

# Studies on Sialic Acids. XVIII. Synthesis of Aryl- $\alpha$ -glycosides of 3-Deoxy-D-glycero-D-galacto-2-nonulopyranosonic Acid (KDN)

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Condensation of sodium phenoxide, sodium *p*-nitrophenoxide, and sodium 4-methylumbelliferonate with benzyl (4,5,7,8,9-penta-*O*-acetyl-3-deoxy-D-glycero- $\beta$ -D-galacto-2-nonulopyranosyl chlorid)onate under the Williamson reaction conditions gave the corresponding  $\alpha$ -glycosides in good yields. Deprotection reaction of these  $\alpha$ -glycosides gave sodium salts of phenyl-, *p*-nitrophenyl-, and 4-methylumbelliferonyl- $\alpha$ -glycosides of 3-deoxy-D-glycero-D-galacto-2-nonulopyranosonic acid (KDN). The structure and stereochemistry of the glycosylation products were determined by proton nuclear magnetic resonance ( $^1\text{H-NMR}$ ) and circular dichroism (CD) spectral analysis.

**Keywords** KDN; sialic acid; aryl glycoside;  $^1\text{H-NMR}$ ; CD

Recently, we have reported the synthesis of 3-deoxy-D-glycero-D-galacto-2-nonulopyranosonic acid (**1**, KDN) by base-catalyzed condensation and the synthesis of 2-*O*-methyl and 2-*O*-cholesterol derivatives of **1** under Koenigs-Knorr-like reaction conditions.<sup>1)</sup> In studies of sialic acids, the aryl- $\alpha$ -glycosides of sialic acid are useful chromogenic substrates for tracing and quantifying sialidase (EC 3.2.1.18,  $\alpha$ -*N*-acetylneuraminosyl glycohydrolase) in biological materials.<sup>2,3)</sup> The phenols released in the enzymatic hydrolysis can be determined spectrophotometrically, either directly or after a specific transformation. The *p*-nitrophenyl-,<sup>4)</sup> the *m*-methoxyphenyl-,<sup>5)</sup> and the 4-methylumbelliferonyl- $\alpha$ -glycosides<sup>6)</sup> of sialic acid are known as appropriate substrates. In this paper, we wish to report the synthesis of aryl- $\alpha$ -glycosides of **1** under Williamson reaction conditions.<sup>7)</sup> The structure and stereochemistry of the glycosylation products were determined by proton nuclear magnetic resonance ( $^1\text{H-NMR}$ ) and circular dichroism (CD) spectral analysis.

In the previous paper,<sup>1)</sup> we examined the Koenigs-Knorr-like reaction, using benzyl (4,5,7,8,9-penta-*O*-acetyl-3-deoxy-D-glycero- $\beta$ -D-galacto-2-nonulopyranosyl bromid)onate (**3**) as a glycosyl donor without purification, because of its instability. This time, we prepared the corresponding 2-chloro derivative, benzyl (4,5,7,8,9-penta-*O*-acetyl-3-

deoxy-D-glycero- $\beta$ -D-galacto-2-nonulopyranosyl chlorid)onate (**4**), as a new glycosyl donor which is stable enough to purify on a silica gel column. Compound **4** was synthesized by treatment of benzyl 2,4,5,7,8,9-hexa-*O*-acetyl-3-deoxy-D-glycero- $\beta$ -D-galacto-2-nonulopyranosonate (**2**) with acetic acid saturated with dry hydrogen chloride gas. The structure of **4** was elucidated by  $^1\text{H-NMR}$  comparison with **3**. Condensation of **4** with sodium salts<sup>8)</sup> of phenol, *p*-nitrophenol, and 4-methylumbelliferone in *N,N*-dimethylformamide gave the corresponding  $\alpha$ -glycosides, benzyl (phenyl 4,5,7,8,9-penta-*O*-acetyl-3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**5a**), benzyl (*p*-nitrophenyl 4,5,7,8,9-penta-*O*-acetyl-3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**5b**), and benzyl (4-methylumbelliferonyl 4,5,7,8,9-penta-*O*-acetyl-3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**5c**) in 31–77% yields, respectively. These compounds were deprotected with 0.1 *N* sodium hydroxide to give sodium (phenyl 3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**6a**), sodium (*p*-nitrophenyl 3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**6b**), and sodium (4-methylumbelliferonyl 3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (**6c**) in 80–87% yields, respectively.

In the  $^1\text{H-NMR}$  spectra, the chemical shifts at the 3- $\text{H}''$  (eq) double-doublet resonance of the protected derivatives

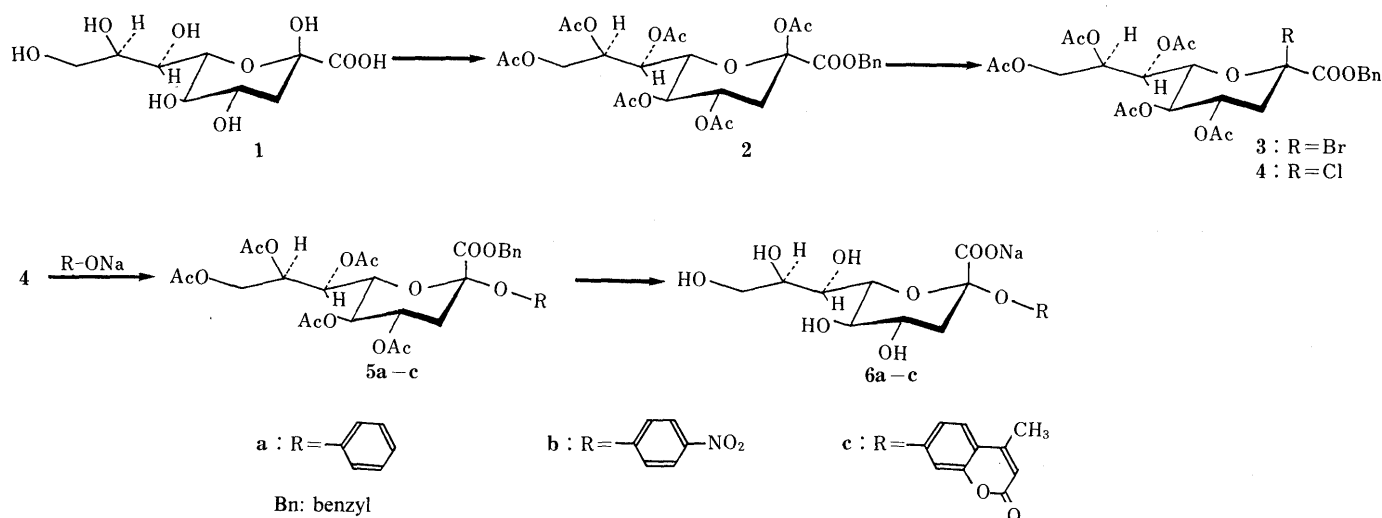


Chart 1

TABLE I. Proton Chemical Shifts and Spin-Coupling Data for **5a**,<sup>a)</sup> **5b**,<sup>a)</sup> **5c**,<sup>a)</sup> **6a**,<sup>b)</sup> **6b**,<sup>b)</sup> and **6c**<sup>b)</sup>

Compound	Spin couplings (Hz)									
	KDN moiety									
	$J_{3a,3e}$	$J_{3a,4}$	$J_{3e,4}$	$J_{4,5}$	$J_{5,6}$	$J_{6,7}$	$J_{7,8}$	$J_{8,9}$	$J_{8,9'}$	$J_{9,9'}$
<b>5a</b>	13.0	11.8	4.5	9.6	9.6	—	—	4.1	2.0	12.8
<b>5b</b>	13.0	11.9	4.6	9.5	9.5	—	—	4.2	2.0	12.5
<b>5c</b>	13.0	11.9	4.6	9.6	9.6	—	—	3.8	2.0	12.1
<b>6a</b>	12.5	11.5	4.8	9.5	10.0	0.8	—	—	—	—
<b>6b</b>	12.8	11.8	4.8	9.5	9.5	1.0	—	—	—	—
<b>6c</b>	12.8	11.8	4.7	9.5	10.0	1.0	—	—	—	—

Compound	Chemical Shifts ( $\delta$ )											
	KDN moiety											
	3-H <sub>ax</sub>	3-H <sub>eq</sub>	4-H	5-H	6-H	7-H	8-H	9-H	9'-H	COCH <sub>3</sub>	CH <sub>2</sub> Ph	Aromatic
<b>5a</b>	2.21	2.83	4.97	4.91	4.56	5.40	5.40	4.19	4.29	2.01—2.14	5.03	6.97—7.41
<b>5b</b>	2.27	2.84	4.98	5.20	4.74	5.38	5.38	4.12	4.21	2.03—2.06	4.92, 5.18	6.96—8.02
<b>5c</b>	2.26	2.85	4.99	4.94	4.67	5.40	5.40	4.16	4.27	2.03—2.20	4.96, 5.18	6.83—7.35
<b>6a</b>	1.81	2.76	—	3.52	3.75	—	—	—	—	—	—	—
<b>6b</b>	1.90	2.72	—	3.56	4.04	—	—	—	—	—	—	—
<b>6c</b>	1.89	2.74	—	3.56	3.96	—	—	—	—	—	—	—

a) Recorded for a solution in CDCl<sub>3</sub> at 300 MHz. b) Recorded for a solution in D<sub>2</sub>O at 400 MHz.

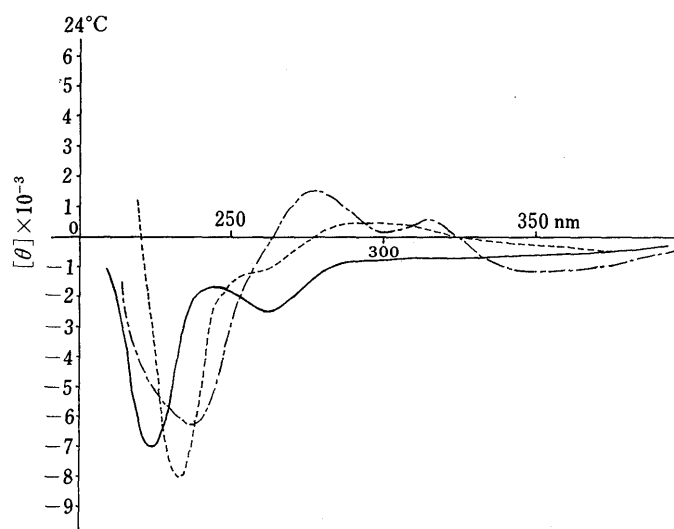


Fig. 1. CD Curves of **6a** (—), **6b** (---), and **6c** (— · —) in MeOH

(**5a**, **5b**, and **5c**) were 2.83—2.85 ppm, and those of the deprotected compounds (**6a**, **6b**, and **6c**) were 2.72—2.76 ppm as shown in Table I. Empirical studies of *N*-acetyl-D-neuraminic acid derivatives<sup>9,10)</sup> indicated that for  $\alpha$ -anomers the chemical shift of 3-H'' (eq) varies between 2.6 and 2.8 ppm. For  $\beta$ -anomers, the range is 2.1—2.5 ppm. The chemical shifts of the above compounds were therefore considered to be those of the  $\alpha$ -anomers.

Figure 1 shows the CD spectra of the deprotected derivatives (**6a**, **6b**, and **6c**). In the previous paper, we reported the CD spectra of several derivatives of **1**,<sup>1)</sup> and the peak around 220—230 nm was assigned to the  $n\text{-}\pi^*$  Cotton effect of the carboxyl group. The negative sign of the Cotton effect was assigned to the  $\alpha$ -anomer and the positive sign to the  $\beta$ -anomer. In Fig. 1, all curves show a negative  $n\text{-}\pi^*$  Cotton effect around 220—230 nm, and this result supports the assignment of **6a**, **6b**, and **6c** as the  $\alpha$ -anomers.

In conclusion, the glycosylation of **4** with several sodium phenoxides under the Williamson reaction conditions stereospecifically gave  $\alpha$ -glycosides in high yields. The stereochemistry of these derivatives could be determined on the basis of <sup>1</sup>H-NMR and CD spectral analysis.

#### Experimental

**Melting points** were measured with a Yamato melting point apparatus and are uncorrected. Optical rotations were measured with a JASCO DIP-181 digital polarimeter. Thin layer chromatography (TLC) was performed on Silica gel GF254 (Merck) plates; detection was done under ultraviolet (UV) irradiation and by spraying 5% sulfuric acid solution. Fast atom bombardment mass spectra (FAB-MS) and infrared (IR) spectra were measured with JEOL JMA-3100 and JASCO IR-A2 instruments, respectively. CD spectra were measured in a 0.1 cm cell with a JASCO J-20 spectrometer. <sup>1</sup>H-NMR spectra were measured with Varian VXR-300 and XL-400 spectrometers. Tetramethylsilane (TMS) in CDCl<sub>3</sub> or sodium 3-(trimethylsilyl)-1-propanesulfonate (DSS) in D<sub>2</sub>O was used as an internal reference. Column chromatography was conducted on Silica gel 60 (70—230 mesh).

**Benzyl (4,5,7,8,9-Penta-O-acetyl-3-deoxy-D-glycero- $\beta$ -D-galacto-2-nonulopyranosyl chlorid)onate (4)** A solution of **2** (3.0 g, 4.9 mmol) in acetic acid (30 ml) and acetyl chloride (1 ml) was saturated with dry hydrogen chloride in an ice-water bath. After 18 h at room temperature, the reaction mixture was evaporated *in vacuo*, and the residue was purified by silica gel column chromatography with ether:hexane = 2:1 to give **4** (2.8 g, 97%) as a colorless amorphous solid.  $[\alpha]_D^{25}$  -61.9° (*c* = 0.32, CHCl<sub>3</sub>). MS (EI) *m/z*: 586 (M<sup>+</sup>). Anal. Calcd for C<sub>26</sub>H<sub>31</sub>ClO<sub>13</sub>: C, 53.20, H, 5.32. Found: C, 53.12; H, 5.42. IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 2950, 1740, 1430, 1360. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): 2.00, 2.03, 2.03, 2.05, 2.07 (3H  $\times$  5, *s*  $\times$  5, COCH<sub>3</sub>), 2.24 (1H, dd, *J* = 14.0, 11.3 Hz, 3-H<sub>ax</sub>), 2.84 (1H, dd, *J* = 14.0, 5.0 Hz, 3-H<sub>eq</sub>), 4.10 (1H, dd, *J* = 12.5, 5.0 Hz, 9-H), 4.34 (1H, dd, *J* = 12.5, 2.5 Hz, 9'-H), 4.48 (1H, dd, *J* = 10.5, 2.3 Hz, 6-H), 4.95 (1H, t, *J* = 10.0 Hz, 5-H), 5.22 (1H, m, 8-H), 5.22, 5.34 (1H  $\times$  2, *d*  $\times$  2, *J* = 12.0 Hz, CH<sub>2</sub>Ph), 5.44 (1H, m, 4-H), 5.47 (1H, dd, *J* = 8.0, 2.3 Hz, 7-H), 7.40 (5H, m, aromatic H).

**Benzyl (Phenyl 4,5,7,8,9-Penta-O-acetyl-3-deoxy-D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (5a)** A solution of sodium phenoxide (100 mg, 0.85 mmol) and **4** (500 mg, 0.85 mmol) in dry *N,N*-dimethylformamide (20 ml) was stirred for 2 h at room temperature. Then the reaction mixture was poured into water (100 ml) and extracted with ethyl acetate. The combined extracts were evaporated *in vacuo* to give a syrup, which was chromatographed on a silica gel column with ethyl acetate:hexane = 1:1 to give **5a** (170 mg, 31%) as an amorphous powder.  $[\alpha]_D^{25}$  -15.3° (*c* = 0.31,

$\text{CHCl}_3$ ). MS (EI)  $m/z$ : 644 ( $\text{M}^+$ ). *Anal.* Calcd for  $\text{C}_{32}\text{H}_{36}\text{O}_{14}$ : C, 59.62; H, 5.63. Found: C, 59.52; H, 5.75. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 2945, 1750, 1375. The  $^1\text{H}$ -NMR data are summarized in Table I.

**Benzyl (*p*-Nitrophenyl 4,5,7,8,9-Penta-*O*-acetyl-3-deoxy- $\alpha$ -D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (5b)** A solution of sodium *p*-nitrophenoxide (165 mg, 0.85 mmol) and **4** (500 mg, 0.85 mmol) in dry *N,N*-dimethylformamide (20 ml) was stirred for 2 h at room temperature. The mixture was processed as described for **5a** to give **5b** (450 mg, 77%) as an amorphous powder.  $[\alpha]_{\text{D}}^{25} -16.2^\circ$  ( $c=0.33$ ,  $\text{CHCl}_3$ ). MS (EI)  $m/z$ : 689 ( $\text{M}^+$ ). *Anal.* Calcd for  $\text{C}_{32}\text{H}_{35}\text{NO}_{16}$ : C, 55.73; H, 5.12; N, 2.03. Found: C, 55.70; H, 5.13; N, 1.89. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 2970, 1740, 1600, 1520, 1370, 1350. The  $^1\text{H}$ -NMR data are summarized in Table I.

**Benzyl (4-Methylumbelliferonyl 4,5,7,8,9-Penta-*O*-acetyl-3-deoxy- $\alpha$ -D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (5c)** A solution of sodium 4-methylumbelliferonate (170 mg, 0.85 mmol) and **4** (500 mg, 0.85 mmol) in *N,N*-dimethylformamide (20 ml) was stirred for 2 h at room temperature. The mixture was processed as described for **5a** to give **5c** (410 mg, 66%) as an amorphous powder.  $[\alpha]_{\text{D}}^{25} -31.4^\circ$  ( $c=0.28$ ,  $\text{CHCl}_3$ ). MS (EI)  $m/z$ : 727 ( $\text{M}^+$ ). *Anal.* Calcd for  $\text{C}_{36}\text{H}_{38}\text{O}_{16}$ : C, 59.50; H, 5.27. Found: C, 59.74; H, 5.43. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 2940, 1750, 1620, 1380. The  $^1\text{H}$ -NMR data are summarized in Table I.

**Sodium (Phenyl 3-Deoxy- $\alpha$ -D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (6a)** A stirred solution of **5a** (150 mg, 0.23 mmol) in methanol (15 ml) was treated with 0.1 N sodium hydroxide (15 ml) at room temperature. After 1 h, the solution was cooled to  $0^\circ\text{C}$ , neutralized with Dowex-50 ( $\text{H}^+$ ) resin, filtered and concentrated. The residue was freeze-dried to give **6a** (65 mg, 81%) as an amorphous powder.  $[\alpha]_{\text{D}}^{25} -12.6^\circ$  ( $c=0.13$ , MeOH). MS (FAB)  $m/z$ : 367 ( $\text{M}^+ + 1$ ). *Anal.* Calcd for  $\text{C}_{15}\text{H}_{19}\text{NaO}_9 \cdot 2\text{H}_2\text{O}$ : C, 44.78; H, 5.76. Found: C, 44.55; H, 5.66. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3370, 1600, 1400. The  $^1\text{H}$ -NMR data are summarized in Table I.

**Sodium (*p*-Nitrophenyl 3-Deoxy- $\alpha$ -D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (6b)** A stirred solution of **5b** (300 mg, 0.44 mmol) in methanol (30 ml) was treated with 0.1 N sodium hydroxide (30 ml) at room temperature and the mixture was processed as described for **6a** to give **6b** (147 mg, 87%) as an amorphous powder.  $[\alpha]_{\text{D}}^{25} +27.4^\circ$  ( $c=0.39$ , MeOH). MS (FAB)  $m/z$ : 412 ( $\text{M}^+ + 1$ ). *Anal.* Calcd for  $\text{C}_{15}\text{H}_{18}\text{NNaO}_{11} \cdot 2\text{H}_2\text{O}$ : C, 40.28; H, 4.96; N, 3.13. Found: C, 40.26; H, 4.89; N, 3.00. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ :

3370, 1720, 1590, 1490, 1340. The  $^1\text{H}$ -NMR data are summarized in Table I.

**Sodium (4-Methylumbelliferonyl 3-Deoxy- $\alpha$ -D-glycero- $\alpha$ -D-galacto-2-nonulopyranosid)onate (6c)** A stirred solution of **5c** (300 mg, 0.41 mmol) in methanol (30 ml) was treated with 0.1 N sodium hydroxide (30 ml) at room temperature and the mixture was processed as described for **6a** to give **6c** (140 mg, 80%) as an amorphous powder.  $[\alpha]_{\text{D}}^{25} +21.5^\circ$  ( $c=0.54$ , MeOH). MS (FAB)  $m/z$ : 449 ( $\text{M}^+ + 1$ ). *Anal.* Calcd for  $\text{C}_{19}\text{H}_{21}\text{NaO}_{11} \cdot 2\text{H}_2\text{O}$ : C, 47.11; H, 5.20. Found: C, 47.64; H, 4.88. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400, 1690, 1610, 1390. The  $^1\text{H}$ -NMR data are summarized in Table I.

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## References and Notes

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