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Three New Neolignans, Fargesones A, B and C, from the Flower Buds of Magnolia fargesii

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Three new neolignans, fargesone A (I), fargesone B (II) and fargesone C (III), together with two known compounds, denudatin B (IV) and syringin (V), have been isolated from the flower buds of *Magnolia fargesii*, and their structures determined.

Keywords -- Magnolia fargesii; hsin-i; fargesone A; fargesone B; fargesone C; neolignan

The flower buds of *Magnolia fargesii* (Chinese name hsin-i) have been used for nasal empyema and headache. From this herb we have isolated four lignans: pinoresinol dimethyl ether, lirioresinol-B dimethyl ether, magnolin and fargesin.¹⁾ Recently, we have been searching for Ca²⁺-antagonizing activity in Chinese herbs. We found that the extract of hsin-i exhibited a Ca²⁺-antagonizing activity on the taenia coli of the guinea pig. Meanwhile three new neolignans, fargesone A (I), fargesone B (II) and fargesone C (III), together with two known compounds, denudatin B (IV) and syringin (V), have been isolated from the CHCl₃ extracts of the herb,²⁾ and we present here the structural elucidation of the three new neolignans.

Fargesone A (I) was obtained as a viscous oil, $[\alpha]_D - 150^\circ$ (c = 1.1, CHCl₃), with a molecular weight of 372 as determined from the mass (MS) spectrum. Its ultraviolet (UV) spectrum (250, 270 nm) and infrared (IR) spectrum (1665, 1612, 1490 cm⁻¹) suggested the presence of a substituted benzene and an α,β -unsaturated carbonyl group. Fargesone A was considered to be a hydrobenzofuranoid neolignan on the basis of the proton nuclear magnetic resonance (¹H-NMR) spectrum (Table I) and MS spectrum (m/z 162, 46%), which clearly indicated the presence of a 3,4-methylenedioxyphenyl function and an Me-CH-CH-Ar

moiety with Ar and Me groups in a *trans*-relationship.³⁻⁵⁾ The chemical shift of the methyl group at δ 1.05 (d, J=7 Hz) also indicated a *trans*-relationship with the phenyl group.³⁻⁵⁾ Irradiation of the methyl doublet at δ 1.05 (H-9) caused the multiplet at δ 2.17—2.24 to collapse to a doublet δ 2.18 (d, $J_{H-7,H-8}=9.5$ Hz, H-8). Irradiation of the methine doublet at δ 4.67 (H-7) simplified the multiplet at δ 2.17—2.24 to a quartet at δ 2.18 (J=7 Hz, H-8). The additional C_6 – C_3 unit could be accounted for by a cyclohexenone having two methoxyls (δ 3.20 and 3.76), one linked to the fully substituted sp^3 -carbon and the other to the sp^2 -carbon, and an allyl group (δ 2.51—2.58, m, H_A -7'; 2.66—2.76, m, H-1' and H_B -7'; 5.06—5.16, 2H, m, H-9'; 5.81—5.90, m, H-8') linked to the trisubstituted sp^3 -carbon. The chemical shift of H-8 (δ 2.17—2.24) indicated OMe-3' and H-8 to be in a *trans*-relationship.^{6,7)} The OMe-3' must be in the β -orientation, due to the coupling constant of H-7 and H-8 being 9.5 Hz.⁶⁾ The presence of an nuclear Overhauser effect (NOE) between OMe-3' and H-2' established the *cis* junction of

TABLE I.	¹ H-NMR Data for Fargesone A (I), Fargesone B (II)
	and Fargesone C (III) (300 MHz, CDCl ₃)

Proton No.	I	II	III
2	6.78 br s		7.29 s
5 6	6.74 br s	6.65—6.72 m	6.76 d (8) ^{a)} 7.47 d (8) ^{a)}
7	4.67 d (10) ^{a)}	4.20—4.23 m	
8	2.17—2.24 m	2.01—2.11 m	$4.18 \text{ q } (7)^{a}$
9	1.05 d (7) ^{a)}	1.05 d (7) ^{a)}	$1.32 d (7)^{a}$
1′	2.66—2.76 m	2.63—2.70 m	
2′	4.40 d (9) ^{a)}	4.20—4.23 m	6.42 s
5′	5.53 s	5.52 s	5.39 s
7 _A ′ 7 _B ′	2.51—2.58 m 2.66—2.76 m	2.29—2.39 m 2.79—2.87 m	3.13 d (6.6) ^{a)}
8′	5.81—5.90 m	5.83—5.97 m	5.85—5.91 m
9′	5.06—5.16 m	5.03—5.15 m	5.105.12 m
O-CH ₂ -O	5.92 s	5.91 s	6.00 s
OMe-3'	3.20 s	3.29 s	3.06 s
OMe-4'	3.76 s	3.70 s	3.42 s

a) Figures in parentheses are coupling constants in Hz.

TABLE II. ¹³C-NMR Data (δ Values) for Fargesone A (I), Fargesone B (II) and Fargesone C (III)

Carbon No.	I	II	III
1	134.06 s	133.68 s	131.65 s
2	105.35 d	103.21 d	105.46 d
3	147.74 s	147.65 s	151.76 s
4	147.26 s	147.19 s	148.08 s
5	106.47 d	106.47 d	107.48 d
6	120.12 d	119.81 d	124.33 d
7	86.94 d	85.79 d	197.36 s
8	51.00 d	49.24 d	47.82 s
9	8.56 q	10.99 q	12.50 q
1'	52.47 d	53.60 d	140.32 s
2′	79.70 d	80.89 d	138.07 d
3′	84.39 s	82.04 s	76.53 s
4′	172.12 s	173.12 s	170.84 s
5′	107.90 d	107.80 d	108.10 d
6′	196.07 s	196.59 s	185.97 s
7′	30.94 t	28.93 t	32.86 t
8′	135.20 d	136.13 d	135.09 d
9′	117.55 t	116.70 t	116.98 t
O-CH ₂ -O	100.92 t	100.99 t	101.01 t
OMe-3'	51.21 q	53.14 q	52.05 q
OMe-4'	56.11 q	55.92 q	55.41 q

Run in CDCl₃ at 300 MHz. Assignment established by off-resonance and DEPT methods.

the two fused rings, and therefore H-2' must be in the β -axial orientation. The chemical shift of the H-2' doublet at δ 4.40 (J=9 Hz) confirmed the *trans*-diaxial relationship of H-1' and H-2'. Therefore the allyl group must be in the β -equatorial orientation. The structure of I was further confirmed by carbon-13 nucler magnetic resonance (13 C-NMR) spectroscopy (Table

No. 5

II); the methoxyl group, linked to the sp^2 -carbon, located at the β -position (C-4') from the carbonyl group showed a downfield singlet at δ 172.12.8 Based on the above evidence, the structure of fargesone A can be unambiguously assigned as formula I.

Fargesone B (II) was also obtained as a viscous oil, $[\alpha]_D - 190^\circ$ (c = 1.0, CHCl₃), with a molecular weight of 372. The UV, IR and ¹H-NMR (Table I) spectra of I and II were very similar. Fargesone B also had an α,β -unsaturated carbonyl group and the partial structure Me-CH-CH-Ar (Ar=3,4-methylenedioxyphenyl) with the Ar and Me group in a *trans*-

relationship (δ 1.05, d, J=7 Hz, H-9).³⁻⁵⁾ This was confirmed by irradiation of the methyl double doublet at δ 2.66 (J=9.5 and 4.8 Hz). Irradiation at δ 2.06 (H-8) caused the H-7 signal (J=9 Hz, H-8). The multiplet at δ 4.20—4.23 (2H) was the overlapped signals of H-7 and H-2'. Since irradiation of the multiplet at δ 4.22 caused (i) the multiplet at δ 2.06 (H-8) to collapse to a quartet (J=7 Hz), and (ii) the multiplet at δ 2.63—2.70 (H-1') to collapse to a double doublet at δ 2.66 (J=9.5 and 4.8 Hz). Irradiation at δ 2.06 (H-8) caused the H-7 signal to collapse to a singlet at δ 4.21 and H-2' to appear as a doublet (J=4.2 Hz) at δ 4.22. Therefore the allyl group must be in the α -axial orientation due to the small coupling constant (J=4.2 Hz) between H-1' and H-2'. The structure II was further confirmed by ¹³C-NMR (Table II), and the evidence that the methoxyl group linked to the sp^2 -carbon was located at the β -position (C-4') from the carbonyl group. From the above evidence, fargesone B must be an epimer of fargesone A. Chemical correlation between fargesone A and fargesone B was achieved as follows. When fargesone B was treated with 1% methanolic NaOH at room temperature overnight, fargesone A was obtained quantitatively. This result also confirmed the orientation of the allyl group in fargesone A or fargesone B.

Fargesone C (III) was obtained as colorless prisms, mp $126-127\,^{\circ}$ C, $[\alpha]_D - 206.5^{\circ}$ (c = 1.0, CHCl₃) with a molecular weight of 370 (from the MS spectrum). Its UV spectrum (234, 280, 310 nm) and IR spectrum (1670, 1642, 1610 cm⁻¹) suggested the presence of a substituted aromatic ketone and an α,β -unsaturated carbonyl group. The ¹H-NMR (δ 6.00, 6.76, 7.29 and 7.47) (Table I) and MS spectrum (m/z 149) clearly indicated the presence of a 3,4-methylenedioxyphenyl ketone. The chemical shifts at δ 1.32 (3H, d, J=7 Hz) and δ 4.18 (1H, q, J=7 Hz) were assigned to H-9 and H-8, respectively. The additional C₆-C₃ unit could be accounted for by a cyclohexadienone having two methoxyls (δ 3.06 and 3.42), one linked to the fully substituted sp^3 -carbon and the other to the sp^2 -carbon, and an allyl group [δ 3.13

(2H, d, H-7'); δ 5.12 (2H, m, H-9'); δ 5.91 (1H, m, H-8')] linked to the sp^2 -carbon, with two isolated olefinic proton signals at δ 5.39 and 6.42. The structure III was further confirmed by ¹³C-NMR (Table II), and the evidence that the methoxyl group linked to the sp^2 -carbon was located at the β -position (C-4') from the carbonyl group. The gross structure of fargesone C can be assigned as formula III. An analysis of the stereochemistry is in progress.

In addition to the three new neolignans discussed above, two known compounds, namely denudatin B (IV) and syringin (V), were isolated from the same source. Their structures were confirmed by comparison of their spectral data with those in the literature. $^{4,9-11}$)

Experimental

Melting points were determined on a Yanagimoto micro melting point apparatus and are uncorrected. IR spectra were recorded on a Hitachi 260-30 instrument. UV spectra were taken on a Hitachi 124 spectrophotometer. Optical rotations were measured with JASCO DIP-140 polarimeter. ¹H- and ¹³C-NMR spectra were taken on a Brucker AM 300 at 300 MHz, using tetramethylsilane (TMS) as an internal standard.

Extraction and Isolation—Commercial Chinese crude drug (1.5 kg), the flower buds of Magnolia fargesii (hsini), was extracted with CHCl₃ (31×2) and EtOH (31×2), successively. The CHCl₃ extract (110 g) was chromatographed on a silica gel (800 g) column. The following fractions were eluted, in order, with the indicated solvent systems: Fr. I (n-hexane: EtOAc = 10:1), fr. II (EtOAc) and fr. III (MeOH). Fr. II was further chromatographed on a silica gel column, using a linear gradient of increasing concentrations of EtOAc in n-hexane, to give six fractions (fr. II-a \rightarrow fr. II-f). Fr. II-a was purified by preparative thin layer chromatography (PTLC) to obtain denudatin B (IV). Fr. II-b was purified on a Lichroprep Si60 column (Merck Co.) with CH₂Cl₂. Two new compounds, fargesone A (I) and fargesone B (II), were isolated from this fraction. Fr. II-c was purified by PTLC to give a new crystalline compound, fargesone C (III). The EtOH extract was chromatographed on a silica gel column and eluted exhaustively with a CHCl₃-MeOH gradient, then further purified by PTLC to give a glucoside, syringin (V).

Fargesone A (I)—A viscous oil, $[\alpha]_D - 150^\circ$ (c = 1.1, CHCl₃). UV λ_{max}^{MeOH} nm (ϵ): 250 (17798), 290 (4683) IR (KBr) cm⁻¹: 3000—2800, 1665, 1612, 1490, 1445, 1365, 1245, 1195, 1040, 930, 810, 750. MS m/z (%): 372 (100, M⁺), 195 (60), 190 (50), 167 (48), 162 (46), 149 (98). ¹H-NMR: Table II. ¹³C-NMR: Table II.

Fargesone B (II)—A viscous oil, $[\alpha]_D - 196^\circ$ (c = 1.0, CHCl₃). UV λ_{max}^{MeOH} nm (ϵ): 245 (13924), 290 (2531). IR (KBr) cm⁻¹: 3000—2800. 1670, 1610, 1490, 1440, 1365, 1245, 1195, 1040, 932, 810, 750. MS m/z (%): 372 (100, M⁺), 195 (60), 190 (60), 167 (48), 162 (46), 149 (98). ¹H-NMR: Table I. ¹³C-NMR: Table II.

Conversion of II to I—Compound II (2 mg) was added to 10 ml of 1% methanolic NaOH and left to stand at room temperature overnight. Then the reaction mixture was diluted with H_2O (10 ml) and extracted with CHCl₃ (20 ml). The extracts were dried (Na₂SO₄), and the solvent was evaporated off under reduced pressure. The product was identical with compound I as judged by comparison with an authentic sample.

Fargesone C (III)—Colorless prisms, mp 126—127 °C, $[\alpha]_D$ –206.5° $(c=1.0, CHCl_3)$. UV λ_{max}^{MeOH} nm (ϵ): 234 (22222), 280 (8994), 310 (9523). IR (KBr) cm⁻¹: 3100—2300, 1670, 1640, 1610, 1440, 1355, 1245, 1225, 1168, 1080, 1030, 995, 970, 920, 875, 830. MS m/z(%): 370 (19, M⁺), 193 (100), 162 (26), 149 (57). ¹H-NMR: Table I. ¹³C-NMR: Table II.

Denudatin B (IV)—A viscous oil, [α]_D +83.2° (c = 1.1, CHCl₃). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ε): 208 (18010), 235 (17820), 285 (6900), 300 (6350). IR (KBr) cm⁻¹: 3000—2800, 1675, 1645, 1625, 1505, 1460, 1350, 1275, 1180, 1140, 1060, 1030, 920. MS m/z: 356 (M⁺), 341, 325, 178. ¹H-NMR (CDCl₃) δ : 1.10 (3H, d, J = 7 Hz, H-9), 2.16 (1H, m, H-8), 3.10 (3H, s, OMe-5′), 3.14 (2H, m, H-7′), 3.85 (6H, s, Ar-OMe), 5.08—5.14 (2H, m, H-9′), 5.32 (1H, d, J = 9.5 Hz, H-7). 5.78 (1H, s, H-3′), 5.81—5.92 (1H, m, H-8′), 6.23 (1H, s, H-6′), 6.72—6.88 (3H, m, Ar-H).

Syringin (V)—White prisms, mp 187—189 °C. IR (KBr) cm⁻¹: 3600—3100, 2900, 1640, 1589, 1505, 1415, 1240, 1130, 1090, 1025, 965. MS m/z: 210, 182, 167.

Syringin Pentaacetate (Va) — Acetylation of syringin (Ac₂O/pyr.) resulted in the formation of prisms, mp 110—111 °C. IR (KBr) cm⁻¹: 3000—2850, 1740, 1580, 1220, 1020. ¹H-NMR (CDCl₃) δ : 2.00 (3H, s), 2.02 (6H, s), 2.05 (3H, s), 2.08 (3H, s), 3.66 (1H, m, H-5'), 3.81 (6H, s, Ar-OMe), 4.09 (1H, dd, J=12, 3 Hz, H_A-6'), 4.22 (1H, dd, J=12, 5 Hz, H_B-6'), 4.68 (2H, d, J=6 Hz, =C-CH₂-O), 5.05 (1H, d, J=7 Hz, H-1'), 5.18—5.31 (3H, m, H-2', 3' and 4'), 6.14—6.23 (1H, m, C=CH-CH₂), 6.54 (1H, d, J=18 Hz, Ar-CH=C), 6.57 (2H, d, Ar-H).

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