

Crystallographic report

Bis(O-methoxyethyldithiocarbonato)(4,7-dimethyl-1,10-phenanthroline)cadmium(II)

Danlin Chen, Chian Sing Lai and Edward R. T. Tiekink*

Department of Chemistry, National University of Singapore, Singapore 117543, Singapore

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The structure of mononuclear $\text{Cd}(\text{S}_2\text{COCH}_2\text{CH}_2\text{OCH}_3)_2(4,7\text{-Me}_2\text{phen})$ shows an N_2S_4 donor set about cadmium that defines a coordination geometry intermediate between octahedral and trigonal prismatic. Copyright © 2003 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; cadmium; dithiocarbonate; organometallic; imine

COMMENT

Two distinct structural motifs are known for the cadmium xanthates.¹ In one motif, adopted by $\text{R} = \text{Et}$,² $\text{R} = i\text{-Pr}$,^{3–5} and $\text{R} = n\text{-Bu}$,⁶ two-dimensional sheets, comprising interconnected 16-membered rings, are formed as a result of all xanthates being bridging. This results in four-coordinate, distorted tetrahedral cadmium centres. In the other motif, adopted by $\text{R} = \text{Me}$ ⁷ and $\text{R} = \text{CH}_2\text{CH}_2\text{OCH}_3$,⁸ square-planar cadmium geometries are found as a result of the presence of chelating xanthate ligands; mononuclear units aggregate into loosely associated chains facilitated by supramolecular $\text{Cd} \cdots \text{S}$ interactions. Despite the different structures found for the parent compounds, adduct formation leads to similar mononuclear species as seen in the title compound and related species.^{9,10} The cadmium atom in $\text{Cd}(\text{S}_2\text{COCH}_2\text{CH}_2\text{OCH}_3)_2(4,7\text{-Me}_2\text{phen})$ exists within an N_2S_4 donor set with a coordination geometry intermediate between octahedral and trigonal prismatic (Fig. 1).

EXPERIMENTAL

To a stirred chloroform–acetonitrile (50 ml) solution of $\text{Cd}(\text{S}_2\text{COCH}_2\text{CH}_2\text{OCH}_3)_2$ (0.2 g)⁸ was added a stoichiometric amount of 4,7-Me₂phen (Aldrich). The mixture was refluxed for 2 h, filtered and the solvent removed *in vacuo*. The precipitate was recrystallized by the slow evaporation of a chloroform solution of the compound to yield colourless crystals suitable for X-ray analysis; m.p. 137–138 °C.

*Correspondence to: Edward R. T. Tiekink, Department of Chemistry, National University of Singapore, Singapore 117543, Singapore. E-mail: chmtert@nus.edu.sg

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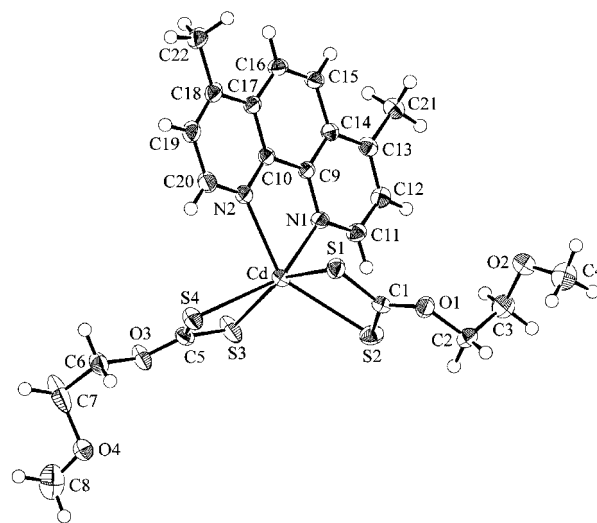


Figure 1. Molecular structure of $\text{Cd}(\text{S}_2\text{COCH}_2\text{CH}_2\text{OCH}_3)_2(4,7\text{-Me}_2\text{phen})$. Key geometric parameters: $\text{Cd}-\text{S}1$ 2.6028(6), $\text{Cd}-\text{S}2$ 2.7960(7), $\text{Cd}-\text{S}3$ 2.6401(7), $\text{Cd}-\text{S}4$ 2.7018(6), $\text{Cd}-\text{N}1$ 2.3394(18), $\text{Cd}-\text{N}2$ 2.4064(18) Å; $\text{S}1-\text{Cd}-\text{S}2$ 66.880(18), $\text{S}1-\text{Cd}-\text{S}3$ 154.79(2), $\text{S}1-\text{Cd}-\text{S}4$ 102.750(19), $\text{S}1-\text{Cd}-\text{N}1$ 102.53(4), $\text{S}1-\text{Cd}-\text{N}2$ 92.59(4), $\text{S}2-\text{Cd}-\text{S}3$ 97.41(2), $\text{S}2-\text{Cd}-\text{S}4$ 121.06(2), $\text{S}2-\text{Cd}-\text{N}1$ 88.02(5), $\text{S}2-\text{Cd}-\text{N}2$ 145.96(4), $\text{S}3-\text{Cd}-\text{S}4$ 67.517(19), $\text{S}3-\text{Cd}-\text{N}1$ 96.20(4), $\text{S}3-\text{Cd}-\text{N}2$ 109.84(5), $\text{S}4-\text{Cd}-\text{N}1$ 147.29(4), $\text{S}4-\text{Cd}-\text{N}2$ 88.89(4), $\text{N}1-\text{Cd}-\text{N}2$ 69.55(6)°.

IR (KBr): $\nu(\text{C}-\text{S})$ 1058 and $\nu(\text{C}-\text{O})$ 1180 cm^{-1} . Intensity data were collected at 183 K on a Bruker AXS SMART CCD diffractometer for a block $0.31 \times 0.39 \times 0.52 \text{ mm}^3$. $\text{C}_{22}\text{H}_{26}\text{CdN}_2\text{O}_4\text{S}_4$, $M = 623.1$, monoclinic, $P2_1/n$, $a = 8.9770(8)$, $b = 23.381(2)$, $c = 11.7006(10)$ Å,

$\beta = 91.519(2)^\circ$, $V = 2455.0(4) \text{ \AA}^3$, $Z = 4$, 7157 unique data ($\theta_{\text{max}} 30.1^\circ$), $R = 0.043$ (all data), $wR = 0.092$ (all data), $\rho_{\text{max}} = 1.41 \text{ e}^- \text{ \AA}^{-3}$ (near cadmium). Two conformations were detected for the O4 atom, in the ratio 0.55:0.45 (from refinement). Programs used: teXsan, DIRDIF, SHELXL, and ORTEP. CCDC deposition number: 199840.

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