

Crystallographic report

**Bis[bis(*N*-methyl-*N*-phenyldithiocarbamato)-
(methylxanthato)bismuth(III)]**

Han-Dong Yin*, Chuan-Hua Wang and Yong Wang

Department of Chemistry, Liaocheng University, Liaocheng 252059, People's Republic of China

Received 6 December 2003; Revised 22 December 2003; Accepted 23 December 2003

The centrosymmetric structure of $\{\text{Bi}[\text{S}_2\text{CN}(\text{Me})\text{Ph}]_2(\text{S}_2\text{COMe})_2\}$ features chelating dithiocarbamate and xanthate ligands, as well as intermolecular $\text{Bi} \cdots \text{S}$ interactions, so that a distorted pentagonal bipyramidal S_7 coordination geometry results. Copyright © 2004 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; bismuth; dithiocarbamate; xanthate

COMMENT

The structural chemistry of bismuth dithiocarbamates and bismuth xanthates is both rich and diverse, so that monomeric,¹ dimeric,^{2–6} and linear polymeric species are known.^{7–9} In the title structure (Fig. 1), a mixed dimeric structure containing both dithiocarbamate and xanthate ligands is found. The bismuth atoms are seven-coordinated distorted S_7 pentagonal bipyramidal geometry owing to one of the dithiocarbamate ligands chelating one bismuth centre and via the S3 atom simultaneously bridging a centrosymmetrically related bismuth atom.

EXPERIMENTAL

An aqueous solution of $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ (1.0 mmol) and mannite (1.0 mmol) was added to an aqueous solution of sodium *N*-methyl-*N*-phenyldithiocarbamate (2.0 mmol) and sodium methylxanthate (1.0 mmol) and stirred for 2.0 h at 50 °C. A yellow solid was obtained by filtration. The product was recrystallized from acetonitrile to give yellow crystals, m.p. 274 °C (dec.). IR (KBr) ν : 3045, 2933, 2851, 1468, 1372, 1256, 1153, 1001, 954, 553, 464 cm^{-1} . Intensity data were collected at 298 K on a Bruker Smart 1000 CCD for a block $0.10 \times 0.10 \times 0.30 \text{ mm}^3$. $\text{C}_{18}\text{H}_{19}\text{BiN}_2\text{OS}_6$, $M = 680.69$, triclinic, $P\bar{1}$, $a = 10.272(2)$, $b = 10.523(2)$, $c = 12.700(3) \text{ \AA}$, $\alpha = 99.821(3)^\circ$, $\beta = 92.494(3)^\circ$, $\gamma = 116.729(3)^\circ$, $V = 1196.8(4) \text{ \AA}^3$, $Z = 2$, 4393 unique data ($\theta_{\text{max}} = 26.4^\circ$), $R_1 = 0.035$ (3262 data with $I > 2\sigma(I)$), $wR_2 = 0.052$

*Correspondence to: Han-Dong Yin, Department of Chemistry, Liaocheng University, Liaocheng 252059, People's Republic of China. E-mail: handongyin@lctu.edu.cn

Contract/grant sponsor: National Natural Foundation; Contract/grant number: 20271025.

Contract/grant sponsor: Natural Foundation of Shandong Province; Contract/grant number: L2003B01.

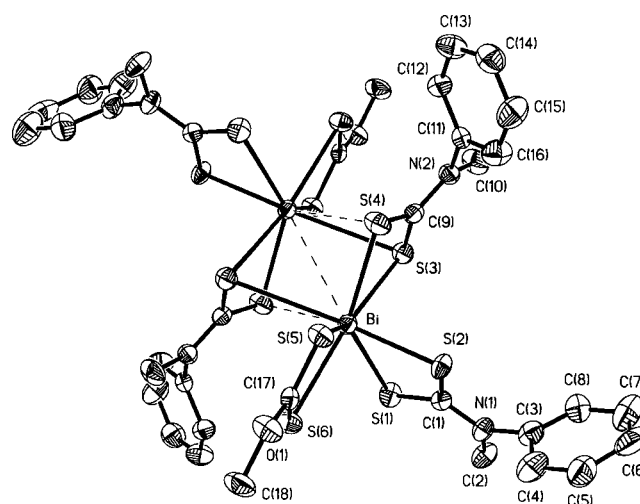


Figure 1. The molecular structure of $\{\text{Bi}[\text{S}_2\text{CN}(\text{Me})\text{Ph}]_2(\text{S}_2\text{COMe})_2\}$. Key geometric parameters: Bi–S1 2.9193(16), Bi1–S2 2.6122(14), Bi–S3 2.9392(18), Bi–S4 2.7925(17), Bi–S5 2.7470(15), Bi–S6 2.8984(19), Bi–S3' 3.2490(16) Å; S1–Bi–S2, 64.55(4), S1–Bi–S3 79.49(5), S1–Bi–S4 135.84(5), S1–Bi–S5 132.40(5), S1–Bi–S6 80.44(5), S2–Bi–S3 87.90(5), S2–Bi–S4 91.29(5), S2–Bi–S5 85.50(5), S2–Bi–S6 92.19(5), S3–Bi–S4 62.35(4), S3–Bi–S5 138.29(4), S3–Bi–S6 157.70(4), S4–Bi–S5 76.67(5), S4–Bi–S6 139.91(4), S5–Bi–S6 63.84(5), S1–Bi–S3' 124.83(4), S2–Bi–S3' 166.73(4), S3–Bi–S3' 102.66(4), S4–Bi–S3' 86.65(5), S5–Bi–S3' 81.27(4), S6–Bi–S3' 81.10(5)°. Symmetry code i : $-x + 1, -y + 2, -z + 1$.

(all data); $\rho_{\max} = 1.08 \text{ e}^- \text{ \AA}^{-3}$. Programs used: SHELXL and ORTEP. CCDC deposition number: 179929.

Acknowledgements

The National Natural Foundation People's Republic of China (20271025) and the National Natural Foundation of Shandong Province are thanked for support.

REFERENCES

1. Raston CL, Rowbottom GL, White AH. *J. Chem. Soc. Dalton Trans.* 1981; 1379.
2. Yin HD, Wang CH, Xing QJ. *Chin. J. Inorg. Chem.* 2003; **19**: 955.
3. Venkatachalam V, Ramalingam K, Casellato U, Graziani R. *Polyhedron* 1997; **16**: 1211.
4. Snow MR, Tiekink ERT. *Aust. J. Chem.* 1987; **40**: 743.
5. Tiekink ERT. *J. Cryst. Spectrosc. Res.* 1992; **22**: 231.
6. Tiekink ERT. *Main Group Met. Chem.* 1994; **17**: 727.
7. McKie G, Raston CL, Rowbottom GL, White AH. *J. Chem. Soc. Dalton Trans.* 1981; 1360.
8. Hoskins BF, Tiekink ERT, Winter G. *Inorg. Chim. Acta* 1984; **99**: 177.
9. Koh YW, Lai CS, Du AY, Tiekink ERT, Loh KP. *Chem. Mater.* 2003; **15**: 4544.