

Crystallographic report

Bis[(nitro)bis(dithiotetrahydropyrrolocarbamato)]
bismuth(III)]

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The loosely associated centrosymmetric structure of $[\text{Bi}(\text{S}_2\text{CNC}_4\text{H}_8)_2(\text{NO}_3)]_2$ features chelating dithiocarbamate and nitrate ligands, as well as weak intermolecular Bi–S interactions, so that a distorted pentagonal bipyramidal S_5O_2 coordination geometry results. Copyright © 2005 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; bismuth; dithiocarbamate; nitrate

COMMENT

The structural chemistry of bismuth 1,1-dithiolates is both rich and diverse, so that monomeric,¹ dimeric,² and linear polymeric species are known.³ In the title structure (Fig. 1), a

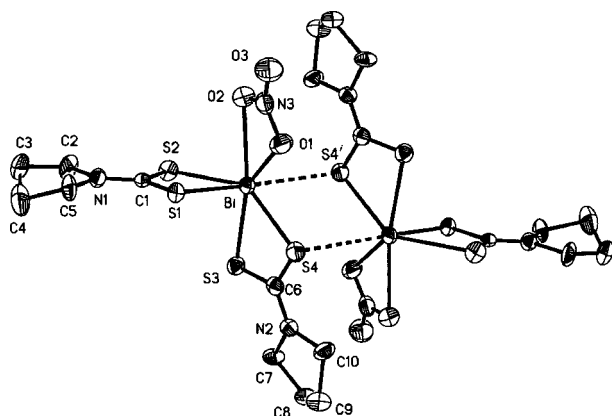


Figure 1. The molecular structure of $[\text{Bi}(\text{S}_2\text{CNC}_4\text{H}_8)_2(\text{NO}_3)]_2$; hydrogen atoms are omitted for clarity. Key geometric parameters: Bi–S1 2.5727(15), Bi–S2 2.7786(17), Bi–S3 2.6656(16), Bi–S4 2.7727(17), Bi–O1 2.630(5), Bi–O2 2.738(4), Bi–S4ⁱ 3.4320(18) Å. Symmetry operation *i*: 1 – *x*, –*y*, 1 – *z*.

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mixed-ligand loosely associated dimeric structure containing both dithiocarbamate and nitrate ligands is found. The bismuth atom is in a seven-coordinated, distorted S_5O_2 pentagonal bipyramidal geometry owing to one of the dithiocarbamate ligands chelating one bismuth center and, via the S4 atom, simultaneously bridging a centrosymmetrically related bismuth atom.

EXPERIMENTAL

$[\text{Bi}(\text{NO}_3)(\text{S}_2\text{CNC}_4\text{H}_8)_2]_2$ was obtained from the 1 : 2 reaction between $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ and the sodium salt of the ligand as per the literature method.² Crystals were isolated from the slow evaporation of a CH_3CN solution of the compound; m.p. 335 °C (dec.). IR (KBr) ν : 1631, 1500, 1443, 1385, 1148, 1021, 993, 693, 553, 450 cm^{-1} . Data were collected at 299 K on a Bruker Smart 1000 CCD for a block $0.10 \times 0.10 \times 0.20 \text{ mm}^3$. $\text{C}_{20}\text{H}_{32}\text{Bi}_2\text{N}_6\text{O}_6\text{S}_8$, $M = 1126.96$, monoclinic, $P2_1/n$, $a = 6.4718(13)$, $b = 26.738(6)$, $c = 10.423(2)$ Å, $\beta = 106.384(3)^\circ$, $V = 1730.3(6)$ Å³, $Z = 2$, $R = 0.029$ (2040 data with $I \geq 2\sigma(I)$; $\theta_{\text{max}} 25.0^\circ$), $wR = 0.045$ (all 3047 data). Programs used: SHELXL and ORTEP. CCDC deposition number: 179 925.

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REFERENCES

1. Yin HD, Wang CH. *Appl. Organometal. Chem.* 2004; **18**: 195.
2. Yin HD, Wang CH, Xing QJ. *Chin. J. Inorg. Chem.* 2003; **19**: 955.
3. Koh YW, Lai CS, Du AY, Tiekink ERT, Loh KP. *Chem. Mater.* 2003; **15**: 4544.