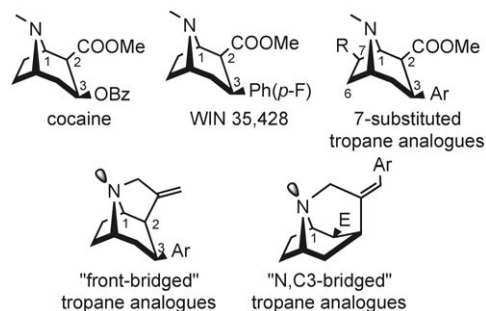


# Structural Remodeling of Cocaine: Design and Synthesis of Trisubstituted Cyclopropanes as Selective Serotonin Reuptake Inhibitors

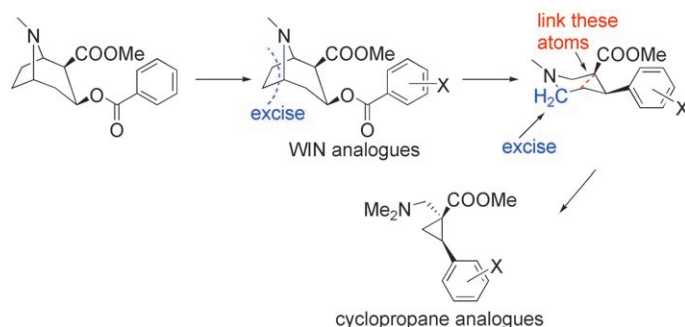
Alan P. Kozikowski,<sup>[a]</sup> Lianyun Zhao,<sup>[a]</sup> Ao Zhang,<sup>[a]</sup> Cheng Z. Wang,<sup>[b]</sup> Judith Flippen-Anderson,<sup>[c]</sup> and Kenneth M. Johnson<sup>[b]</sup>

Cocaine, one of the most powerful stimulants of the mammalian central nervous system, binds all monoamine transporters (dopamine transporter (DAT), serotonin transporter (SERT) and norepinephrine transporter (NET)) with similar potency.<sup>[1–3]</sup> As the reinforcing effects of cocaine are believed to be due primarily to its activity toward DAT, over the past decades, several different classes of DAT-selective ligands have been designed for use as cocaine treatment medications. However, compounds that are selective solely for the DAT and NET have not yet resulted in clinically effective medications against cocaine abuse.<sup>[4–11]</sup> For this reason, ligands selective for the SERT transporter or various central nervous system (CNS) receptors have been investigated for use in the treatment of cocaine abuse as well.<sup>[9–17]</sup> The selective serotonin reuptake inhibitors (SSRIs) have been long known to be valuable in the treatment of certain affective disorders including depression.<sup>[18]</sup>

Our recent work in this area has led to the discovery of several different series of compounds with novel structures that are distinct from those of the traditional tropane analogues belonging to the Sterling–Winthrop (WIN) series.<sup>[19–27]</sup> Structure–activity relationship (SAR) studies have shown that some of these newer ligands exhibit good to excellent potency and selectivity for the DAT, SERT, or NET. For example, conformationally constrained tricyclic tropane analogues of the (Z)-9-(aryl-methylene)-7-azatricyclo[4.3.1.0<sup>3,7</sup>]decane<sup>[25–27]</sup> class are highly potent and selective for the SERT over the DAT and NET, whereas several piperidine-based ligands display selectivity for the DAT and NET.<sup>[21–24]</sup> One of these piperidine ligands ("Nocaine")



has been found to antagonize cocaine-induced locomotor effects in rodents while acting as a weak positive reinforcer compared with cocaine in primates.<sup>[24,28]</sup> Such results encouraged us to explore further structural modifications of cocaine without alteration to its essential pharmacophore. Specifically, we envisioned the excision of C6 from the piperidine structure and the connection of C3 with C5 by a single bond to form a cyclopropane ring (Figure 1). Herein we report the synthesis of



**Figure 1.** General plan for the synthesis of cyclopropane analogues that retain the pharmacophore of the parent cocaine.

representative compounds of this class and their SAR at monoamine transporters. Related compounds in which the ester function is located at the same position on the cyclopropane ring as the aryl group were previously investigated by Bonnaud et al., and several congeners were found to exhibit activity profiles suggestive of their suitability as antidepressants.<sup>[29]</sup>

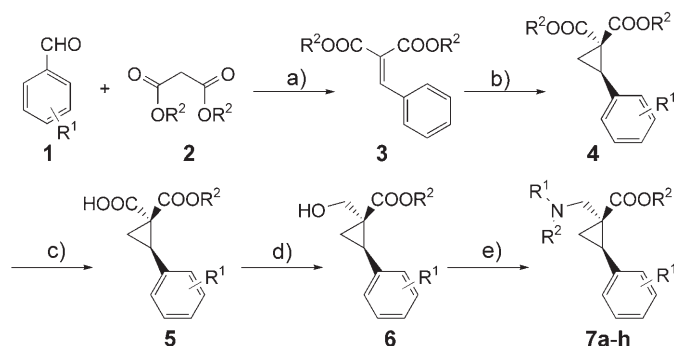
The first step of this project involved a literature search for methods of synthesis of the required cyclopropane scaffold. Although the synthesis of this particular scaffold has been reported,<sup>[30,31]</sup> there is a general lack of detailed procedures, and little substantive biological activity has been reported for such ligands. The chemistry that we developed to gain access to compounds **7a–h** is illustrated in Scheme 1. Dimethyl or dibenzyl (*para*-substituted benzylidene)malonates **3** were synthesized by the Knoevenagel condensation of *para*-substituted benzaldehydes **1** and malonates **2**.<sup>[32]</sup> Treatment of compounds **3** with dimethylsulfoxonium methylide in *N,N*-dimethylformamide (DMF) gave cyclopropanes **4**.<sup>[33]</sup> The sterically more accessible ester group *trans* to the aromatic ring was selectively hydrolyzed in high yield with methanolic NaOH to give monoesters **5**. The carboxyl group of these intermediates was then con-

[a] Prof. Dr. A. P. Kozikowski, Dr. L. Zhao, Dr. A. Zhang  
Drug Discovery Program, Department of Neurology  
Georgetown University Medical Center  
3900 Reservoir Road, NW, Washington, DC 20007 (USA)  
Fax: (+1) 312-996-7107  
E-mail: Kozikowa@uic.edu

[b] Dr. C. Z. Wang, Prof. Dr. K. M. Johnson  
Department of Pharmacology and Toxicology  
University of Texas Medical Branch  
Galveston, TX 77555 (USA)

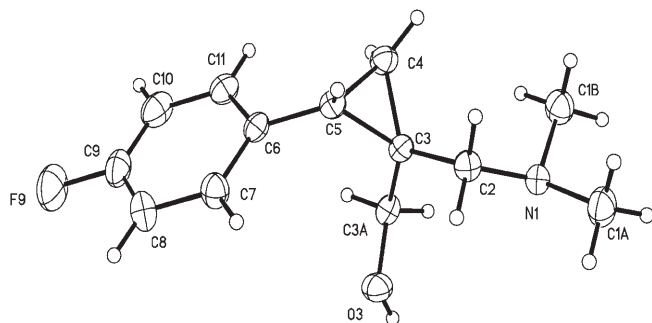
[c] Dr. J. Flippen-Anderson  
Laboratory for the Structure of Matter, Code 6030  
Naval Research Laboratory  
4555 Overlook Avenue, SW, Washington, DC 20375 (USA)

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**Scheme 1.** Reagents and conditions: a) piperidine, AcOH, benzene, reflux, 6 h; b) NaH, Me<sub>3</sub>S(O)I, DMF, RT→−40 °C, 2 h; c) NaOH (1 N), MeOH, RT, 2 h; d) *N*-methylmorpholine, isobutyl chloroformate, THF, 0 °C, 15 min, then NaBH<sub>4</sub>, THF/MeOH (2:1), −78 °C, 2 h; e) Tf<sub>2</sub>O, 2,4,6-trimethylpyridine, CH<sub>2</sub>Cl<sub>2</sub>, −78 °C, 40 min, then amine, −78 °C→−40 °C, overnight. Tf = trifluoromethanesulfonyl.

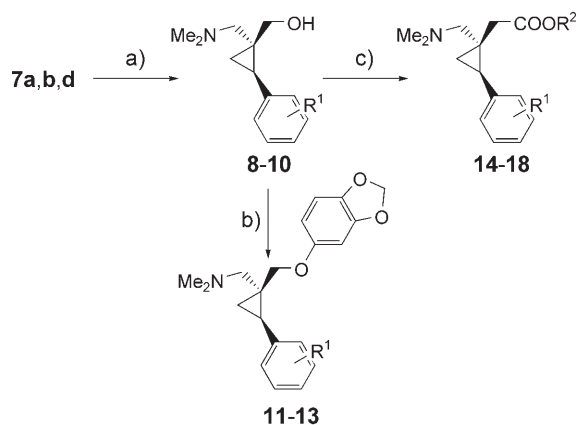
verted into a mixed anhydride with isobutyl chloroformate and *N*-methylmorpholine followed by reduction with sodium borohydride to give the corresponding alcohols **6** in high yield.<sup>[34,35]</sup> Alcohols **6** were converted into the corresponding triflates, which underwent nucleophilic displacement with amines to yield the desired compounds **7a–h**. Amino alcohols **8–10** were obtained by reduction of the esters **7** with LiAlH<sub>4</sub>. The *trans* relationship between the aryl and amino groups in this series of compounds was confirmed by X-ray crystallographic analysis of compound **8** (Figure 2). Further treatment of intermediates



**Figure 2.** View of compound **8** showing the labeling of non-hydrogen atoms. Displacement ellipsoids are shown at the 30% probability level.

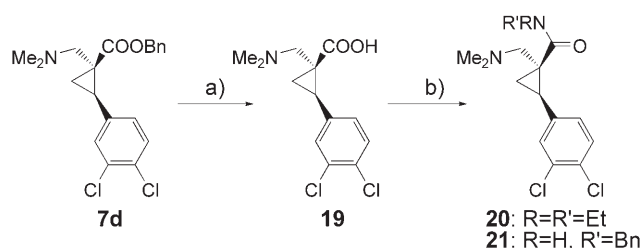
**8–10** with methanesulfonyl chloride, followed by reflux of the resulting mesylate with the sodium salt of sesamol in THF/DMF (5:1) containing a small amount of 18-crown-6, afforded the ethers **11–13** (Scheme 2).<sup>[36]</sup> The reverse esters **14–18** were prepared by esterification of the alcohols **8–10** with acetic anhydride or acid chlorides.<sup>[25]</sup> The amides **20** and **21** were obtained by hydrolysis of the ester **7d** with HCl (6 N) followed by reaction of the crude acid with oxalyl chloride to give the intermediate acid chloride, which was subsequently treated with the appropriate amines (Scheme 3).

For racemate resolution, acid **19** was treated with oxalyl chloride; the resulting acid chloride reacted with (−)-8-phenyl-



**Scheme 2.** Reagents and conditions: a) LiAlH<sub>4</sub>, THF; b) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C→RT, 3 h, then sodium salt of sesamol, 18-crown-6, THF/DMF (5:1), reflux, overnight; c) acetic anhydride or acid chloride, Et<sub>3</sub>N, THF, 0 °C→RT. Ms = methanesulfonyl.

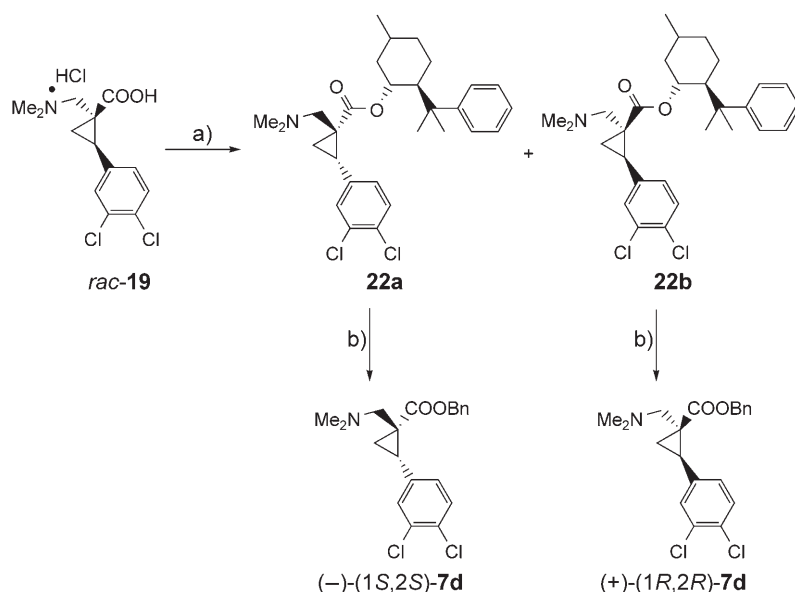
menthol in ether to give a mixture of esters **22a** and **22b** (Scheme 4), which are readily separable by column chromatography on silica gel. The pure diastereomers were hydrolyzed in HCl (6 N, reflux), and the resulting enantiomeric acids were



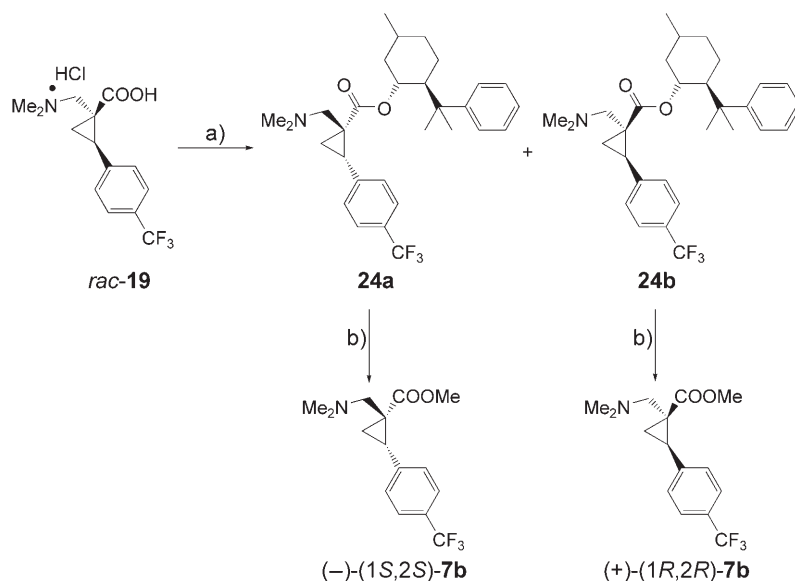
**Scheme 3.** Reagents and conditions: a) HCl (6 N), reflux; b) oxalyl chloride, CH<sub>2</sub>Cl<sub>2</sub>, RT, 2 h, then amine, CH<sub>2</sub>Cl<sub>2</sub>, RT, overnight. Bn = benzyl.

converted into the corresponding benzyl esters (−)-(1*S*, 2*S*)-**7d** and (+)-(1*R*, 2*R*)-**7d** (Scheme 4). The diastereomeric esters **24a** and **24b** (Scheme 5) were prepared in the same manner, and the relative configuration of **24a** was established by X-ray crystallographic analysis (Figure 3). The relative configuration of **22a**, and thus the absolute configuration of (−)-**7d**, follow by analogy.

All final compounds were tested for their ability to inhibit the high-affinity reuptake of dopamine (DA), 5-hydroxy-L-tryptamine (5-HT), and norepinephrine (NE) into nerve endings (synaptosomes) prepared from brain regions enriched in transporters for these biogenic amine neurotransmitters.<sup>[37,38]</sup> The uptake data and selectivity profiles (based on the *K<sub>i</sub>* values) of these compounds are listed in Table 1. All data are mean values ± SEM from at least three independent experiments, each of which consisted of six drug concentrations in triplicate. Nearly all compounds exhibit weak or no inhibition of reuptake at the DA and NE transporters, but modest to good potency toward the 5-HT transporter. These results contrast with the transporter inhibition data reported for the piperidine ana-



**Scheme 4.** Reagents and conditions: a) oxalyl chloride,  $\text{CH}_2\text{Cl}_2$ , RT, 2 h, then (-)-8-phenylmenthol, *n*BuLi, ether, RT, overnight; b) HCl (6 N), reflux, 12 h, then oxalyl chloride,  $\text{CH}_2\text{Cl}_2$ , RT, 2 h, then  $\text{PhCH}_2\text{OH}$ , *n*BuLi, THF, RT, overnight.



**Scheme 5.** Reagents and conditions: a) oxalyl chloride,  $\text{CH}_2\text{Cl}_2$ , RT, 2 h, then (-)-8-phenylmenthol, *n*BuLi, ether, RT, overnight; b) HCl (6 N), reflux, 12 h, then oxalyl chloride,  $\text{CH}_2\text{Cl}_2$ , RT, 2 h, then  $\text{CH}_3\text{OH}$ , *n*BuLi, THF, RT, overnight.

logues, which are potent DA and NE reuptake inhibitors.<sup>[21–24]</sup> By using compound 7a as a starting point, we investigated the effects of structural modifications made to the tropane series similar to modifications previously reported by us or other research groups.<sup>[6,25–27]</sup> Compound 7a displayed no activity at the DAT and NET and only weak inhibitory activity at the SERT ( $K_i = 1.6 \mu\text{M}$ ). Replacement of the 4-fluoro substituent in 7a with a trifluoromethyl group gave compound 7b, which exhibited a fourfold increase in potency against SERT. The benzyl ester 7c was even more potent toward the SERT, with a

$K_i$  value of 98 nM. Replacement of the 4-substituent with the 3,4-dichloro substitution pattern gave the analogue 7d, which displayed fairly good inhibitory activity at the SERT ( $K_i = 55 \text{ nM}$ ). Introduction of a larger group in the ester moiety (compounds 7e and 7f) resulted in a ninefold decrease in potency for the SERT.<sup>[6,39,40]</sup>

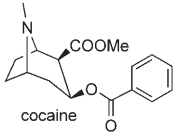
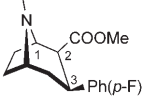
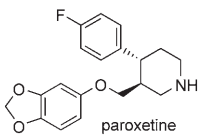
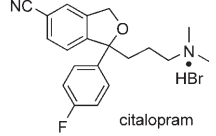
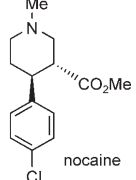
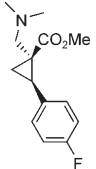
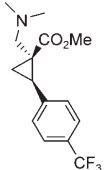
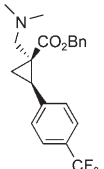
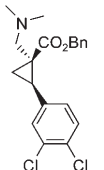
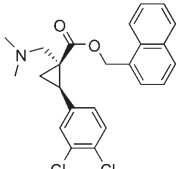
Modification of the amino group revealed that the primary and secondary amines 7h and 7g are eight- and fivefold less active at the SERT than is compound 7d. Interestingly, in the WIN series, N-demethylation generally leads to an improvement in SERT activity.<sup>[23,41,42]</sup>

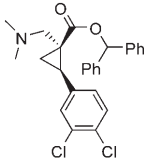
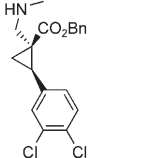
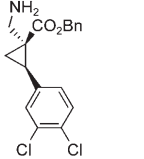
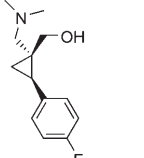
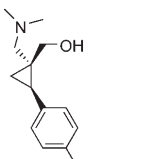
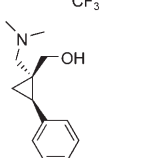
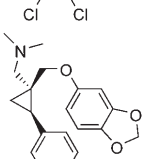
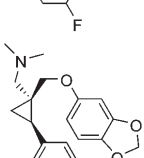
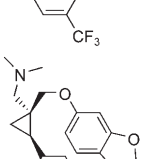
As the ester group in the WIN series of tropanes is generally tolerant to a variety of structural changes,<sup>[6,7]</sup> we examined the effect of modification at this site in the cyclopropanes. First, esters 7a, c, and d were reduced to the corresponding alcohols 8–10. Alcohol 8 displayed the same low activity at the SERT as the parent compound 7a, alcohol 9 displayed a 12-fold decreased potency at the SERT compared with the parent compound 7c, and alcohol 10 was fourfold less potent than its parent 7d. On the other hand, the ethers 11–13 prepared from their corresponding alcohols had improved potency at the SERT. Compound 11 showed a sixfold improvement in activity over ester 7a, ether 12 exhibited a twofold increase relative to ester 7c, and ether 13 was twofold more potent than the parent 7d, with a  $K_i$  value of 22 nM. Compound 13 also displayed some activity at the NET ( $K_i = 451 \text{ nM}$ ); the DAT/SERT uptake ratio is 81:1, and the NET/SERT uptake ratio is 20:1.

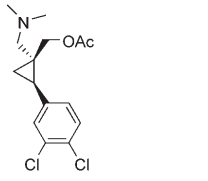
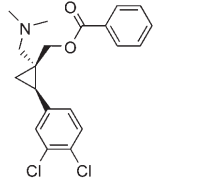
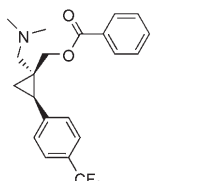
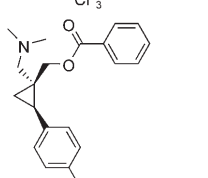
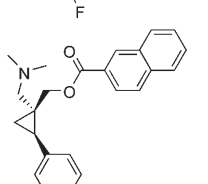
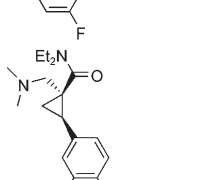
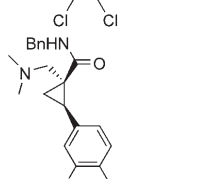
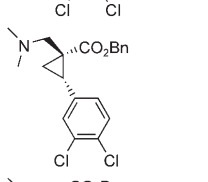
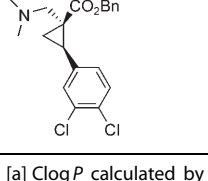
In recent studies on some conformationally constrained tropane analogues,<sup>[25–27]</sup> we found that replacement of the ester group with a benzyloxy-methyl group resulted in the most potent and selective SERT ligands. It was therefore reasonable to apply this type of modification to the current series as well. Acetate 14 is almost equipotent to ester 7d at the SERT, but it is somewhat less potent than ether 13. Benzoate 15 is more potent than acetate 14 and almost equipotent to ether 13. The SERT selectivity of benzoate 15 is, however, higher than that of 13 or 14. Benzoate 16 also displayed a twofold increase in SERT inhibition relative to the ester 7c, and it is more potent than the ether 12. The benzoate 17 was even more potent at the SERT, with a  $K_i$  value of 20 nM, which is 10-fold higher than that of ether 11. The most potent compound in the current series is naphthoate 18, which displayed a  $K_i$  value of 13 nM, a DAT/SERT uptake ratio of 366:1, and a NET/SERT uptake ratio of 248:1. However, its calculated log *P* is 5.7 (determined with Daylight software<sup>[43]</sup>), and thus this compound may not be the most suitable drug candidate in this series. The Clog *P* of benzoate 17 is 4.6, which falls within the range of typical druglike molecules.

Replacement of the ester group by an amide group caused a significant decrease in potency of SERT inhibition. Although

**Table 1.** Inhibition of Reuptake at Monoamine Transporters.

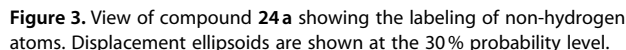
Compound	Clog $P^{[a]}$	[ $^3$ H]DA Uptake [nM] <sup>[b]</sup>	[ $^3$ H]5-HT Uptake [nM] <sup>[b]</sup>	[ $^3$ H]NE Uptake [nM] <sup>[b]</sup>	$K_{i(\text{DA})}/K_{i(\text{5-HT})}$	$K_{i(\text{NE})}/K_{i(\text{5-HT})}$
 cocaine	2.57	259 ± 19.9	155 ± 0.40	108 ± 3.50	1.7	0.7
 WIN 35,428	2.96	26.1	31.9	127	0.8	4.0
 paroxetine	4.24	–	0.25	312	–	1248
 citalopram	3.13	10 000	5.4	> 1000	4000	> 185
 nocaïne	3.22	233 ± 62	8490 ± 1430	252 ± 43	0.03	0.03
 (±)- <b>7 a</b>	2.54	> 10 000	1600 ± 290	> 10 000	> 6.3	> 6.3
 (±)- <b>7 b</b>	3.28	> 10 000	424 ± 42.1	> 10 000	> 23.6	> 23.6
 (±)- <b>7 c</b>	5.00	> 10 000	98.3 ± 7.2	> 10 000	> 10.2	> 10.2
 (±)- <b>7 d</b>	5.43	5950 ± 1030	55.4 ± 4.3	4560 ± 170	107	82.3
 (±)- <b>7 e</b>	6.60	9730 ± 1580	511 ± 38.0	5910 ± 10	19	11.5

Compound	Clog $P^{[a]}$	[ $^3$ H]DA Uptake [nM] <sup>[b]</sup>	[ $^3$ H]5-HT Uptake [nM] <sup>[b]</sup>	[ $^3$ H]NE Uptake [nM] <sup>[b]</sup>	$K_{i(\text{DA})}/K_{i(\text{5-HT})}$	$K_{i(\text{NE})}/K_{i(\text{5-HT})}$
 (±)- <b>7 f</b>	6.78	> 10000	539 ± 44.7	> 10000	> 18.6	> 18.6
 (±)- <b>7 g</b>	4.83	> 10000	297 ± 0.8	2020 ± 190	> 33.7	6.8
 (±)- <b>7 h</b>	4.53	1760 ± 90	435 ± 57.0	3730 ± 540	4.0	8.6
 (±)- <b>8</b>	1.88	> 10000	1750 ± 243	> 10000	5.7	5.7
 (±)- <b>9</b>	2.62	> 10000	1230 ± 144	> 10000	> 8.1	> 8.1
 (±)- <b>10</b>	3.04	7650 ± 211	221 ± 36	1740 ± 263	36.2	7.9
 (±)- <b>11</b>	4.53	1090 ± 100	272 ± 0.9	816 ± 60.1	4.0	3.0
 (±)- <b>12</b>	5.27	10200 ± 290	59.8 ± 11.7	2140 ± 200	170	35.8
 (±)- <b>13</b>	5.70	1800 ± 200	22.1 ± 4.5	451 ± 44.0	81.4	20.4

Compound	Clog <i>P</i> <sup>[a]</sup>	[ <sup>3</sup> H]DA Uptake [nM] <sup>[b]</sup>	[ <sup>3</sup> H]5-HT Uptake [nM] <sup>[b]</sup>	[ <sup>3</sup> H]NE Uptake [nM] <sup>[b]</sup>	<i>K</i> <sub>i(DA)</sub> / <i>K</i> <sub>i(5-HT)</sub>	<i>K</i> <sub>i(NE)</sub> / <i>K</i> <sub>i(5-HT)</sub>	
	(±)- <b>14</b>	3.95	9231 ± 993	53.7 ± 2.1	3039 ± 158	172	56.6
	(±)- <b>15</b>	5.70	9590 ± 1740	29.2 ± 4.4	2650 ± 116	328	90.8
	(±)- <b>16</b>	5.28	> 10 000	48.8 ± 5.6	> 10 000	205	205
	(±)- <b>17</b>	4.53	> 10 000	20.5 ± 2.4	4780 ± 215	> 488	233
	(±)- <b>18</b>	5.71	4900 ± 59	13.4 ± 3.0	3330 ± 326	366	248
	(±)- <b>20</b>	4.23	> 10 000	3780 ± 550	> 10 000	> 2.6	> 2.6
	(±)- <b>21</b>	4.68	> 10 000	1690 ± 100	> 10 000	> 5.9	> 5.9
	(-)- <b>7 d</b>	5.43	> 10 000	184 ± 1.0	> 10 000	> 54.3	> 54.3
	(+)- <b>7 d</b>	5.43	7720 ± 170	64.6 ± 6.9	3450 ± 220	119	53.4

[a] Clog*P* calculated by using the web-based program from Daylight (ref. [43]). [b] Data are mean *K*<sub>i</sub> ± SEM from two to four independent experiments, each consisting of six drug concentrations (in triplicate) that were selected on the basis of preliminary screening experiments to bracket the approximate IC<sub>50</sub> value.





In summary, a series of cyclopropane analogues was synthesized through a cyclopropanation reaction of a benzyldienemalonate. These cyclopropanes bear the same pharmacophore elements as those found in the WIN series of cocaine analogues. Through modifications of ester, aryl, and amino groups, ligands were identified that show good selectivity and potency at the SERT. The ether **13** and the benzoates **15** and **17** are roughly equipotent at the SERT with  $K_i$  values around 20 nM. The 2-naphthyl-bearing ligand **18** represents the most potent ligand in this series, with  $K_i$ =13 nM, a DAT/SERT uptake ratio of 366:1, and a NET/SERT uptake ratio of 248:1. Ether **13** likely represents one of the more promising candidates of the series presented herein for in vivo study, owing to its relative metabolic stability. As both selective and mixed monoamine transporter reuptake inhibitors have proven useful as medications in the treatment of depression, anxiety and eating disorders, schizophrenia, Alzheimer's disease, and other CNS disorders,<sup>[44, 45]</sup> we believe that the compounds described herein are worthy of continued investigation. Accordingly, further pharmacological experiments are planned along with assessments of the behavioral activity of selected compounds.

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