected at 173 K on a Rigaku AFC7R

diffractometer for a colourless crystal $0.23 \times 0.23 \times 0.29 \text{ mm}^3$. $\text{C}_{26}\text{H}_{58}\text{Cl}_4\text{Si}_6\text{Sn}_2$, $M = 918.5$, monoclinic, $P2_1/c$, $a = 11.380(1)$, $b = 11.209(2)$, $c = 17.133(1) \text{ \AA}$, $\beta = 100.968(8)^\circ$, $V = 2145.6(4) \text{ \AA}^3$, $Z = 2$, 4927 unique data ($\theta_{\text{max}} 27.5^\circ$), 3786 data with $I \geq 2\sigma(I)$, $R(\text{obs.}) = 0.026$, $wR(\text{all data}) = 0.072$, $\rho_{\text{max}} = 0.38 \text{ e}^- \text{ \AA}^{-3}$. Programs used: *teXsan*, *DIRDIF*, *SHELXL*, *PLATON*, and *ORTEP*. CCDC deposition number: 186483.

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