## Communications to the Editor

Chem. Pharm. Bull. 31(2) 756—759 (1983)

SYNTHESIS OF DIOXAZOLOI3.31CYCLOPHANES BY CYCLIZATION OF BIS(2-ISOCYANO-2-TOSYLETHYL)BENZENES WITH BENZENEDICARBALDEHYDES

Hideaki Sasaki\* and Tokujiro Kitagawa
Faculty of Pharmaceutical Sciences, Kobe Gakuin University
Ikawadani, Nishi-ku, Kobe 673, Japan

Bis(2-isocyano-2-tosylethyl)benzenes  $(\underline{4})$  reacted with benzenedicarbaldehydes  $(\underline{9})$  in the presence of sodium ethoxide as a base in refluxing ethanol to form dioxazolo-I3.3lcyclophanes (10).

KEYWORDS ———— cyclophane; tosylmethyl isocyanide; isocyanide; aldehyde; oxazole; cyclization

The condensation of tosylmethyl isocyanide (TosMIC;  $\underline{1a}$ ) or mono-substituted TosMIC ( $\underline{1b}$ ) with aldehydes ( $\underline{2}$ ) leading to 5-substituted or 4,5-disubstituted oxazoles ( $\underline{3a}$  or  $\underline{3b}$ ) has been reported in detail. 1)

X : o-, m-, and p-xylylene

In continuation of our study on the preparation of [3.3]cyclophanes (8) by the reaction of bis(2-isocyano-2-tosylethyl)benzenes  $(\underline{4})^2$  with bis(bromomethyl)benzenes (5) followed by the hydrolysis of  $\underline{6}$  and Wolff-Kishner reduction of  $\underline{7}$ , we have effected the first preparation of dioxazolo[3.3]cyclophanes ( $\underline{10}$ ) from the intermolecular cyclization of  $\underline{4}$  with benzenedicarbaldehydes ( $\underline{9}$ ) in the presence of sodium ethoxide as a base in refluxing ethanol.

Thus, when 1,2- and 1,3-bis(2-isocyano-2-tosylethyl)benzenes ( $\frac{40}{2}$  and  $\frac{4m}{2}$ ) were caused to react with 1,3-benzenedicarbaldehyde ( $\frac{9m}{2}$ ), the corresponding dioxazolot3.31cyclophanes ( $\frac{100m}{2}$  and  $\frac{10mm}{2}$ ) were produced in 84% and 77% yields, respectively. Similarly, 1,4-bis(2-isocyano-2-

Y : o-, m-, and p-phenylene

tosylethyl)benzene ( $\frac{4p}{}$ ) was treated with  $\frac{9m}{}$  to give the corresponding cyclophane of type [3.3],  $\frac{10pm}{}$ , in 15% yield along with a 2:2 adduct 3) in 52% yield.

With respect to the reaction mechanism, we have rationalized as follows: the first nucleophilic addition of a bis-TosMIC derivative anion ( $\underline{11}$ ) to one of carbonyl carbons of  $\underline{9}$  takes place under basic condition to provide an intermediate ( $\underline{12}$ ). Subsequent conversion of  $\underline{12}$  to an oxazole ( $\underline{13}$ ) is induced and the following intramolecular cyclization of  $\underline{13}$  occurs to prepare  $\underline{10}$  as a final product.

Attempts to ascertain the sensitivity toward the structural changes of the reaction species were carried out in a practicable combination of the bis-TosMIC derivatives (4) with 1,2- or 1,4-benzenedicarbaldehydes (90 or 9p). Namely, when 90 was used as a carbonyl component, 40 led to 1000 in 2% yield along with a large amount of polymerized materials. 4m and 4p, however, did not afford the corresponding dioxazolo[3.3]cyclophanes (10m0 and 10p0), probably because the structure of the intermediate (13) is too strained to form a [3.3]cyclophane ring. 4) In the choice of 9p as a carbonyl component, neither 40 nor 4p afforded our desired compounds (100p and 10pp) under the above-mentioned condition, whereas 4m gave the corresponding dioxazolo[3.3]metaparacyclophane (10mp) in 85% yield.

The structural assignments of the synthesized dioxazolot 3.31 cyclophanes ( $\underline{1000}$ ,  $\underline{-om}$ ,  $\underline{-mm}$ ,  $\underline{-pm}$ , and  $\underline{-mp}$ ) were achieved mainly by elemental analyses and  $^1\text{H-NMR}(\text{CDCl}_3)$  and mass spectroscopies. The  $^1\text{H-NMR}$  spectrum of each compound of type  $\underline{10}$  exhibited a C-2 proton of each oxazole ring as a singlet at around  $\delta$  7.70-7.86. These  $\delta$  values were similar to those of the 4,5-disubstituted oxazoles already reported. Furthermore, the inner aryl proton (Hc) of  $\underline{10pm}$  was observed as a broad singlet at  $\delta$  5.30, whereas the corresponding protons (Ha and Hb) of  $\underline{10om}$  and  $\underline{10mm}$  were observed at  $\delta$  7.70 and  $\delta$  6.75, respectively. This up-field shift of Hc is presumably attributed

Table I			
CH <sub>2</sub> CH \sqrt{N=C} Tos		СНО	
	<u>90</u>	<u>9m</u>	<u>9p</u>
<u>40</u>		Han	( <u>10op</u> )
	1000 yield: 2 % mp: 205-6°C	10om yield: 84 % mp: 263-4°C NMR: δ 7.70(Ha)	
<u>4m</u>	( <u>10mo</u> ) <sup>a)</sup>	Hb Hd	He
		10mm yield: 77 % mp: 195-6°C NMR: δ 6.75(Hb) δ 7.50(Hd)	10mp yield: 85 % mp : 246-8°C NMR : δ 6.10(He)
		Hc	
<u>4p</u>	( 10po ) <sup>a)</sup>		( <u>10pp</u> )
		10pm	
		yield: 15 % mp: 190-2°C NMR: δ 5.30(Hc)	

to shielding of Hc by the opposite benzene ring. Incidentally, the inner aryl proton (He) of  $\underline{10mp}$ , which is isomeric with  $\underline{10pm}$ , was observed as a broad singlet at  $\delta$  6.10. The 0.80 ppm down-field shift compared with the  $\delta$  value (5.30) of Hc for  $\underline{10pm}$  suggests that the meta-substituted benzene ring of  $\underline{10mp}$  stands on a site in somewhat gapped surroundings due to the greater pliability of the methylene group than that of the double bond of the oxazole ring linked up with the corresponding meta-substituted benzene ring of 10pm.

To the best of our knowledge, this facile preparation in one step provides a route to a new class of [3.3]cyclophanes, and a typical experiment for the synthesis of 10mm is described as follows: a solution of 1,3-benzenedicarbaldehyde (9m) (10 mmol) in ethanol (50 ml) was added dropwise to a suspension of 1,3-bis(2-isocyano-2-tosylethyl)benzene (4m) (10mmol) in ethanol (550 ml) containing sodium ethoxide (Na; 20 mmol) with stirring at room temperature. The resulting mixture was refluxed for 2 h, and then ethanol was removed under reduced pressure. The resulting residue was poured into water containing acetic acid (20 mmol), then extracted with ethyl acetate. After the organic solvent was removed under reduced pressure, the resultant residue was purified by recrystallization from benzene to give 10mm (mp 195-6°C) in 77% yield. 10mm showed the following spectral data;  $1\text{H-NMR}(\text{CDCl}_3)$   $\delta$ :  $4.16(4\text{H}, \text{br s}, -\text{CH}_2-)$ , 6.75(4H, m, ph-H) contained Hb), 7.15(3H, m, ph-H), 7.50(1H, br s, Hd), 7.86(2H, s, oxazole C2-H); 1R(KBr)  $\nu$ :  $3140 \text{ cm}^{-1}(\text{oxazole C2-H})$ ; MS(m/e):  $314(\text{M}^+)$ .

## REFERENCES AND NOTES

- a) A. M. van Leusen, B. E. Hoogenboom, and H. Siderius, Tetrahedron Lett., 1972, 2369;
   b) H. Saikachi, T. Kitagawa, H. Sasaki, and A. M. van Leusen, Chem. Pharm. Bull., 27, 793(1979);
   c) O. Possel and A. M. van Leusen, Heterocycles, 7, 77(1977).
- 2) a) H. Sasaki and T. Kitagawa, Chem. Pharm. Bull., accepted; b) K. Kurosawa, M. Suenaga, T. Inazu, and T. Yoshino, Tetrahedron Lett., 23, 5335(1982).
- 3) The structure of this 2:2 adduct was assigned as tetraoxazolo[3.3.3.3]metaparametaparacyclophane, which was confirmed by elemental analysis and the following spectral data; IR(KBr) v: 3140 cm<sup>-1</sup> (oxazole C2-H) MS(m/e): 628(M<sup>+</sup>).
- 4) Instead of 10mo and 10po, 1:2 adducts (14a and 14b) were isolated in 74% and 84% yields, which were calculated on the basis of 9. 14a and 14b were fully characterized by  $^{1}$ H-NMR, IR, and mass spectroscopies and elemental analyses. Data for 14a;  $^{1}$ H-NMR(CDCl $_{3}$ )  $\delta$ : 8.60-8.70 and 7.50-8.20(14H, m, aromatic-H), 8.24(2H, s, oxazole C2-H); IR(KBr)  $\nu$ : 3130 cm $^{-1}$ (oxazole C2-H); MS(m/e): 412(M $^{+}$ ). Data for 14b;  $^{1}$ H-NMR(CDCl $_{3}$ )  $\delta$ : 7.50-8.25(14H, m, aromatic-H), 8.27(2H, s, oxazole C2-H); IR(KBr)  $\nu$ : 3130 cm $^{-1}$ (oxazole C2-H); MS(m/e): 412(M $^{+}$ ).

$$\begin{array}{cccc}
R & & & & R \\
R & & & & & R \\
\hline
 & & & & & & R \\
\hline
 & & & & & & & R \\
\hline
 & & & & & & & & & R \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & & \\
\hline
 & & & & & & & & & & \\
\hline
 & & & & & & & & & & \\
\hline
 & & & & & & & & & & \\
\hline
 & & & & & & & & & \\
\hline
 & & & & & & & & & \\
\hline
 & & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 & & & & & & & \\
\hline
 &$$

(Received December 10, 1982)