

## FACILE SYNTHESIS OF ISOFLAVONES BY THE CROSS-COUPLING REACTION OF 3-iodochromone WITH ARYLBORONIC ACIDS

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We describe a convenient preparation of isoflavones by the cross-coupling reactions of 3-iodochromone (1) with arylboronic acids (2) catalyzed by tetrakis(triphenylphosphine)palladium(0).

**KEYWORDS** isoflavone; 3-iodochromone; phenylboronic acid; cross-coupling reaction; tetrakis(triphenylphosphine)palladium

Isoflavonoids are widely distributed in plants, especially in Leguminosae, Moraceae, Iridaceae and Rosaceae. Some of these isoflavonoids have physiological activities such as estrogenic, antispastic, antibacterial, antiprotozoal, and insecticidal.<sup>1)</sup>

Synthesis of isoflavones is needed to confirm the structures of natural products, and for this the long-established methods of ring closure of a C<sub>1</sub> unit onto a deoxybenzoin are still widely used. While the addition of one-carbon unit to a deoxybenzoin can be accomplished efficiently, synthesis of the deoxybenzoin itself may proceed in rather poor yields, and starting materials are not readily available.

Recently the cross-coupling reactions of arylhalides with alkenes, alkynes, and arylboronic acids catalyzed by Pd-complex have been studied extensively.<sup>2)</sup> We have applied the reaction to the novel and efficient synthesis of isoflavones (3) by using tetrakis(triphenylphosphine)palladium as a catalyst.

A typical experiment can be described as follows. Isoflavone (3a): To a solution of 3-iodochromone (1) (0.5 mmol), Pd(PPh<sub>3</sub>)<sub>4</sub> (0.05 mmol), and 1 ml of 2M-Na<sub>2</sub>CO<sub>3</sub> in benzene (20 ml) was added phenylboronic acid (2a) (2 mmol) in 2 ml of ethanol under argon with stirring at room temperature. The reaction mixture was refluxed for 15 h followed by the addition of 1 ml of 30% H<sub>2</sub>O<sub>2</sub> to oxidize excess phenylboronic acid. After being stirred for 1 h at room temperature the mixture was poured into ice and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 15 ml). The organic layer was washed with water and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Removal of the solvent gave a pale yellow product which was purified by silica gel chromatography with benzene as an eluent to give isoflavone (3a), mp 135 °C, in 97.8% yield. Palladium(0) catalyzed the cross-coupling reactions of other arylboronic acids (2b-2e), prepared by the reported method<sup>3)</sup> through the corresponding Grignard reagents, with 3-iodochromone (1) using the above general procedure, are summarized in Table I. The results show that the cross-coupling reaction of substituted arylboronic acids (2) with 1 is a quite general and simple method for the synthesis of isoflavones (3).

When palladium acetate was used as the catalyst instead of Pd(PPh<sub>3</sub>)<sub>4</sub>, the yield of isoflavone was decreased to ca. 30%.

Thiophen-2-boronic acid (4) was also coupled with 1 under the same condition to afford 5, mp 138-139 °C, in 70.2% yield. This implies that we could introduce aromatic and heterocyclic ring systems into the 3-position of chromone under the mild condition in the presence of "Pd" catalyst.

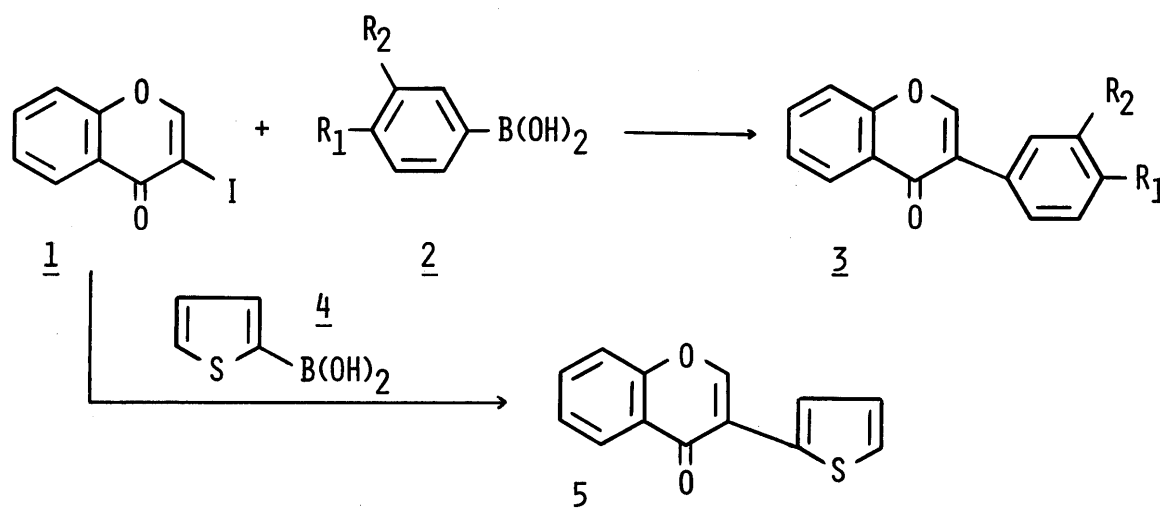


Table I. Reactions of 3-Iodochromone (**1**) with Arylboronic Acids (**2**) Catalyzed by  $\text{Pd(PPh}_3)_4$

	Arylboronic acid ( <b>2</b> )		Yield of <b>3</b> (%) <sup>a</sup>	mp of <b>3</b> (°C)
	R <sub>1</sub>	R <sub>2</sub>		
a	H	H	97.8	135
b	CH <sub>3</sub>	H	93.2	154–156
c	OCH <sub>3</sub>	H	94.1	144
d	OCH <sub>3</sub>	OCH <sub>3</sub>	95.8	144–145
e	–O–CH <sub>2</sub> –O–		92.8	158

a) Satisfactory spectroscopic data were obtained for all **3** and yields were after the isolation by the silica gel column chromatography.

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