Asernestioside C, a New Minor Saponin from the Roots of *Astragalus ernestii* COMB.; First Example of Negative Nuclear Overhauser Effect in the Saponins

Hui Kang Wang,*,a Kan He, Li Ji,b Yasuhiro Tezuka,b Tohru Kikuchi,*,b and Isao Kitagawac

Department of Chinese Material Medica, Shanghai College of Traditional Chinese Medicine, 530 Linlin Road, Shanghai, China, Research Institute for Wakan-Yaku (Oriental Medicines), Toyama Medical and Pharmaceutical University, Sugitani 2630, Toyama 930–01, Japan, and Faculty of Pharmaceutical Sciences, Osaka University, Yamada-oka 1–6, Suita, Osaka 565, Japan. Received February 2, 1989

A minor saponin, asernestioside C (1), was isolated from the roots of *Astragalus ernestii* COMB. (Leguminosae) and its structure was determined from the two-dimensional nuclear magnetic resonance spectra and nuclear Overhauser effect (NOE) difference spectra. In this case, negative NOE was observed between the 3-H and the anomeric proton of 3-O-sugar moiety and between the 24-H and the 26-H₃. Examples of negative NOE were also found in some other saponins.

Keywords Asernestioside C; Astragalus ernestii; Astragali Radix; saponin; Asernestioside C; cycloastragenol; 2-D NMR; negative NOE

Dried roots of Astragalus membranaceus BUNGE and other Astragalus spp. (Leguminosae) are used as a crude drug "Huang qi" (Astragali Radix), which is prescribed in several Chinese traditional medicines as an antiperspirant, a diuretic, or a tonic.¹¹ Among Astragalus species, A. membranaceus BUNGE and A. sieversianus PALL were investigated chemically by several groups of authors, and lipids,²¹ flavonoids,³¹ γ-aminobutyric acid,⁴¹ L-canavanine,⁵¹ polysaccarides,⁶¹ and saponins²¹,৪⟩ have so far been characterized.

In the course of our study on the constituents of Chinese Astragalus plants, we have isolated three new saponins named asernestioside A (2), B (3), and C (1) from the roots of A. ernestii COMB. and determined the structures of the major constituents, 2 and 3,90 based on the chemical and spectral evidence. In this paper we wish to report the isolation and structure determination of the minor constituent, asernestioside C (1), by means of proton, carbon, and two-dimensional nuclear magnetic resonance (1H-, 13C-, and 2-D NMR) spectroscopies without a derivatiza-

tion or degradation.

The 95% ethanolic extract from the dried roots of A. ernestii was treated with methanol and the methanol-soluble fraction was partitioned between butanol and water. The crude saponin obtained from the butanol-soluble portion was further separated by a combination of reversed-phase column chromatography and silica gel column chromatography to give a small amount of asernestioside C (1) together with asernestioside D (2) and D (3).

Asernestioside C (1), mp 204—207 °C, showed $[\alpha]_D$ – 13.2° (MeOH) and its molecular formula was determined to be $C_{49}H_{80}O_{19}$ by fast-atom bombardment mass spectroscopy (FAB-MS) measurement (M⁺+1, m/z 973). The infrared (IR) spectrum of 1 (in KBr) exhibited a strong hydroxyl absorption band at 3375 cm⁻¹ and an estercarbonyl absorption band at 1732 cm⁻¹. In the ¹H-NMR spectrum (in pyridine- d_5), it showed signals due to cyclopropane-methylene protons at δ 0.26 and 0.56 ppm

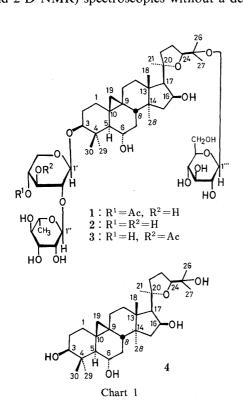


Fig. 1. Partial Structures of Asernestioside C (1) Deduced from the ¹H- and ¹³C-NMR Data

: long-range coupling observed in ¹H-¹H COSY.

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Table I. ¹H-NMR Chemical Shifts (in ppm) and Coupling Constants (in Hz, in parentheses) of Asernestioside C and Cycloastragenol in C_5D_5N

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	Asernestioside C (1)	Cycloastragenol (4)	Sugar part of 1			
1-Ηα	1.55 m	1.64 ddd	Xylose			
		(13, 12, 4.5)	1′-H	4.90 d		
1-Hβ	1.18 m	1.25 ddd		(6)		
		(13, 4.5, 3)	2'-H	4.28 dd		
2-Ηα	2.23 m	2.055 dtd		(8.5, 6)		
		(12, 4.5, 3)	3'-H	4.24 t		
2-Ηβ	1.91 m	1.96 tdd		(8.5)		
<i>-</i> -		(12, 11.5, 4.5)	4'-H	5.25 td		
3-H	3.48 dd	3.66 dd		(8.5, 5)		
	(11.5, 4.5)	(11.5, 4.5)	5'-H ₂	3.53 dd		
5-H	1.65 d	1.72 d	2	(11.5, 8.5)		
J 11	(9)	(9.5)		4.31 dd		
6-H	3.72 td	3.80 td		(11.5, 5)		
0-11	(9, 4)	(9.5, 4)		(11.5, 5)		
7-Ηα	1.64	1.65 td	Rhamnose			
/-110c	1.07	(12, 9.5)	1''-H	6.32 d		
7-Hβ	1.76 dt	1.83 dt	. 11	(1)		
, 11ρ	(12, 4)	(12, 4)	2′′-H	4.74 dd		
8-H	1.89 dd	1.96 dd	<u>د</u> -11	(3.5, 1)		
0-11	(12, 4)	(12, 4)	3′′-H	(3.3, 1) 4.61 dd		
11-Ηα	1.88 m	1.98 dt	J -11			
11-Πα	1.00 III	(10, 3.5)	4′′-H	(9.5, 3.5) 4.28 t		
11-Hβ	1.16 m	1.23 td	T -11	(9.5)		
	1.10 111	(10, 4.5)	5′′-H	(9.3) 4.72 dq		
12-H ₂	1.59 m	1.63 m	J -Π	(9.5, 6)		
	1.57 111	1.67 m	6′′-H ₃	1.68 d		
15-H ₂	1.61 m	1.78 dd	0 -113	(6)		
112	I.OI III	(12.5, 7.5)		(0)		
	1.99 dd	2.13 dd	Glucose			
	(12.5, 8)	(12.5, 7.5)	1'''-H	5.01 d		
16-H	4.89 td	5.02 qd	1 -11	(8)		
10-11	(8, 6)	(7.5, 2)	2′′′-H	3.96 t		
17-H	2.43 d	2.55 d	2 -11	(8)		
	(8)	(7.5)	3′′′-H	ca. 4.14		
18-H ₃	1.32 s	1.45 s	J -11	cu. 7.17		
10-113	1.52 8	1.700	4′′′-H	ca. 4.12		
19-H ₂	0.26 d	0.35 d	7 -11	cu. 7.12		
17-112	(4)		5′′′-H	3.83 ddd		
	0.56 d	(4) 0.62 d	J -N			
	0.36 d (4)		6′′′-H ₂	(8.5, 5, 2.5 4.24 dd		
21-H ₃	1.27 s	(4) 1.33 s	0 -n ₂	4.24 dd (11.5, 5)		
21-113	1.2/3	1.22 8		(11.3, 3) 4.37 dd		
22-H ₂	1 50	1 60 444				
∠∠- ⊓ 2	1.58 m	1.69 ddd		(11.5, 2.5)		
	2 70 +4	(11.5, 9, 2.5)	CH CO	1.01 -		
	2.79 td	3.12 td	CĤ³CO	1.91 s		
22 LJ	(11.5, 8)	(11.5, 9)				
23-H ₂	1.93 m	2.06 dq				
	2 20	(11.5, 9)				
	2.30 m	2.32 tdd				
24-H	2 00 11	(11.5, 5.5, 2.5)				
	3.88 dd	3,89 dd				
26 11	(8.5, 7)	(9, 5.5)				
26-H ₃	1.40 ^{a)} s	1.30^{a} s				
27-H ₃	1.64 ^{a)} s	1.58 ^{a)} s				
28-H ₃	0.94 s	1.03 s				
$29-H_3$	1.87 s 1.41 ^{b)} s	1.89 s 1.37 ^{b)} s				
$30-H_{3}$						

a) The signal at the high-field side was assigned arbitrarily to 26-H₃ and the other to 27-H₃. b) Assignment was done based on NOE experiments with 19-H (Fig. 3).

(each d, J=4 Hz, 19-H₂), seven tertiary methyls at δ 0.94, 1.27, 1.32, 1.40, 1.41, 1.64, and 1.87 (28-, 21-, 18-, 26-, 30-, 27-, and 29-methyl, respectively), and an acetyl methyl at δ 1.91 along with signals due to oxymethine and/or oxymethylene protons at around δ 3.4—6.4 ppm (Table I).

The ¹H-NMR signals of 1 could be analyzed reasonably

by the use of ${}^{1}H^{-1}H$ shift correlation spectroscopy (${}^{1}H^{-1}H$ COSY) coupled with ${}^{1}H^{-13}C$ COSY, which indicated the presence of the partial structures shown in Fig. 1. Also it was suggested from the ${}^{1}H^{-}$ and ${}^{13}C^{-}NMR$ data that the sugar moieties in 1 may be β -xylopyranose, α -rhamnopyranose, and β -glucopyranose (Tables I and II) and that the double triplet at δ 5.25 (J=8.5, 5 Hz) due to a proton geminal to the ester group is ascribed to the 4'-proton of xylopyranose moiety. Therefore, the acetyl group should be linked to the 4'-position of the xylose residue.

Next, we measured the long-range ¹H-¹³C COSY in order to clarify the connectivities of these partial structures. As shown in Fig. 2, the methyl signals at δ 0.94 ppm (28-H₃) showed long-range correlations with the carbons at δ 45.1 (s, C-13), 46.0 (s, C-14 and t, C-15), and 46.7 ppm (d, C-8) and the methyl signal at δ 1.32 ppm (18-H₃) showed longrange correlations with the carbons at δ 33.4 (t, C-12), 45.1 (s, C-13), 46.0 (s, C-14), and 58.1 ppm (d, C-17), indicating that C-12, C-14, C-17, and C-18 were connected with C-13, and C-8, C-13, C-15, and C-28 were connected with C-14. Also, the methyl signal at δ 1.27 ppm (21-H₃) showed longrange correlations with the carbons at δ 35.0 (t, C-22), 58.1 (d, C-17), and 87.2 ppm (s, C-20), indicating that C-17, C-21, and C-22 were connected with C-20. On the other hand, the methyl signal at $\delta 1.40$ (26-H₃) and 1.64 ppm (27-H₃) showed correlations with the carbons at $\delta 25.6$ (q, C-27), 78.5 (s, C-25), and 82.0 ppm (s, C-24) and at δ 23.0 (q, C-26), 78.5 (s, C-25), and 82.0 ppm (d, C-24), respectively, suggesting that C-26 and C-27 were connected with C-25, and C-25 with C-24. Furthermore, the methyl signals at δ 1.87 (29-H₃) and 1.41 ppm (30-H₃) showed correlations with the carbons at δ 28.6 (q, C-29), 42.6 (s, C-4), and 54.0 ppm (d, C-5) and at δ 16.6 (q, C-30), 42.6 (s, C-4), and 54.0 ppm (d, C-5), respectively, indicating the connectivities of C-4 and C-29, C-30, and C-5. These data and the other long-range correlations shown in Table II led us to suppose that 1 is a triglycoside of cycloastragenol (4). This was supported by the close similarity of its ¹³C-NMR spectrum with that of 4 (Table II).¹⁰⁾

The glycosidation sites of 1 were suggested by the ¹³C-NMR spectrum compared with that of 4. As shown in Table II, the signals of the oxygenated carbons C-3 and C-25 appeared at lower field and the carbons C-26 and C-27 at higher field than the corresponding signals of the aglycone (4), while the oxymethine carbons C-6 and C-16 appeared at almost the same chemical shifts. Therefore, the sugar moieties must be located at the C-3 and C-25 positions. On the other hand, the anomeric carbon of the β -glucopyranose residue resonated at higher field as compared with that of methyl β -glucopyranoside.¹¹⁾ It has been reported that the anomeric carbon signal of tertiary alcoholic β glucosides appears at significantly higher field (δ about 99) than those of primary (δ about 104) and secondary (δ about 102) alcoholic β -glucosides. 12) Thus, the glucopyranose must be attached to the C-25 position of the aglycone. As to the xylose and rhamnose moieties, the carbon C-2' of xylopyranose showed the glycosidation shift (3 ppm), while none of the carbons of rhamnopyranose showed a glycosidation shift. Therefore, it followed that xylopyranose was attached to the 3-hydroxyl group of the aglycone and rhamnopyranose to the 2'-hydroxyl group of xylopyranose.

In order to confirm these conclusions, nuclear

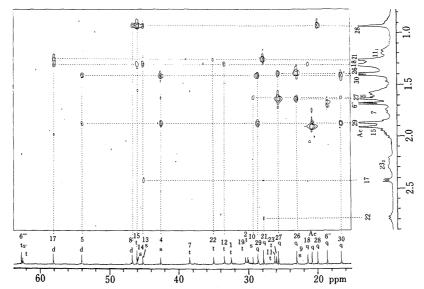


Fig. 2. Long-Range ¹H-¹³C Shift-Correlated Spectrum of Asernestioside C (1) in the Upfield Region (J=10 Hz; Sample, 15 mg; 35 h Run)

Table II. 13C Chemical Shifts (in ppm) and Long-Range-Coupled 1H Signals of Asernestioside C and Cycloastragenol in C5D5Na)

	δ	Asernestioside C (1) ¹ H L-r. coupled (${}^{3}J_{C-H}$; ${}^{2}J_{C-H}$)		δ	Cycloastragenol (4) ¹ H L-r. coupled ($^{3}J_{C-H}$; $^{2}J_{C-H}$)		Sugar part of 1		Methyl glycoside ^{b)}		
									α	β	
C-1	32.4 t			32.8 t	19		Xylose				
C-2	30.3 t			31.4 t			C-1′	105.4	d	101.5	106.0
C-3	88.4 d			78.3 d	29, 30		C-2′	77.6	d	73.7	74.0
C-4	42.6 s		5, 29, 30	42.4 s		5, 29, 30	C-3′	74.6	d	75.5	78.
C-5	54.0 d	29, 30		53.9 d	1, 29, 30		C-4'	73.3	d	71.4	70.
C-6	67.7 d	,		68.3 d		5	C-5′	62.5	t	63.1	67.
C-7	38.4 t			38.8 t		8					
C-8	46.7 d	28		47.2 d	15, 19, 28	7	Rhamnose				
C-9	20.8 s			20.9 s	7	8	C-1′′	101.9	d	102.6	102.
C-10	29.9 s		2	29.9 s	8	1, 5	C-2''	72.3	d	72.1	72.
						,	C-3′′	72.5	d	72.7	75.
C-11	26.2 t			26.3 t			C-4''	74.0	d	73.8	73.
C-12	33.4 t	18		33.4 t	17, 18		C-5''	69.9	d	69.5	73.
C-13	45.1° s	28	17, 18	45.0° s	15, 28	17, 18	C-6′′	18.6	a	18.6	18.
C-14	46.0° s	18	28	46.2° s	18	8, 28			•		
C-15	46.0 t	28		46.7 t	28	-,	Glucose				
C-16	73.5 d	20		73.4 d			C-1'''	98.8	d	101.3	105.
C-17	58.1 d	15, 18, 21		58.4 d	15, 18, 21		C-2′′′	75.1		73.7	74.
C-18	21.4 q	.0, .0,		21.6 g			C-3′′′	78.4		75.3	78.
C-19	30.4 t			31.0 t	5		C-4'''	71.3	d	72.0	71.
C-20	87.2 s		17, 21	87.2 s	· ·	17, 21, 22	C-5'''	77.9		74.0	78.
C-20	07.2 3		17, 21	07.2 3		11, 21, 22	C-6'''	62.7		62.7	62.
C-21	27.8 q	22		28.6 q	17, 22						
C-22	35.0 t	21		34.9 t	17, 21		Acetyl grou	ın			
C-23	25.9 t	21		26.4 t	,		CH3CO	20.8	а		
C-24	82.0 d	26, 27		81.7 d	26, 27		211300		7		
C-25	78.5 s	20, 27	26, 27	71.2 s	20, 27	26, 27	CH ₃ CO	170.5	s		
C-26	$\frac{76.5}{23.0^{d}}$ q	27	20, 27	27.2 ^d) q	27	,	3=0		-		
C-20 C-27	$\frac{25.0}{25.6^{d}}$ q	26		28.2^{d} q	26						
C-27 C-28	23.6° q 20.0 q	20		20.2 q							
C-28 C-29		30		29.4 q							
C-29 C-30	28.6 q 16.6 q	5, 29		16.1 q							

a) Carbon signals affected by glycosylation or acetylation are underlined. b) Taken from the literature (ref. 11). c) Assignments were done by comparisons with those of 24-methylenecycloartenol (ref. 13). d) These signals correspond to the 26- and 27-H₃ signals, respectively.

Overhauser effect (NOE) difference spectra were measured. Irradiation of both the 26-methyl and 27-methyl protons caused an increase of the intensity of 1'''-H of the glucopyranose residue, supporting the location of glucopyranose at the C-25 position (Fig. 3). It should be noted here that

irradiation of the 26-methyl protons concomitantly caused a decrease of the intensity of 24-H. Also, it was observed that irradiation of 3-H caused a decrease of the intensity of 1'-H and vice versa (Fig. 3). These phenomena may be ascribed to the negative NOE as reported in the case of

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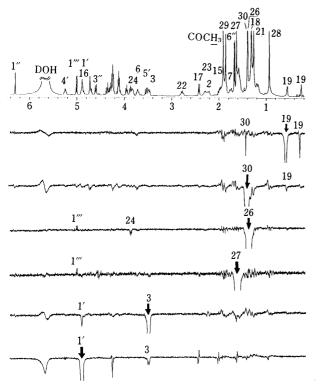


Fig. 3. NOE Difference Spectra of Asernestioside C (1) in C_5D_5N at $25\,^{\circ}C$

some peptides¹⁴⁾ and a flavonoid glucoside.¹⁵⁾

Based on the foregoing evidence, and by assuming that xylose, rhamnose, and glucose residue have the D, L, and D absolute configuration, respectively, the structure of asernestioside C (1) was concluded to be 25-O- β -D-glucopyranosyl cycloastragenol 3-O- $[\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)]$ -(4'-O-acetyl)- β -D-xylopyranoside.

Then we measured the NOE difference spectra of some other saponins, such as astragalosides I (5), II (6), III (7), IV (8), V (9), and VI (10), Io saikosaponins a (11) and c (12), and ginsenoside R_{b1} (13), to examine whether negative NOE is common in saponins or not. As shown in Fig. 4 and Table III, negative NOE was observed between the proton at the C-3 position and the anomeric proton of the 3-O-glucoside residue in all the compounds examined. On irradiation of the 21-methyl protons of 13, negative NOE was also observed at the anomeric proton of the 20-O-glucoside residue, while positive NOE was observed on irradiation of 6-H of 5, 6, and 8. Our present result is the first example of negative NOE in the saponin field¹⁷⁾ and it will be useful for the structure elucidation of saponins.

Experimental

The melting point was measured on a micromelting point apparatus and is uncorrected. $^1\text{H-}$ and $^{13}\text{C-NMR}$ spectra were taken on a JEOL JNM-GX400 spectrometer in $\text{C}_5\text{D}_5\text{N}$ with tetramethylsilane as an internal

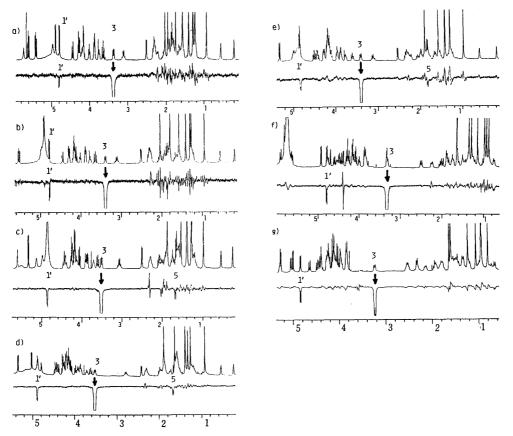


Fig. 4. NOE Difference Spectra of Some Saponins in C_5D_5N at 25 °C

Normal (Upper) and Difference (Lower) Spectra of a) Astragaloside I (5), b) Astragaloside II (6), c) Astragaloside III (7), d) Astragaloside V (9), e) Astragaloside VI (10), f) Saikosaponin c (12), and g) Ginsenoside R_{b1} (13)

TABLE III. NOE Observed in NOE Difference Spectra of Some Saponins^{a)}

Proton(s) irradiated	Proton(s) which showed NOE ^{b)}								
	5	6	7	8	9	10	11	12	13
3	1' (-)	1' (-)	5 (-) 1' (-)	1' (-)	5 (-) 1'·(-)	5 (-) 1'(-)	c)	1' (-)	1′ (-
6 18	1"(+)	1"(+)	,	1''(+)	` ,	, ,			13 (-
19									19 (- 18 (- 29 (-
21 28									1''' (- 3 (- 5 (- 29 (-
29 30									1' (- 19 (- 17 (-
1′	3' (-)	3 (-) 3' (-)	3 (-) 3' (+) 2'' (+)	3 (-)	3 (-)	3 (-)	3 (-) 5" (-)	3 (-) 3' (-) 5' (-)	3 (- 3' (- 5' (-
1''	3''(+)	2" (+) 3" (+) 6" (+)	2' (+)	3′′ (+)	d)	<i>d</i>)	3′′ (−)	2' (-) 3' (-) 4' (-)	2' (-
1′′′					d)			2''(-)	17 (- 3''' (-

a) Measured in C_5D_5N at 25 °C. b) (+) and (-) indicate positive and negative NOE, respectively. c) NOE experiment was not done because of the overlapping of the 3-H and 3"-H signals. d) Significant NOE was not observed.

standard, and chemical shifts are recorded in δ values. Multiplicities of $^{13}\text{C-NMR}$ signals were determined by the distortionless enhancement by polarization transfer (DEPT) method. The 2-D NMR spectra were

measured by the use of the JEOL standard pulse sequences ($^{1}H^{-1}H$ COSY, VCOSYN, 45° mixing pulse; $^{1}H^{-13}C$ COSY, VBDCHSHF, J=140 Hz; long-range $^{1}H^{-13}C$ COSY, VCHSHF, J=10 Hz) and the collected data

were processed with the standard JEOL software. NOE difference spectra were also measured by the use of the JEOL standard pulse sequence (DIFNOE2) with 5 s irradiation.

Extraction and Separation Dried roots (8 kg) of Astragalus ernestii COMB. were cut into small pieces and percolated with 95% EtOH. The extract was concentrated in vacuo to a syrup. Ten volumes of MeOH was added to this syrup and the insoluble material was filtered off. The filtrate was concentrated in vacuo and the residue was dissolved again in ten volumes of water and extracted with BuOH. The BuOH extract was concentrated in vacuo to give a syrup (300 g), which was then subjected to silica gel (2.4 kg) column chromatography with CHCl₃-MeOH-H₂O (7:3:1, lower layer) to give a mixture of crude saponins (98g). This mixture was subjected to reversed-phase column chromatography (Waters, Bondapak C_{18} , 500 g) and eluted with H_2O , H_2O -MeOH (9:1, 8:2, 6:4, 4:6, and 2:8), and MeOH. The fractions eluted with H_2O- MeOH (6:4 and 4:6) were combined and further separated by repeated silica gel column chromatography. Elution with CHCl₃-MeOH-H₂O (7:3:1, lower layer) gave asernestioside C (1, 45 mg) along with asernestiosides A (2, 408 mg)9) and B (3, 1.5 g).9)

Asernestioside C (1) Colorless fine crystals (from MeOH), mp 204—207 °C, $[\alpha]_D$ –13.22° (c=0.32, MeOH). IR ν_{max}^{KBr} cm⁻¹: 3375 (OH), 1732 (CO), 1623, 1369, 1241, 1039. H- and 13 C-NMR: see Tables I and II. FAB-MS m/z: 973 $[M+H]^+$.

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