

## Crystallographic report

**Bis[bis(*N,N*-dibenzylthiocarbamato)cadmium(II)]**Xia Yin<sup>1</sup>, Weiguang Zhang<sup>1\*</sup>, Qijiao Zhang<sup>1</sup>, Jun Fan<sup>1</sup>, Chian Sing Lai<sup>2</sup> and Edward R. T. Tiekink<sup>2\*\*</sup><sup>1</sup>Department of Chemistry, South China Normal University, Guangzhou 510631, People's Republic of China<sup>2</sup>Department of Chemistry, National University of Singapore, Singapore 117543

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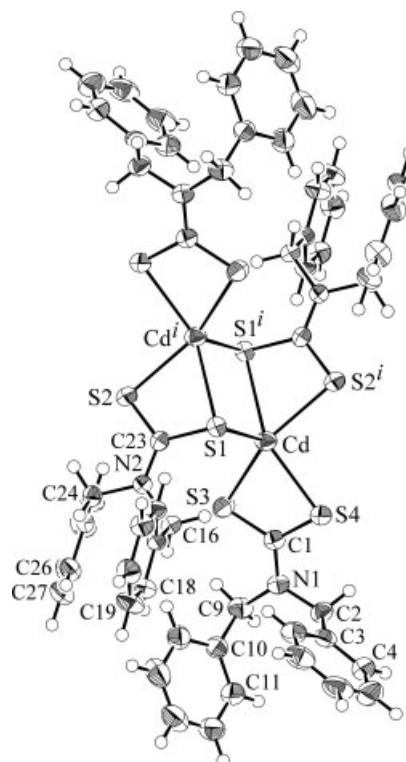
The centrosymmetric structure of  $\{Cd[S_2CN(CH_2Ph)_2]_2\}_2$  features both bridging and chelating dithiocarbamate ligands so that a square pyramidal  $S_5$  coordination geometry results. Copyright © 2004 John Wiley & Sons, Ltd.

**KEYWORDS:** crystal structure; cadmium; dithiocarbamate**COMMENT**

The centrosymmetric dinuclear structure of  $\{Cd[S_2CN(CH_2Ph)_2]_2\}_2$ , Fig. 1, features two tridentate dithiocarbamate ligands that  $\mu_2$ -bridge via one sulfur atom each to two cadmium atoms leading to the formation of a  $Cd_2S_2$  core; the other ligands are chelating. This arrangement leads to five-coordinate cadmium centres that are best described as existing in square pyramidal geometries. The overall structure conforms to the predominant motif found for the other binary cadmium dithiocarbamates that have been characterized crystallographically;<sup>1</sup> the exception is that of  $[Cd(S_2CNnBu_2)_2]_2$ , which has two fold symmetry, so that the chelating dithiocarbamate ligands lie to one side of the molecule.<sup>2</sup>

**EXPERIMENTAL AND RESULTS**

A solution of  $CdCl_2$  (0.5 mmol) in water (10 ml) was added to a solution of sodium dibenzylthiocarbamate (1 mmol), prepared by standard methods, in ethanol (20 ml). The mixture was stirred for 2 h at room temperature and the precipitated compound was filtered off, washed with ethanol then with diethyl ether and dried in



**Figure 1.** Molecular structure of  $\{Cd[S_2CN(CH_2Ph)_2]_2\}_2$ . Key geometric parameters: Cd–S1 2.6200(6), Cd–S1<sup>i</sup> 2.7875(7), Cd–S2<sup>i</sup> 2.5834(7), Cd–S3 2.5491(7), Cd–S4 2.6030(7) Å; S1–Cd–S1<sup>i</sup> 92.461(19), S1–Cd–S2<sup>i</sup> 101.17(2), S1–Cd–S3 106.64(2), S1–Cd–S4 118.20(2), S1<sup>i</sup>–Cd–S2<sup>i</sup> 67.060(19), S1<sup>i</sup>–Cd–S3 112.60(2), S1<sup>i</sup>–Cd–S4 147.53(2), S2<sup>i</sup>–Cd–S3 152.15(2), S2<sup>i</sup>–Cd–S4 94.88(2), S3–Cd–S4 70.70(2)°. Symmetry operation *i*:  $-1 - x, -y, 1 - z$ .

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vacuum. Colourless crystals were obtained from the slow evaporation of an acetonitrile solution of the compound; m.p. 195 °C. IR (KBr):  $\nu(\text{C-S})$  972 and  $\nu(\text{C-N})$  1454, 1480, 1495  $\text{cm}^{-1}$ .  $^1\text{H}$ NMR (300.13 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 5.16 [s, 2H,  $\text{CH}_2$ ], 7.21–7.46 ppm [complex pattern, 5H, aromatic]. Intensity data were collected at 223 K on a Bruker AXS SMART CCD for a colourless block  $0.15 \times 0.34 \times 0.34 \text{ mm}^3$ .  $\text{C}_{30}\text{H}_{28}\text{CdN}_2\text{S}_4$ ,  $M = 657.2$ , monoclinic,  $P2_1/n$ ,  $a = 11.1098(4)$ ,  $b = 15.6325(5)$ ,  $c = 16.6695(6)$  Å,  $\beta = 97.9220(10)^\circ$ ,  $V = 2867.43(17)$  Å<sup>3</sup>,  $Z = 4$ , 8313 unique data ( $\theta_{\text{max}}$  30.0°), 6404 data with  $I \geq 2\sigma(I)$ ,  $R = 0.044$  (obs. data),  $wR = 0.099$  (all data). Programs used: teXsan, DIRDIF, SHELXL-97 and ORTEP. CCDC deposition number: 213433.

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## REFERENCES

1. Tiekink ERT. *Cryst. Eng. Commun.* 2003; **5**: 101.
2. Casas JS, Sánchez A, Bravo J, García-Fontán S, Castellano EE, Jones MM. *Inorg. Chim. Acta* 1989; **158**: 119.