





Fig. 1. Fluorescence Development vs. ACE Amount

Assays were performed as described under Experimental Procedure. Substrate, **1d**. ACE amount ranged from  $1.1 \times 10^{-6}$  to  $4.4 \times 10^{-5}$  units (in 2.22 ml of assay solution).

those for **1a** and **1b**. Higher affinity of **1c** and **1d** to ACE is probably due to the presence of the same amino acid (phenylalanine) at the  $P_1$  site as in angiotensin I. Therefore, the values of  $k_{cat}/K_m$  for **1c** and **1d** are higher than those for the known fluorogenic substrates (**3**, **5**, **6**). From these kinetic results, substrate **1d** was selected and the linearity of the relationship between ACE concentration and fluorescence intensity was examined. As shown in Fig. 1, substrate **1d** can be used to detect as little as around  $10^{-6}$  unit of ACE (in 2.22 ml of assay solution) using a substrate concentration of  $4.30 \times 10^{-6}$  M.

When the hydrolytic activity for **1d** was measured in the presence of *p*-chloromercuribenzoic acid in buffer solution, the activity was not decreased. Therefore, this ACE used for the assay of the bimane substrates is free from cysteine proteinase. Although hydrolyses of **1c** and **1d** containing the phenylalanine residue by a chymotrypsin-like enzyme might occur, hydrolytic activity of ACE was not affected by the treatment of ACE with *p*-tosyl-L-phenylalanyl chloromethyl ketone, which is a specific inhibitor of chymotrypsin. Enkephalinase has a similar substrate specificity to ACE, but the following assay conditions discriminate between them. Although enkephalinase is inhibited in the presence of chloride ion, ACE is rather activated<sup>11)</sup> and maximum activity is observed with 300 mM sodium chloride.

It has thus been demonstrated that the bimane system can be used in the fluorometric assay for ACE. Compared with other fluorogenic substrates for ACE, the bimane system has several advantages. 1) Bimane substrates **1c** and **1d** have higher  $k_{cat}/K_m$  values than those for previously reported fluorescent substrates (**3**, **5** and **6**). 2) Bimane substrates can provide a continuous and rapid fluorometric assay, though a known conventional procedure [spectrofluorometric (using OPA) determination of the hydrolyzed product (his-leu) from **3**] is an end-point assay. 3) Substrates **4–6** can also be employed for continuous assay, but the fluorescence intensity of the bimane group is higher than that of the aminobenzoyl group (of **5**) or tryptophan (of **4**). 4) Bimane derivatives have a longer excitation maximum wavelength (399 nm) than other fluorogenic substrates (365 nm for **3** with OPA, 360 nm for **5**, 290 nm for **4** and **6**), so that interference from most biological compounds is low.<sup>12)</sup> 5) The bimane system has a constant fluorescence intensity over a wide pH range of 1–9.

Because of these advantages, these bimane substrates are expected to be potent fluorogenic substrates for the micro-determination of ACE activity. Further application of these substrates to biochemical studies or clinical diagnoses, including monitoring of therapy with antihypertensive drugs such as captopril (which acts by inhibiting ACE) or monitoring or elevated serum ACE level in sarcoidosis, are being examined.

#### Experimental

Melting points were determined with a Yamato MP-21 melting point apparatus and are uncorrected. Infrared (IR) spectra were measured with a JASCO IRA-1 spectrophotometer. Optical rotations were obtained with a JASCO DIP-4 polarimeter. Ultraviolet (UV) spectra were measured with a Hitachi 200-10 spectrophotometer. Fluorescence spectra were recorded with a Hitachi 650-60 fluorescence spectrophotometer.

**General Procedure for the Preparation of Substrate (1)** To a stirred solution of 0.6 mmol of the corresponding tripeptide (or amino acid) and 101 mg (1.2 mmol) of sodium bicarbonate in 20 ml of water, 190 mg (0.5 mmol) of 1,7-dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethanecarboxylic acid hydroxysuccinimide ester (bimane-succinimide ester)<sup>13)</sup> in 10 ml of acetonitrile was added. After being stirred overnight at room temperature, the solution was concentrated to about two-thirds of the original volume under reduced pressure, diluted with 30 ml of water, and then washed twice with ethyl acetate. The aqueous solution was acidified with concentrated hydrochloric acid, then extracted with ethyl acetate. The extract was washed with water and dried over anhydrous sodium sulfate.

**1,7-Dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethylcarbonyl-glycyl-L-tryptophyl-L-leucine (**1a**)** Obtained from ethyl acetate by condensation as a pale yellow fine powder of mp 190–195 °C. 227 mg, 71%. IR (Nujol): 1730, 1650, 1635  $\text{cm}^{-1}$ .  $[\alpha]_D^{25} -11.1^\circ$  ( $c=0.738$ , DMF). Anal. Calcd for  $\text{C}_{31}\text{H}_{38}\text{N}_6\text{O}_7\text{S} \cdot 1/2\text{H}_2\text{O}$ : C, 57.48; H, 6.07; N, 12.97; S, 4.95. Found: C, 57.35; H, 6.13; N, 12.76; S, 4.76.

**1,7-Dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethylcarbonyl-glycyl-L-tryptophyl-L-proline (**1b**)** Obtained from ethanol-ethyl acetate by condensation; 162 mg, 52%. IR (Nujol): 1730, 1625  $\text{cm}^{-1}$ .  $[\alpha]_D^{25} -17.6^\circ$  ( $c=1.04$ , DMF). Anal. Calcd for  $\text{C}_{30}\text{H}_{34}\text{N}_6\text{O}_7\text{S} \cdot \text{H}_2\text{O}$ : C, 56.24; H, 5.66; N, 13.12; S, 5.00. Found: C, 56.53; H, 5.63; N, 13.03; S, 5.17.

**1,7-Dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethylcarbonyl-L-phenylalanyl-L-tryptophyl-L-leucine (**1c**)** Obtained from methanol-ethyl acetate by condensation as a pale yellow powder of mp 156–166 °C (dec.). 220 mg, 60%. IR (Nujol): 1730, 1650  $\text{cm}^{-1}$ .  $[\alpha]_D^{25} -0.042^\circ$  ( $c=0.497$ , DMF). Anal. Calcd for  $\text{C}_{38}\text{H}_{44}\text{N}_6\text{O}_7\text{S} \cdot \text{H}_2\text{O}$ : C, 61.10; H, 6.20; N, 11.25; S, 4.29. Found: C, 61.04; H, 5.96; N, 11.16; S, 4.40.

**1,7-Dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethylcarbonyl-L-phenylalanyl-L-tryptophyl-L-proline (**1d**)** The peptide dissolved in water (20 ml) and dimethyl sulfoxide (30 ml) as a cosolvent was treated with the ester. **1d** was obtained as a pale yellow powder from ethyl acetate by condensation, mp 153.5–160.5 °C. 259 mg, 73%. IR (Nujol): 1740, 1640, 1605  $\text{cm}^{-1}$ .  $[\alpha]_D^{25} -0.27^\circ$  ( $c=0.275$ , DMF). Anal. Calcd for  $\text{C}_{37}\text{H}_{40}\text{N}_6\text{O}_7\text{S} \cdot \text{H}_2\text{O}$ : C, 60.80; H, 5.79; N, 11.50; S, 4.39. Found: C, 60.64; H, 5.62; N, 11.21; S, 4.50.

**1,7-Dioxo-2,5,6-trimethyl-1*H*,7*H*-pyrazolo[1,2-*a*]pyrazol-3-yl-methylthiomethylcarbonyl-L-phenylalanine (**1f**)** Prepared from 40 mg (0.24 mmol) of phenylalanine and 76 mg (0.20 mmol) of bimane-succinimide ester.<sup>1)</sup> Recrystallization from acetonitrile gave pale yellow needles of mp 169–170 °C. 63 mg, 61%. IR (Nujol): 3300, 1725, 1630, 1615, 1600  $\text{cm}^{-1}$ .  $[\alpha]_D^{25} +0.15^\circ$  ( $c=0.213$ , DMF). Anal. Calcd for  $\text{C}_{21}\text{H}_{23}\text{N}_3\text{O}_5\text{S}$ : C, 58.73; H, 5.40; N, 9.79; S, 7.45. Found: C, 58.61; H, 5.31; N, 9.67; S, 7.40.

**Assay Procedure for ACE with Substrates 1a–1d** a) Kinetic Parameter ( $K_m$  and  $k_{cat}$ ) Measurement: A solution of ACE (50  $\mu\text{l}$ ) was prepared at  $9.5 \times 10^{-2}$  unit/ml (concentration of ACE was estimated from the optical density at 280 nm based on a molar absorptivity of  $204000 \text{ M}^{-1} \text{cm}^{-1}$  and a molecular weight of 130000<sup>13)</sup>) (ACE, EC 3.4.15.1, from rabbit lung (Sigma Chem. Co.); Specific activity, 2.2 units/mg protein with hippuryl-L-histidyl-L-leucine; one unit of enzyme activity is the amount required to catalyze the formation of 1.0  $\mu\text{mol}$  of hippuric acid per min at 37 °C). To this solution, 30–100  $\mu\text{l}$  of the substrate solution [ $1.91 \times 10^{-3}$  M for **1d**, 25% dimethyl sulfoxide (DMSO) (final concentration of DMSO was 1.2%) in 50 mM Tris-HCl buffer (pH 8.0) containing 300 mM NaCl and

2.0 ml of buffer solution were added, and the increase in emission at 483 nm (appearance of Bim-SCH<sub>2</sub>CO-L-Phe-OH, emission at 399 nm) was measured. Rates of hydrolysis were established from the rate of increase in fluorescence intensity based on the fluorescence intensities of various concentrations of standard bimeane derivative (**1f**). Kinetic parameters for the hydrolysis were obtained from Lineweaver-Burk plots. Kinetic parameters for other substrates (**1a**–**1c**) were also obtained by similar procedures.

b) Linear Relation of Fluorescence Intensity vs. Enzyme Concentration: A solution (5–200  $\mu$ l) of ACE ( $2.2 \times 10^{-4}$  unit/ml) in buffer solution was added to a solution of 2.0 ml of buffer solution and 20  $\mu$ l ( $1.91 \times 10^{-4}$  M; 2.5% DMSO, finally 0.023% DMSO) of substrate (**1d**) (final concentration:  $4.30 \times 10^{-6}$  M) at 37°C, and measurement was carried out in the manner described in a).

#### References

- 1) Part XVI of "Organic Fluorescent Reagents." Part XV: E. Sato and Y. Kanaoka, *Chem. Pharm. Bull.*, **36**, 4494 (1988).
- 2) Y. Kanaoka, T. Takahashi, H. Nakayama and K. Tanizawa, *Chem. Pharm. Bull.*, **33**, 1721 (1985) and references cited therein.
- 3) E. Sato, M. Miyakawa and Y. Kanaoka, *Chem. Pharm. Bull.*, **32**, 336 (1984).
- 4) E. Sato, A. Matsuhisa, M. Sakashita and Y. Kanaoka, *Chem. Pharm. Bull.*, **36**, 3496 (1988).
- 5) E. M. Kosower and B. Pazhenchevski, *J. Am. Chem. Soc.*, **100**, 6515 (1978).
- 6) E. Sato, M. Sakashita, Y. Kanaoka and E. M. Kosower, *Bioorg. Chem.*, **16**, 723 (1988).
- 7) a) H.-S. Cheung, F.-L. Wang, M. A. Ondetti, E. F. Sabo and D. W. Cushman, *J. Biol. Chem.*, **255**, 401 (1980); b) D. W. Cushman and H.-S. Cheung, *Biochem. Pharmacol.*, **20**, 1637 (1971).
- 8) A. Persson and I. B. Wilson, *Anal. Biochem.*, **83**, 296 (1977).
- 9) A. Carmel and A. Yaron, *Eur. J. Biochem.*, **87**, 265 (1978).
- 10) G. Fleminger, D. Goldenberg and A. Yaron, *FEBS Lett.*, **135**, 131 (1981).
- 11) J.-C. Schwartz, B. Malfroy and S. De La Baume, *Life Sci.*, **29**, 1715 (1981).
- 12) E. Koller and O. S. Wolfbeis, *Anal. Biochem.*, **143**, 146 (1984).
- 13) P. Bunning, B. Holmquist and J. F. Riordan, *Biochemistry*, **22**, 103 (1983).