

[Chem. Pharm. Bull.]  
29(9) 2526—2530 (1981)

# The Chemistry of Lactim Ethers. V.<sup>1)</sup> Reaction of Lactim Thioethers with $\beta$ -Aminoesters<sup>2)</sup>

HIROKI TAKAHATA, AKIRA TOMOGUCHI, and TAKAO YAMAZAKI\*

Faculty of Pharmaceutical Sciences, Toyama Medical and Pharmaceutical University, Sugitani 2630, Toyama 930-01, Japan

(Received March 9, 1981)

Reaction of a cyclic  $\beta$ -aminoester such as 2-ethoxycarbonylmethylpiperidine (**5**) with the lactim thioethers **1b** and **1c** gave a 10-membered cyclic diamide (**7**) and an 11-membered ring compound (**9**), respectively. On the other hand, though acyclic  $\beta$ -aminoesters (**6**) reacted with **1** to afford methyl 3-methylthiopropionate (**11**), N-alkyl 3-methylthiopropionamide (**12**), lactams (**13**), and amidines (**14**), no cyclic diamide was obtained.

**Keywords**—lactim thioethers; cyclic  $\beta$ -aminoesters; acyclic  $\beta$ -aminoesters; cyclic diamide; medium-ring; methyl 3-methylthiopropionate; N-alkyl 3-methylthiopropionamide; amidines; lactams;  $\beta$ -lactams

In recent publications,<sup>1,3,4)</sup> we reported the annulation reaction of lactim thioethers (**1**) and lactim ethers (**2**) with cyclic  $\beta$ -aminoesters. Interestingly, the annulation of **1** with  $\beta$ -aminoesters (**3**) and (**4**) gave exclusively the imine type products and provided two kinds of products, the cyclic diamide (**I**) and the enaminoamide (**II**). On the basis of these results, we examined the cyclization of lactim thioethers (**1**) with  $\beta$ -aminoesters lacking an aromatic moiety, such as 2-ethoxycarbonylmethylpiperidine (**5**), and acyclic  $\beta$ -aminoesters (**6**) in anticipation of the predominant formation of the type **I** product having a medium-sized ring.

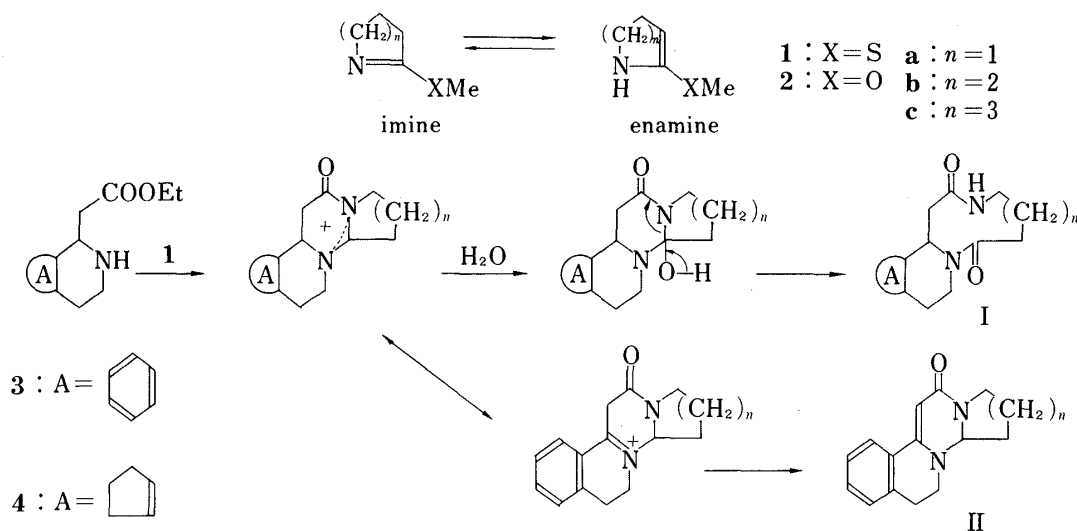


Chart 1

First, treatment of **1b** with **5** in a sealed tube at 100—110°C for 10 days afforded a 10-membered ring diamide (**7**) (mp 186—188°C) in 42% yield together with **8** (mp 297—298°C)<sup>3)</sup> in 3% yield, attributable to the enamine form of **1b**; these products were isolated by alumina column chromatography. The spectral data of **7** [ $\nu_{\text{max}}^{\text{KBr}}$  NH (3300  $\text{cm}^{-1}$ ) and C=O (1650  $\text{cm}^{-1}$ ); PMR  $\delta$  (DMSO- $d_6$ ) 7.5—9 (NH, 1H, broad singlet)], MS  $m/e$  224 ( $\text{M}^+$ ), and elemental analysis (molecular formula  $\text{C}_{12}\text{H}_{20}\text{N}_2\text{O}_2$ ) supported this structure. A similar annulation of **1c** with **5** gave only an 11-membered diamide (**9**) (mp 190—192°C)<sup>3)</sup> in 53% yield. However, reaction of with **5** did not give a 9-membered ring product but furnished a non-cyclized product (**10**)

(mp 35—38°C) in a low yield. This compound (**10**) gave appropriate MS [ $m/e$  286 ( $M^+$ )] and elemental analysis (molecular formula  $C_{14}H_{26}N_2O_2S$ ). In the IR spectrum of **10**, a band due to NH ( $3330\text{ cm}^{-1}$ ) and one attributable to C=O ( $1710\text{ cm}^{-1}$ ) were observed. The PMR spectrum of **10** exhibited a triplet (3H,  $J=7\text{ Hz}$ ) at  $\delta$  1.27 and a quartet (2H,  $J=7\text{ Hz}$ ) at  $\delta$  4.17, a singlet (3H) at  $\delta$  2.03, and a broad singlet (1H) at  $\delta$  5.40 indicating ethylester, SMe, and NH, respectively.

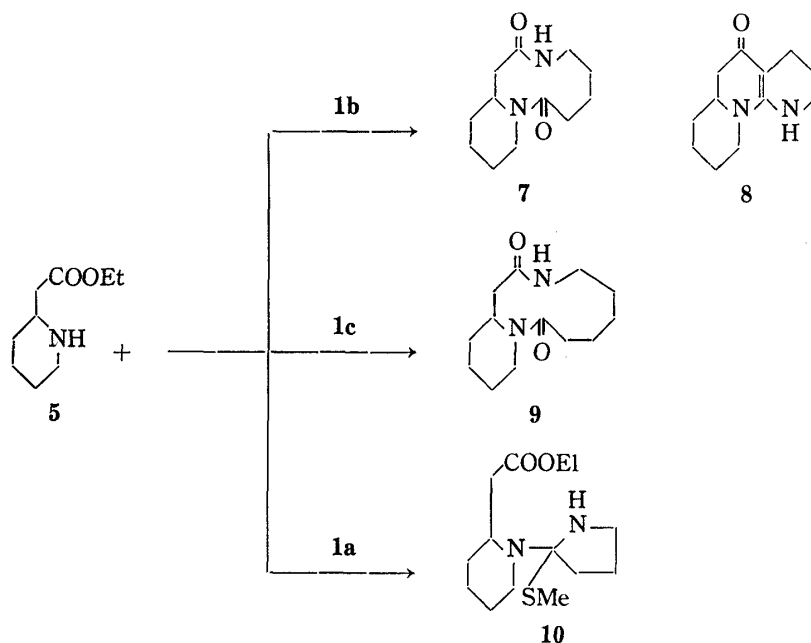


Chart 2

Next, the reaction of acyclic  $\beta$ -aminoester (**6**) with **1** was carried out in a similar manner. However, the desired products could not be obtained. In the reaction of **6a** with **1a**, a mixture of methyl 3-methylthiopropionate (**11**) (bp  $65^\circ\text{C}/6\text{ mmHg}$ ), N-benzyl 3-methylthiopropionamide (**12a**) (mp  $50\text{--}51^\circ\text{C}$ ), butyrolactam [**13**( $n=1$ )], and N-benzyl 2-iminopyrrolidine [**14a**( $n=1$ )] was obtained. Both **11** and **12a** were characterized spectroscopically and gave satisfactory elemental analyses. All reactions of **1a**, **b**, **c** with acyclic  $\beta$ -aminoesters (**6a**, **b**, **c**) afforded similar products. The results are summarized in Table I.

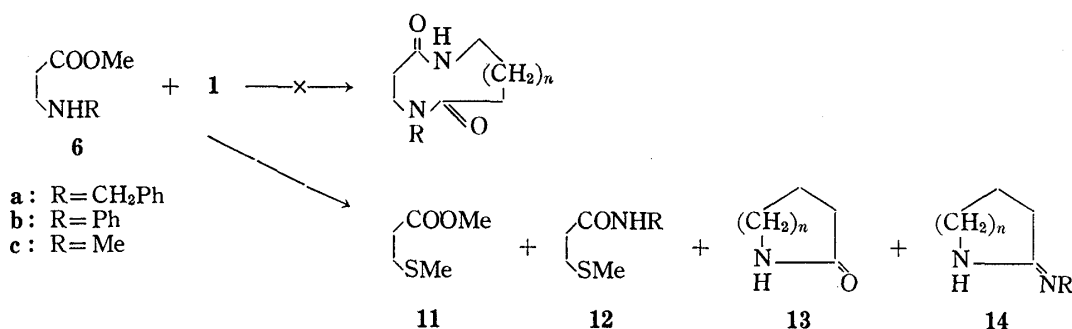


Chart 3

In order to investigate the mechanism of these reactions, we carried out the following experiment. The reaction of amidine [**14a**( $n=1$ )] with methyl 3-ethylthiopropionate (**15**) afforded an amide (**16**) (mp  $49\text{--}50^\circ\text{C}$ ) and **13** ( $n=1$ ) in 56% and 62% yields, respectively. A possible mechanism for the formation of the products obtained by the reaction of acyclic  $\beta$ -aminoesters with **1** is shown in Chart 4; *i.e.*, the first stage might involve the formation of a

TABLE I. Reaction of **1** with **6** Yields (%)<sup>a</sup> and Melting Points (mp) or Boiling Points (bp) for Products

| <b>6</b>                        | <b>1</b>    | <b>11</b><br>(bp 65°C/6mmHg) | <b>12</b><br>(bp 65°C/6mmHg) | (mp)   | <b>13</b> | <b>14</b> | (mp)                 |
|---------------------------------|-------------|------------------------------|------------------------------|--------|-----------|-----------|----------------------|
| <b>a</b> , R=CH <sub>2</sub> Ph | <i>n</i> =1 | 36                           | 26                           | (51°C) | 66        | 31        | (79°C) <sup>b</sup>  |
|                                 | <i>n</i> =2 | 21                           | 48                           | (51°C) | 67        | 28        | (95°C)               |
|                                 | <i>n</i> =3 | 45                           | 39                           | (51°C) | 55        | 41        | (71°C) <sup>c</sup>  |
| <b>b</b> , R=Ph                 | <i>n</i> =1 | 22                           | —                            | —      | 43        | 40        | (113°C) <sup>d</sup> |
|                                 | <i>n</i> =3 | 56                           | —                            | —      | 28        | 50        | (107°C) <sup>e</sup> |
| <b>c</b> , R=Me                 | <i>n</i> =1 | 14                           | 32                           | (35°C) | 34        | 36        | (95°C) <sup>f</sup>  |

<sup>a</sup>) The yields of (**11** and **12**) and (**13** and **14**) are based on **6** and **1**, respectively.<sup>b</sup>) Lit.<sup>6</sup>) mp 82°C. <sup>c</sup>) Lit.<sup>5</sup>) mp 76°C. <sup>d</sup>) Lit.<sup>6</sup>) mp 115° <sup>e</sup>) Lit.<sup>6</sup>) mp 107°C. <sup>f</sup>) Lit.<sup>7</sup>) mp 95°C.

1:1 adduct (**17**) as an intermediate. This view is supported by the formation of compound (**10**) in Chart 2. Subsequently, the intermediate (**17**) would be concertedly cleaved to produce the amidine (**14**) and **11**. The next stage would involve the condensation of **14** with **11** to afford **12a**, **c** and **13** (preliminary results). However, because the nucleophilicity of nitrogen of **14b** would be weakened by a phenyl group, N-phenyl 3-methylthiopropionamide (**12b**) could not be obtained. Next, the  $\beta$ -aminoester (**6a**) was treated with ethanethiol in both the presence and absence of triethylamine, but no replacement of the benzylamino group by an ethylthio group took place. This suggested that the  $\beta$ -aminoester did not undergo self retro-Michael reaction to the acrylate.

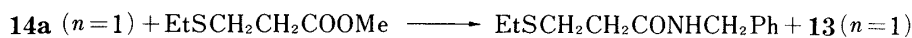
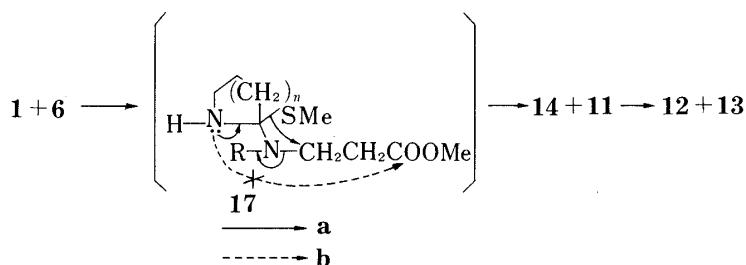
**15****16**

Chart 4

Reaction of ethyl 3-benzylamino-3-phenylpropionate (**18**) with **1a** gave a similar product; *i.e.*, a mixture of ethyl 3-methylthio-3-phenylpropionate (**19**), N-benzyl 3-methylthio-3-phenylpropionamide (**20**), **13a** (*n*=1), and **14a** (*n*=1). In the case of acyclic  $\beta$ -aminoesters

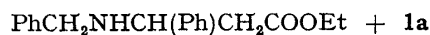
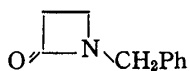
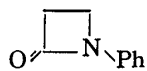
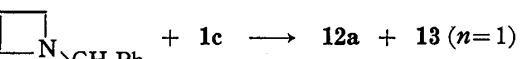
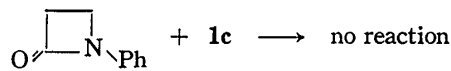
**18****19****20****21****22**

Chart 5

it was found that annulation as shown by the arrow (---->b) never occurred, in contrast to the results for the cyclic  $\beta$ -aminoesters.

Next,  $\beta$ -lactam rings are well known to react with an amino group to cleave the  $N_1$ - $C_2$  bond.<sup>8)</sup> 1-Benzyl-2-azetidinone (21) was thus allowed to react with 1c, but the unexpected amide (12a)<sup>9)</sup> (60%) and 13 ( $n=1$ ) were obtained. On the other hand, reaction of 1-phenyl-2-azetidinone (22) with 1c resulted in recovery of the  $\beta$ -lactam in spite of the use of a higher temperature.

### Experimental

Infrared spectra were determined on a Jasco IRA 1 spectrophotometer and are given in units of  $\text{cm}^{-1}$ . Proton NMR spectra were determined in the indicated solvent on a JEOL C-60H machine and chemical shifts are given in  $\delta$  units downfield from internal  $\text{Me}_4\text{Si}$ . Splitting patterns are designated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; brs, broad singlet. Mass spectra were obtained on a JEOL OLSG spectrometer. Melting points were determined on a Yanaco MP apparatus and are uncorrected.

**Piperidino[1,2-*a*][1,5]diazecane-2,8-dione (7) and 1,2,3,4,6,6a,7,8,9,10-Decahydro-5H-pyrido[1,2-*a*][1,8]-naphthyldin-5-one (8)**—A mixture of 1b (2.58 g) and 5<sup>10)</sup> (1.71 g) was heated in a sealed tube at 100–110°C for 10 d. The resulting precipitate was collected by filtration and washed with  $\text{CH}_2\text{Cl}_2$  to give 8 (65 mg, 3.2%). The filtrate was concentrated *in vacuo* to leave an oil, which was purified by column chromatography on alumina with  $\text{CHCl}_3$ :EtOH (30:1) as an eluant to afford 7 (937 mg, 41.8%). 7: mp 186–188°C (recrystallized from ethyl acetate-isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3300, 1650. NMR ( $\text{DMSO}-d_6$ ): 7.5–7.9 (1H, brs, NH). MS  $m/e$ : 224 ( $M^+$ ). Anal. Calcd for  $\text{C}_{12}\text{H}_{20}\text{N}_2\text{O}_2$ : C, 64.25; H, 8.99; N, 12.49. Found: C, 64.47; H, 9.23; N, 12.23. 8: mp 292–294°C (recrystallized from ethyl acetate-MeOH) (lit.<sup>3)</sup> mp 290–292°C). The product was identical with an authentic sample prepared from methyl valerolactim (2b),<sup>3)</sup> with respect to spectroscopic data.

**Piperidino[1,2-*a*][1,5]diazacycloundecane-2,9-dione**—A mixture of 5 (1 g) and 1c (1.68 g) was heated in a sealed tube at 100–110°C for 10 d. The reaction mixture was separated by column chromatography on alumina to give 9 (730 mg, 52.9%) using  $\text{CHCl}_3$ :EtOH (30:1) as an eluant. mp 170–172°C (recrystallized from ethyl acetate) (lit.<sup>3)</sup> 170–172°C). The product was identical with an authentic sample, with respect to spectral data.

**2-Ethoxycarbonylmethyl-1-(2'-methylthiopyrrolidinyl)piperidine (10)**—A mixture of 1a (1.30 g) and 5 (1.55 g) was heated in a sealed tube at 100°C for 7 d. The reaction mixture was purified by column chromatography on alumina to afford 10 (190 mg, 6.6%) using  $\text{CHCl}_3$ :EtOH (10:1) as an eluant. mp 35–38°C (recrystallized from petro. ether-isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3330 (NH), 1710 (C=O). NMR ( $\text{CDCl}_3$ ): 1.27 (3H, t,  $J=7$  Hz), 2.03 (3H, s), 4.17 (2H, q,  $J=7$  Hz), 5.4 (1H, brs). MS  $m/e$ : 286 ( $M^+$ ). Anal. Calcd for  $\text{C}_{14}\text{H}_{26}\text{N}_2\text{O}_2\text{S}$ : C, 58.71; H, 9.15; N, 9.78. Found: C, 58.99; H, 8.89; N, 9.45.

**General Procedure for the Reaction of 1a, b, c with Acyclic  $\beta$ -Aminoesters (6a, b, c)**—A mixture of 1 (0.01 mol) and 6 (0.01 mol) was heated in a sealed tube at 100–110°C for 7 d. The reaction mixture was separated by column chromatography on alumina with benzene (A),  $\text{CHCl}_3$  (B), and then  $\text{CHCl}_3$ :MeOH (30:1) (C) as eluents to give methyl 3-methylthiopropionate (11) (A) and N-alkyl 3-methylthiopropionamide (12) (B) in the early fraction, lactams (13) (B) in the later fraction, and the amidine (14) (C). 11: (bp 65°C/6 mmHg) (lit.<sup>11</sup>). bp 170–180°C). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 1740. NMR ( $\text{CDCl}_3$ ): 2.13 (3H, s, SMe), 2.67 (4H, m), 3.73 (3H, s, OMe). N-Benzyl 3-methylthiopropionamide (12a): mp 50–51°C (recrystallized from isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3280, 1630. NMR ( $\text{CDCl}_3$ ): 2.13 (3H, s, SMe), 2.60 (4H, m), 4.43 (2H, d,  $J=6$  Hz), 6.50 (1H, brs), 7.33 (5H, s). Anal. Calcd for  $\text{C}_{11}\text{H}_{15}\text{NOS}$ : C, 63.12; H, 7.22; N, 6.69. Found: C, 63.20; H, 7.08; N, 6.63.

N-Methyl 3-Methylthiopropionamide (12c): mp 33–35°C (recrystallized from isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3280, 1630, 1570. NMR ( $\text{CDCl}_3$ ): 2.10 (3H, s, SMe), 2.8 (3H, d,  $J=5$  Hz), 6.80 (1H, brs). Anal. Calcd for  $\text{C}_5\text{H}_{11}\text{NOS}$ : C, 45.10; H, 8.33; N, 10.52. Found: C, 45.42; H, 8.49; N, 10.14. Spectral data for butyrolactam [13 ( $n=1$ )], valerolactam [13 ( $n=2$ )], and caprolactam [13 ( $n=3$ )] were identical with those of authentic samples. Melting points of 2-benzyliminopyrrolidine [14a ( $n=1$ )], 2-benzyliminoazacycloheptane [14a ( $n=3$ )], 2-phenyliminopyrrolidine [14b ( $n=1$ )], 2-phenyliminoazacycloheptane [14b ( $n=3$ )], and 2-methyliminopyrrolidine [14c ( $n=1$ )] are shown in Table I.

2-Benzyliminopiperidine [14a ( $n=2$ )]: mp 92–95°C (recrystallized from  $\text{CH}_2\text{Cl}_2$ -isopropylether). IR  $\nu_{\text{max}}^{\text{KBr}}$ : 3400, 1620. NMR ( $\text{CDCl}_3$ ): 4.33 (2H, s), 5.3 (1H, s), 7.33 (5H, s). Anal. Calcd for  $\text{C}_{12}\text{H}_{16}\text{N}_2$ : C, 76.55; H, 8.57; N, 14.88. Found: C, 76.86; H, 8.42; N, 14.68.

**Reaction of 1a with 18**—A mixture of 1a (0.6 g) and 18 (1.20 g) was heated in a sealed tube at 100–110°C for 7 d. The reaction mixture was separated by column chromatography on alumina to give ethyl 3-methylthio-3-phenylpropionate (19) (470 mg, 56%) using benzene as an eluant. 19: bp 81°C/6 mmHg. IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 1740. NMR ( $\text{CDCl}_3$ ): 1.17 (3H, t, 6 Hz), 1.91 (3H, s), 2.90 (2H, d,  $J=8$  Hz), 4.10 (1H, t,  $J=8$  Hz), 4.20 (2H, q,  $J=6$  Hz), 7.40 (5H, s). Anal. Calcd for  $\text{C}_{12}\text{H}_{16}\text{O}_2\text{S}$ : C, 64.26; H, 7.19. Found: C, 63.89; H, 7.17. 20: mp 98–100°C (recrystallized from isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3280, 1640, 1560. NMR

(CDCl<sub>3</sub>): 1.90 (3H, s), 2.73 (2H, d,  $J=8$  Hz), 4.23 (3H, m), 7.33 (11H, m). *Anal.* Calcd for C<sub>17</sub>H<sub>19</sub>NOS: C, 71.54; H, 6.71; N, 4.51. Found: C, 71.70; H, 6.87; N, 4.73.

**Reaction of 14a ( $n=1$ ) with 15**—A mixture of 14a ( $n=1$ ) (1.74 g) and 15 (1.48 g) was heated in a sealed tube at 100–110°C for 7 d. The reaction mixture was separated by column chromatography on alumina to afford N-benzyl 3-ethylthiopropionamide (16) (1.25 g, 56%) and 13 ( $n=1$ ) (530 mg, 62%) using CHCl<sub>3</sub> as an eluant. 16: mp 49–50°C (recrystallized from isopropylether). IR  $\nu_{\text{max}}^{\text{Nujol}}$ : 3240, 1630, 1560. NMR (CDCl<sub>3</sub>): 1.20 (3H, t,  $J=6$  Hz), 4.33 (2H, d,  $J=6$  Hz), 7.3 (5H, s). *Anal.* Calcd for C<sub>12</sub>H<sub>17</sub>NOS: C, 64.53; H, 7.67; N, 6.27. Found: C, 64.46; H, 7.53; N, 6.11.

**Reaction of 21 with 1c**—A mixture of 21 (1.61 g)<sup>12)</sup> and 1c (1.43 g) was heated in a sealed tube at 140°C for 3 d. The reaction mixture was separated by column chromatography on alumina to afford 12a (1.25 g, 59.8%) and 13 ( $n=3$ ) (1.01 g, 88.5%) using CHCl<sub>3</sub> as an eluant.

**Acknowledgement** This work was supported in part by a grant from the Foundation for the Promotion of Research on Medicinal Resources.

### References and Notes

- 1) Part IV: H. Takahata, M. Ishikura, K. Nagai, M. Nagata, and T. Yamazaki, *Chem. Pharm. Bull.*, **29**, 366 (1981).
- 2) This work was presented at the 51th Meeting of the Hokuriku Branch of the Pharmaceutical Society of Japan, Kanazawa, November, 1980.
- 3) H. Takahata, M. Ishikura, and T. Yamazaki, *Chem. Pharm. Bull.*, **28**, 220 (1980).
- 4) H. Takahata, A. Tomiguchi, and T. Yamazaki, *Chem. Pharm. Bull.*, **28**, 1000 (1980).
- 5) J.R. Geigy, *Fr.* 1367799.
- 6) H. Breadreck and K. Breadreck, *Chem. Ber.*, **94**, 2278 (1961).
- 7) A. Etienne and Y. Corria, *Compt. Rend.*, **259**, 2660 (1964).
- 8) a) B.J.R. Nicolaus, E. Bellasio, G. Pagan, L. Mariani, and E. Testa, *Helv. Chim. Acta.*, **48**, 1867 (1965); b) D. Bormann, *Chem. Ber.*, **103**, 1797 (1970); c) M.S. Manhas, S.G. Amin, and A.J. Bose, *Heterocycles*, **5**, 669 (1976); d) H.H. Wassermann, R.P. Robinson, and H. Matsuyama, *Tetrahedron Lett.*, **21**, 3493 (1980).
- 9) The formation of 12a would be effected by cleavage of the N<sub>1</sub>–C<sub>4</sub> bond of the  $\beta$ -lactam instead of that of the N<sub>1</sub>–C<sub>2</sub> bond, since the amidine [14a ( $n=1$ )] was not isolated.
- 10) H. Takahata, M. Hara, A. Tomiguchi, T. Yamazaki, and R.N. Castle, *J. Heterocycl. Chem.*, **17**, 403 (1980).
- 11) E. Adlerora and M. Protiva, *Collect. Czech. Chem. Commun.*, **24**, 1268 (1959).
- 12) H. Takahata, Y. Ohnishi, H. Takehara, K. Tsuritani, and T. Yamazaki, *Chem. Pharm. Bull.*, **29**, 1063 (1981).