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A Novel Oxidation Product Formed by the Oxidation of Di(1-propenyl)tetramethoxybiphenyl with CrO₃-HBF₄-MeCN

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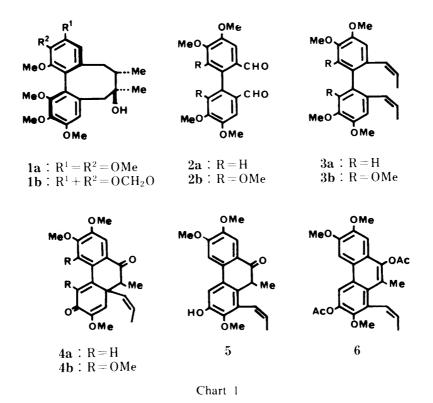
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Oxidation of the di(1-propenyl)-biphenyl 3a with the CrO₃-HBF₄-MeCN reagent system gave a novel oxidation product 4a, which was subsequently transformed to the phenanthrenes 5 and 6

Keywords—coxidation; chromium trioxide; di(1-propenyl)-biphenyl; cross-conjugated dienone; phenanthrene

There is considerable interest in the synthesis of dibenzocyclooctadiene lignans, schizandrin 1a and gomisins (1b: gomisin A), in connection with their interesting pharmacological activities, ¹⁾ namely, antitussive, tonic, inhibition of gastric ulceration, and tranquilizing activities. Although several syntheses of dibenzocyclooctadiene lignans have been reported already, ²⁾ we investigated oxidation of the tetramethoxy-(Z)-di(1-propenyl)biphenyl 3a with a new type of CrO₃ reagent system, ³⁾ CrO₃-HBF₄-MeCN, continuing our previous work on the syntheses of the tetrahydronaphthalene and tetrahydrofuran neo-lignans⁴⁾ and podophyllum lignans⁵⁾ with the use of this reagent system, and we found that 3a gives a novel oxidation product 4a instead of the expected dibenzocyclooctenone.



No. 1

The tetramethoxy-(Z)-di(1-propenyl)biphenyl 3a, mp 90—92 °C, was synthesized from the corresponding dialdehyde 2a, which was prepared by Ullmann reaction of 6-bromoveratraldehyde with ethyltriphenyl phosphonium bromide in the presence of K₂CO₃ and 18-crown-6.⁶⁾ Oxidation of 3a with the CrO₃-HBF₄-MeCN system gave a cross-conjugated dienone 4a, mp 115—117 °C, in 46% yield. The structure of 4a was assigned from its spectral data and the following chemical transformations. Treatment of 4a with SnCl₄ in AcOH gave the phenol 5, mp 170—172 °C, by dienone-phenol rearrangement in 63% yield.⁷⁾ The phenol 5 gave the diacetate 6, mp 157—159 °C, on acetylation with Ac₂O in pyridine. On the other hand, oxidation of hexamethoxy-(Z)-di(1-propenyl)-biphenyl 3b with the same reagent system gave no identifiable products, presumably because of the instability of the corresponding dienone 4b in acidic media, or *peri* interaction of the methoxy group.

Experimental

All melting points are uncorrected. Infrared (IR) spectra were recorded with a Hitachi 260-10 spectrometer, nuclear magnetic resonance (NMR) spectra with a JEOL JNM-FX 100 spectrometer with tetramethylsilane as an internal standard (CDCl₃ soln.) and mass spectra (MS) with a JEOL JMS-D 300 spectrometer. Elemental analyses were done by Ms. M. Takeda and Ms. S. Okamura, Kissei Pharmaceutical Company, Matsumoto, Japan. Mallinckrodt silica gel (100 mesh) and Merck Kieselgel 60 F₂₅₄ were used for column chromatography and thin-layer chrometography (TLC), respectively.

Oxidation of 3a with CrO₃-HBF₄-MeCN—A solution which was prepared by adding MeCN (17.6 ml) to a mixture of CrO₃ (220 mg, 2.2 mmol) and aqueous 42% HBF₄ (4.4 ml) was added at room temperature to a solution of 3a (354 mg, 1 mmol) in MeCN (12 ml), and the whole was stirred at room temperature for 30 s. The reaction mixture was poured into ice-water and extracted with ether. The organic layer was washed with sat. NaHCO₃ and H₂O, then dried and concentrated. The residue was subjected to silica gel chromatography. The eluate with 40% hexane in chloroform gave 162.8 mg (46%) of 2,6,7-trimethoxy-10-methyl-10a-(Z)-(1-propenyl)phenanthrene-3,9(10H, 10aH)-dione 4a as colorless crystals (chloroform–ether), mp 115—117 °C. IR (Nujol): 1680, 1650, 1630, 1590 cm⁻¹. NMR (CDCl₃) δ : 1.36 (3H, d, J = 5.61 Hz, C = C-Me), 1.52 [3H, d, J = 6.59 Hz, C(10)-Me], 2.62 [1H, q, J = 6.59 Hz, C(10)-H], 3.78, 3.97, 4.00 (each 3H, s, 3 × OMe), 5.22—5.57 (2H, m, olefinic H), 5.99 (1H, s, C(1)-H), 6.77, 7.04, 7.52 (each 1H, s, C(4)-H, and aromatic H). *Anal.* Calcd for C₂₁H₂₂O₅: C, 71.17; H, 6.26. MS m/e, 354.1468. Found: C, 71.08; H, 6.24. MS m/e: 354.1496.

3-Hydoxy-2,6,7-trimethoxy-10-methyl-(Z)-1-(1-propenyl)phenanthrene-9(10H)-one (5)——SnCl₄ (624 mg, 2.4 mmol) was added to a solution of 4a (213 mg, 0.6 mmol) in AcOH (15 ml), and the whole was stirred overnight at room temperature. The reaction mixture was poured into ice-water and extracted with chloroform. The organic layer was washed with sat. NaHCO₃ and H₂O, then dried and concentrated. The residue was subjected to silica gel chromatography. The eluate with 10% hexane in chloroform gave 134 mg (63%) of 5 as colorless crystals (chloroform-ether), mp 170—172 °C. IR (Nujol): 3400, 1660, 1600 cm⁻¹. NMR (CDCl₃) δ : 1.29 (3H, d, J=7.32 Hz, C(10)–Me), 1.60 (3H, dd, J=6.59 and 1.46 Hz, -C=C-Me), 3.74 (1H, q, J=7.32 Hz, C(10)–H), 3.75, 3.97, 4.05 (each 3H, s, 3 × OMe), 5.97 (1H, dq, J=11.23 and 6.59 Hz, olefinic H), 6.41 (1H, dd, J=11.23 and 1.46 Hz, olefinic H), 7.30, 7.44, 7.50 (each 1H, s, aromatic H). *Anal.* Calcd for C₂₁H₂₂O₅. MS m/e: 354.1466. Found: 354.1461.

3,9-Diacetoxy-2,6,7-trimethoxy-10-methyl-(*Z*)**-1-(1-propenyl)phenanthrene (6)**——Ac₂O (1.5 ml) was added to a solution of **5** (45 mg, 0.127 mmol) in dry pyridine (0.5 ml) and the whole was stirred overnight at room temperature. The reaction mixture was worked up in the usual manner to yield 52.9 mg (95%) of **6** as colorless crystals from ethanol, mp 157—159 °C. IR (Nujol): 1740 cm⁻¹. NMR (CDCl₃) δ : 1.37 (3H, dd, J=6.84 and 1.71 Hz, -C=C-Me), 2.42, 2.47, 2.49 (each 3H, s, C(10)–Me and 2×OCOMe), 3.73, 4.00, 4.08 (each 3H, s, 3×OMe), 5.88 (1H, dq, J=11.23 and 6.84 Hz, olefinic H), 6.96 (1H, dd, J=11.23 and 1.71 Hz, olefinic H), 7.08, 7.77, 8.16 (each 1H, s, aromatic H). *Anal.* Calcd for C₂₅H₂₆O₇: C, 68.48; H, 5.98. MS m/e: 438.1676. Found: C, 68.42; H, 5.95. MS m/e: 438.1669.

References and Notes

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