

Articles

Improved Non-chromatographic Purification of a Recombinant Protein by Cationic Elastin-like Polypeptides

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This paper reports an improvement in the purification of thioredoxin (Trx) expressed from *E. coli* by inverse transition cycling (ITC) using cationic elastin-like polypeptides (ELPs). Two ELP libraries having 2% and 5% lysine residues and molecular weights ranging from 4 to 61.1 kDa showed greater salt sensitivity in their inverse transition behavior than purely aliphatic ELPs. Expression yield of Trx-ELP fusions was an unpredictable function of guest residue composition, but reducing the molecular weight of the ELP tag generally increased Trx yield. A cationic 4.3 kDa ELP is the shortest ELP used to purify any protein by ITC to date. A 15.9 kDa ELP with a guest residue composition of K:V:F of 1:7:1 was found to be the optimal cationic tag to purify Trx, as it provided 50% greater Trx yield and only required one-fifth the added NaCl for purification of Trx as compared to previously used aliphatic ELP tags.

Introduction

A variety of affinity-based chromatographic purification schemes have been developed to simplify protein purification at the laboratory scale in a serial format.¹ By expressing recombinant proteins fused to a peptide or protein having moderate affinity and high specificity for a particular ligand, recombinant proteins can be purified in a single step by binding to resins to which the ligand has been immobilized.² In practice, however, affinity chromatography is costly, requiring specialized equipment and customized resins, and scale-up to industrial scale is empirical, expensive, and often results in significant purification losses.

We have previously developed inverse transition cycling (ITC) of elastin-like polypeptide (ELP) fusion proteins as an alternative to affinity-based purification schemes.^{3,4} ELPs are artificial biopolymers comprised of the pentapeptide repeat motif, Val-Pro-Gly-Xaa-Gly (VPGXG), which is derived from the hydrophobic domain of tropoelastin. At low temperatures, ELPs are soluble in aqueous solution, but as the solution temperature is raised, they become insoluble and aggregate at a critical temperature, termed the inverse transition temperature (T_i).^{5–7} This process is reversible so that subsequent cooling of the solution below the T_i results in the resolubilization of the ELP. The inverse transition can also be isothermally triggered by the addition of salt, and the response of ELPs to different salts follows the Hofmeister series.^{5,8–10}

We were the first group to demonstrate that the inverse transition behavior of ELPs was imparted to ELP fusion proteins,³ and we exploited this finding to develop a protein purification process that we have termed inverse transition cycling (ITC).^{3,4,11–15} This approach for protein purification has

subsequently been validated by other groups.^{16–24} The ITC method is inexpensive, as it requires no specialized equipment or resins; instead, it uses inexpensive reagents such as sodium chloride to trigger the inverse phase transition and readily available centrifuges to separate the ELP fusion protein from other cellular contaminants. Multiple rounds of ITC increase the purity of the ELP fusion protein, and target proteins can be liberated from their ELP purification tag either by proteolytic cleavage at engineered polypeptide sequences^{3,4,14,15} or by self-cleaving inteins.^{24,25}

Although previous studies have shown that ITC purification of ELP fusion proteins is a promising alternative to chromatography for purification of recombinant proteins, the ELP tags most commonly used for ITC have not been optimized for maximal protein expression or purification efficiency. To date, nearly all published examples of ITC purification of ELP fusion proteins have only utilized two types of ELP tags that were introduced in our original articles on ITC.^{3,4} These tags were not optimal, because they required at least ~2 M NaCl to induce the phase transition. This concentration of salt was necessary to trigger the inverse phase transition of ELP fusion proteins because these ELP tags had aliphatic guest residues that are only modestly sensitive to changes in salt concentration.^{3,4,10} Previous studies have shown that incorporation of ionizable guest residues in an ELP increases its sensitivity to salt.^{5,26} Thus, we hypothesized that the addition of ionizable groups in the ELP tag should allow the purification of ELP fusion proteins from *E. coli* soluble lysate with less added NaCl than is needed to purify ELP fusions containing only aliphatic guest residues.

In this study, we report on the phase transition behavior of charged ELPs and their thioredoxin (Trx) fusions. These cationic ELPs have guest residues composed of Lys (K), Val (V), and Phe (F) in 1:2:1 and 1:7:1 ratios and span a range of molecular weights (MWs) from 7.7 to 61.1 kDa. We chose ionizable Lys residues to enhance the salt sensitivity of these ELPs and

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particulate contaminants. This aggregation and resolubilization process was repeated 3–4 times until the purity of the fusion proteins was approximately 95%, as ascertained by SDS-PAGE.

Characterization of ELPs and Trx-ELP Fusion Proteins. The concentration of ELPs and Trx-ELP fusion proteins was determined by the molar extinction coefficients at 280 nm ($5690 \text{ M}^{-1} \text{ cm}^{-1}$ for ELPs and $19\,870 \text{ M}^{-1} \text{ cm}^{-1}$ for Trx-ELPs) calculated from their primary amino acid sequence with the software program, Protean (DNA Star). Absorption at 280 nm was measured by UV–visible spectrophotometry (UC-1601, Shimadzu Scientific Instruments). Purity and molecular weight of ELPs and Trx-ELP fusion proteins were characterized by SDS-PAGE (BioRad, Inc., Hercules, CA) with copper staining and matrix-assisted laser desorption/ionization mass spectroscopy (MALDI-MS), performed on a PE Biosystems Voyager-DE instrument equipped with a nitrogen laser (337 nm). The ELP samples for MALDI-MS measurements were prepared in an aqueous 50% (v/v) acetonitrile solution containing 0.1% (v/v) trifluoroacetic acid, using a sinapinic acid matrix.

The effect of increasing temperature on the turbidity of the ELP solutions was measured at 350 nm on a Cary 300 UV–visible spectrophotometer equipped with a multicell thermoelectric temperature controller (Varian Instruments, Walnut Creek, CA) between 10 and 90 °C at a rate of 1 °C/min. The inverse transition temperature (T_i) of each ELP fusion protein was defined as the temperature at which the first derivative of the turbidity as a function of temperature was the maximum. The sensitivity of T_i to ionic strength was defined by the slope of a linear fit of the T_i as a function of NaCl concentration ($\Delta T_i/\text{M NaCl}$) in phosphate-buffered saline (PBS).

Statistical Analysis. To quantitatively evaluate the linearity of the ELP and Trx-ELP salt sensitivity trends, the T_i data for each ELP as a function of salt were fit by linear regression. The residual at each data point was determined from the difference between the observation and the best-fit line, and the squared residuals were log transformed to meet equality of variance assumptions. The transformed fit residuals were compared by a three-way ANOVA using SPSS version 14.0.0 (Chicago, IL). Three variables were considered: (1) the ELP library (KV₇F or KV₂F); (2) the presence of Trx (yes or no); and (3) the polymer length (short, medium, or long).

Results and Discussion

Purification of Cationic ELPs and Their Trx Fusions.

Figure 2 shows copper-stained SDS-PAGE gels of purified ELP-[KV₂F] and ELP-[KV₇F] of differing molecular weights as well as their corresponding Trx-ELP fusion proteins. The gels show that the purified ELPs and their Trx-ELP fusions migrate to positions that are approximately consistent with molecular weights 20% greater than the molecular weights calculated from their engineered DNA sequences; this observation is typical of previously characterized ELPs.^{12,27} However, MALDI-MS indicated that for all ELPs and their Trx fusions, experimentally measured molecular weights were within 0.1–0.5% of their calculated molecular weights.

ELP-[KV₂F-8] and ELP-[KV₇F-9], with only 40 and 45 amino acids (4.3 and 4.6 kDa), respectively, are the shortest tags, to date, that have been used to purify an ELP fusion protein by ITC; however, their corresponding free ELPs could not be purified by ITC. Although the exact reasons for this remain unclear, two problems, the decrease of ELP expression (data not shown) and the increase in ELP T_i with decreasing molecular weight, compound the purification of very low molecular weight ELPs (Table 1). These combined effects are likely responsible for the observations that the inverse phase transition of ELP-[KV₂F-8] and ELP-[KV₇F-9] could not be detected in cell lysate and no ELP could be retrieved by ITC. However, Trx fusions to these very short ELPs were successfully purified by ITC,

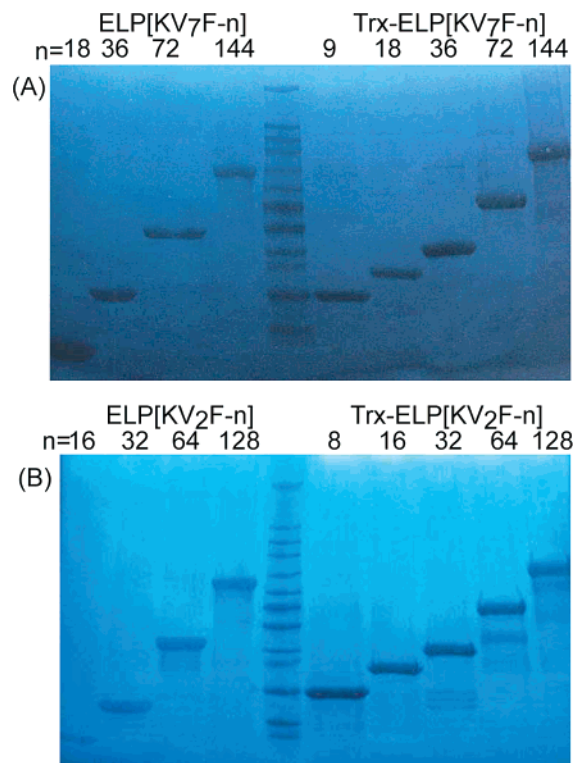


Figure 2. Copper-stained SDS-PAGE gels of purified (A) ELP-[KV₇F] (lanes 1–4) and Trx-ELP-[KV₇F] (lanes 6–10) and (B) ELP-[KV₂F] (lanes 1–4) and Trx-ELP-[KV₂F] (lanes 6–10). Lane 5 on both gels contains molecular size markers (205, 116, 97, 84, 66, 55, 45, 36, 29, 24, 20, 14, 7 kDa from top to bottom). Each lane is marked with the number of pentapeptide repeats (n) in the ELPs and ELP fusion proteins. Lower MW fragments observed in Trx-ELP-[KV₂F] protein samples are the result of proteolytic cleavage within the linker peptide located between Trx and ELP domains.

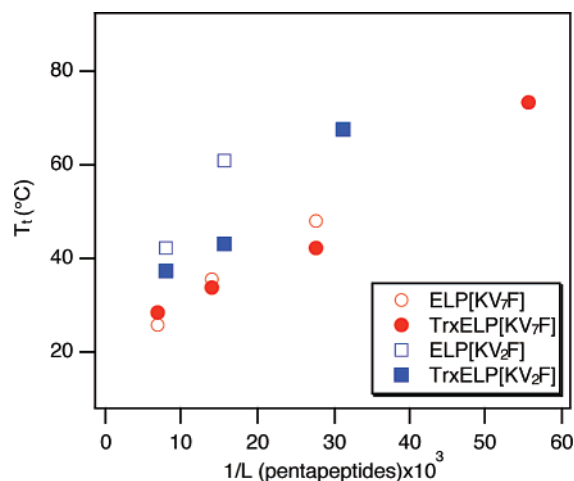
which is consistent with the facts that Trx is an overexpressed protein in *E. coli*, and fusion to Trx lowers the T_i of these ELPs, such that the addition of 3 M NaCl is enough to trigger the phase transition of the fusion protein.

The copper-stained SDS-PAGE gel in Figure 2B shows several smaller MW bands in the purified Trx-ELP-[KV₂F-32, -64, and -128] products. Analysis of MALDI-MS data indicates that all of the Trx-ELP-[KV₂F] fusions, except Trx-ELP-[KV₂F-8], contain minor contaminants, the largest of which is consistently 13.4 ± 0.45 kDa smaller than the Trx-ELP fusion protein from which it is derived. This difference in size approximately corresponds to the MW of Trx that would be liberated from the ELP by enzymatic cleavage with thrombin (13.9 kDa), suggesting that there are two or three sites within the SSGLVPRGS linker region of the Trx-ELP fusion protein that are susceptible to enzymatic attack by *E. coli* proteases. Because this minor contamination product likely consists of the cleaved ELP fragment of the fusion, it is co-purified with the Trx-ELP fusion through multiple rounds of ITC and hence is present in the final purified product. Furthermore, we hypothesize that ELP-[KV₂F] fusions are more susceptible to proteolysis than ELP-[KV₇F] fusions because of conformational differences caused by different interactions between Trx and ELP domains in the fusion proteins. Clues to these conformational differences can be elucidated from analysis of the thermal properties of these ELPs and their Trx fusions (vide infra).

Thermal Characterization of ELPs and Trx-ELP Fusion Proteins. Figure 3 and Table 1 show the T_i 's of all expressed ELPs and Trx-ELP fusion proteins at a concentration of 25 μM in PBS over the experimentally accessible temperature range

Table 1. Molecular Weight, Lysine (K) and Phenylalanine (F) Content, Transition Temperature (T_i), and Fusion ΔT_i Parameters for Purified ELPs and Trx-ELPs at 25 μ M in PBS (N/A: Not Available)

ELPs	no. of pentapeptides	no. of K or F residues	purified ELPs		purified Trx-ELPs		fusion ΔT_i parameter
			observed MW (kDa)	T_i ($^{\circ}$ C)	observed MW (kDa)	T_i ($^{\circ}$ C)	
[KV ₇ F]	9	1	N/A	N/A	18.5	>90.0	N/A
	18	2	8.4	>90.0	22.2	73.1	>-16.9
	36	4	15.8	48.0	29.7	42.1	-5.9
	72	8	31.0	35.7	44.9	34.0	-1.7
	144	16	61.1	26.1	74.9	28.4	2.3
[KV ₂ F]	8	2	N/A	N/A	18.1	>90.0	N/A
	16	4	7.7	>90.0	21.5	>90.0	N/A
	32	8	14.5	>90.0	28.4	67.4	>-22.6
	64	16	28.4	60.7	42.1	43.1	-17.6
	128	32	55.7	42.4	69.6	37.3	-5.1

**Figure 3.** Plot of transition temperature as a function of 1/length of ELP pentapeptide $\times 10^3$ for ELPs and thioredoxin-ELPs. The transition temperature was measured from 25 μ M solutions of ELP and Trx-ELP in PBS, and the data support a linear relationship between T_i and 1/ELP length similar to that observed for other ELPs.³⁰ Linear regression shows $r^2 = 0.986$ for ELP[KV₇F], $r^2 = 0.983$ for Trx-ELP[KV₇F], and $r^2 = 0.977$ for Trx-ELP[KV₂F]. Goodness of fit could not be established for ELP[KV₂F] polymers because only two of the four expressed ELP[KV₂F] proteins have measurable T_i (below 90 $^{\circ}$ C) at 25 μ M in PBS.

of 20–90 $^{\circ}$ C. The T_i 's of a number of ELPs and Trx-ELP fusions could not be accurately measured under these solution conditions because they were higher than 90 $^{\circ}$ C (Table 1). In general, the T_i 's of members of the ELP[KV₂F] series are 16–42 $^{\circ}$ C higher than those of the ELP[KV₇F] series of similar MW, reflecting their >2 -fold greater lysine content. ELP[KV₂F] series ELPs have T_i 's that are very similar to ELPs of similar MW having a guest residue composition of 50% valine, 20% alanine, and 30% glycine (ELP[V₅A₂G₃]),^{12,13} the most widely used ELP fusion protein purification tag.^{3,4,14,15} In contrast, the ELP[KV₇F] series exhibit T_i 's approximately 20 $^{\circ}$ C lower than their size-matched ELP[V₅A₂G₃] counterparts.

We have previously reported that the T_i of a set of aliphatic ELPs could be predicted by a simple equation, which accounts for the ELP concentration and its length, where the T_i is inversely proportional to the length of the ELP.¹³ Figure 3 shows that the T_i 's of these cationic ELPs and Trx-ELP fusions also exhibit a linear relationship with 1/length of ELP pentapeptides. ELP[KV₂F] polypeptides and Trx fusions have steeper slopes than their corresponding ELP[KV₇F] counterparts, which indicates that the fraction of Lys residues in these cationic ELPs dominates their change in T_i with ELP length. Although we

were only able to measure the T_i at 25 μ M in PBS for a small subset of the ELP and Trx-ELP proteins ($T_i < 90$ $^{\circ}$ C), limiting our ability to measure the goodness of the linear fits, these data suggest that this linear correlation is not limited to aliphatic ELPs. It appears that cationic ELPs and their Trx fusions also exhibit linear correlations with 1/ELP length as well. Thus, the T_i 's of ELPs and Trx-ELPs can be reasonably approximated by these correlations, allowing the T_i 's for new ELP lengths to be predicted for these amino acid compositions. This knowledge provides a predictive tool to control the phase transition behavior of ELPs and their fusion proteins for different applications.

Figure 3 and Table 1 also show the effect that fusion of Trx to these ELPs has on the ELP T_i . In general, fusion of Trx to ELP[KV₂F] and ELP[KV₇F] results in a depression in the ELP T_i relative to the free ELP (negative fusion ΔT_i parameter), and the magnitude of this depression depends on the ELP sequence. ELP[KV₂F] sequences show a greater fusion ΔT_i parameter than ELP[KV₇F] sequences. Only ELP[KV₇F-144] exhibits a positive fusion ΔT_i parameter, as the T_i of its Trx fusion is elevated by approximately 2 $^{\circ}$ C. The negative fusion ΔT_i parameter, observed for most of these ELPs, contrasts with the results of a previous study of Trx-ELP fusion proteins, in which fusion of Trx to an ELP having only aliphatic guest residues resulted in an elevation of the ELP T_i by ~ 6 $^{\circ}$ C.¹⁵

These previous studies showed that the fusion ΔT_i parameter exhibits a negative linear correlation with the fraction of hydrophobic surface area of the fused protein. These studies were performed using a single ELP tag, ELP[V₅A₂G₃-90], and different fusion partners.¹⁵ In this study, however, we have a single protein, Trx, but the composition and the hydrophobicity of the ELP are altered by variations in both guest residue composition and ELP length.

There are two plausible explanations for the observed changes in the sign of the fusion ΔT_i parameter: enhanced interaction between Trx and ELP domains via hydrophobic contacts or electrostatic attraction between positively charged Lys residues in the ELP and negatively charged acidic residues on the surface of Trx. We note that the latter possibility is unique to the cationic ELPs studied here, as the aliphatic ELP[V₅A₂G₃-90] in the previous study only had Val, Ala, and Gly at the guest residue position. The T_i 's of ELP[KV₇F] proteins are 16–42 $^{\circ}$ C lower than ELP[KV₂F] polypeptides, indicating that for the same concentration and similar molecular weight they are more hydrophobic; however, when fused to Trx, ELP[KV₂F] polypeptides exhibit a greater negative fusion ΔT_i parameter, indicating that in this case it is not the net hydrophobic character of the ELP that is responsible for the changes in the fusion ΔT_i parameter. Instead, the sign and magnitude of their fusion ΔT_i

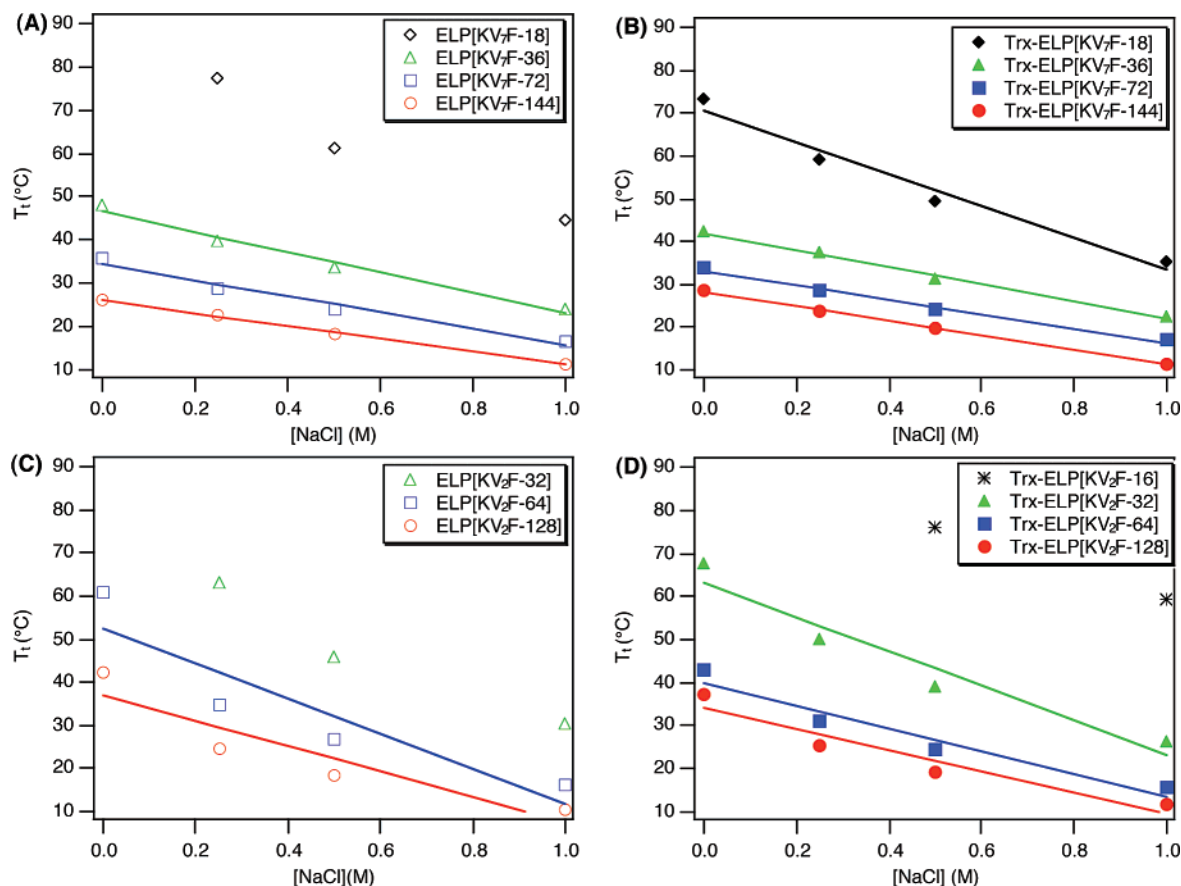


Figure 4. Plot of transition temperature as a function of NaCl concentration and molecular weight for (A) ELP[KV₇F], (B) Trx-ELP[KV₇F], (C) ELP[KV₂F], and (D) Trx-ELP[KV₂F]. The transition temperature was measured from 25 μ M ELP and Trx-ELP solutions in PBS supplemented with the appropriate NaCl concentration. ELP[KV₇F] (r^2 from 0.980 to 0.998) and Trx-ELP[KV₇F] (r^2 from 0.974 to 0.999) exhibit largely linear trends in T_t as a function of NaCl concentration, while ELP[KV₂F] (r^2 from 0.840 to 0.860) and Trx-ELP[KV₂F] (r^2 from 0.918 to 0.941) proteins show more nonlinear character consistent with their higher ELP charge densities. Best fit linear correlations to salt sensitivity data were plotted only for ELPs and Trx-ELPs where the T_t could be accurately measured at all four tested NaCl concentrations in PBS (0, 0.25, 0.5, and 1 M NaCl). These correlations are summarized in Figure 5.

parameter are consistent with electrostatic attraction between the positively charged lysine residues on these ELPs and negatively charged Asp and Glu residues present on the surface of Trx. The isoelectric point of Trx, calculated from its primary sequence, is 4.5, indicating that, at pH 7, each Trx molecule has a formal charge of nearly -5 , so that it presents a negatively charged surface with which positively charged ELPs may favorably interact. Electrostatic interaction between the cationic ELPs and Trx then results in the neutralization of charged moieties in the ELP, which lowers the T_t of the fused ELP, resulting in a negative ΔT_t parameter. As ELP[KV₂F] tags have more than twice the fraction of charged lysine residues, their T_t 's are depressed to greater extent by fusion to Trx than ELP[KV₇F], so that they have greater negative ΔT_t parameters.

Salt Sensitivity of Cationic ELPs and Their Trx Fusion Proteins. Figure 4 shows the effect of NaCl concentration on the T_t of the ELPs (Figure 4A and C) and their Trx fusions (Figure 4B and D). As is typical of all ELPs, the T_t decreases with NaCl concentration, and this trend holds true for their Trx fusions as well. Comparison of panels A and C in Figure 4 shows that the T_t values of ELP[KV₂F] proteins are more greatly affected by NaCl concentration than ELP[KV₇F] proteins, as indicated by their steeper slopes. These trends can be quantitatively evaluated by fitting linear correlations to the data in Figure 4A–D.

Figure 5A plots the slopes of these linear correlations as a function of ELP length for both the ELPs and their Trx fusions, and Figure 5B and C shows the average of the squared fit

residuals for each correlation for ELP[KV₇F] and ELP[KV₂F] proteins with and without the presence of Trx. Only proteins whose T_t could be measured ($T_t < 90^\circ\text{C}$) in PBS for both the ELP and the corresponding Trx-ELP protein have been plotted and statistically analyzed to determine the effects of ELP composition and added Trx on the error in the linear fit.

Free ELP[KV₂F] polypeptides exhibit the greatest sensitivity to NaCl, followed by Trx-ELP[KV₂F], ELP[KV₇F], and Trx-ELP[KV₇F]. Although the slopes plotted in Figure 5A are derived from linear fits to the data in Figure 4, it is clear from Figure 4 as well as Figure 5B and C that not all of the ELPs and their Trx fusions exhibit a linear change in T_t with NaCl concentration. The salt sensitivity of ELP[KV₇F] polypeptides and their Trx fusion proteins are better approximated by a linear fit than the ELP[KV₂F] and their Trx fusions. Figure 5B and C shows the average squared fit residuals for the ELPs and their corresponding Trx fusions for which T_t could be measured ($T_t < 90^\circ\text{C}$) in PBS without any added salt. Figure 5C shows that ELP[KV₂F] proteins have average squared fit residuals between 20 and 45 as compared to ELP[KV₇F] proteins (Figure 5B), which have average squared fit residuals of less than 1 for the proteins of similar length. When similar length proteins are directly compared, the average squared fit residuals for ELP[KV₂F] proteins are as much as 270-fold higher than ELP[KV₇F] proteins with ELP[KV₂F] proteins. The deviation from linearity is correlated with the net charge on the ELP, as the more highly charged ELP[KV₂F] polypeptides having 5% cationic residues

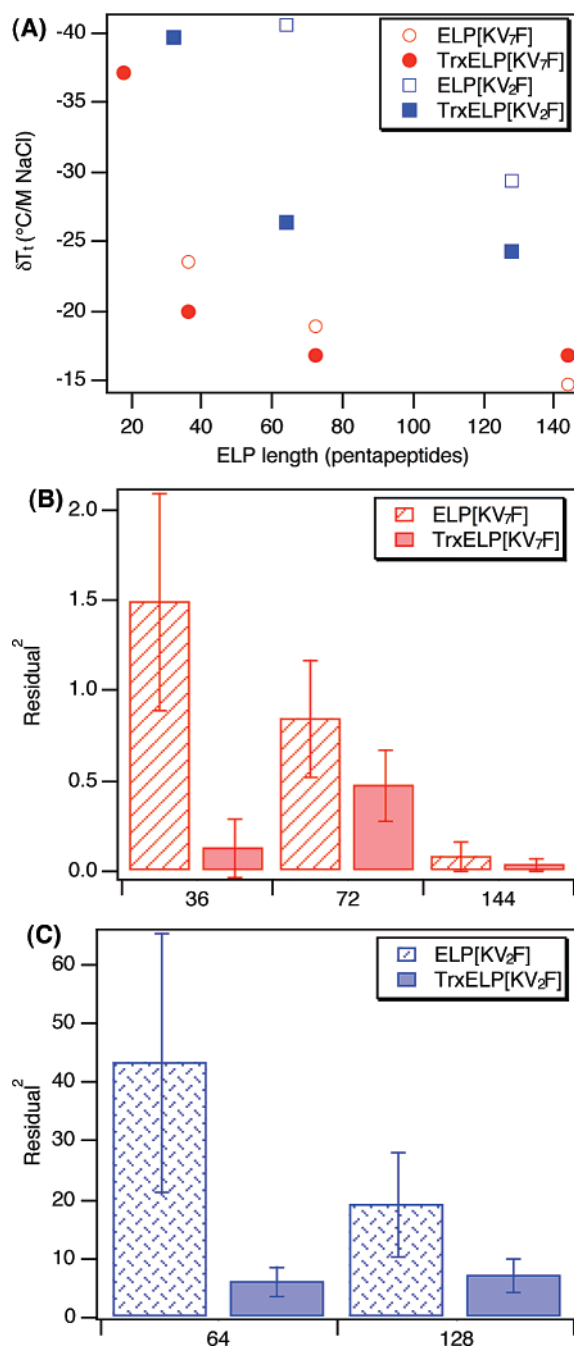


Figure 5. Sensitivity of the ELP and Trx-ELP phase transition to added NaCl as a function of ELP length. (A) δT_1 /M of NaCl taken from the slope of a linear fit to the data in Figure 4 plotted as a function of number of ELP pentapeptides for ELPs and Trx-ELPs. Linear fit error is expressed as average of the squared fit residuals for (B) ELP-[KV₇F] and (C) ELP[KV₂F] proteins and corresponding Trx fusions. ELP[KV₂F] proteins were found to be significantly more nonlinear than ELP[KV₇F] proteins ($p = 1.0 \times 10^{-9}$) in their sensitivity to added salt, and error in the linear fit over all of the ELP fusion proteins is significantly reduced ($p = 0.0033$) upon fusion to Trx. Error bars reflect the standard deviation for $n = 4$ salt concentrations per protein.

are more poorly fit by a linear function than the ELP[KV₇F] polypeptides, which have only 2% cationic content.

We believe that electrostatic repulsion between the positive charges of the ELP is responsible for the nonlinear changes in T_1 with salt concentration. Previous studies of purely aliphatic ELPs (with no ionizable amino acid content) have shown that T_1 is linearly correlated with NaCl (-14 °C/M NaCl).¹⁰ ELP-[KV₇F-72], which has only 2% positively charged lysine

residues, exhibits a 19 °C/M NaCl depression and can be reasonably fit by a linear function having an average squared residual of only 0.84. In contrast, ELP[KV₂F-64], which has more than twice the cationic character (5%), exhibits a 40 °C/M NaCl depression in T_1 and is poorly fit by a linear model, exhibiting an average squared fit residual of 43. Addition of Trx to both of these polymers not only reduces the slope of the T_1 depression to 17 and 26 °C/M NaCl for ELP[KV₇F-72] and ELP[KV₂F-64] proteins, respectively. Furthermore, it also reduces the average squared fit residuals by 1.8- and 7.2-fold, respectively. We believe these changes in slope and linear fit error are a function of Coulombic repulsion, which is a nonlinear effect of the solution ionic strength as described by Debye–Hückel screening. The Debye length, which is a measure of the distance over which electrostatic effects are damped out by mobile ions in an aqueous solution, is inversely proportional to the square root of the solution ionic strength.²⁸ Thus, in low ionic strength buffer, like charges in the ELP would repel one another over greater distances, inhibiting aggregate formation and leading to elevated T_1 's. This charge screening hypothesis is consistent with the fact that the greatest nonlinearity in T_1 as a function of ionic strength is observed for the ELPs with the greatest cationic content and that the addition of the negatively charged Trx reduces the net charge in the fusion, leading to increased linear behavior in the T_1 with NaCl concentration.

Overall trends in the fit linearity and the significance of fused Trx to this linearity were determined by statistical analysis of the squared fit residuals. A global three-way ANOVA was found to significantly fit the entire dataset ($F_{9,30} = 12.9$, $p = 4.0 \times 10^{-8}$, $r^2 = 0.795$). The statistical analysis pointed to two significant main effects of (1) ELP identity ($F_{1,30} = 76.1$, $p = 1.0 \times 10^{-9}$); and (2) the fusion of Trx ($F_{1,30} = 10.2$, $p = 0.0033$), confirming the visually observable differences between the thermal behavior ELP[KV₇F] and ELP[KV₂F] polymers. This model also controlled for an expected significant main effect of ELP length ($F_{2,30} = 9.42$, $p = 0.00067$) and a two-way interaction between ELP and length ($F_{1,30} = 11.2$, $p = 0.0023$). Because the primary goal of this analysis was to compare deviation from linearity between ELP[KV₇F] and ELP-[KV₂F] with and without fused Trx, and because there was no interaction between Trx and ELP, we did not split the dataset to further explore the significance of length on the ELPs and corresponding Trx-ELPs.

The main effects from three-way ANOVA of Trx and ELP on the linearity of T_1 with added NaCl indicate two things. First, the charge density of ELP has a significant effect on linearity between T_1 and added salt. ELP[KV₂F] was significantly ($p = 1.0 \times 10^{-9}$) more nonlinear than ELP[KV₇F]. Second, the fusion of a negatively charged Trx protein to cationic ELPs overall significantly improved the linearity between T_1 and added salt ($p = 0.0033$). The significant improvement in linearity upon fusion to Trx suggests that the fusion of the negatively charged Trx to the cationic ELPs effectively reduces the net cationic charge in the fusion protein, promoting T_1 salt sensitivity that more closely approximates that exhibited by non-ionic ELPs. This indicates that ELP to fused protein interactions are likely important to the formation of aggregates above T_1 , especially at lower ionic strengths. Furthermore, these data suggest that ELP/target protein interactions are improved by matching anionic target proteins with cationic ELPs. By extrapolation, we suggest that cationic proteins should be fused to anionic ELPs to improve ELP–protein interactions in the aggregated state.

Because of the heightened sensitivity of charged ELPs and their Trx fusions to NaCl, we would anticipate that Trx-ELP

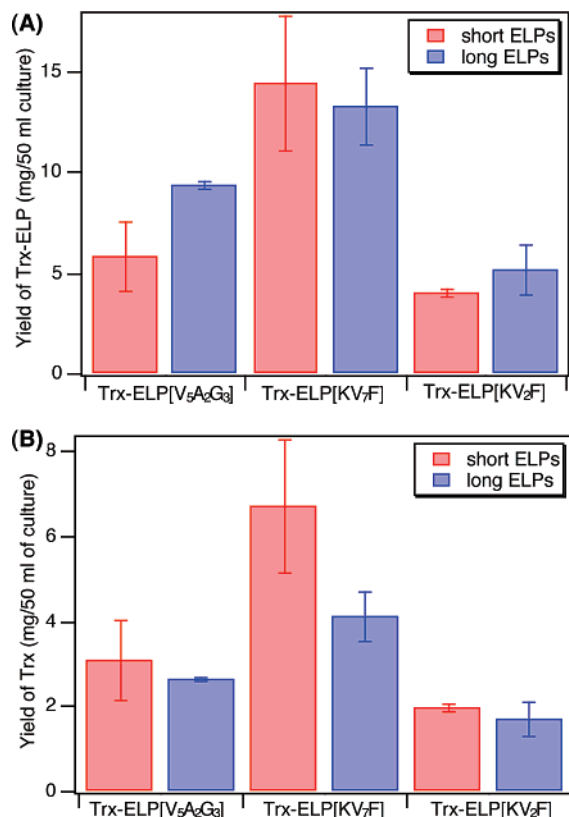


Figure 6. Yield of (A) Trx-ELP fusion proteins and (B) target Trx protein from 50 mL of *E. coli* culture as a function of the ELP length (30–36 pentapeptides for the short tags and 64–90 pentapeptides for the long tags) and guest residue composition of the ELP purification tag. Protein yield is reported as the mean \pm SD ($n = 3$). The gravimetric yield of Trx is higher for the lower molecular weight tag (30–36 pentapeptides) in the case of all three ELP guest residue compositions, and the yield of Trx-ELP is greatest for the ELP[KV₇F] tags. The poorest fusion protein and target protein yield was exhibited by the ELP[KV₂F] proteins, suggesting that expression is highly dependent on the guest residue composition and not easily predictable.

would be purified from soluble *E. coli* lysate with a minimum of added salt by using a relatively short ELP[KV₂F] fusion tag; however, in practice we found that more NaCl was necessary to purify Trx-ELP[KV₂F] proteins from soluble lysate than Trx-ELP[KV₇F] proteins. This unanticipated problem results from the strong effect that ELP composition has on fusion protein yield and therefore, the fusion protein concentration in the soluble lysate. Figure 6 shows the quantitative yield of the Trx-ELP fusion protein and the corresponding mass of Trx target protein calculated on the basis of its mass fraction in each fusion for short (30–36 pentapeptides) and long (64–90 pentapeptides) ELPs from three different series: ELP[KV₇F], ELP[KV₂F], and ELP[V₅A₂G₃]. The last series is included because it is the most commonly used ELP tag for purification of recombinant proteins. Figure 6A shows that the yield of Trx-ELP[KV₇F-36] is approximately 2.5 times higher than that of Trx-ELP[V₅A₂G₃-30] and 3.6 times higher than that of Trx-ELP[KV₂F-32] proteins, suggesting that the yield of Trx-ELP fusion proteins is dependent on the stoichiometry of the guest residue composition of the fused ELP tag, which may arise from changes in mRNA stability and translation efficiency of the ELP tag.²⁹

In addition, decreasing the length of the ELP tag can be advantageous for fusion protein expression: we previously showed that reducing the length of the ELP purification tag from 90 to 20 pentapeptides resulted in a \sim 4-fold increase in the yield of Trx in the ELP fusion.⁴ This is consistent with the data

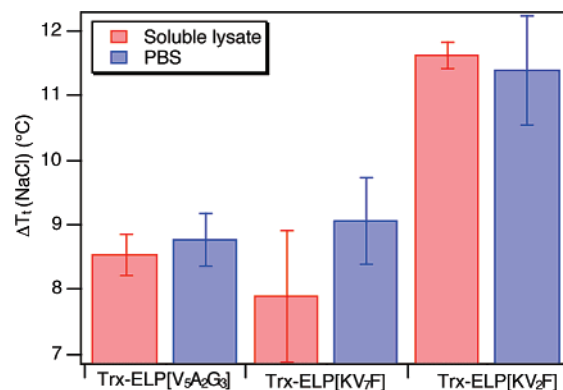


Figure 7. Transition temperature difference ($\Delta T_i(\text{NaCl})$) between 0.5 and 1 M NaCl added to *E. coli* soluble lysate and to PBS solutions of 100 μM Trx-ELP[KV₇F-36], Trx-ELP[KV₂F-32], and Trx-ELP[V₅A₂G₃-90]. There is no significant difference between lysate and PBS in the $\Delta T_i(\text{NaCl})$ in any of the proteins investigated. The transition temperature difference ($\Delta T_i(\text{NaCl})$) is reported as the difference in the mean T_i 's at 0.5 and 1 M NaCl ($n = 3$ for each salt concentration) \pm the standard deviation calculated by the propagation of errors in the measurements of T_i with 0.5 and 1 M NaCl.

in Figure 6B. The yield of Trx from the shorter fusion proteins is 1.1–1.6 times that of the yield from the corresponding longer ELP fusions. Trx-ELP[KV₇F-36] has more than 1.6-fold greater Trx yield than that from Trx-ELP[KV₇F-72]. It is more than 2.5-fold greater than that from Trx-ELP[V₅A₂G₃-90], the most commonly used ELP tag for purification. Furthermore, the Trx yield from Trx-ELP[KV₇F-32] is approximately 1.5-fold greater than that from Trx-ELP[V-20], the fusion boasting the highest Trx published yield to date.⁴ Although ELP[KV₇F-36] is somewhat longer than ELP[V-20], it has a significant advantage as a purification tag because the Trx fusion was completely retrieved from the *E. coli* lysate without any loss during ITC. In contrast, Trx-ELP[V-20] forms nanoparticles, that cannot be easily recovered by centrifugation, which complicates the purification of Trx-ELP[V-20].⁴

To deconvolute the coupled effects that ELP T_i , salt sensitivity, and Trx-ELP expression yield have on the purification efficiency of Trx-ELP fusion proteins, the T_i of Trx-ELP fusion proteins was measured at a fixed concentration in *E. coli* soluble lysate as well as in PBS. Figure 7 shows the difference in T_i between 0.5 and 1.0 M NaCl ($\Delta T_i(\text{NaCl})$) added to 100 μM Trx-ELP solutions in PBS and soluble *E. coli* lysate for Trx-ELP[KV₇F-36], Trx-ELP[KV₂F-32], and Trx-ELP[V₅A₂G₃-90]. These data indicate that the sensitivity of cationic ELP fusion proteins is maintained in the soluble lysate. For each of the proteins investigated, the $\Delta T_i(\text{NaCl})$ in lysate and PBS are significantly different. Trx-ELP[KV₂F-32], which has the highest cationic content in the ELP, exhibits the greatest sensitivity to salt in the soluble lysate, similar to its behavior in PBS buffer. However, this sensitivity to salt does not positively impact its purification as an expression product due to its poor expression yield. Conversely, the improved expression yield combined (Figure 6) with similar salt sensitivity (Figure 7) and a lower T_i (Table 1 and Figure 3) allows Trx-ELP[KV₇F-36] to be purified using one-third of the salt that is needed to purify Trx-ELP[V₅A₂G₃-90] (the most commonly used purification tag) with a 2.6-fold improvement in the gravimetric yield of Trx. Furthermore, Trx-ELP[KV₇F-36] exhibits a 50% increase in Trx gravimetric yield over that purified from Trx-ELP[V-20], the most successful Trx tag previously published, using only 20% of the added salt without the complication of nanoparticle formation exhibited by the ELP[V-20] tag.⁴

Conclusions

The first important finding of this study is that the ELP tag can be significantly shortened without introducing any complications in the purification of a model protein, Trx. ELP[KV₂F-8] with a MW of ~4.3 kDa is the shortest ELP that has been used to purify an ELP fusion protein to date, and ELP[KV₇F-36], the most optimized tag engineered to date for the purification of Trx, has a MW of only 15.9 kDa. The MW of ELP[KV₂F-8] is similar to that of many peptide affinity tags, while that of the optimized ELP[KV₇F-36] is significantly smaller than or similar to many protein affinity tags.

The study of cationic tags for the purification of Trx also uncovered molecular parameters that will be important for the design of future ELP tags for purification of recombinant proteins. These are as follows: (1) The incorporation of ionizable residues into the ELP sequence enhances the sensitivity of the ELP inverse phase transition to added salt and thereby reduces the amount of salt needed to induce the inverse phase transition in the soluble lysate. (2) The incorporation of ionizable groups must be balanced by the addition of hydrophobic residues to maintain a reasonable T_i for the ELP. The optimal T_i should be above room temperature at all working fusion protein concentrations, but should be designed to be reasonably low so that only a minimum of salt must be added at each round of ITC. (3) The ELP molecular weight should be minimized to improve target protein yield; however, it should be high enough that micellization or nanoparticle formation above the phase transition is not observed. (4) The guest residue composition significantly affects the expression yield of the ELP and thus the ELP fusion protein, so that well-expressed ELP sequences should be used as purification tags. Because the cause of sequence-dependent expression is, as yet, unknown, this variable will be difficult to predict and control. (5) Finally, the data suggest that the incorporation of cationic residues into ELPs is effective in promoting ELP–protein interactions in the aggregated state of the anionic Trx-ELP fusion proteins. We believe that opposite charges of the Trx and ELPs aid the formation of large aggregates during fusion protein purification. Bearing this in mind, we caution that the design of ELP fusion proteins having like charge characteristics in both the target protein and the ELP might inhibit aggregate formation and thereby complicate ELP fusion protein purification.

In conclusion, we believe that, for most laboratory scale purification by ITC, optimization of the ELP tag is probably not necessary, as the most commonly used aliphatic ELP tags are sufficiently versatile to provide purified fusion proteins with yields and purities that are typically similar to, if not greater, than those achieved with affinity chromatography.¹⁴ However, the results presented here suggest that large-scale purification of recombinant proteins for pharmaceutical or industrial applications will benefit from optimization of the ELP tag for a specific protein to maximize its expression yield and purification efficiency.

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Note Added after ASAP Publication. There was an error in the conclusions regarding ELP in the version published ASAP April 4, 2007; the corrected version was published ASAP April 6, 2007.

References and Notes

- (1) Nilsson, J.; Stahl, S.; Lundeberg, J.; Uhlen, M.; Nygren, P. A. *Protein Expression Purif.* **1997**, *11*, 1–16.
- (2) Mondal, K.; Gupta, M. N. *Biomol. Eng.* **2006**, *23*, 59–76.
- (3) Meyer, D. E.; Chilkoti, A. *Nat. Biotechnol.* **1999**, *17*, 1112–1115.
- (4) Meyer, D. E.; Trabbic-Carlson, K.; Chilkoti, A. *Biotechnol. Prog.* **2001**, *17*, 720–728.
- (5) Urry, D. W. *J. Phys. Chem. B* **1997**, *101*, 11007–11028.
- (6) Li, B.; Alonso, D. O. V.; Bennion, B. J.; Daggett, V. *J. Am. Chem. Soc.* **2001**, *123*, 11991–11998.
- (7) Li, B.; Alonso, D. O. V.; Daggett, V. *J. Mol. Biol.* **2001**, *305*, 581–592.
- (8) Cacace, M. G.; Landau, E. M.; Ramsden, J. J. *Q. Rev. Biophys.* **1997**, *30*, 241–277.
- (9) Zhang, Y. J.; Trabbic-Carlson, K.; Albertorio, F.; Chilkoti, A.; Cremer, P. S. *Biomacromolecules* **2006**, *7*, 2192–2199.
- (10) Luan, C. H.; Parker, T. M.; Prasad, K. U.; Urry, D. W. *Biopolymers* **1991**, *31*, 465–475.
- (11) Chow, D. C.; Dreher, M. R.; Trabbic-Carlson, K.; Chilkoti, A. *Biotechnol. Prog.* **2006**, *22*, 638–646.
- (12) Meyer, D. E.; Chilkoti, A. *Biomacromolecules* **2002**, *3*, 357–367.
- (13) Meyer, D. E.; Chilkoti, A. *Biomacromolecules* **2004**, *5*, 846–851.
- (14) Trabbic-Carlson, K.; Liu, L.; Kim, B.; Chilkoti, A. *Protein Sci.* **2004**, *13*, 3274–3284.
- (15) Trabbic-Carlson, K.; Meyer, D. E.; Liu, L.; Piervincenzi, R.; Nath, N.; LaBeau, T.; Chilkoti, A. *Protein Eng. Des. Select.* **2004**, *17*, 57–66.
- (16) Kim, J. Y.; Mulchandani, A.; Chen, W. *Biotechnol. Bioeng.* **2005**, *90*, 373–379.
- (17) Kim, J. Y.; O'Malley, S.; Mulchandani, A.; Chen, W. *Anal. Chem.* **2005**, *77*, 2318–2322.
- (18) Kim, J. Y.; Mulchandani, A.; Chen, W. *Anal. Biochem.* **2003**, *322*, 251–256.
- (19) Kostal, J.; Mulchandani, A.; Chen, W. *Biotechnol. Bioeng.* **2004**, *85*, 293–297.
- (20) Kostal, J.; Mulchandani, A.; Chen, W. *Macromolecules* **2001**, *34*, 2257–2261.
- (21) Gao, D.; McBean, N.; Schultz, J. S.; Yan, Y. S.; Mulchandani, A.; Chen, W. F. *J. Am. Chem. Soc.* **2006**, *128*, 676–677.
- (22) Kostal, J.; Yang, R.; Wu, C. H.; Mulchandani, A.; Chen, W. *Appl. Environ. Microbiol.* **2004**, *70*, 4582–4587.
- (23) Shimazu, M.; Mulchandani, A.; Chen, W. *Biotechnol. Bioeng.* **2003**, *81*, 74–79.
- (24) Banki, M. R.; Feng, L. A.; Wood, D. W. *Nat. Methods* **2005**, *2*, 659–661.
- (25) Ge, X.; Yang, D. S. C.; Trabbic-Carlson, K.; Kim, B.; Chilkoti, A.; Filipe, C. D. M. *J. Am. Chem. Soc.* **2005**, *127*, 11228–11229.
- (26) Trabbic-Carlson, K.; Setton, L. A.; Chilkoti, A. *Biomacromolecules* **2003**, *4*, 572–580.
- (27) McPherson, D. T.; Morrow, C.; Minehan, D. S.; Wu, J. G.; Hunter, E.; Urry, D. W. *Biotechnol. Prog.* **1992**, *8*, 347–352.
- (28) Creighton, T. E. *Proteins: Structures and Molecular Properties*, 2nd ed.; W. H. Freeman and Co.: New York, 1993.
- (29) Fexby, S.; Bulow, L. *Trends Biotechnol.* **2004**, *22*, 511–516.
- (30) Meyer, D. E.; Chilkoti, A. *Biomacromolecules* **2004**, *5*, 846–851.

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