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Thermal Rearrangements of Cyclic Amine Ylides. VIII.¹⁾ Intramolecular Cyclization of 2-Ethynylpyridine *N*-Ylides into Indolizines and Cycl[3.2.2]azines

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Treatment of 6-unsubstituted 2-ethynyl-1-phenacylpyridinium bromides (**10a—c**) with 1,5-diazabicyclo[5.4.0]undec-5-ene in refluxing benzene resulted in cyclization to give the 3-benzoyl-indolizines (**11**) *via* the *N*-ylide intermediates **13**, whereas heating the salts **10** in refluxing acetic acid afforded the 1-benzoylindolizines (**12**). Upon similar treatment with the base, 2-ethynyl-6-methylpyridinium salts (**20a—c**) gave the 3-benzoyl-5-methylindolizines (**22**); in the case of the phenylacetylene compound **20c**, 2,3-diphenylcycl[3.2.2]azine (**23**) was also obtained. However, the reaction of 6-amino-2-ethynylpyridines (**27a—c**) with phenacyl bromide produced 5-ethynyl-1-azaindolizines (**28**), which were converted into 1-azacycl[3.2.2]azines (**29**). The mechanisms of these intramolecular cyclizations are discussed.

Keywords—2-ethynylpyridine; 2-ethynyl-1-phenacylpyridinium ylide; indolizine; 1-azaindolizine; 1-azacycl[3.2.2]azine; cycl[3.2.2]azine; intramolecular cyclization; rearrangement

Thermal sigmatropic rearrangements of cyclic allylamine *N*-ylides have been well studied²⁾ in connection with those of related open-chain compounds.³⁾ The propynylamine *N*-ylides (**1a**)⁴⁾ and *N*-oxides (**1b**)⁵⁾ and related *S*-ylides⁶⁾ are also known to undergo thermal [2,3]-sigmatropic rearrangement giving the corresponding allenic compounds **2**. These results prompted us to examine the thermal behavior of cyclic propargylic ylides, and we have already reported several new results. For example, the thermolysis of 6-ethynyl-1,2,5,6-tetrahydropyridine *N*-imides (**3**) resulted in ring expansion to give the 1,2-diazonines (**5**) *via* the allenic intermediates **4**.⁷⁾ Similar ring expansion was also observed for the cyclic sulfonium ylides having an ethynyl group at the α -position of the ring, giving rise to the corresponding three-carbon ring enlargement products such as thiocin, thionin, and thiecin derivatives.⁸⁾

On the other hand, as for aromatic amine ylides, the intermolecular 1,3-dipolar cycloaddition⁹⁾ of azine *N*-ylides and *N*-imides to olefins or acetylenes and the intramolecular 1,5-dipolar cyclization¹⁰⁾ of azine *N*-vinylimides have been widely investigated. However, few such reactions of pyridinium ylides having an unsaturated substituent at the 2-position of the pyridine ring are known. Therefore, we examined the thermal behavior of 2-ethynylpyridine *N*-imides (**6**) and have shown that they undergo intramolecular cyclization to afford 3-azaindolizines (**8**) *via* the intermediates **7**.¹¹⁾ We report here the thermal intramolecular cyclization of 2-ethynylpyridine *N*-ylides.¹²⁾

The starting 2-ethynylpyridines used in the present work were prepared from the corresponding 2-bromopyridines and acetylenes according to the reported method.¹³⁾ Treatment of the 6-unsubstituted 2-ethynylpyridines (**9a—c**)¹¹⁾ with phenacyl bromide gave the 1-phenacylpyridinium bromides (**10**), precursors of the desired pyridine *N*-ylides, in high yields. The salts **10a—c** were treated with 1,5-diazabicyclo[5.4.0]undec-5-ene (DBU) in refluxing benzene, resulting in cyclization to give the 3-benzoylindolizines (**11a—c**) in 60—70% yields. Treatment of **10** with hydroxide or alkoxide ions also afforded **11**, but in lower

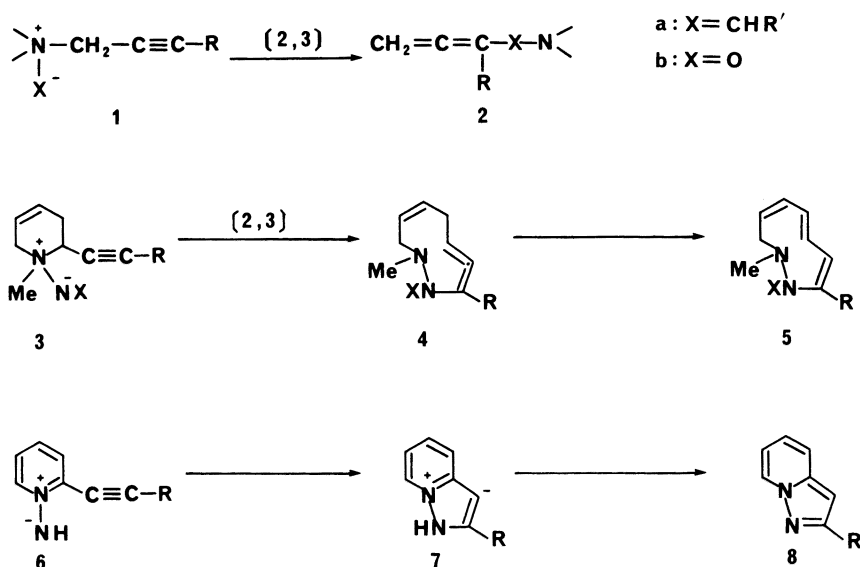


Chart 1

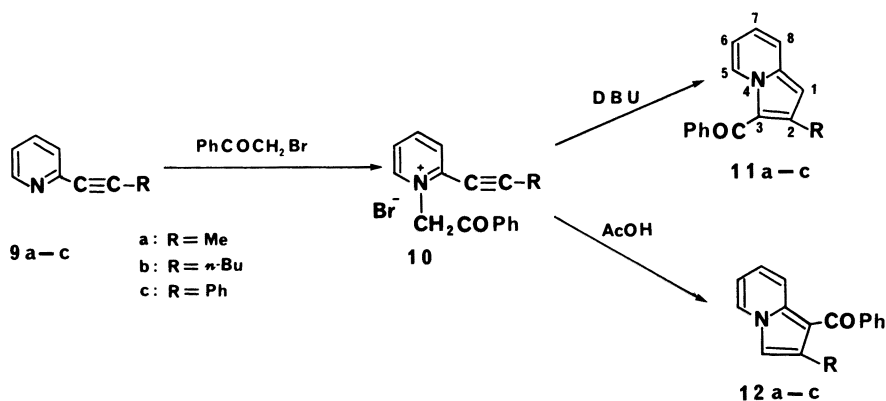


Chart 2

yields. When a polar solvent such as ethanol or dimethylformamide was used instead of benzene, the yields of **11** decreased.

It should be noted that, unexpectedly, heating the salts **10a–c** in refluxing acetic acid also resulted in cyclization giving rise to the 1-benzoylindolizines (**12a–c**) in 50–70% yields, but no 3-benzoyl isomers **11**.

The 2-phenylindolizines (**11c**¹⁴) and **12c**¹⁵) are known and the other products **11a, b** and **12a, b** were characterized by spectral comparison with **11c** and **12c**. The proton nuclear magnetic resonance (¹H-NMR) spectra of the 3-benzoylindolizines (**11**) showed singlet signals due to 1-H at around δ 6.3, whereas those of the 1-benzoyl isomers **12** showed singlet signals assignable to 3-H at around δ 7.2. The doublet signal due to 5-H in **11** appeared at lower field (*ca.* δ 9.5) than that for **12** (*ca.* δ 8.0), probably because of the shielding effect of the 3-benzoyl group.

Possible mechanisms leading to the indolizines are shown in Chart 3, although none of the intermediates could be isolated. The base-induced reaction of **10** may proceed by initial formation of the *N*-ylides **13**, which might cyclize to give the products **11** *via* the zwitter ionic

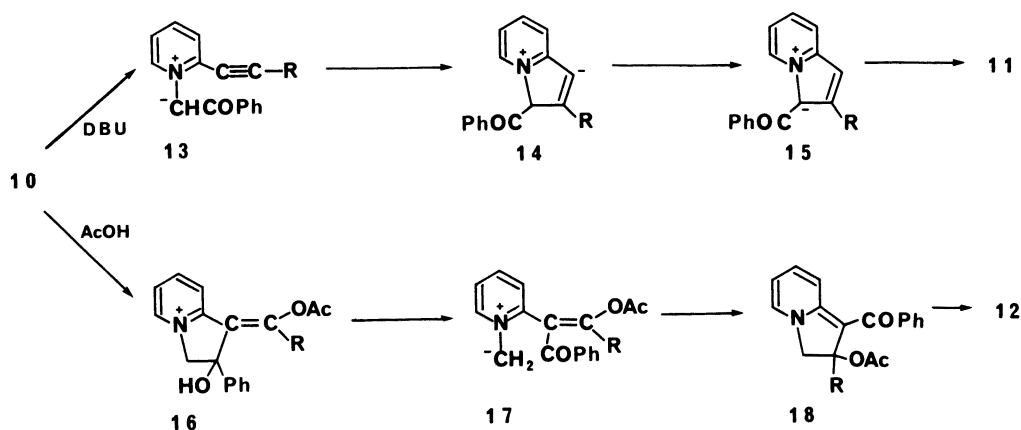


Chart 3

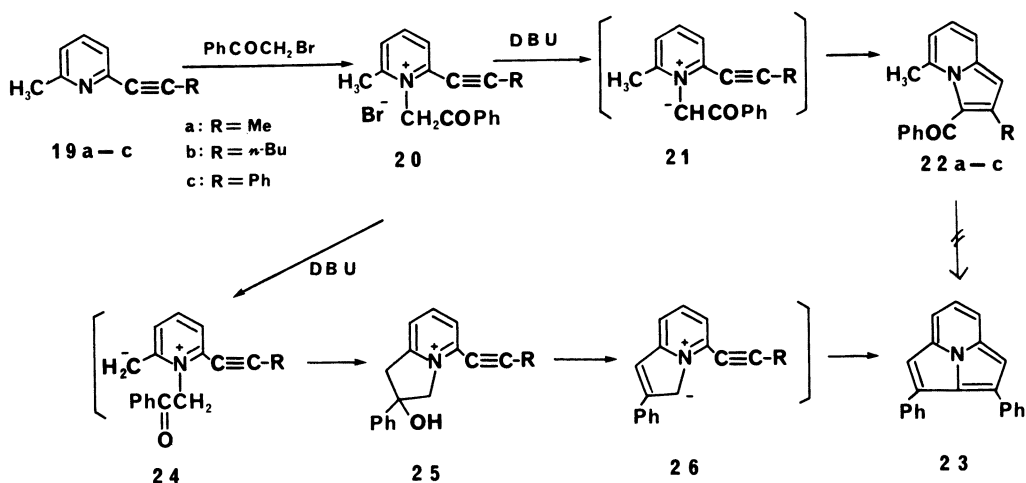


Chart 4

intermediates **14** and the cyclic *N*-ylides **15** successively. In contrast to the cases of the aliphatic amine ylides (**1** and **3**),⁴⁻⁷⁾ a concerted mechanism for the present reaction seems unlikely because of prohibitive ring strain in the five-membered cyclic allenic intermediates. This behavior is analogous to that observed for the 2-ethynylpyridine *N*-imides (**6**). On the other hand, the acid-induced reaction may proceed *via* the addition of an acetoxy anion to the triple bond to produce allenic intermediates followed by cyclization to **16**, which may give the methylides **17** by ring fission. The ylides **17** would undergo [2,3]-sigmatropic rearrangement to the bicyclic intermediates **18**, which then give the 1-benzoyl products **12** by elimination of acetic acid. In addition, when a solution of hydrochloric acid or *p*-toluenesulfonic acid in ethanol or benzene was used instead of acetic acid, the salts **10** were decomposed and gave no indolizines, indicating that acetoxy anion may play an important role in this acid-induced reaction.

Next, the 6-methylpyridinium bromides (**20a-c**), prepared from the corresponding 2-ethynyl-6-methylpyridines (**19**) and phenacyl bromide, were treated with DBU to also produce the 5-methyl-3-benzoylindolizines (**22a-c**) in 30–60% yields, probably *via* the *N*-ylides **21**. In the case of the phenylacetylene compound **20c**, 2,3-diphenylcyclo[3.2.2]azine (**23**)¹⁶⁾ was obtained in 7% yield in addition to **22c**. However, in the cases of **20a, b**, the

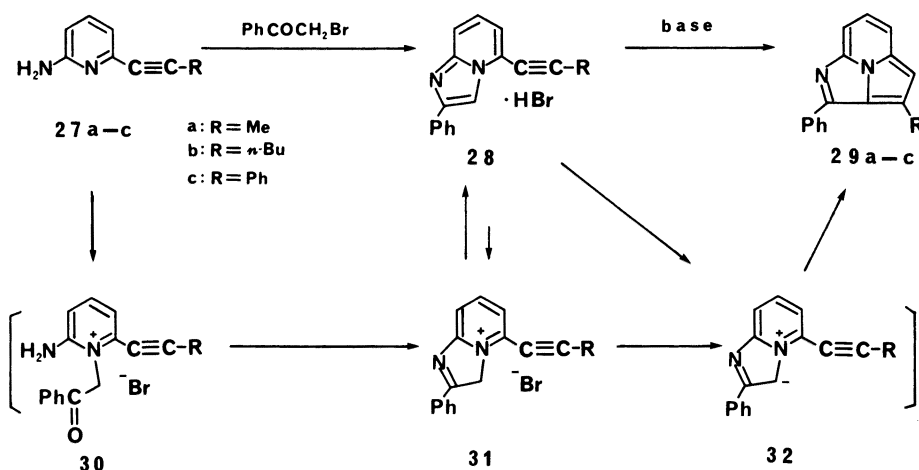


Chart 5

corresponding cyclazines could not be isolated, although the presence of very small amounts of compounds assumed to be cyclazines in product mixtures was confirmed by thin-layer chromatography and ¹H-NMR spectral analysis.

A possible mechanism for the formation of the cyclazine **23** is shown in Chart 4. The initial base-induced deprotonation from the methyl group to **24** may also occur in competition with the deprotonation from the phenacyl group to the ylide **21**. The intermediate **24** may cyclize to give the cyclic *N*-ylide **26** via **25**. The ylide **26** might further cyclize to the tricyclic product **23**, analogously to the formation of **11** from **13**. In addition, further treatment of the indolizines (**22**) with the base gave no cyclazines.

In contrast to the cases of the 6-unsubstituted pyridines (**9**) and 6-methylpyridines (**19**), treatment of 6-amino-2-ethynylpyridines (**27a-c**) with phenacyl bromide did not give the phenacylpyridinium salts **30**, but directly afforded the 5-ethynyl-2-phenyl-1-azaindolizine hydrobromides (**28a-c**) in 80–90% yields. The salts were treated with DBU in refluxing benzene to produce the 2-phenyl-1-azacycl[3.2.2]azines (**29a-c**: 70–80% yields), which were characterized by comparison with the related compounds reported.¹⁷⁾ These compounds **29** were also obtained by treatment of **28** with sodium carbonate in water.

The reaction of **27** with phenacyl bromide may proceed by initial formation of the phenacylpyridinium salts **30**, which are then converted to the products **28** via the Schiff bases **31**, although these key intermediates could not be isolated. The formation of the cyclazines (**29**) from **28** may also involve the cyclic ylide intermediates **32** formed from **28** directly or via **31** by the base treatment.

In conclusion, the present results provide a new route for preparing indolizines¹⁸⁾ and cyclazines,¹⁹⁾ although many synthetic methods for these heterocyclic rings are known.

Experimental

Melting points were measured on a Yanagimoto micro melting hot stage apparatus and are uncorrected. Infrared (IR) spectra were determined with a Hitachi 270-30 spectrometer and mass spectra (MS) were measured with a JEOL DX-300 instrument. ¹H-NMR spectra were recorded on a JEOL JNM-MH100 spectrometer in CDCl₃ using tetramethylsilane as an internal standard unless otherwise stated; spectral assignments were confirmed by spin-decoupling experiments. Microanalyses were carried out in the Microanalytical Laboratory of this school by Mrs. R. Igarashi.

2-Ethynyl-1-phenacylpyridinium Bromides (10a-c)—General Procedure: A solution of a 2-ethynylpyridine

(**9**:¹¹) 3 mmol) and phenacyl bromide (0.6 g, 3 mmol) in CHCl_3 (20 ml) was refluxed for 4–6 h and then evaporated to dryness *in vacuo*. The resulting solid residue was washed with ether and recrystallized from methanol–ethyl acetate to give **10** as pale yellow prisms.

10a: 0.87 g, 92% yield, mp 189–190 °C (dec.). IR (KBr): 2220 ($\text{C}\equiv\text{C}$), 1660 ($\text{C}=\text{O}$) cm^{-1} . Anal. Calcd for $\text{C}_{16}\text{H}_{14}\text{BrNO}$: C, 60.76; H, 4.43; N, 4.43. Found: C, 60.66; H, 4.34; N, 4.48.

10b: 0.83 g, 77% yield, mp 148–149 °C (dec.). IR (KBr): 2220 ($\text{C}\equiv\text{C}$), 1685 ($\text{C}=\text{O}$) cm^{-1} . Anal. Calcd for $\text{C}_{19}\text{H}_{20}\text{BrNO}$: C, 63.69; H, 5.58; N, 3.91. Found: C, 63.61; H, 5.68; N, 3.92.

10c: 0.91 g, 80% yield, mp 196–198 °C (dec.). IR (KBr): 2220 ($\text{C}\equiv\text{C}$), 1675 ($\text{C}=\text{O}$) cm^{-1} . Anal. Calcd for $\text{C}_{21}\text{H}_{16}\text{BrNO}$: C, 66.67; H, 4.23; N, 3.70. Found: C, 66.59; H, 4.27; N, 3.75.

Treatment of the Salts (10a–c) with DBU: Formation of 3-Benzoylindolizines (11a–c)—General Procedure: DBU (0.31 ml, 2 mmol) was added dropwise to a mixture of a salt (**10**: 2 mmol) and benzene (10 ml) with stirring. The mixture was refluxed for 12 h with stirring. After cooling, the mixture was washed with saturated NaCl, dried over MgSO_4 , and evaporated *in vacuo*. The residue was chromatographed on silica gel using hexane– CH_2Cl_2 (1 : 3) as an eluent to give **11**.

11a: 0.32 g, 67% yield, pale yellow viscous oil. MS m/z : 235 (M^+). IR (neat): 1595 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : 1.86 (3H, s, Me), 6.26 (1H, s, 1-H), 6.74 (1H, dd, 6-H), 7.05 (1H, dd, 7-H), 7.3–7.7 (6H, m, 8- and Ph-H), 9.76 (1H, d, 5-H), $J_{5,6}=7$, $J_{6,7}=7$, $J_{7,8}=9$ Hz. Anal. Calcd for $\text{C}_{16}\text{H}_{13}\text{NO}$: C, 81.70; H, 5.53; N, 5.95. Found: C, 81.73; H, 5.54; N, 5.98.

11b: 0.33 g, 59% yield, pale yellow viscous oil. MS m/z : 277 (M^+). IR (neat): 1590 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : *n*-Bu [0.67 (3H, t, $J=7$ Hz), 0.8–1.5 (4H, m), 2.18 (2H, t, $J=8$ Hz)], 6.22 (1H, s, 1-H), 6.63 (1H, dd, 6-H), 6.92 (1H, dd, 7-H), 7.2–7.4 (6H, m, 8- and Ph-H), 9.49 (1H, d, 5-H), $J_{5,6}=7$, $J_{6,7}=7$, $J_{7,8}=8$ Hz. Anal. Calcd for $\text{C}_{19}\text{H}_{19}\text{NO}$: C, 82.31; H, 6.85; N, 5.05. Found: C, 82.28; H, 6.92; N, 5.07.

11c: 0.40 g, 68% yield, mp 136–138 °C (lit.¹⁴) 137–137.5 °C. This compound was identical with an authentic sample prepared by the method of the literature,¹⁴ in which no spectral data were described, so they are reported here. MS m/z : 297 (M^+). IR (KBr): 1590 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : 6.39 (1H, s, 1-H), 6.68 (1H, dd, 6-H), 6.8–7.4 (12H, m, 7-, 8-, 2-Ph-, and CPh-H), 9.59 (1H, d, 5-H), $J_{5,6}=7$, $J_{6,7}=7$ Hz.

Treatment of the Salts (10a–c) with Acetic Acid: Formation of 1-Benzoylindolizines (12a–c)—General Procedure: A solution of a salt (**10**: 1 mmol) in AcOH (10 ml) was refluxed with stirring for 12 h and then evaporated *in vacuo*. The residue was extracted with CH_2Cl_2 and the extract was washed with saturated NaHCO_3 , dried, and concentrated *in vacuo*. The residue was chromatographed on silica gel using hexane– CH_2Cl_2 (1 : 3) as an eluent to give **12**.

12a: 115 mg, 49% yield, pale yellow viscous oil. MS m/z : 235 (M^+). IR (neat): 1600 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : 2.03 (3H, s, Me), 6.72 (1H, dd, 6-H), 7.09 (1H, dd, 7-H), 7.12 (1H, s, 3-H), 7.3–7.5 (5H, m, Ph-H), 7.95 (1H, d, 5-H), 8.48 (1H, d, 8-H), $J_{5,6}=7$, $J_{6,7}=7$, $J_{7,8}=9$ Hz. Anal. Calcd for $\text{C}_{16}\text{H}_{13}\text{NO}$: C, 81.70; H, 5.53; N, 5.95. Found: C, 81.65; H, 5.59; N, 5.86.

12b: 122 mg, 44% yield, mp 68–70 °C, pale yellow needles (from CH_2Cl_2 –hexane). MS m/z : 277 (M^+). IR (KBr): 1610 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : *n*-Bu [0.71 (3H, t, $J=7$ Hz), 0.9–1.6 (4H, m), 2.36 (2H, t, $J=8$ Hz)], 6.76 (1H, dd, 6-H), 7.13 (1H, dd, 7-H), 7.18 (1H, s, 3-H), 7.3–7.4 (5H, m, Ph-H), 8.00 (1H, d, 5-H), 8.51 (1H, d, 8-H), $J_{5,6}=7$, $J_{6,7}=7$, $J_{7,8}=9$ Hz. Anal. Calcd for $\text{C}_{19}\text{H}_{19}\text{NO}$: C, 82.31; H, 6.86; N, 5.05. Found: C, 82.19; H, 6.94; N, 5.07.

12c: 200 mg, 67% yield, mp 133–135 °C (lit.¹⁵) mp 135–136 °C, pale yellow prisms (from CH_2Cl_2 –hexane). This compound was identical with an authentic sample prepared by the method of the literature,¹⁵ in which no spectral data were given, so they are reported here. MS m/z : 297 (M^+). IR (KBr): 1600 ($\text{C}=\text{O}$) cm^{-1} . $^1\text{H-NMR}$ δ : 6.57 (1H, dd, 6-H), 6.9–7.5 (11H, m, 7- and Ph-H), 7.14 (1H, s, 3-H), 7.81 (1H, d, 5-H), 7.90 (1H, d, 8-H), $J_{5,6}=7$, $J_{6,7}=7$, $J_{7,8}=9$ Hz.

2-Ethynyl-6-methylpyridines (19a–c)—These compounds were prepared from 2-bromo-6-methylpyridine and corresponding acetylenes, according to the procedure for the preparation of the 2-ethynylpyridines (**9**).

19a: 81% yield, pale yellow oil, bp 114–116 °C (12 mmHg). MS m/z : 131 (M^+). IR (neat): 2260 ($\text{C}\equiv\text{C}$) cm^{-1} . $^1\text{H-NMR}$ δ : 2.02 (3H, s, Me), 2.48 (3H, s, 6-Me), 6.99 (1H, d, 5-H), 7.13 (1H, d, 3-H), 7.44 (1H, dd, 4-H), $J_{3,4}=8$, $J_{4,5}=7$ Hz. Anal. Calcd for $\text{C}_9\text{H}_9\text{N}$: C, 82.44; H, 6.87; N, 10.68. Found: C, 82.32; H, 6.96; N, 10.71.

19b: 82% yield, pale yellow oil, bp 150–155 °C (20 mmHg). MS m/z : 173 (M^+). IR (neat): 2250 ($\text{C}\equiv\text{C}$) cm^{-1} . $^1\text{H-NMR}$ δ : *n*-Bu [0.92 (3H, t, $J=7$ Hz), 1.2–1.8 (4H, m), 2.2–2.8 (2H, m)], 2.48 (3H, s, 6-Me), 6.99 (1H, d, 5-H), 7.15 (1H, d, 3-H), 7.46 (1H, dd, 4-H), $J_{3,4}=8$, $J_{4,5}=7$ Hz. Anal. Calcd for $\text{C}_{12}\text{H}_{15}\text{N}$: C, 83.23; H, 8.67; N, 8.09. Found: C, 83.19; H, 8.68; N, 7.99.

19c: 79% yield, pale yellow oil, bp 165–168 °C (6 mmHg). MS m/z : 193 (M^+). IR (neat): 2250 ($\text{C}\equiv\text{C}$) cm^{-1} . $^1\text{H-NMR}$ δ : 2.49 (3H, s, 6-Me), 7.01 (1H, d, $J=7$ Hz, 5-H), 7.2–7.7 (7H, m, 3-, 4-, and Ph-H). Anal. Calcd for $\text{C}_{14}\text{H}_{11}\text{N}$: C, 87.04; H, 5.70; N, 7.25. Found: C, 86.94; H, 5.76; N, 7.30.

2-Ethynyl-6-methyl-1-phenacylpyridinium Bromides (20a–c)—Compounds **10a–c** (3 mmol) were treated with phenacyl bromide and worked up as described for the preparation of **10** to give **20a–c**. However, the salts **20** were too hygroscopic to be isolated, and thus were used for the following reaction without purification.

Treatment of the Salts (20a–c) with DBU: Formation of 3-Benzoyl-5-methylindolizines (22a–c) and 2,3-

Diphenylcycl[3.2.2]azine (23c)—The salts **20a–c** were treated with DBU in benzene and worked up as described for **10** to give **22a–c** and **23c**.

22a: 250 mg, 33% yield from **19a**, pale yellow viscous oil. MS m/z : 249 (M^+). IR (neat): 1620 ($C=O$) cm^{-1} . 1H -NMR δ : 1.99 (3H, s, 2-Me), 2.37 (3H, s, 5-Me), 6.37 (1H, s, 1-H), 6.61 (1H, d, 6-H), 7.03 (1H, dd, 7-H), 7.2–7.6 and 7.8–8.0 (4H, m, and 2H, m, 8- and Ph-H), $J_{6,7}=7$, $J_{7,8}=8$ Hz. Anal. Calcd for $C_{17}H_{15}NO$: C, 81.90; H, 6.06; N, 5.62. Found: C, 81.82; H, 6.08; N, 5.61.

22b: 260 mg, 30% yield from **19b**, pale yellow viscous oil. MS m/z : 291 (M^+). IR (neat): 1620 ($C=O$) cm^{-1} . 1H -NMR δ : *n*-Bu [0.73 (3H, t, $J=7$ Hz), 0.9–1.6 (4H, m), 2.2–2.4 (2H, m)], 2.33 (3H, s, 5-Me), 6.45 (1H, s, 1-H), 6.60 (1H, d, 6-H), 7.04 (1H, dd, 7-H), 7.3–7.6 and 7.9–8.0 (4H, m, and 2H, m, 8- and Ph-H), $J_{6,7}=7$, $J_{7,8}=8$ Hz. Anal. Calcd for $C_{20}H_{21}NO$: C, 82.44; H, 7.26; N, 4.81. Found: C, 82.51; H, 7.27; N, 4.74.

22c: 560 mg, 61% yield from **19c**, pale yellow viscous oil. MS m/z : 311 (M^+). IR (neat): 1630 ($C=O$) cm^{-1} . 1H -NMR δ : 2.26 (3H, s, 5-Me), 6.39 (1H, d, 6-H), 6.45 (1H, s, 1-H), 6.7–7.3 and 7.5–7.6 (10H, m, and 2H, m, 7-, 8-, and Ph-H), $J_{6,7}=7$ Hz. Anal. Calcd for $C_{22}H_{17}NO$: C, 84.86; H, 5.50; N, 4.50. Found: C, 84.77; H, 5.56; N, 4.29.

23c: 61 mg, 7% yield from **19c**, mp 123–124 °C, yellow prisms (from benzene–hexane). This compound was identical with an authentic sample prepared by the literature method.¹⁶⁾ 1H -NMR δ : 6.4–6.8 (2H, m), 7.1–7.3 (7H, m), 7.4–7.8 (6H, m).

6-Amino-2-ethynylpyridines (27a–c)—These compounds were prepared from 6-amino-2-bromopyridine²⁰⁾ and the corresponding acetylenes, according to the procedure for the preparation of the 2-ethynylpyridines (**9**).¹¹⁾

27a: 74% yield, mp 97–99 °C, pale yellow prisms (from ether). MS m/z : 132 (M^+). IR (KBr): 3340 and 3150 (NH), 2230 ($C\equiv C$) cm^{-1} . 1H -NMR δ : 1.97 (3H, s, Me), 4.88 (2H, br, NH_2), 6.40 (1H, d, 5-H), 6.70 (1H, d, 3-H), 7.29 (1H, dd, 4-H), $J_{3,4}=7$, $J_{4,5}=8$ Hz. Anal. Calcd for $C_8H_8N_2$: C, 72.73; H, 6.06; N, 21.21. Found: C, 72.70; H, 6.16; N, 21.17.

27b: 80% yield, yellow viscous oil. MS m/z : 174 (M^+). IR (neat): 3470 and 3330 (NH), 2240 ($C\equiv C$) cm^{-1} . 1H -NMR δ : *n*-Bu [0.89 (3H, t, $J=7$ Hz), 1.1–1.7 (4H, m), 2.33 (2H, t, $J=8$ Hz)], 5.27 (2H, br, NH_2), 6.39 (1H, d, 5-H), 6.78 (1H, d, 3-H), 7.27 (1H, dd, 4-H), $J_{3,4}=8$, $J_{4,5}=8$ Hz. Anal. Calcd for $C_{11}H_{14}N_2$: C, 75.86; H, 8.05; N, 16.09. Found: C, 75.77; H, 8.13; N, 16.10.

27c: 85% yield, mp 102–105 °C, yellow prisms (from ether). MS m/z : 194 (M^+). IR (KBr): 3470 and 3330 (NH), 2220 ($C\equiv C$) cm^{-1} . 1H -NMR δ : 5.04 (2H, br, NH_2), 6.34 (1H, d, 5-H), 6.86 (1H, d, 3-H), 7.31 (1H, dd, 4-H), 7.2–7.4 and 7.5–7.7 (3H, m, and 2H, m, Ph-H), $J_{3,4}=8$, $J_{4,5}=8$ Hz. Anal. Calcd for $C_{13}H_{10}N_2$: C, 80.41; H, 5.15; N, 14.43. Found: C, 80.45; H, 5.31; N, 14.33.

Treatment of 27a–c with Phenacyl Bromide: Formation of 5-Ethynyl-2-phenyl-1-azaindolizine Hydrobromides (28a–c)—Compounds **27a–c** (2 mmol) were treated with phenacyl bromide and worked up as described for **10** to give **28**.

28a: 490 mg, 79% yield, mp 213–215 °C (dec.), colorless prisms. IR (KBr): 2240 ($C\equiv C$), 1640 ($C=N$) cm^{-1} . 1H -NMR (CD_3OD) δ : 2.37 (3H, s, Me), 7.5–7.7 and 7.9–8.1 (each 4H, m, 6-, 7-, 8-, and Ph-H), 8.72 (1H, s, 3-H). Anal. Calcd for $C_{16}H_{13}BrN_2$: C, 61.34; H, 4.15; N, 8.95. Found: C, 61.20; H, 4.18; N, 8.76.

28b: 550 mg, 78% yield, mp 114–116 °C (dec.), colorless prisms. IR (KBr): 2230 ($C\equiv C$), 1640 ($C=N$) cm^{-1} . 1H -NMR (CD_3OD) δ : *n*-Bu [1.02 (3H, m), 1.4–1.9 (4H, m), 2.81 (2H, t, $J=7$ Hz)], 7.5–7.7 and 7.9–8.1 (each 4H, m, 6-, 7-, 8-, and Ph-H), 8.68 (1H, s, 3-H). Anal. Calcd for $C_{19}H_{19}BrN_2$: C, 64.22; H, 5.35; N, 7.88. Found: C, 63.90; H, 5.39; N, 7.70.

28c: 650 mg, 87% yield, mp 211–215 °C (dec.), colorless needles. IR (KBr): 2210 ($C\equiv C$), 1635 ($C=N$) cm^{-1} . 1H -NMR (CD_3OD) δ : 7.5–7.9 and 8.0–8.1 (9H, m, and 4H, m, 6-, 7-, 8-, and Ph-H), 8.76 (1H, s, 3-H). Anal. Calcd for $C_{21}H_{15}BrN_2$: C, 67.20; H, 4.00; N, 7.47. Found: C, 67.03; H, 4.02; N, 7.47.

Treatment of 28a–c with DBU: Formation of 2-Phenyl-1-azacycl[3.2.2]azines (29a–c)—The salts **28a–c** (1 mmol) were treated with DBU and worked up as described for the treatment of **10** with DBU to give **29**.

29a: 160 mg, 72% yield, mp 121–124 °C, pale yellow needles (from CH_2Cl_2 –hexane). MS m/z : 232 (M^+). 1H -NMR δ : 1.99 (3H, s, 3-Me), 6.7–7.0, 7.2–7.6, and 7.8–8.1 (2H, m, 4H, m, and 2H, m, 5-, 6-, 7-, and Ph-H), 7.95 (1H, s, 4-H). Anal. Calcd for $C_{16}H_{12}N_2$: C, 82.73; H, 5.21; N, 12.06. Found: C, 82.74; H, 5.09; N, 12.12.

29b: 190 mg, 70% yield, pale yellow viscous oil. MS m/z : 274 (M^+). 1H -NMR δ : 3-*n*-Bu [0.98 (3H, t, $J=7$ Hz), 1.3–1.9 (4H, m), 2.54 (2H, t, $J=8$ Hz)], 6.8–7.1, 7.2–7.6, and 7.8–8.0 (2H, m, 4H, m, and 2H, m, 5-, 6-, 7-, and Ph-H), 8.03 (1H, s, 4-H). Anal. Calcd for $C_{19}H_{18}N_2$: C, 83.17; H, 6.61; N, 10.21. Found: C, 83.05; H, 6.65; N, 10.08.

29c: 230 mg, 79% yield, mp 124–126 °C, pale yellow needles (from CH_2Cl_2 –hexane). MS m/z : 294 (M^+). 1H -NMR δ : 6.9–7.1, 7.3–7.7, and 7.4–8.1 (2H, m, 9H, m, and 2H, m, 5-, 6-, 7-, and Ph-H), 8.06 (1H, s, 4-H). Anal. Calcd for $C_{21}H_{14}N_2$: C, 85.69; H, 4.79; N, 9.52. Found: C, 85.61; H, 4.78; N, 9.38.

References and Notes

- 1) Part VII: H. Sashida and T. Tsuchiya, *Chem. Pharm. Bull.*, **34**, 3682 (1986).
- 2) E. Vedjes, M. J. Arco, D. W. Powell, J. M. Renga, and S. P. Singer, *J. Org. Chem.*, **43**, 4831 (1978); T. Tsuchiya, H. Sashida, and H. Sawanishi, *Chem. Pharm. Bull.*, **26**, 2880 (1978); T. Tsuchiya and H. Sashida, *ibid.*, **29**, 1887 (1981).

- (1981); *idem*, *ibid.*, **32**, 4117 (1984).
- 3) For reviews, see M. J. McKillip, E. A. Sedor, B. M. Culbertson, and S. Wawzonek, *Chem. Rev.*, **73**, 255 (1973); E. C. Taylor, and I. J. Turchi, *ibid.*, **79**, 181 (1979); T. Nakai and K. Mikami, *Yuki Gosei Kagaku Kyokai Shi*, **38**, 381 (1980).
 - 4) W. D. Ollis, I. O. Sutherland, and Y. Thebaranonth, *J. Chem. Soc., Chem. Commun.*, **1973**, 657; S. Mageswaran, W. D. Ollis, D. A. Suthan, I. O. Sutherland, and Y. Thebaranonth, *J. Chem. Soc., Perkin Trans. 1*, **1981**, 1969.
 - 5) A. H. Khuthier, M. A. Al-Iraq, G. Hallatröm, and B. Lindeke, *J. Chem. Soc., Chem. Commun.*, **1979**, 9; J. C. Craig, N. N. Ekwuribe, and L. D. Gruenke, *Tetrahedron Lett.*, **1979**, 4025; G. Hallström, B. Lideke, A. H. Khuthier, and M. A. Al-Iraq, *ibid.*, **1980**, 667.
 - 6) J. E. Baldwin, R. E. Hackler, and D. P. Kelly, *Chem. Commun.*, **1968**, 1109; A. Terada and Y. Kishida, *Chem. Pharm. Bull.*, **17**, 966 (1969).
 - 7) H. Sashida and T. Tsuchiya, *Chem. Pharm. Bull.*, **32**, 4600 (1984).
 - 8) H. Sashida and T. Tsuchiya, *Chem. Pharm. Bull.*, **34**, 3644 (1986).
 - 9) For reviews, see Y. Tamura, *Yakugaku Zasshi*, **100**, 1 (1980); Y. Tamura and M. Ikeda, "Advances in Heterocyclic Chemistry," Vol. 29, ed. by A. R. Katritzky and A. J. Boulton, Academic Press, London, 1981, p. 71.
 - 10) Y. Tamura, N. Tsujimoto, Y. Sumide, and M. Ikeda, *Tetrahedron*, **28**, 21 (1971); T. Sasaki, K. Kanematsu, and A. Kakehi, *J. Org. Chem.*, **37**, 3106 (1972); A. Kakehi, S. Ito, K. Uchiyama, Y. Konno, and K. Kondo, *ibid.*, **42**, 443 (1977).
 - 11) T. Tsuchiya and H. Sashida, *J. Chem. Soc., Chem. Commun.*, **1980**, 1109; T. Tsuchiya, H. Sashida, and A. Konoshita, *Chem. Pharm. Bull.*, **31**, 4568 (1983).
 - 12) A part of this work has been reported in a preliminary communication: T. Tsuchiya, M. Kato, and H. Sashida, *Chem. Pharm. Bull.*, **32**, 4666 (1984).
 - 13) D. Leaver, W. K. Gibson, and J. D. R. Vass, *J. Chem. Soc.*, **1963**, 6053; K. Sonogashira, Y. Tohda, and N. Hagihara, *Tetrahedron Lett.*, **1975**, 4467.
 - 14) J. Melton and D. G. Wibberley, *J. Chem. Soc. (C)*, **1967**, 983.
 - 15) D. R. Bragg and D. G. Wibberley, *J. Chem. Soc.*, **1963**, 3277.
 - 16) V. Boekelheide and R. J. Windgassen, Jr., *J. Am. Chem. Soc.*, **80**, 2020 (1958); R. J. Windgassen, Jr., W. H. Saunders, Jr., and V. Boekelheide, *ibid.*, **81**, 1459 (1959).
 - 17) V. Boekelheide and A. Miller, *J. Org. Chem.*, **26**, 431 (1961); O. Fuentes and W. W. Paudler, *ibid.*, **40**, 1210 (1975).
 - 18) F. J. Swinbourne, J. H. Hunt, and G. Klinkert, "Advances in Heterocyclic Chemistry," Vol. 23, ed. by A. R. Katritzky and A. J. Boulton, Academic Press, London, 1978, p. 103; T. Uchida and K. Matsumoto, *Synthesis*, **1976**, 209.
 - 19) W. Flitsch and U. Krämer, "Advances in Heterocyclic Chemistry," Vol. 22, ed. by A. R. Katritzky and A. J. Boulton, Academic Press, London, 1978, p. 321.
 - 20) F. Johnson, J. P. Panella, A. A. Carlson, and D. H. Hunneman, *J. Org. Chem.*, **27**, 2473 (1962).