

Total Synthesis of (±)-Carpesiolin

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The synthetic scheme is shown in Chart 1. The features of this route are; i) complete control of the C-10 methyl group in α -orientation in a perhydroindanone, ii) a regiospecific



ring expansion to a perhydroazulenone and iii) complete control of other chiralities on the 7-membered ring.

The readily available α -methylene ketone (2)⁴⁾ was hydrogenated [H_2 (3 atm)/PtO₂/MeOH] and equilibrated (MeONa/MeOH) to give stereospecifically the desired α -methyl ketone (3)⁵⁾ in 77% yield. The crucial regiospecific ring expansion was examined using 3 and the derivatives thereof, but the best result was obtained by using Nozaki's procedure;⁶⁾ the reaction of 3 with dibromomethyl lithium generated *in situ* [lithium diisopropylamide (LDA)-CH₂Br₂/THF/-78°C], followed by treatment with butyllithium, gave the perhydroazulenone (4) in 40% yield.^{7,8)}

The regiospecific introduction of the double bond into 4 was accomplished using phenylselenenylation-deselenenylation sequence [i) LDA/THF, PhSeBr, ii) H₂O₂/THF] in 68% yield. The stereospecific reduction of the enone (5) thus obtained was achieved with LiAlH₄ in THF to give the allylic alcohol (6) in 90% yield. Its stereochemistry was deduced from the results of reduction in similar cases.¹⁾ Henbest-type epoxidation of 6 (*m*-chloroperbenzoic acid/CH₂Cl₂), followed by nucleophilic epoxide ring opening by excess dilithioacetate (DME-HMPA/55°C), gave the desired γ -lactone (7) in 44% yield.

Treatment of 7 with trifluoroacetic anhydride in pyridine followed by *p*-TsOH in boiling toluene (to remove the *tert*-butyl group), afforded the unexpected hydroxy-ester (8) in 89% yield. Its structure was established as follows: in its PMR spectrum the methine proton at C-6 was found to resonate at δ 3.47 (t, $J=8$ Hz; d, $J=8$ Hz upon D₂O addition), in contrast to the value of 5.10 (d, $J=8$ Hz) for the trifluoroacetate of 7.⁹⁾ Although the occurrence of 8 was quite surprising, its functionalities were suitable for the subsequent steps.

The conversion of 8 to the lactone (9) was achieved in the usual manner [i) dihydropyran-pyridinium *p*-toluenesulfonate/CH₂Cl₂, ii) 5% K₂CO₃/THF] in 95% yield. Grieco's methylenation sequence [i) LDA/THF, CH₂O, ii) MsCl/pyridine, iii) DBU/C₆H₆] was successfully applied to 9 to give the α -methylene lactone (10) in 56% yield. The final sequence of reactions [i) pyridinium chlorochromate (PCC)-AcONa/CH₂Cl₂, ii) *p*-TsOH/C₆H₆-H₂O] gave ($\pm$)-1 in 50% yield, whose IR and PMR spectra were completely identical with those of the natural product.^{10,11)} Application of the scheme to the synthesis of other helenanolides is in progress.

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References and Notes

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- 2) M. Maruyama and S. Omura, *Phytochemistry*, **16**, 782 (1977). The structure was established by synthesis; see ref. 1d).
- 3) F. Bohlmann and P.K. Mahanta, *Phytochemistry*, **18**, 887 (1979).
- 4) Z.G. Hajos, D.R. Parrish, and E.P. Oliveto, *Tetrahedron*, **24**, 2039 (1968); R.A. Micheli, Z.G. Hajos, N. Cohen, D.R. Parrish, L.A. Portland, W. Sciamanna, M.A. Scott, and P.A. Wehrli, *J. Org. Chem.*, **40**, 675 (1975).
- 5) All new compounds formulated in this report showed satisfactory spectral data and elemental analyses.
- 6) H. Taguchi, H. Yamamoto, and H. Nozaki, *Bull. Chem. Soc. Jpn.*, **50**, 1588, 1592 (1977).
- 7) The structure of this compound was ascertained by comparison with the C-9 oxo-regioisomer, prepared in an alternative way.
- 8) During this work Schlessinger *et al.* synthesized this compound *via* a different route, which was converted into helenalin; see ref. 1c).
- 9) Further supporting evidence for the structure of 8 were: i) cycloheptanone formation (IR: 1700 cm⁻¹) upon PCC oxidation, ii) acetonide formation [Me₂C(OMe)₂-*p*-TsOH/Me₂CO] from its corresponding diol,

indicating the α -orientation of hydroxyl group at C-4. This type of acyl migration with concomitant inversion of the hydroxyl group may best be explained by the acyl group participation as A shown in Chart 1.

- 10) Our synthetic compound showed the following characteristics. Mp 197—200°C (CHCl₃-hexane).¹¹⁾ IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3470, 1765, 1730. PMR (CDCl₃) δ : 6.22 (1H, d, J =3.6 Hz), 5.99 (1H, d, J =3.2 Hz), 4.38 (1H, ddd, J =2.9, 9.8, 10.7 Hz), 4.01 (1H, dd, J =8.8, 3.0 Hz; d, J =8.8 Hz on D₂O addition), 2.94 (1H, d, J =3.0 Hz, disappeared on D₂O addition), 1.09 (3H, d, J =6.1 Hz), 1.03 (3H, s).
- 11) The mp of ($\pm$)-carpesiolin reported in ref. 1*d*) is 145—148°C (sublimes). In a private communication Vandewalle explained that their sample probably includes water of crystalization.

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