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SYNTHESIS OF NEW ANTINEOPLASTIC PROSTAGLANDINS1)

Satoshi Sugiura, ^a Takeshi Toru, ^a Toshio Tanaka, ^a Atsuo Hazato, ^a Noriaki Okamura, ^a Kiyoshi Bannai, ^a Kenji Manabe, ^a Seizi Kurozumi, *, ^a and Ryoji Noyori^b

Institute for Bio-Medical Research, Teijin Ltd., a
4-3-2 Asahigaoka, Hino, Tokyo 191, Japan and
Department of Chemistry, Nagoya University, b
Chikusa, Nagoya 464, Japan

The three-component coupling process allows a single-pot entry to protected 7-hydroxy-PGE $_1$ derivatives, leading to a variety of new PGs functionalized or unsaturated at C-7. The enone and dienone derivatives exhibit potent inhibitory activity on L1210 tumor cell growth <u>in vitro</u>.

KEYWORDS — 7-hydroxy-PGA₁; Δ^7 -PGA₁; Δ^7 -PGE₁; 12-<u>epi</u>- Δ^7 -PGA₁; prostaglandin; antineoplastic

The relationship between tumor cell growth and prostaglandin (PG) formation is an attractive problem for scientists in a variety of areas. Recent in vitro and in vivo studies indicate that some naturally occurring PGs such as PGE_1 , PGE_2 ,

Synthesis of these analogs starts from the protected 7-hydroxy-PGE $_1$ derivatives (8), aldol products easily accessibly by the three-component coupling process mediated by organocopper chemistry. $^{12,13)}$ The absolute configurations of the hydroxylated carbons of 8a and 8b were substantiated by chemical correlation with the 5,6-unsaturated derivatives 9^{14} and 10, coupled with the exciton chirality method. 16 The less polar $7\underline{S}$ -isomer 9a and the more polar $7\underline{R}$ -isomer 9b were

$$5a: R^1 = H, R^2 = OH$$

 $5b: R^1 = OH, R^2 = H$

 $7a: R^1 = Si(CH_3)_2 - \underline{t} - C_4H_9,$ $R^2 = CH_3$

 $7b: R^1 = H, R^2 = CH_3$

 $7c: R^1 = H, R^2 = H$

10a: $R^1 = H$, $R^2 = OCOC_6H_5$, $R^3 = Si(CH_3)_2 - \underline{t} - C_4H_9$ 10b: $R^1 = OCOC_6H_5$, $R^2 = H$,

 $R^3 = Si(CH_3)_2 - t - C_4H_9$

COOR

 $6a: R = CH_3$ 6b: R = H

8a: $R^1 = H$, $R^2 = OH$, $R^3 = Si(CH_3)_2 - t - C_4H_9$

8b: $R^1 = OH$, $R^2 = H$, $R^3 = Si(CH_3)_2 - \underline{t} - C_4H_9$

 $8c: R^1 = H, R^2 = OH, R^3 = H$ $8d: R^1 = OH, R^2 = H, R^3 = H$

 $\underbrace{11}: R = Si(CH_3)_2 - \underline{t} - C_4H_9$

 $\frac{6c}{6d}: R = CH_3$

 $\frac{9a: R^1 = H, R^2 = OH,}{R^3 = Si(CH_3)_2 - t - C_4H_9}$

9b: $R^1 = OH$, $R^2 = H$, $R^3 = Si(CH_3)_2 - t - C_4H_9$

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converted to the corresponding allyl benzoates, $10a\ (44\%)^{15}$ and $10b\ (84\%)$, respectively, by catalytic half hydrogenation of the triple bond over the Lindlar catalyst (CH₃OH, room temp, 20 h) followed by benzoylation (C₆H₅COC1, pyridine, room temp, 4 h). Measurements of the CD spectra of the benzoates indicated that $10a\ [\Delta \epsilon = +10.5\ (226\ nm,\ cyclohexane)]$ and $10b\ [\Delta \epsilon = -7.37\ (226\ nm,\ cyclohexane)]$ have $7S\ and\ 7R\ configuration$, respectively, confirming the C-7 configuration of the precursors $9a\ and\ 9b$. Catalytic hydrogenation of the triple bond of the $7S\ isomer\ 9a\ and\ the\ 7R\ isomer\ 9b\ over\ Pd\ CaCO_3\ catalyst\ (CH_3OH,\ room\ temp,\ 30\ h)$ afforded the 5,6-saturated alcohols, $8a\ (50\%)^{15}$ and $8b\ (84\%)$, having $7R\ and\ 7S\ configuration$, respectively. Both stereoisomers serve as useful intermediates for the preparation of antineoplastic artificial PGs.

The first targets, $7\underline{R}$ -hydroxy-PGA $_1$ methyl ester (5a) and the $7\underline{S}$ -hydroxy isomer (5b), were obtained from 8a and 8b. The starting materials have two kinds of hydroxyl groups β to the 9-keto function. In a 2:1:1 CH $_3$ COOH-THF-H $_2$ O mixture (70°C, 16 h), dehydration of 8a and 8b occurred preferentially in such a way as to create the 2-cyclopentenone structure rather than the 2-alkylidenecyclopentanone moiety to give 5a (35%) and 5b (36%), respectively. ¹⁵⁾ In order to compare the biological activities of the 2-cyclopentenones and ketones conjugated with an exocyclic double bond, Δ^7 -PGE $_1$ derivatives of type 7 were then prepared by aldol dehydration using CH $_3$ SO $_2$ Cl (1 eq) and 4-dimethylaminopyridine (2 eq) (CH $_2$ Cl $_2$, room temp, 6 h). ¹²⁾ Dehydration of 8b gave the 7E-isomer 7a (80%) (1 H-NMR, H-7, δ 6.88) as a major

product accompanied by a small amount of the $7\underline{Z}$ -isomer 11 (<5%) (1H -NMR, H-7, 6 5.84). The reaction of 8a gave the stereoisomers (7 2a, 11) in a similar ratio (6 4%). 17) Deprotection of 7a was made by HF-pyridine (6 CH₃CN, room temp, 1 h) to afford 6 PGE₁ methyl ester (7 b) (8 1%), which was then hydrolyzed with porcine liver esterase. 19) Acid treatment (6 2:1:1 CH₃COOH-THF-H₂O mixture, 6 0°C, 15 h) of 7b gave 6 PGA₁ methyl ester (6 4a) (7 2%) having a cross-conjugated dienone structure. As a by-product, the 7 2-isomer 12 was isolated (7 5%). Enzymatic hydrolysis of 6 4a afforded 7 -PGA₁ (6 6b) (6 7%). The most stable products among the compounds (7 5), (6 6), (7 7) was the cross-conjugated dienone (6 6).

As expected, when the three-component coupling process was carried out using the ω -side chain unit of the "natural" C-15 S configuration and (±)-4- \underline{t} -butyldimethylsilyloxy-2-cyclopentenone instead of the optically active 4R compound, a 1:1 mixture of 6a and 12- \underline{epi} - Δ^7 -PGA₁ methyl ester (6c) was obtained after dehydration. Enzymatic hydrolysis of the latter gave the free carboxylic acid 6d (63%).

The preliminary biological tests indicated that these newly synthesized PG analogs having a functionalized or unsaturated carbon at the 7-position exhibited significant inhibitory activity (IC₅₀ 0.3 μ g/ml) on L1210 tumor cell growth. Particularly, Δ^7 -PGA₁ methyl ester (6a) and its C-12 epimer (6c) were the most active among the analogs synthesized. They were more potent than natural PGs such as PGA₁, PGE₁, and PGD₂, and were equipotent to clavulone in this evaluation.²⁰⁾

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- 13) Reaction of (\underline{R}) -4- \underline{t} -butyldimethylsilyloxy-2-cyclopentenone (1 eq) and the mixed cuprate formed from $(\underline{E},\underline{S})$ -3- \underline{t} -butyldimethylsilyloxy-1-octen-1-yllithium (1 eq), 1-pentynylcopper (1 eq), and hexamethylphosphorous triamide (2 eq) at -78°C followed by condensation of the resultant enolate with methyl 7-oxoheptanoate yielded 8a (12%) and 8b (42%).
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- 17) There was no equilibrium between 7a and 11 under the reaction condition. This stereoselectivity, leading to the same isomer 7a, is explainable by the consideration that the elimination of the methanesulfonate group occurred via the ElcB mechanism¹⁸⁾ involving the less hindered enolate conformations, A and B. The allylic type repulsion between the α -side chain and the oxygen at C-9 is avoided in such conformations.

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