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Synthesis of Condensed Quinoxalines. VI.¹⁾ Synthesis of 1*H*-Pyrazolo[3,4-*b*]quinoxaline N-Oxides and Related Compounds

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Oxidation of 1H-pyrazolo[3,4-b]quinoxalines (1a: R = H, 1b: $R = CH_3$) with m-chloroperbenzoic acid (MCPBA) gave the 4-oxides (2a, b). The structures of 2a, b were confirmed by synthesis, by condensing 2-chloroquinoxaline-3-carbaldehyde 4-oxide (6) with appropriate hydrazines. Further oxidation of 2a with MCPBA gave the 4,9-dioxide (8).

Treatment of 1,2-dihydro-2-oxoquinoxaline-3-carboxamide (10) and 1,2-dihydro-2-oxoquinoxaline-3-carbonitrile 4-oxide (13) with a mixture of POCl₃ and PCl₅ or POCl₃-dimethyl-formamide afforded 2-chloroquinoxaline-3-carbonitrile (11) and its 4-oxide (14), respectively. When 11 and 14 were reacted with hydrazines, the corresponding 3-amino-1*H*-pyrazolo[3,4-*b*]-quinoxalines (12a, b) and their 4-oxides (15a, b) were obtained in high yields.

The reaction of ethyl 2-chloroquinoxaline-3-carboxylate (16) with hydrazine hydrate afforded a mixture of uncyclized products, N,N'-bis(2-ethoxycarbonyl-3-quinoxalinyl)hydrazine (17), ethyl 2-hydrazinoquinoxaline-3-carboxylate (18) and 2-hydrazinoquinoxaline-3-carbohydrazide (19).

Keywords—1*H*-pyrazolo[3,4-*b*]quinoxaline 4-oxide; 1*H*-pyrazolo[3,4-*b*]quinoxaline 4,9-dioxide; 3-amino-1*H*-pyrazolo[3,4-*b*]quinoxaline 4-oxide; 2-chloroquinoxaline-3-carbaldehyde 4-oxide; hydrazine; N,N'-bis(2-ethoxycarbonyl-3-quinoxalinyl)hydrazine; ethyl 2-hydrazinoquinoxaline-3-carboxylate; 2-hydrazinoquinoxaline-3-carbohydrazide

As a part of our continuing studies on the synthesis of condensed quinoxalines, 1H-pyrazolo[3,4-b]quinoxalines and their N-oxides were synthesized with a view to evaluating their biological activities. Some 1-substituted 1H-pyrazolo[3,4-b]quinoxalines (1: R=H, CH₃, C_6H_5) were prepared from 2-bromomethyl-3-chloroquinoxaline by treatment with hydrazines by Beyer,²⁾ but the N-oxides were not obtained. The present paper deals with the N-oxidation of 1a, b and with synthesis of this ring system by taking advantage of the reactive functionalities at the 2 and 3 positions of quinoxaline.

The oxidation of 1a with m-chloroperbenzoic acid (MCPBA) in chloroform solution formed a sole mono-oxide (2a) in 74% yield. The structure of 2a was confirmed by the three-step synthesis outlined in Chart 1.

The similar oxidation of 2-chloro-3-methylquinoxaline (3) using MCPBA gave a single mono-oxide (4) in 82% yield. Treatment of 4 with sodium ethoxide afforded the 2-ethoxy derivative (5), which was identical with 2-ethoxy-3-methylquinoxaline 4-oxide prepared by the oxidation of 2-ethoxy-3-methylquinoxaline with hydrogen peroxide in acetic acid. The structural assignment of 5 was established by Newbold *et al.*³⁾ and Otomasu *et al.*⁴⁾ When the resulting product (4) was oxidized with selenium oxide in benzene, a yellow product (6) was obtained in 72% yield. The infrared (IR) spectrum of 6 showed a C=O absorption band at $1700 \, \text{cm}^{-1}$, and the nuclear magnetic resonance (NMR) spectrum of 6 indicated the presence of an aldehyde proton, δ 10.54 (1H, s).

These spectral and analytical data were consistent with the 2-chloroquinoxaline-3-

carbaldehyde 4-oxide structure (6).

Compound 6 was easily reacted with hydrazine hydrate to give a yellow solid (2a). The NMR spectrum of 2a showed signals at δ 8.74 (1H, s) and 14.8 (1H, br), which could be assigned to 3-H and 1-H, respectively, of the pyrazole ring. The compound thus obtained 2a was identical with the foregoing sample prepared by the oxidation of 1a with MCPBA on the basis of mixed fusion and comparison of NMR spectra. Similarly, the oxidation of 1b with MCPBA afforded a mono-oxide (2b), which was identical with 1-methyl-H-pyrazolo[3,4-b]quinoxaline 4-oxide prepared by the condensation of 6 with methylhydrazine. The reaction of 6 with phenylhydrazine produced only the phenylhydrazone (7), but this was cyclized into a condensed pyrazolo compound (2c) in good yield by treatment with potassium hydroxide in dimethylformamide (DMF).

Chart 1

The mono-oxide 2a was oxidized with MCPBA to give a slightly soluble dioxide (8) in 90% yield. The mass spectrum (MS) of 8 showed the molecular ion peak (M⁺) at m/e 202 and two deoxygenated fragment peaks at 186 (M⁺-O) and 170 (M⁺-O₂). In the NMR spectrum of 8, the signals of 1-H and 8-H appeared at δ 15.01 and 8.47, respectively, being shifted 0.2 ppm downfield from those of 2a. These results were rationalized in terms of the magnetic anisotropy effect of the N-O bond at the 9-position. Based on these spectral data, the structure of 8 was concluded to be 1H-pyrazolo[3,4-b]quinoxaline 4,9-dioxide. Similar oxidations of 2b, c did not give their dioxides.

The title compounds were also synthesized by alternative procedures. In a previous paper,⁵⁾ we reported that 2-chloro-3-cyano-substituted pyridine and quinoline compounds reacted with hydrazine hydrate to form 3-amino-1*H*-pyrazolo[3,4-*b*]quinoline compounds, respectively. Using this procedure, 3-amino-1*H*-pyrazolo[3,4-*b*]quinoxaline and their N-oxides were prepared.

Ethyl 1,2-dihydro-2-oxoquinoxaline-3-carboxylate⁶⁾ (9) reacted with ammonium carbonate to form the carboxamide (10) in ca. 90% yield. When 10 was heated with a mixture of PCl₅ and POCl₃, chlorination and dehydration occurred and 2-chloroquinoxaline-3-

carbonitrile (11) was obtained in 85% yield. Refluxing of an ethanolic solution of 11 with hydrazine hydrate gave an orange-red solid (12a) in 83% yield. The IR spectrum of 12a showed absorption bands due to NH₂ and NH groups at 3400, 3340 and 3200 cm⁻¹, and these were also apparent in the NMR spectrum as singlets (exchangeable with D₂O) at δ 6.12 (2H) and 12.21 (1H), respectively. Similarly, the reaction of 11 with methylhydrazine gave 12b in 77% yield.

Confirmation of structure of 12a, b was obtained by the following experiments. On treatment of 1,2-dihydro-2-oxoquinoxaline-3-carbonitrile 4-oxide⁷⁾ (13) with a mixture of POCl₃ and DMF in toluene solution followed by chromatography, the major product (14) was obtained in 45% yield. The reaction of 14 with appropriate hydrazines gave the products 15a, b as in the reaction of 11. Deoxygenation of 15a, b with PCl₃ afforded the products 12a, b, which were shown to be identical with the products obtained from the condensation of 11 with hydrazines.

Finally, the condensation of ethyl 2-chloroquinoxaline-3-carboxylate⁸⁾ (16) with hydrazine hydrate, which was expected to produce the homologous compound in a similar manner, was carried out. The reaction did not give the corresponding product (20) but a mixture of uncyclized products, N,N'-bis(2-ethoxycarbonyl-3-quinoxalinyl) hydrazine (17), ethyl 2-hydrazinoquinoxaline-3-carboxylate (18) and 2-hydrazinoquinoxaline-3-carbohydrazide (19). The formation ratio of these products depended on the reaction conditions. The structures of 17—19 were supported by the spectral and analytical data. Attempts to obtain a cyclized product under various conditions failed.

Experimental

All melting points were determined in a capillary and are uncorrected. The following instruments were used; IR spectra, Hitachi 215 spectrometer; MS, Hitachi RMS-4 and JEOL D-300 machines at 70 eV; NMR spectra, JEOL FX-100 spectrometer with tetramethylsilane as the internal standard.

- **1H-Pyrazolo[3,4-b]quinoxaline 4-Oxide (2a)**—i) Oxidation of **1a**. MCPBA (85%, 0.12 g) was added to a suspension of **1a**²¹ (0.1 g) in CHCl₃ (5 ml), and the mixture was stirred at 40 °C for 3 h. The precipitate was collected and recrystallized from MeOH to give 0.08 g (74%) of **2a** as yellow needles, mp > 300 °C. NMR (DMSO- d_6) δ : 7.84—7.94 (2H, m, H-6 and H-7), 8.22—8.63 (2H, m, H-5 and H-8), 8.75 (1H, s, H-3), 14.79 (1H, br, NH). *Anal*. Calcd for $C_9H_6N_4O$: C, 58.06; H, 3.25; N, 30.10. Found: C, 57.93; H, 3.10; N, 30.02.
- ii) Condensation of 6 with hydrazine hydrate. Compound 6 (0.13 g) was added to hydrazine hydrate (1.5 ml) and the mixture was heated at 70 °C with stirring for 15 min. The separated product was recrystallized from MeOH to give yellow needles of 2a. Yield, 70%. The IR and NMR spectra were superimposable on those of the above sample.
- **1-Methyl-1***H*-pyrazolo[3,4-*b*]quinoxaline 4-Oxide (2*b*)—i) Oxidation of $1b^2$) (0.1 g) with MCPBA (85%, 0.12 g) as described for the preparation of 1a, followed by recrystallization from MeOH gave yellow flakes, mp 189—190 °C. Yield, 78%. NMR (DMSO- d_6) δ : 4.12 (3H, s, CH₃), 7.65—7.99 (2H, m, H-6 and H-7), 8.05—8.55 (2H, m, H-5 and H-8), 8.65 (1H, s, H-3). *Anal*. Calcd for $C_{10}H_8N_4O$: C, 59.99; H, 4.03; N, 27.99. Found: C, 60.23; H, 3.92; N, 27.94.
- ii) Condensation of 6 (0.18 g) with methylhydrazine (2 ml) as described for the preparation of 1a gave yellow flakes of 2b, which were identical with the above sample.
- **2-Chloro-3-methylquinoxaline 4-Oxide (4)**—MCPBA (85%, 1.14g) was added to a solution of **3** (1.0 g) in CHCl₃ (100 ml), and the mixture was stirred at room temperature for 6 h. The CHCl₃ solution was washed with aqueous NaHCO₃, dried over MgSO₄ and evaporated to dryness. The product was recrystallized from MeOH to give 0.89 g (82%) of **4** as colorless needles, mp 95—96 °C. NMR (CDCl₃) δ : 2.80 (3H, s, CH₃), 7.69—7.86 (2H, m, H-6 and H-7), 7.89—8.56 (2H, m, H-5 and H-8). *Anal.* Calcd for C₉H₇ClN₂O: C, 55.17; H, 3.46; N, 14.56. Found: C, 55.54; H, 3.62; N, 14.39.
- 2-Ethoxy-3-methylquinoxaline 4-Oxide (5)——A solution of Na (0.03 g) in EtOH (5 ml) was added to a solution of 5 (0.2 g) in EtOH (3 ml), and the mixture was refluxed for 30 min. The salt that separated was removed and the solution was evaporated to dryness. The product was recrystallized from EtOH to give colorless prisms, mp 86—87 °C, which were identical with an authentic sample⁴⁾ (mixed mp and IR spectral comparison).
- **2-Chloroquinoxaline-3-carbaldehyde 4-Oxide (6)**—A mixture of **4** (1.0 g) and SeO₂ (0.7 g) and in benzene (30 ml) was refluxed under stirring, during which time a black precipitate formed. The mixture was filtered, the filtrate was evaporated, and the residual product was recrystallized from benzene to give 0.77 g (72%) of **6** as yellow needles, mp 165—166 °C. IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1700 (C=O). NMR (CDCl₃) δ : 7.70—8.12 (3H, m, H-6, H-7 and H-8), 8.44—8.54 (1H, m, H-5), 10.54 (1H, s, CHO). *Anal.* Calcd for C₉H₅ClN₂O₂: C, 68.69; H, 3.84; N, 21.34. Found: C, 68.85; H, 3.74; N, 21.50.
- **2-Chloroquinoxaline-3-carbaldehyde Phenylhydrazone 4-Oxide (7)**—Compound **6** (1.0 g) was added to phenylhydrazine (5 ml) with stirring at room temperature. The precitate that formed immediately was corrected and recrystallized from MeOH to give 1.35 g (94%) of 7 as dark red needles, mp 220—221 °C (dec.). NMR (DMSO- d_6) δ : 6.89—8.02 (8H, m, Ar-H), 8.33—8.43 (1H, m, H-5), 8.53 (1H, s, CH=N), 11.45 (1H, s, NH). *Anal.* Calcd for $C_{15}H_{11}ClN_4O$: C, 60.36; H, 3.71; N, 18.77. Found: C, 60.21; H, 3.64; N, 18.74.
- 1-Phenyl-1*H*-pyrazolo[3,4-*b*] quinoxaline 4-Oxide (2c)—An aqueous solution of KOH (20%, 4 ml) was added to an ice-cooled solution of 7 (1.0 g) in DMF (20 ml) with stirring, and the stirring was continued for 30 min. The reaction mixture was poured into ice-water, and the resulting precipitate was collected and recrystallized from EtOH to give 0.8 g (91%) of 2c as yellow needles, mp 190—191 °C. NMR (DMSO- d_6) δ : 7.40—8.36 (8H, m, Ar-H), 8.52—8.60 (1H, m, H-5), 8.89 (1H, s, H-3). *Anal.* Calcd for $C_{15}H_{10}N_4O$: C, 68.69; H, 3.84; N, 21.34. Found: C, 68.85; H, 3.74; N, 21.50.
- 1H-Pyrazolo[3,4-b]quinoxaline 4,9-Dioxide (8)—MCPBA (85%, 0.4 g) was added to a suspension of 2a (0.1 g) in CHCl₃ (5 ml), and the mixture was warmed at 40 °C for 30 min. The precipitate was recrystallized from MeOH to give red needles, mp > 300 °C (dec.). Yield, 90% MS m/e: 202 (M⁺), 186 (M⁺-O), 170 (M⁺-O₂). NMR (DMSO- d_6) δ : 7.79—7.82 (2H, m, H-6 and H-7), 8.49—8.59 (2H, m, H-5 and H-8), 8.71 (1H, s, H-3), 15.01 (1H, br, NH). Good agreement with the calculated values was not obtained on elemental analysis, possibly because of explosive burning.
- 1,2-Dihydro-2-oxoquinoxaline-3-carboxamide (10)—A mixture of 9^{6} (2.0 g) and (NH₄)₂CO₃ (4.0 g) in MeOH (20 ml) was allowed to stand overnight at room temperature under stirring. The solvent was removed, and the residual product was washed with water to give 1.6 g (89%) of 10 as pale yellow needles, mp 265 °C (dec.). (MeOH). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3350—3150 (NH, NH₂), 1690 (C=O). *Anal.* Calcd for C₉H₇N₃O₂: C, 57.14; H, 3.73; N, 22.21. Found: C, 56.72; H, 3.99; N, 22.15.
- **2-Chloroquinoxaline-3-carbonitrile (11)**—A mixture of **10** (5.4 g), PCl₅ (12 g) and POCl₃ (4.4 g) was heated at 110 °C for 1.5 h. The reaction mixture was poured into ice-water and the resulting precipitate was collected. The product was taken up in benzene and run through an Al₂O₃ column to give 4.5 g (85%) of **11** as colorless needles, mp 159—160 °C. MS m/e: 189 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 2220 w (CN). *Anal*. Calcd for C₉H₄ClN₃: C, 57.01; H, 2.12; N, 22.16. Found: C, 56.77; H, 2.45; N, 21.96.
- 3-Amino-1*H*-pyrazolo[3,4-*b*]quinoxaline (12a)—Hydrazine hydrate (0.26g) was added to a solution of 11 (0.5g) in EtOH (30 ml), and the mixture was refluxed for 30 min, then allowed to cool. The precipitate was collected, washed with water and recrystallized from EtOH to give 0.4g (83%) of 12a as orange-red leaflets, mp 292—293 °C.

MS m/e: 185 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3400, 3340, 3210 (NH, NH₂). NMR (DMSO- d_6) δ : 6.12 (2H, s, NH₂), 12.2 (1H, s, NH). Anal. Calcd for C₉H₇N₅: C, 58.37; H, 3.81; N, 37.82. Found: C, 58.28; H, 3.57; N, 37.34.

3-Amino-1-methyl-1*H*-pyrazolo[3,4-*b*]quinoxaline (12b)—Treatment of 11 (0.5 g) with methylhydrazine (0.28 g) as described above gave 0.4 g (77%) of 12b as orange-red needles, mp 178 °C (EtOH). MS m/e: 199 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3410, 3340, 3210 (NH, NH₂). *Anal.* Calcd for $C_{10}H_9N_5$: C, 60.29; H, 4.55; N, 35.16. Found: C, 60.63; H, 4.55; N, 34.81.

2-Chloroquinoxaline-3-carbonitrile 4-Oxide (14) —A mixture of 13^{7} (3.0 g), POCl₃ (16.7 g) and DMF (1.8 g) in toluene (40 ml) was refluxed for 2 h. After cooling, the reaction mixture was poured into ice-water, and the toluene layer was collected and evaporated to dryness. The residual product was taken up in benzene and run through an Al₂O₃ column to give 2.1 g (45%) of **14** as pale yellow leaflets, mp 192 °C. MS m/e: 205 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 2220 w (CN). NMR (CDCl₃) δ : 7.81—8.16 (3H, m, Ar-H), 8.42—8.53 (1H, m, H-5). *Anal*. Calcd for C₉H₄ClN₃O: C, 52.57; H, 1.96; N, 20.43. Found: C, 52.79; H, 2.49; N, 20.31.

3-Amino-1*H*-pyrazolo[3,4-*b*] quinoxaline 4-Oxide (15a) — Hydrazine hydrate (0.26 g) was added to a solution of 14 (0.5 g) in EtOH (30 ml) and the mixture was refluxed for 30 min, then allowed to cool. The precipitate was filtered off and recrystallized from EtOH to give 15a as dark red needles, mp 276 °C. Yield, 0.46 g (93%). MS m/e: 201 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3420, 3290, 3170 (NH, NH₂). NMR (DMSO- d_6) δ : 6.08 (2H, br, NH₂), 12.41 (1H, s, NH). *Anal*. Calcd for $C_8H_7N_5O$: C_7 : 53.73; H, 3.51; N, 34.81. Found: C_7 : 53.54; H, 3.42; N, 34.77.

3-Amino-1-methyl-1*H*-pyrazolo[3,4-*b*]quinoxaline **4-Oxide** (15b) — Treatment of **14** (0.5 g) with methylhydrazine (0.28 g) as described above gave 0.39 g (75%) of **15b** as dark red needles, mp 203 °C (EtOH). MS m/e: 215 (M⁺). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3470, 3290, 3180 (NH₂). NMR (DMSO- d_6) δ : 3.80 (3H, s, CH₃), 6.18 (2H, s, NH₂). *Anal.* Calcd for $C_{10}H_9N_5O$: C, 55.81; H, 4.22; N, 32.54. Found: C, 55.94; H, 4.45; N, 32.73.

Deoxygenation of 15a, b—A solution of **15a** (0.1 g) in PCl₃ (5 ml) was refluxed for 1 h. After cooling, the mixture was poured into ice-water. The aqueous solution was made alkaline with Na₂CO₃ and a brown precipitate was collected. Recrystallization from EtOH gave 0.03 g of orange-red leaflets, which were identical with an authentic sample of **12a**.

By a similar procedure, 0.045 g of 12b was obtained from 0.1 g of 15b.

Reaction of Ethyl 2-Chloroquinoxaline-3-carboxylate (16) with Hydrazine Hydrate — Hydrazine hydrate (0.32 g) was added to a solution of 16⁸⁾ (1.0 g) in EtOH (40 ml) and the mixture was warmed at 70 °C for a while. Slightly soluble red crystals (17) immediately precipitated, and were collected. Yield, 0.45 g (25%). The mother liquor was concentrated to give 18. Yield, 0.25 g (25%). In order to control the formation of 17, the following procedure was carried out: a solution of 16 (1.0 g) in EtOH (20 ml) was added to a solution of hydrazine hydrate (0.63 g) in EtOH (20 ml) in small portions at 70 °C over a period of 30 min under stirring. In this case, the reaction mixture gave 0.86 g (95%) of another product (19) along with a small amount of 17.

N,N'-Bis(2-ethoxycarbonyl-3-quinoxalinyl)hydrazine (17)—Red needles, mp 183—184 °C (THF). MS m/e: 432 (M⁺). IR $v_{\rm max}^{\rm Nujol}$ cm⁻¹: 3350 (NH), 1760 (C=O). NMR (DMSO- d_6) δ: 1.44 (3H, t, J=7 Hz, C $\underline{\rm H}_3$ CH₂), 4.51 (2H, q, J=7 Hz, CH₃C $\underline{\rm H}_2$), 10.71 (1H, br, NH). *Anal*. Calcd for C₂₂H₂₀N₆O₄: C, 61.10; H, 4.61; N, 19.44. Found: C, 61.14; H, 4.50; N, 19.21.

Ethyl 2-Hydrazinoquinoxaline-3-carboxylate (18)—Yellow needles, mp 141—142 °C (EtOH). MS m/e: 232 (M⁺). IR $\nu_{\max}^{\text{Nujol}}$ cm⁻¹: 3390. 3310, 3200 (NH, NH₂). NMR (DMSO- d_6) δ: 1.37 (3H, t, J=7 Hz, CH₃CH₂), 4.41 (2H, q, J=7 Hz, CH₃CH₂), 4.66 (2H, br, NH₂), 8.75 (1H, br, NH). Anal. Calcd for C₁₁H₁₂N₄O₂: C, 56.89; H, 5.21; N, 24.14. Found: C, 56.96; H, 5.50; N, 24.15.

2-Hydrazinoquinoxaline-3-carbohydrazide (19)—Orange needles, mp 281—282 °C (dec.) (EtOH). MS m/e: 218 (M⁺). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3330—3190 (br, NH₂, NH), 1662 (C=O). NMR (DMSO- d_6) δ : 4.66 (4H, br, 2NH₂), 9.39 (1H, br, NH), 10.23 (1H, br, NH). *Anal*. Calcd for C₉H₁₀N₆O: C, 49.53; H, 4.62; N, 38.52. Found: C, 49.96; H, 4.69; N, 38.32.

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