Chem. Pharm. Bull. 36(12)4711—4721(1988)

Studies on Taxane Synthesis. I. Synthesis of 3,8,11,11-Tetramethyl-4-oxobicyclo[5.3.1]undecane as a Model for Taxane Synthesis

YASUO OHTSUKA* and TAKESHI OISHI*

RIKEN (The Institute of Physical and Chemical Research), Hirosawa, Wako-shi, Saitama 351, Japan

(Received May 16, 1988)

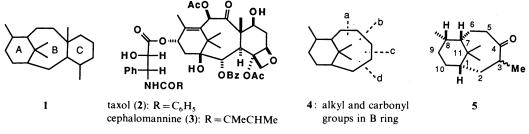
The bicyclo[5.3.1]undecanone 5 corresponding to the A and B rings in taxane diterpenes was synthesized from β -ionone. Reduction of the tosylhydrazone 11 with catecholborane afforded the trisubstituted olefin 12 having trans substituents at the C-1 and C-7 positions. The requisite cisarrangement of these substituents for eight-membered ring formation was induced by epimerization of the aldehyde 13 to give the bicyclic hemiacetal 14. The lithium diisopropylamide-promoted intramolecular cyclization of the twelve-membered lactam sulfoxide 32 proceeded quite smoothly, affording the eight-membered keto sulfoxide 33 in quantitative yield.

Keywords—taxane; eight-membered ring; intramolecular cyclization; twelve-membered lactam sulfoxide; β -ionone; epimerization; tosylhydrazone; catecholborane; Michael addition; bicyclo[5.3.1]undecane

The taxanes, a group of naturally occurring diterpenes isolated from various species of *Taxus*, possess a unique tricyclo[9.3.1.0^{3,8}]pentadecane structure (1) involving a sterically congested eight-membered B-ring.¹⁾ Among them, taxol (2) and cephalomannine (3) were recently found to exhibit antileukemic and antitumor activities.²⁾

The crucial step in the synthesis of 1 is obviously in the construction of the eight-membered B-ring, and several approaches based on fragmentation, ^{3a)} rearrangement, ^{3b)} intramolecular Diels-Alder reaction ^{3c)} or direct ring closure ^{3d)} have been reported. Recently, we developed an efficient general method for the construction of medium-ring ketones based on the intramolecular cyclization of large-membered lactam sulfoxides or sulfones ⁴⁾ and this method was successfully applied for the total syntheses of caryophyllene and isocaryophyllene, possessing a bicyclo [7.2.0] undecane ring system. ⁵⁾ We now report a series of model experiments aimed at examining whether our new strategy can be applied for the construction of the sterically extremely congested eight-membered taxane B-ring having a functionality for C-ring formation.

Initially, we intended the synthesis of the bicyclo[5.3.1]undecane derivative 4 having a keto group on the eight-membered B-ring. In our strategy for eight-membered ketone



formation, a key intermediate is a twelve-membered lactam sulfoxide or sulfone. However, in the present cases, the oxidation-labile tetrasubstituted double bond is present in the A-ring of the natural products and thus the preferential oxidation of sulfide to sulfone is presumed to be difficult, while oxidation to sulfoxide does take place without any difficulty. Therefore, the pathway which proceeds *via* lactam sulfoxides was chosen.

As a route for the synthesis of compounds having a bicyclo[5.3.1]undecane skeleton 4, eight variants can be formulated (Fig. 2). Among them, the routes 1, 2, 4 and 7 are eliminated since the keto groups in the products are located inadequately for the subsequent C-ring formation. The routes 5 and 6 should also be avoided for the following reasons. In the previous paper,⁴⁾ it was shown that the intramolecular cyclization of lactam sulfoxides proceeded smoothly only when an alkyl group was present α to the lactam carbonyl group.

According to this limitation, only the ketones having an unnecessary alkyl group on the C-2 or C-5 position can be prepared by the routes 5 or 6, respectively, while alkyl groups in the ketones prepared by the routes 3 and 8 are situated in positions where they can be used directly for formation of the C-ring provided they are properly functionalized. In the present paper the synthesis of the 3-methyl-4-oxo compound 5 by the route 3 is described and in the following paper the synthesis of the 4-alkyl-8-en-3-oxo derivative of 4 by the route 8 will be reported.

The acetate 7, easily prepared from β-ionone (6) by selective ozonolysis of the side chain, half 1 chain, half 1 chain, half 2 chain, affording the keto acetate 8 (73% yield). To introduce a functional group at the C-1 position, half 2 chain and 3 chain a

The tetrasubstituted α,β -unsaturated ketone in 10 was then converted into the trisubstituted olefin 12 in 95.3% yield by reduction of the corresponding tosylhydrazone 11 with catecholborane. The substituents at the C-1 and C-7 positions in 12 can be assigned as *trans* since it is presumed that the reducing agent approaches the C=N double bond of the tosylhydrazone 11 from the less hindered α -side producing the diimine ii *via* i, and then double bond migration, nitrogen gas extrusion and intramolecular hydride transfer take place concomitantly, affording *trans* 12 as shown in Fig. 5. The nitromethyl group in 12 was then converted into an aldehyde by MeONa and TiCl₃ treatment⁸) producing 13. In order that the eight-membered ring formation might be achieved at a later stage, substituents as the C-1 and C-7 positions should be *cis*. Therefore, base-induced epimerization of the formyl group in 13 was examined at this stage. 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU) (benzene, 80 °C) treatment of 13 led only to the recovery of the starting material, and lithium diisopropylamide (LDA) (tetrahydrofuran (THF), -10 °C) treatment gave a complex mixture. Quite unexpectedly, however, treatment of 13 with MeONa in MeOH afforded the bicyclic hemiacetal

CO₂H | Me O | Me O | N-C-CH₂ | N-C-CH₂ | SCH₂(CH₂)_{n-3} |
$$m = 1$$
, R = alkyl | m = 2, R = H or alkyl | $m = 2$, R = H or alkyl | $m = 2$, R = H or alkyl | $m = 2$, R = $m = 1$,

14. Sodium borohydride reduction of 14 followed by acetylation produced the cis diacetate 15. On the other hand, direct lithium aluminum hydride reduction of the aldehyde 13 followed by acetylation yielded the isomeric diacetate 16. Therefore, the substituents at C-1 and C-7 in 16 should be assigned as trans, which clearly shows that the above stereochemical assumption leading to 12 from 11 is correct. Initial methanolysis of the acetate in 13 by MeONa-MeOH and the subsequent intramolecular attack of the resulting alkoxide anion on the nearby C-1 hydrogen atom may account for this unique epimerization of the formyl group. The fact that the hydrolysis of the acetate is essential for the initiation of the reaction is supported by the following observations; 1) an attempt at epimerization under anhydrous conditions using DBU failed, and 2) even when the pivalate 17 or the methoxymethyl ether 18 were treated with tert-BuOK (or KH)-THF or MeONa-MeOH, respectively, epimerization did not take place. These compounds are known to resist hydrolysis under the above specific conditions. Next, the bicyclic hemiacetal 14 was treated with (carbethoxymethylene)triphenylphos-

Fig. 5

phorane, affording the unsaturated ester 19. However, an attempted selective hydrogenation of the α,β -unsaturated ester of 19 to 20 with NaTeH,⁹⁾ or lithium tri-sec-butylborohydride (L-Selectride)–tert-BuOH¹⁰⁾ was unsuccessful. Only the starting 19 was recovered unchanged. Thus, 19 was converted by catalytic hydrogenation into the fully saturated compound 21, which was still expected to serve as a model compound in testing the crucial B-ring formation. The proton nuclear magnetic resonance (¹H-NMR) spectrum of 21 showed that substituents at the C-7 and C-8 positions of 21 could be assigned as cis from the small coupling constant ($J_{7,8} = 4.6 \, \text{Hz}$). One-carbon elongation of the C-7 side chain in 21 was then carried out. Mesylation followed by NaCN treatment (DMSO, 100 °C) yielded the cyanide 22 (45% yield from 21) along with the elimination product 23 (45% yield from 21). Selective reduction of 22 with LiBH₄ afforded the alcohol 24 in quantitative yield.

Then, the tetrahydropyranyl ether 25 derived from 24 was treated successively with diisobutylaluminum hydride, NaBH₄ and Ac₂O to give an acetate, whose acidic hydrolysis followed by Jones oxidation afforded the acetoxy acid 26 in 46% yield. The acetoxy acid 26 was converted, via the acetate 28, into the amide tosylate 29 by amidation (i, (COCl)₂, ii 27), hydrolysis (K₂CO₃-MeOH) and tosylation (p-TsCl-pyridine). Slow addition of a dioxane solution of 29 into a mixture of tert-BuOK in tert-BuOH brought about retro-Michael reaction of the cyanoethyl protecting group liberating sulfide anion, which attacked the tosylate intramolecularly, producing the twelve-membered lactam sulfide 30 in 26% yield from 26. Formation of the corresponding dimer or polymer was not detected in this reaction. Here, an alkyl group should be introduced to the position of the lactam carbonyl group to promote the subsequent cyclization to the eight-membered ketone. In the present model experiment, the methyl group was introduced in order to simplify the subsequent step. Peracid oxidation of 31 gave the lactam sulfoxide 32 in 91% yield.

The LDA-promoted intramolecular cyclization of the lactam sulfoxide 32 proceeded quite smoothly affording the eight-membered keto sulfoxide 33 in quantitative yield. Re-

ductive removal of the methylaminophenylsulfinyl group of 33 with Na–Hg in MeOH in the presence of Na₂HPO₄ produced the desired bicyclic ketone 5 in 68% yield as a sole product but the stereochemistry at the C-3 position remained unknown. The analytical and spectral data including ¹³C-nuclear magnetic resonance (¹³C-NMR) were consistent with the proposed structure.

Fig. 7

Experimental

All melting points are uncorrected. ¹H-NMR spectra were taken on a JEOL FX-60, FX-90, or GX-400 instrument and ¹³C-NMR spectra on a JEOL FX-100 in CDCl₃ solution with Me₄Si as an internal standard. A JEOL FX-60 instrument was routinely used. Infrared (IR) spectra were measured in CCl₄ solution with a JASCO A-3 spectrometer. Mass spectra (MS) were obtained with a Hitachi RMU-6M mass spectrometer and high resolution mass spectra were recorded on a Hitachi M-80 GC-MS spectrometer.

2-Acetoxymethyl-1,3,3-trimethylcyclohexene (7)— β -Ionone (6) was subjected to partial ozonolysis according to the procedure reported by Müller and Hoffmann. ⁶⁾ A solution of 6 (23.65 g, 123.2 mmol) in MeOH was treated with ozone at -27° C until the starting material was no longer detectable (about 4 h). After removal of an excess of ozone at the same temperature, NaBH₄ (22 g) was added portionwise to the stirred solution below 10 °C. The solvent was evaporated off under reduced pressure from the mixture, and the residue was dissolved in Et₂O–H₂O. The organic layer was separated, washed with brine, dried (MgSO₄), and concentrated. The resulting oily alcohol (21.81 g) was subjected to the next acetylation without purification. ¹H-NMR δ : 1.04 (6H, s), 1.75 (3H, br), 4.14 (2H, s).

The crude alcohol (21.81 g) obtained above was treated with acetic anhydride (120 ml) in pyridine (240 ml) in the presence of 4-dimethylaminopyridine (100 mg) for 3 d at room temperature and then concentrated under reduced pressure. SiO_2 column chromatography (hexane–AcOEt (19:1)) of the resulting oil afforded 7 (13.30 g, 55.1% yield from 6) as a pale yellow oil. IR cm⁻¹: 1735. ¹H-NMR δ : 0.99 (6H, s), 1.67 (3H, br), 2.05 (3H, s).

2-Acetoxymethyl-1,3,3-trimethyl-6-oxocyclohexene (8)—A mixture of 7 (9.03 g, 46.1 mmol) and SeO₂ (5.21 g) in dioxane (145 ml) was refluxed for 45 min and, after cooling, filtered through a short column packed with SiO₂. The solvent was evaporated off under reduced pressure. Jones reagent (14.7 ml) was added dropwise to a stirred solution of the resulting oil in acetone (245 ml) at 0 °C and stirring was continued for 5 min. After addition of 2-propanol (12 ml), usual work-up of the mixture and subsequent SiO₂ column chromatography (hexane–AcOEt (9:1)) afforded **8** (7.09 g, 73.3%) as a yellow oil. IR cm⁻¹: 1740, 1675. ¹H-NMR δ : 1.18 (6H, s), 1.81 and 2.09 (each 3H, s), 4.74 (2H, s). MS m/z: 210 (M⁺). High-resolution MS Calcd for C₁₂H₁₈O₃ (M⁺) m/z: 210.1256. Found m/z: 210.1279.

2-Acetoxymethyl-1,3,3-trimethyl-6-oxo-1,4-cyclohexadiene (9)—2,3-Dichloro-5,6-dicyanobenzoquinone (DDQ, 7.07 g) was added to a solution of **8** (8.65 g, 41.2 mmol) in benzene (290 ml) and the mixture was refluxed for 3 d with stirring. More DDQ (4.02 g) was added and reflux was continued for 3 d. The precipitate was filtered off and the filtrate was diluted with Et₂O. The solution was washed with aqueous Na₂CO₃ solution, water and brine, and dried (MgSO₄). Removal of the solvent gave an oil, which was subjected to SiO₂ column chromatography. Elution with hexane–AcOEt (9:1—4:1) afforded successively **9** (7.48 g, 87.3%) as a pale yellow oil and the starting keto acetate **8** (0.83 g). IR cm⁻¹: 1740, 1660. ¹H-NMR δ : 1.26 (6H, s), 1.94 and 2.09 (each 3H, s), 4.84 (2H, s), 6.22 and 6.78 (each 1H, d, J = 10 Hz). MS m/z: 208 (M⁺). High-resolution MS Calcd for C₁₂H₁₆O₃ (M⁺) m/z: 208.1099. Found m/z: 208.1116.

2-Acetoxymethyl-1,3,3-trimethyl-4-nitromethyl-6-oxocyclohexene (10)—i) Diisopropylamine (31.5 ml) and nitromethane (39 ml) were added to a solution of **9** (11.86 g, 57 mmol) in DMSO (17.5 ml) and the mixture was stirred for 20 d at 65 °C under nitrogen. Excess nitromethane and diisopropylamine were removed under reduced pressure from the mixture and the residue was extracted with Et₂O-AcOEt (3:1). The extract was washed with water, dried (MgSO₄), and concentrated to give an oil, which was subjected to SiO₂ column chromatography. The fraction eluted with hexane-AcOEt (4:1—7:3) afforded **10** (13.55 g, 88.3%) as crystals, which were recrystallized from CHCl₃-hexane to yield colorless prisms, mp 74—75 °C. IR cm⁻¹: 1740, 1680, 1375. ¹H-NMR δ : 1.13, 1.30 and 2.10 (each 3H, s), 1.85 (3H, br), 4.23 (1H, dd, J=8.8, 12.5 Hz), 4.67 (1H, dd, J=4.2, 12.5 Hz), 4.76 (2H, s). *Anal.* Calcd for C₁₃H₁₉NO₅: C, 57.98; H, 7.11; N, 5.20. Found: C, 57.90; H, 7.19; N, 5.02.

The starting dienone 9 (1.04 g, 8.8%) was recovered from the fraction eluted with hexane-AcOEt (4:1).

ii) In a sealed tube, a mixture of 9 (7.43 g, 35.7 mmol), DMSO (11 ml), diisopropylamine (20 ml) and nitromethane (24.5 ml) was stirred for 3 d at 110 °C under nitrogen. The same work-up as described above and subsequent SiO₂ chromatography gave 10 (5.10 g, 53.1%) and the starting dienone 9 (1.81 g, 24.4%).

6α-Acetoxymethyl-1,5,5-trimethyl-4β-nitromethylcyclohexene (12)—A mixture of 10 (4.785 g, 17.8 mmol) and p-toluenesulfonylhydrazide (4.02 g, 1.2 eq) in EtOH (15 ml) was stirred for 1 h at 65 °C under nitrogen. Removal of the solvent afforded 11 as yellow crystals (10.20 g), which were subjected to the next reductive degradation without purification. IR cm⁻¹: 3200, 1740, 1165. 1 H-NMR δ : 1.00, 1.18, 2.05 and 2.44 (each 3H, s), 1.88 (3H, br), 4.15 (1H, dd, J=7.8, 14.4 Hz), 4.68 (2H, s), 4.75 (1H, dd, J=4.8, 14.4 Hz), 7.33 and 7.87 (each 2H, d, J=8.4 Hz).

Catecholborane (3.66 ml, 34.1 mmol) was added dropwise to a stirred solution of 11 (10.20 g) obtained above in CHCl₃ (20 ml) at 0 °C, and the mixture was stirred for 2 h at room temperature under nitrogen. After careful addition of AcONa · 3H₂O (11.44 g, 84.1 mmol) with stirring, the mixture was refluxed for 2 h under nitrogen. The precipitate was filtered off and washed with CHCl₃. The solvent of the combined filtrate and extract was evaporated to give a yellow oil. SiO₂ column chromatography (hexane–AcOEt (9:1)) of the oil gave 12 (4.325 g, 95.3% from 10) as a colorless oil. IR cm⁻¹: 2850, 1740. ¹H-NMR δ : 0.88, 1.05 and 2.05 (each 3H, s), 1.75 (3H, br), 3.7—4.5 (2H, m), 4.21 (1H, dd, J=1.4, 11.5 Hz), 4.54 (1H, dd, J=4.8, 11.5 Hz), 5.3—5.6 (1H, m). MS m/z: 256 (M⁺+1), 195 (M⁺-60). High-resolution MS Calcd for C₁₃H₂₁NO₄ (M⁺) m/z: 255.1469. Found m/z: 255.1431.

Preparation of 6α-Acetoxymethyl-4β-formyl-1,5,5-trimethylcyclohexene (13)—Sodium methoxide (96%, 540 mg, 9.6 mmol) was dissolved in a solution of 12 (2.170 g, 8.51 mmol) in MeOH (15 ml) under nitrogen. To this solution, a mixture of aqueous TiCl₃ solution (16%, 36 ml) and AcONH₄ (18.70 g) in water (15 ml) was added rapidly with stirring on a water bath (bath temperature, 15 °C) and stirring was continued for 30 min under nitrogen. The reaction mixture was diluted with 1% HCl (about 300 ml) and extracted with AcOEt-Et₂O (1:1). The extract was washed with aqueous NaHCO₃ solution and brine, dried (MgSO₄), and concentrated. The crude aldehyde 13 (1.38 g) was obtained as a pale yellow oil, which was used in the next reaction without purification. IR cm⁻¹: 2900, 1740,

1720. ¹H-NMR δ : 1.00, 1.20 and 2.04 (each 3H, s), 1.74 (3H, br), 3.7—4.5 (2H, m), 5.3—5.6 (1H, m), 9.85 (1H, d, J = 2 Hz).

The Bicyclic Hemiacetal of 4α -Formyl- 6α -hydroxymethyl-1,5,5-trimethylcyclohexene (14)—A solution of the crude aldehyde 13 (1.38 g) prepared from 12 (2.170 g) in MeOH (12 ml) was treated with MeONa (690 mg) and the mixture was stirred for 15 h at room temperature. The solvent was evaporated off from the mixture. After addition of an excess of diluted HCl, the mixture was extracted with AcOEt. The extract was washed with aqueous NaHCO₃ solution and brine, dried (MgSO₄), and concentrated. SiO₂ column chromatography (hexane–AcOEt (4:1)) of the resulting solid gave 14 (761 mg, 48.5% from 12) as colorless crystals, which were recrystallized from CHCl₃-hexane to give colorless prisms, mp 49—50 °C. Thin layer chromatography (TLC, SiO₂) of the product afforded single spot with various solvent systems, but the ¹H-NMR spectrum showed that 14 was a diastereomeric mixture (about 5:2) in CDCl₃ solution. IR cm⁻¹: 3600, 3400. ¹H-NMR (signals due to a major component) δ : 1.00 and 1.15 (each 3H, s), 1.71 (3H, br), 3.51 (1H, dd, J=1, 11.3 Hz), 4.00 (1H, dd, J=1.5, 11.3 Hz), 5.21 (1H, br), 5.4—5.7 (1H, m); (signals due to a minor component) δ : 0.93 and 1.35 (each 3H, s), 1.67 (3H, br), 3.2—4.3 (2H, m), 5.06 (1H, br), 5.4—5.7 (1H, m). *Anal*. Calcd for C₁₁H₈O₂: C, 72.49; H, 9.96. Found: C, 72.28; H, 10.18.

 4α ,6α-Bisacetoxymethyl-1,5,5-trimethylcyclohexene (15)—The bicyclic hemiacetal 14 (168 mg) was treated with NaBH₄ (70 mg) in EtOH (5 ml) for 15 h at room temperature. Usual work-up of the mixture afforded a diol (177 mg) as a colorless oil. 1 H-NMR (90 MHz) δ : 0.85 and 1.08 (each 3H, s), 1.80 (3H, br), 3.43 (1H, dd, J=8, 10.4 Hz), 3.6—4.0 (1H, m), 5.56 (1H, br d, J=3.3 Hz). A part of the crude diol (18 mg) obtained above was acetylated with Ac₂O (0.5 ml) in pyridine (1 ml) (1 h, room temperature). Removal of the solvent gave an oil, whose SiO₂ column chromatography (hexane–AcOEt (19:1)) afforded 15 (23 mg) as a colorless oil. IR cm⁻¹: 1740. 1 H-NMR (90 MHz) δ : 0.80 and 1.10 (each 3H, s), 1.71 (3H, br), 2.05 (6H, s), 3.75—4.1 (1H, m), 4.15—4.4 (3H, m), 5.4—5.6 (1H, m). MS m/z: 268 (M⁺), 208 (M⁺-60). High-resolution MS Calcd for $C_{13}H_{20}O_{2}$ (M⁺-60) m/z: 208.1461. Found m/z: 208.1428.

 4β ,6α-Bisacetoxymethyl-1,5,5-trimethylcyclohexene (16)—A solution of the crude aldehyde 13 (28 mg) prepared from 12 (38 mg) in THF (2 ml) was treated with LiAlH₄ (15 mg) for 15 min on an ice bath. Usual work-up of the mixture gave a diol (24 mg) as a colorless gum. 1 H-NMR (90 MHz) δ : 0.82 and 1.10 (each 3H, s), 1.73 (3H, br), 3.37 (1H, dd, J=7.5, 10.3 Hz), 3.6—3.95 (1H, m), 5.62 (1H, br d, J=3.8 Hz). The crude diol (24 mg) obtained above was treated with Ac₂O (0.8 ml) in pyridine (1.5 ml) (1 h, room temperature). The mixture was concentrated under reduced pressure. SiO₂ column chromatography (hexane–AcOEt (19:1)) of the resulting gum afforded 16 (30 mg) as a colorless oil. IR cm⁻¹: 1740. 1 H-NMR (90 MHz) δ : 0.88, 1.04, 2.04 and 2.05 (each 3H, s), 1.74 (3H, d, J=1.3 Hz), 3.7—4.0 (1H, m), 4.05—4.4 (3H, m), 5.4—5.55 (1H, m). MS m/z: 208 (M $^+$ -60). High-resolution MS Calcd for C₁₂H₁₆O₃ (M $^+$ -60) m/z: 208.1098. Found m/z: 208.1083.

Ethyl (2*E*)-3-(5α-Hydroxymethyl-4,6,6-trimethyl-3-cyclohexen-1α-yl)acrylate (19)—A mixture of 14 (2.52 g, 13.8 mmol) and (carbethoxymethylene)triphenylphosphorane (14.48 g) in toluene (50 ml) was stirred for 24 h at 100 °C under argon. After cooling, the mixture was filtered and the filtrate was concentrated under reduced pressure. The residue was subjected to column chromatography on SiO_2 (hexane–AcOEt (4:1)), affording 19 (2.45 g, 70.5%) as a colorless oil. IR cm⁻¹: 3620, 1715, 1645. ¹H-NMR δ: 0.87 and 1.02 (each 3H, s), 1.29 (3H, t, J=7.1 Hz), 1.81 (3H, br), 3.7—4.0 (2H, m), 4.19 (2H, q, J=7.1 Hz), 5.4—5.6 (1H, m), 5.81 (1H, d, J=15.4 Hz), 6.98 (1H, dd, J=8, 15.4 Hz). MS m/z: 220 (M⁺ –28), 222 (M⁺ –30), 207 (M⁺ –45).

Ethyl 3-(3α-Hydroxymethyl-2,2,4α-trimethylcyclohexan-1α-yl)propionate (21) — A solution of 19 (2.45 g, 9.72 mmol) in AcOEt (50 ml) was stirred for 2 h at room temperature in the presence of PtO₂ (1.16 g) under hydrogen. The filtrate of the mixture was concentrated to give an oil, which was subjected to SiO₂ column chromatography. Elution with hexane–AcOEt (9:1) afforded 21 (2.42 g, 97.3%) as a colorless oil. IR cm⁻¹: 3620, 1715. 1 H-NMR (400 MHz) δ : 0.73 and 1.00 (each 3H, s), 0.95 (3H, d, J=7.6 Hz), 1.28 (3H, t, J=7.1 Hz), 1.40 (1H, ddd, J=4.2, 4.6, 10 Hz; 7β-H), 2.21 (1H, ddd, J=7.2, 8.9, 15.6 Hz, CH₂CO₂), 2.41 (1H, ddd, J=5.4, 9.4, 15.6 Hz, CH₂CO₂), 3.64 (1H, dd, J=10, 10.5 Hz, CH₂OH), 3.91 (1H, dd, J=4.2, 10.5 Hz, CH₂OH), 4.13 (2H, q, J=7.1 Hz).

Ethyl 3-(3 α -Cyanomethyl-2,2,4 α -trimethylcyclohexan-1 α -yl)propionate (22) — Methanesulfonyl chloride (1.1 ml) was added dropwise to a stirred solution of 21 (628 mg, 2.45 mmol) and Et₃N (3.4 ml) in CH₂Cl₂ (22 ml) at 0 °C, and stirring was continued for 20 min at 0 °C. Usual work-up of the reaction mixture gave the crude mesylate (820 mg, 100%) as a colorless oil, which was used for the next reaction without purification. ¹H-NMR δ : 0.77, 1.03 and 3.01 (each 3H, s), 0.97 (3H, d, J=7 Hz), 1.26 (3H, t, J=7.1 Hz), 4.13 (2H, q, J=7.1 Hz), 4.30 (1H, d, J=10.2 Hz), 4.50 (1H, dd, J=4.8, 10.2 Hz).

A mixture of the crude mesylate (820 mg) obtained above and NaCN (635 mg) in DMSO (22 ml) was stirred for 7 h at 100—105 °C under nitrogen and, after cooling, diluted with water. The ethereal extract of the mixture was washed with water, dried (MgSO₄), and concentrated. The resulting oil was chromatographed on SiO₂. Elution with hexane–AcOEt (19:1) afforded ethyl 3-(3-methylene-2,2,4 α -trimethylcyclohexan-1 α -yl)propionate (23) (235 mg, 44.8% from 21) as a colorless oil which tended to color at room temperature. ¹H-NMR δ : 1.02 (3H, d, J = 6.7 Hz), 1.08 and 1.17 (each 3H, s), 1.25 (3H, t, J = 7.1 Hz), 4.11 (2H, q, J = 7.1 Hz), 4.71 and 4.74 (each 1H, s). MS m/z: 238 (M⁺), 223 (M⁺ – 15). High-resolution MS Calcd for C₁₅H₂₆O₂ (M⁺) m/z: 238.1932. Found m/z: 238.1943.

Elution with hexane-AcOEt (9:1) afforded 22 (290 mg, 44.6% from 21) as a colorless oil. IR cm⁻¹: 2230, 1730.

¹H-NMR δ: 0.72 and 0.96 (each 3H, s), 0.96 (3H, d, J = 7.3 Hz), 1.26 (3H, t, J = 7.1 Hz), 2.19 (1H, d, J = 16.6 Hz), 2.65 (1H, dd, J = 5.5, 16.6 Hz), 4.13 (2H, q, J = 7.1 Hz). MS m/z: 250 (M⁺ – 15), 178 (M⁺ – 87). High-resolution MS Calcd for $C_{16}H_{27}NO_2$ (M⁺) m/z: 265.2040. Found m/z: 265.2017.

The intermediary mesylate (29 mg, 3.5%) was recovered from the last fraction eluted with hexane–AcOEt (4:1). 3-(3 α -Cyanomethyl-2,2,4 α -trimethylcyclohexan-1 α -yl)propan-1-ol (24)——A mixture of 22 (290 mg, 1.09 mmol) and excess LiBH₄ (700 mg) in Et₂O (40 ml) was refluxed for 3 h with stirring. The mixture was poured into ice-water, acidified with diluted HCl, and extracted with Et₂O–AcOEt (1:1). The extract was washed with aqueous NaHCO₃ solution and brine, and dried (MgSO₄). Removal of the solvent afforded 24 (240 mg) as a colorless oil, which was used for the next reaction without purification. IR cm⁻¹: 3610, 2230. ¹H-NMR δ : 0.70 and 0.94 (each 3H, s), 0.96 (3H, d, J=7.3 Hz), 2.19 (1H, d, J=16.4 Hz), 2.64 (1H, dd, J=5.5, 16.4 Hz), 3.64 (2H, t, J=6.1 Hz).

The Tetrahydropyranyl Ether 25 of 24—The crude alcohol 24 (240 mg) prepared from 22 (290 mg) was treated with dihydropyran (1 ml) and pyridinium p-toluenesulfonate (56 mg) in CH₂Cl₂ (15 ml) for 13 h at room temperature. The mixture was diluted with Et₂O, washed with aqueous NaHCO₃ solution and brine, dried (MgSO₄), and concentrated. The resulting oil was subjected to SiO₂ column chromatography, affording 25 (321 mg, 95.5% from 22) as a colorless oil by elution with hexane–AcOEt (19:1). ¹H-NMR δ : 0.69 and 0.93 (each 3H, s), 0.96 (3H, d, J= 7.1 Hz), 4.56 (1H, br).

3-[3 α -(2-Acetoxyethyl)-2,2,4 α -trimethylcyclohexan-1 α -yl]propionic Acid (26)—A solution of diisobutylaluminum hydride in hexane (1.76 M, 0.40 ml) was added to a solution of 25 (180 mg, 0.586 mmol) in toluene (6 ml) at -78 °C under argon. The mixture was stirred for 30 min at -78 °C and then for 1 h at room temperature. Saturated aqueous NH₄Cl solution (2 ml) and 5% $_{\odot}$ H₂SO₄ (2 ml) were added successively to the reaction mixture with stirring at 0 C, and stirring was continued for 10 min at room temperature. The mixture was diluted with ice-water and extracted with Et₂O-AcOEt (1:1). The extract was washed with aqueous NaHCO₃ solution and brine, dried (MgSO₄), and filtered through a short column packed with SiO₂. The solvent of the filtrate was removed to give the aldehyde (185 mg) as a colorless oil, which was used immediately for the next reaction without purification. 1 H-NMR $_{\odot}$: 0.73 and 0.90 (each 3H, s), 0.90 (3H, d, $_{\odot}$ J=6.6 Hz), 4.57 (1H, br), 9.72 (1H, dd, $_{\odot}$ J=1.9, 3.7 Hz).

The crude aldehyde (185 mg) obtained above was treated with LiAlH₄ (44 mg) in Et₂O (10 ml) at $-15\,^{\circ}$ C for 30 min followed by successive addition of water (2 drops) and 3 N NaOH (3 drops). The mixture was stirred for 10 min at room temperature, diluted with Et₂O, dried (MgSO₄), and filtered using Celite. Removal of the solvent yielded an alcohol (176 mg) as a colorless oil. ¹H-NMR δ : 0.71 and 0.91 (each 3H, s), 0.91 (3H, d, J=7.1 Hz), 4.57 (1H, br).

Treatment of the resulting alcohol (176 mg) with Ac₂O (1.5 ml) in pyridine (3 ml) (room temperature, 1 h) afforded the acetate (179 mg) as a colorless oil. 1 H-NMR δ : 0.71, 0.90 and 2.04 (each 3H, s), 0.91 (3H, d, J=6.8 Hz), 4.56 (1H, br).

A solution of the acetate (179 mg) obtained above in EtOH (6 ml) was stirred for 3 h at room temperature in the presence of p-TsOH (20 mg), and concentrated under reduced pressure. The extract of the residue with AcOEt was washed with aqueous Na₂CO₃ solution and brine, dried (MgSO₄), and concentrated. SiO₂ column chromatography (hexane–AcOEt (3:1)) of the resulting oil gave an acetoxy alcohol (74 mg, 46.7% from 25) as a colorless oil. IR cm⁻¹: 3610, 1735. 1 H-NMR δ : 0.71, 0.91 and 2.05 (each 3H, s), 0.91 (3H, d, J=7 Hz), 3.64 (2H, br t, J=5.8 Hz), 3.8—4.4 (2H, m). MS m/z: 252 (M⁺ – 18), 210 (M⁺ – 60). High-resolution MS Calcd for C₁₆H₃₀O₃ (M⁺ – H₂O) m/z: 252.2088. Found m/z: 252.2092.

The acetoxy alcohol (70 mg, 0.26 mmol) was treated with Jones reagent (0.3 ml) in acetone (4 ml) at room temperature for 30 min. Usual work-up of the mixture afforded **26** (80 mg) as a colorless oil, which was subjected to the next amidation without purification. IR cm⁻¹: 1735, 1705. ¹H-NMR δ : 0.73, 0.93 and 2.05 (each 3H, s), 0.92 (3H, d, J=7 Hz), 3.65—4.45 (2H, m).

N-Methyl-2'-[(2-cyanoethyl)thio]-3-[3α-(2-acetoxyethyl)-2,2,4α-trimethylcyclohexan-1α-yl]propionanilide (28)—A mixture of 26 (80 mg) obtained above and oxalyl chloride (0.1 ml) in benzene (1.5 ml) was stirred for 1 h at room temperature and then heated for 1 h at 60 °C. The solvent and excess oxalyl chloride were removed under reduced pressure to give an acid chloride as an oil, which was dissolved in THF (2 ml). This solution was added dropwise to a stirred suspension of 2-cyanoethyl 2-(methylamino)phenyl sulfide (27) (98 mg) and anhydrous K_2CO_3 (176 mg) in THF (4 ml) at 0 °C under nitrogen and stirring was continued for 10 min at 0 °C. The mixture was poured into ice and extracted with Et_2O -AcOEt. The extract was washed with brine, dried (MgSO₄), and concentrated. SiO₂ column chromatography (hexane-AcOEt (3:2)) of the resulting oil gave 28 (108 mg, 91% from the acetoxy alcohol) as a colorless gum. IR cm⁻¹: 1735, 1660. ¹H-NMR δ: 0.68, 2.03 and 3.18 (each 3H, s), 0.86 and 0.89 (each 1.5H, s), 0.86 (3H, d, J=7 Hz), 2.5—2.85 (2H, m), 3.1—3.4 (2H, m), 3.8—4.4 (2H, m), 7.1—7.6 (4H, m). MS m/z: 458 (M⁺), 443 (M⁺ – 15). High-resolution MS Calcd for $C_{25}H_{35}N_2O_3S$ (M⁺ – CH₃) m/z: 443.2366. Found m/z: 443.2349.

N-Methyl-2'-[(2-cyanoethyl)thio]-3-[3α-(2-p-toluenesulfonyloxyethyl)-2,2,4α-trimethylcyclohexan-1α-yl]propionanilide (29)—A mixture of acetoxy amide 28 (99 mg, 0.22 mmol) and K_2CO_3 (300 mg) in MeOH (5 ml) was stirred for 1.5 h at 0°C under nitrogen, and then diluted with water. The extract with CHCl₃ of the mixture was washed with brine, dried (MgSO₄), and concentrated to give a hydroxy amide (99 mg) as a colorless oil. IR cm⁻¹: 3600, 1660. 1 H-NMR δ: 0.68 and 3.18 (each 3H, s), 0.85 (3H, d, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.85 (2H, t, J = 7 Hz), 0.86 and 0.89 (each 1.5H, s), 3.59 (2H, t, J = 7 Hz), 0.85 (2H, t, J = 7

7.5 Hz).

The hydroxy amide (99 mg) obtained above was treated with p-toluenesulfonyl chloride (82 mg) in pyridine (2 ml) for 12 h in refrigerator. Usual work-up of the mixture and subsequent column chromatography on SiO₂ (hexane–AcOEt (3:2)) gave **29** (111 mg, 90.1% from **28**) as a colorless gum. IR cm⁻¹: 1660, 1370, 1175. ¹H-NMR δ : 0.62, 2.44 and 3.18 (each 3H, s), 0.75 (3H, d, J=6.5 Hz), 0.78 and 0.81 (each 1.5H, s), 2.5—2.9 (2H, m), 3.0—3.4 (2H, m), 7.1—7.5 (4H, m), 7.21 and 7.78 (each 2H, d, J=8.4 Hz).

3,10 α -Dimethyl-7 β ,11 β -dimethylmethano-14-thia-3-aza-1,2-benzocyclotetradecen-4-one (30)—A solution of 29 (19 mg, 0.033 mmol) in degassed dioxane (3 ml) was added slowly to a stirred mixture of *tert*-BuOK (97%, 20 mg) in DMF-dioxane (1:1, 10 ml) at 65—70 °C over 7 h under nitrogen and, after complete addition, the mixture was stirred for 1 h at the same temperature. The mixture was neutralized with dilute HCl and then concentrated. The ethereal extract of the residue was washed with brine, dried (MgSO₄), and concentrated. The resulting oil was subjected to SiO₂ column chromatography. The first fraction eluted with hexane-AcOEt (4:1) afforded 30 (3 mg, 26%) as colorless prisms (CHCl₃-hexane), mp 152—160 °C. IR cm⁻¹: 1650. ¹H-NMR δ : 0.23 (3H, br), 0.88 and 3.26 (each 3H, s), 0.81 (3H, d, J=6.6 Hz), 6.9—7.5 (3H, m), 7.6—7.9 (1H, m). MS m/z: 345 (M⁺), 330 (M⁺ – 15). Highresolution MS Calcd for C₂₁H₃₁NOS (M⁺) m/z: 345.2126. Found m/z: 345.2141. *Anal.* Calcd for C₂₁H₃₁NOS: C, 72.99; H, 9.04; N, 4.09. Found: C, 72.80; H, 9.02; N, 4.09.

The second fraction eluted with hexane-AcOEt (3:2) gave the starting amide 29 (7.5 mg, 39.5%).

3,5 ξ ,10 α -Trimethyl-7 β -,11 β -dimethylmethano-14-thia-3-aza-1,2-benzocyclotetradecen-4-one (31)—A solution of butyl lithium in hexane (1.45 M, 0.46 ml, 0.67 mmol) was added dropwise to a stirred solution of 30 (23 mg, 0.067 mmol) and diisopropylamine (0.093 ml, 0.66 mmol) in THF (1 ml) at -78 °C under argon. After 5 min, methyl iodide (0.2 ml) was added, and the mixture was stirred for 0.5 h at -78 °C and for 1 h at 0 °C. The reaction was quenched with saturated aqueous NH₄Cl solution and the mixture was extracted with AcOEt. The extract was washed with brine, dried (MgSO₄), and concentrated. SiO₂ column chromatography (hexane-AcOEt (17:3)) of the resulting oil gave 31 (21 mg, 87.8%) as a colorless gum. This product is a mixture of two stereoisomers (3:2), which were inseparable by SiO₂ column chromatography and TLC (SiO₂). IR cm⁻¹: 1650. ¹H-NMR (major component) δ : 0.24, 0.96 and 3.25 (each 3H, s), 0.80 (3H, d, J = 6.6 Hz), 1.19 (3H, d, J = 6.3 Hz), 6.9—7.45 (3H, m), 7.6—7.9 (1H, m); (minor component) δ : 0.24, 0.88 and 3.20 (each 3H, s), 0.49 (3H, d, J = 6.7 Hz), 1.32 (3H, d, J = 6.6 Hz), 6.9—7.45 (3H, m), 7.6—7.9 (1H, m). MS m/z: 359 (M⁺), 344 (M⁺ – 15). High-resolution MS Calcd for C₂₂H₃₃NOS (M⁺) m/z: 359.2280. Found m/z: 359.2250.

3,5 ξ ,10 α -Trimethyl-7 β ,11 β -dimethylmethano-14-thia-3-aza-1,2-benzocyclotetradecen-4-one 14-Oxide (32)—The lactam sulfide 31 (20 mg, 0.056 mmol) was treated with m-chloroperbenzoic acid (85%, 13 mg, 0.064 mmol) in CH₂Cl₂ (2 ml) for 10 min at 0 °C. The reaction mixture was diluted with Et₂O, washed with aqueous K₂CO₃ solution and brine, dried (MgSO₄), and concentrated. SiO₂ column chromatography (hexane-AcOEt (1:1)) of the resulting oil gave a 3:2 mixture of lactam sulfoxides 32 (19 mg, 90.9%) as a colorless caramel, which was inseparable by column chromatography and TLC (SiO₂). IR cm⁻¹: 1660, 1035. ¹H-NMR (major component) δ : 0.19, 0.97 and 3.32 (each 3H, s), 0.86 (3H, d, J=6.6 Hz), 1.19 (3H, d, J=6.5 Hz), 6.9—7.25 (1H, m), 7.45—7.8 (2H, m), 8.0—8.35 (1H, m); (minor component) δ : 0.19, 0.91 and 3.29 (each 3H, s), 0.52 (3H, d, J=6.7 Hz), 1.33 (3H, d, J=6.5 Hz), 6.9—7.25 (1H, m), 7.45—7.8 (2H, m), 8.0—8.35 (1H, m). MS m/z: 375 (M⁺), 359 (M⁺ – 16). High-resolution MS Calcd for C₂₂H₃₃NO₂S (M⁺) m/z: 375.2231. Found m/z: 375.2252.

Preparation of 3,8α,11,11-Tetramethyl-4-oxobicyclo[5.3.1]undecane (5) from 32 via 33—A solution of 32 (15 mg, 0.04 mmol) in THF (1 ml) containing hexamethylphosphoramide (0.067 ml) was added dropwise to a solution of LDA (0.70 mmol) prepared from diisopropylamine (0.10 ml) and n-butyl lithium (1.45 m hexane solution, 0.483 ml) in THF (1.5 ml) at -78 °C under argon. The mixture was stirred for 0.5 h at -78 °C and then for 1 h at 0 °C. The reaction was quenched with saturated aqueous NH₄Cl solution at -78 °C and the mixture was neutralized with dilute HCl. The ethereal extract of the mixture was washed with brine, dried (MgSO₄) and filtered through a short column packed with SiO₂. The solvent was removed to give 33 (15 mg) as a pale yellow oil, which was used in the next desulfurization without purification. IR cm⁻¹: 3300, 1725, 1700. ¹H-NMR δ: 2.89 (3H, d, J=5.1 Hz), 5.36 (1H, dd, J=2.7, 12.4 Hz).

Pulverized 5% Na–Hg (370 mg) was added to a suspension of 33 (15 mg) obtained above and Na₂HPO₄ (150 mg) in MeOH (2 ml) at 0 °C, and the mixture was stirred for 1 h at room temperature under nitrogen. Further Na₂HPO₄ (100 mg) and Na–Hg (250 mg) were added and stirring was continued for 0.5 h at room temperature. The mixture was diluted with Et₂O, washed with cold 3 N HCl and brine, dried (MgSO₄), and concentrated. The resulting oil was subjected to SiO₂ column chromatography (hexane–AcOEt (49:1)), affording 5 (6 mg, 67.6% from 32) as a colorless oil. IR cm⁻¹: 1725 (sh), 1695. ¹H-NMR δ : 0.88 and 1.08 (each 3H, s), 0.96 (3H, d, J=6.6 Hz), 1.03 (3H, d, J=6.8 Hz), 2.7—3.3 (3H, m). ¹³C-NMR δ : 19.26, 20.12, 27.84 and 34.58 (each –CH₃), 26.04, 27.33, 30.76, 42.38 and 43.32 (each –CH₂), 30.14, 40.43, 43.32 and 46.32 (each –CH₃), 35.44 (–C,–), 223.68 (C=O). MS m/z: 222 (M⁺), 207 (M⁺ – 15). High-resolution MS Calcd for C₁₅H₂₆O (M⁺) m/z: 222.1982. Found m/z: 222.1978.

Acknowledgment This work was supported in part by the Ministry of Education, Science and Culture (Grant-in-Aid for Scientific Research No. 567373 (1980)).

References and Notes

- 1) For recent reviews, see: R. W. Miller, J. Nat. Prod., 43, 425 (1980).
- M. C. Wani, H. L. Taylor, M. E. Wall, P. Coggon and A. T. McPhail, J. Am. Chem. Soc., 93, 2325 (1971); R. G. Powell, R. W. Miller and C. R. Smith, Jr., J. Chem. Soc., Chem. Commun., 1979, 102; R. W. Miller, R. G. Powell and C. R. Smith, Jr., J. Org. Chem., 46, 1469 (1981).
- 3) a) B. M. Trost and H. Hiemstra, J. Am. Chem. Soc., 104, 886 (1982); B. M. Trost and M. J. Fray, Tetrahedron Lett., 25, 4605 (1984); H. Neh, S. Blechert, W. Schnick and M. Jansen, Angew. Chem. Int. Ed. Engl., 23, 905 (1984); R. A. Holton, J. Am. Chem. Soc., 106, 5731 (1984); C. S. Swindell and S. J. deSolms, Tetrahedron Lett., 25, 3801 (1984); H. Nagaoka, K. Ohsawa, T. Takata and Y. Yamada, ibid., 25, 5389 (1984); T. Kojima, Y. Inouye and H. Kakisawa, Chem. Lett., 1985, 323; W. F. Berkowitz, J. Perumattam and A. Amarasekara, Tetrahedron Lett., 26, 3665 (1985); R. Kaczmarek and S. Blechert, ibid., 27, 2845 (1986); M. Fetizon, I. Hanna and R. Zeghdoudi, Synthetic Commun., 16, 1 (1986); H. Cervantes, D. Do Khac, M. Fetizon, F. Guir, J.-C. Beloeil, J.-Y. Lallemand and T. Prange, Tetrahedron, 42, 3491 (1986); J. D. Winkler, J. P. Hey and P. G. Williard, J. Am. Chem. Soc., 108, 6425 (1986); C. S. Swindell, B. P. Patel, S. J. deSolms and J. P. Springer, J. Org. Chem., 52, 2346 (1987); C. S. Swindell and B. P. Patel, Tetrahedron Lett., 28, 5275 (1987); b) S. F. Martin, J. B. White and R. Wagner, J. Org. Chem., 47, 3192 (1982); c) K. J. Shea and P. D. Davis, Angew. Chem. Int. Ed. Engl., 22, 419 (1983); K. J. Shea and J. W. Gilman, Tetrahedron Lett., 24, 657 (1983); K. Sakan and B. M. Craven, J. Am. Chem. Soc., 105, 3732 (1983); P. A. Brown, P. R. Jenkins, J. Fawcett and D. R. Russell, J. Chem. Soc., Chem. Commun., 1984, 253; K. Sakan and D. A. Smith, Tetrahedron Lett., 25, 2081 (1984); K. J. Shea and J. W. Gilman, J. Am. Chem. Soc., 107, 4791 (1985); P. A. Wender and N. C. Ihle, ibid., 108, 4678 (1986); K. J. Shea, J. W. Gilman, C. D. Haffner and T. K. Dougherty, ibid., 108, 4953 (1986); P. A. Brown and P. R. Jenkins, J. Chem. Soc., Perkin Trans. 1, 1986, 1303; R. V. Bonnert and P. R. Jenkins, J. Chem. Soc., Chem. Commun., 1987, 1540; P. A. Wender and M. L. Snapper, Tetrahedron Lett., 28, 2221 (1987); P. A. Wender and N. C. Ihle, ibid., 28, 2451 (1987); d) A. S. Kende, S. Johnson, P. Sanfilippo, J. C. Hodges and L. N. Jungheim, J. Am. Chem. Soc., 108, 3513 (1986); cf. T. Kato, H. Takayanagi, T. Suzuki and T. Uyehara, Tetrahedron Lett., 1978, 1201; A. S. Kende, M. Benechie, D. P. Curran, P. Fludzinski, W. Swenson and J. Clardy, ibid., 1979, 4513; I. Kitagawa, H. Shibuya, H. Fujioka, A. Kajiwara, S. Tsuji, Y. Yamamoto and A. Takagi, Chem. Lett., 1980, 1001; Y. Inouye, C. Fukaya and H. Kakisawa, Bull. Chem. Soc. Jpn., 54, 1117 (1981); R. Z. Andriamialisoa, M. Fetizon, I. Hanna, C. Pascard and T. Prange, Tetrahedron, 40, 4285 (1984); G. R. Clark, J. Lin and M. Nikaido, Tetrahedron Lett., 25, 2645 (1984); C. B. Jackson and G. Pattenden, ibid., 26, 3393 (1985); M. J. Bergley, C. B. Jackson and G. Pattenden, ibid., 26, 3397 (1985); W. F. Berkowitz and A. S. Amarasekara, ibid., 26, 3663 (1985); J. D. Winkler, J. P. Hey and S. D. Darling, ibid., 27, 5959 (1986); C. S. Swindell and S. F. Britcher, J. Org. Chem., 51, 793 (1986); B. M. Trost and H. Hiemstra, Tetrahedron, 42, 3323 (1986); G. Berthiaume, J.-F. Lavallee and P. Deslongchamps, Tetrahedron Lett., 27, 5451 (1986); J.-F. Lavallee, G. Berthiaume and P. Deslongchamps, ibid., 27, 5455 (1986); L. Pattersson, T. Frejd and G. Magnusson, ibid., 28, 2753 (1987); D. H. Hua, W.-Y. Gung, R. A. Ostrander and F. Takusagawa, J. Org. Chem., 52, 2509 (1987).
- 4) Y. Ohtsuka and T. Oishi, Tetrahedron Lett., 1979, 4487; idem, Chem. Pharm. Bull., 31, 443, 454 (1983).
- 5) Y. Ohtsuka, S. Niitsuma, H. Tadokoro, T. Hayashi and T. Oishi, J. Org. Chem., 49, 2326 (1984).
- 6) N. Müller and W. Hoffmann, Synthesis, 1975, 781.
- 7) In order to avoid confusion, the numbering for the bicyclo[5.3.1]undecane to be synthesized as the final compound has been used for the cyclohexane derivatives in this paper.
- 8) J. E. McMurry and J. Melton, J. Org. Chem., 38, 4367 (1973).
- 9) M. Yamashita, Y. Kato and R. Suemitsu, Chem. Lett., 1980, 847.
- 10) J. M. Fortunato and B. Ganem, J. Org. Chem., 41, 2194 (1976).