CHEMICAL CONVERSION OF GARDNERINE INTO KOUMIDINE BY INVERTING THE ETHYLIDENE SIDE CHAIN WITH PALLADIUM CATALYST

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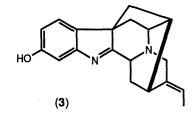
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A minor Gelsemium alkaloid, koumidine (1), was synthesized from gardnerine (2) by the demethoxylating the indole ring and the inverting the configuration of the ethylidene side chain with palladium catalyst.

KEYWORDS Gelsemium alkaloid; koumidine; 19-(E)-koumidine; gardnerine; chemical conversion; olefin inversion; palladium catalyst

Koumidine (1), a minor constituent of $Gelsemium\ elegans$ Benth., was first isolated by a Chinese group¹⁾ and the structure, which had an unusual 19-(Z) ethylidene side chain compared with that of the conventional sarpagine class of indole alkaloids, was revised by us^2) and the Cordell group.³⁾ Koumidine (1) is considered to be one of the important biosynthetic intermediates of some Gelsemium alkaloids, such as 19-(Z)-taberpsychine, humantenine, koumine, gelsemine, etc.⁴⁾ During course of studies on the biomimetic synthesis of Gelsemium alkaloids,^{5, 6)} we needed a large quantity of koumidine (1). Recently we established the synthetic route of koumidine (1) from ajmaline in 15 steps.⁷⁾ However, a more concise way to get (1) was highly required. Therefore, we have developed a synthesis of koumidine (1) from gardnerine (2), which was readily obtainable from $Gardneria\ nutans$.⁸⁾ The transformation involves two structural changes of the starting material (2): 1) removing the methoxy group from the aromatic ring and 2) inverting the 19-(E) ethylidene side chain to 19-(Z) form. Here we report an efficient alternative synthesis of koumidine (1) that features the inversion of the ethylidene side chain assisted by palladium catalyst.

The first attempt to cleave the methyl ether on the indole ring of gardnerine (2) using sodium thioethoxide in dimethyl formamide (DMF) gave the abnormal indolenine derivative (3). Demethylation of the aryl methyl ether was cleanly achieved by applying Fujita's procedure⁹) to the Na-tosyl derivative (4), which was prepared in 98% yield from 17-O-acetylgardnerine by treatment with p-toluenesulfonyl chloride in an aqueous 50% KOH and benzene (1:1) solution in the presence of tetra-n-butylammonium hydrogensulfate. The reaction of (4) with aluminum chloride (4 eq.) in ethanethiol and dry CH2Cl2 at -18°C for 3 h afforded the desired phenolic compound (5) [mp 248-255°C (dec.)] in 91% yield. Next we used the palladium catalyzed triethylammonium formate reduction of aryl triflate to deoxygenate phenol. 10) Triflate (6), prepared in 97% yield from (5) with trifluoromethanesulfonic anhydride and triethylamine in dry CH₂Cl₂, was treated with 0.2 eq. of palladium acetate, 0.4 eq. of 1,1'-bis(diphenylphosphino)ferrocene (DPPF), triethylamine, and formic acid in DMF at 60°C for 2 h to produce deoxygenated compound (7) in 98% yield. Reductive deprotection of Na-tosyl and 17-O-acetyl group in (7) with lithium aluminum hydride in THF (reflux, 6 h) gave rise to 19-(E)-koumidine (8), mp 170-172°C, in 95% yield. The structure of (8) was confirmed by the comparison of the spectroscopic data (¹H and ¹³C-NMR, IR, and MS) with those of koumidine (1). On the other hand, by treating (7) at room temperature for 50 h with a large excess of magnesium (turnings) in dry methanol in the presence of a 0.1 eq. of palladium chloride and 0.2 eq. of triphenylphosphine, deprotection and the inversion of the configuration of the double bond proceeded simultaneously to afford koumidine (1) as the major product (48% yield) along with the isomer (8) (34% yield) and 19,20-dihydrokoumidine (7% yield). Semisynthetic koumidine (1), mp 200-201°C, had spectral properties (¹H-NMR, IR, UV, and MS) in accord with those of a natural sample. Palladium catalyst, phosphine ligand, magnesium, and methanol altogether is essential for the olefin inversion reaction. The plausible mechanism of this reaction would involve the metal hydride 1, 2 addition/β-hydride elimination process. 11) The optimization of the reaction condition of palladium-catalyzed olefin inversion and its application to the indole alkaloid synthesis are in progress in this laboratory.



Reagents and conditions

i, Ac₂O, pyridine, r.t., 8 h, 97%; TsCl, n-Bu₄NHSO₄, 50%KOH-benzene, r.t., 3 h, 98%.

ii, AlCl₃, EtSH, CH₂Cl₂, -18°C, 3 h, 91%.

iii, (CF₃SO₂)₂O, Et₃N, CH₂Cl₂, -20°C, 10 min, 97%.

iv, Pd(OAc)₂, DPPF, Et₃N, HCOOH, DMF, 60°C, 2 h, 98%.

v, Mg, PdCl₂, PPh₃, MeOH, r.t., 50 h, 48% (1), 34% (8).

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