5-Alkoxyimidazoquinolones as Potential Antibacterial Agents. Synthesis and Structure–Activity Relationships¹⁾

Masahiro Fujita,* Hiroshi Egawa, Teruyuki Miyamoto, Junji Nakano, and Jun-ichi Matsumoto

Exploratory Research Laboratories, Dainippon Pharmaceutical Company, Ltd., Enoki 33–94, Suita, Osaka 564, Japan. Received November 22, 1995; accepted January 22, 1996

4-Substituted 6-cyclopropyl-6,9-dihydro-5-methoxy-9-oxo-1*H*-imidazo[4,5-*f*]quinoline-8-carboxylic acids (6) and 8-substituted 1,5,6,11-tetrahydro-5-methyl-1-oxo-imidazo[4,5-*g*]pyrido[1,2,3-*de*][1,4]benzoxazine-2-carboxylic acids (7) were prepared as potential antibacterial quinolone derivatives. The appendages at C-4 of -6 and at C-8 of -7 were selected from 1-piperazinyl, 4-methylpiperazinyl, 3-aminopyrrolidinyl, and 3-aminomethylpyrrolidinyl groups. The 5-methoxyimidazoquinolones 6 were superior to the corresponding ofloxacin type analogues 7 in *in vitro* antibacterial activity. The activity of 6 was equipotent against *S. aureus*, but 2 to 16 times less potent against *E. coli* and *P. aeruginosa* compared to that of the 5-fluoro analogue 3.

Key words quinolone; 1*H*-imidazo[4,5-*f*]quinoline; synthesis; imidazo[4,5-*g*]pyrido[1,2,3-*de*][1,4]benzoxazine; antibacterial activity; structure–activity relationship

Synthetic antibacterial quinolones are widely accepted as useful and indispensable antibacterial agents for the current chemotherapy against bacterial infectious disease, owing to their potent activity and broad spectrum.²⁾ As represented by enoxacin (1)³⁾ and sparfloxacin (2)⁴⁾ (Chart 1), most of the prevailing quinolones possess a fluorine atom at C-6 and a cyclic amino group at C-7 of the 1-substituted 4-oxoquinoline- and 4-oxo-1,8-naphthyridine-3-carboxylic acids.

As part of our study of tricyclic quinolone derivatives without a fluorine atom at C-6 of the quinoline ring, we previously reported the synthesis and antibacterial activity of 4,5-disubstituted 6-cyclopropyl-6,9-dihydro-9-oxo-1H-imidazo[4,5-f]quinoline-8-carboxylic acids, including 3 \mathbf{b} and 3 \mathbf{c} . The quinolones 3 \mathbf{b} , \mathbf{c} exhibited a comparable in vitro antibacterial activity to enoxacin (1)³⁾ and oflox-

acin (5),6 whereas 3b and 3c were inefficacious in in vivo when orally administered in infected mice, probably owing to their limited absorption from animal digestive tracts. Among the quinolones currently utilized and under clinical trials, AM-1155 (4)7) and offoxacin (5),6) which append an alkoxy group or its equivalent at C-8 of their quinolone rings, have been reported to have sufficient bioavailability.^{8,9)} This prompted us to undertake the present study of 4-substituted 6-cyclopropyl-6,9-dihydro-5-methoxy-9-oxo-1*H*-imidazo[4,5-*f*]quinoline-8-carboxylic acids (6) and 8-substituted 1,5,6,11-tetrahydro-5methyl-1-oxo-imidazo[4,5-g]pyrido[1,2,3-de][1,4]benzoxazine-2-carboxylic acids (7). A change of the C-5 fluorine of 3b, c to a methoxy group (giving 6) or replacement of the C-5 and C-6 substituents with a 3-methyloxazine ring (giving 7) was expected to improve both the

* To whom correspondence should be addressed.

© 1996 Pharmaceutical Society of Japan

988 Vol. 44, No. 5

Reagents: a (NH₄)₂CO₃; b 1) H₂, Pd-C, 2) HC(OEt)₃ HCOOH; c R-H.

For 6 and 7:
$$a$$
, $R = HN N -; b$, $R = MeN N -; c$, $R = \frac{H_2N}{N} N -; d$, $R = \frac{H_2N}{N} N -; d$

Chart 2

Table 1. Physical Data for the Substituted Imidazoquinolones

Compd.	mp (°C) (Recryst. solvent)	Yield (%)		Analysis (%)							
			Formula	Calcd			Found				
				С	Н	Cl	N	C	Н	Cl	N
6a	243—246 (aq. HCl)	10	C ₁₉ H ₂₁ N ₅ O ₄ ·2HCl·3/4H ₂ O	48.57	5.26	15.09	14.91	48.74	5.10	15.38	14.89
6b	283—286 (dec.) (aq. HCl)	38	$C_{20}H_{23}N_5O_4 \cdot 2HCl \cdot 1/2H_2O$	50.11	5.47	14.79	14.61	49.91	5.56	15.03	14.55
6c	236-238 (dec.) (aq. HCl-EtOH)	41	$C_{19}H_{21}N_5O_4 \cdot 2HCl \cdot H_2O$	48.11	5.31	14.95	14.76	48.05	5.34	14.84	14.49
6d	193—195 (aq. HCl)	11	$C_{20}H_{23}N_5O_4 \cdot 2HCl \cdot 9/4H_2O$	47.02	5.82	13.88	13.71	47.25	5.67	13.47	13.67
7a	290-294 (dec.) (aq. HCl-EtOH)	10	$C_{18}H_{19}N_5O_4 \cdot 2HCl \cdot 2H_2O$	45.20	5.27	14.82	14.64	45.26	5.22	14.55	14.56
7b	286-290 (dec.) (aq. HCl-EtOH)	43	$C_{19}H_{21}N_5O_4 \cdot 2HCl \cdot 2H_2O$	46.35	5.53	14.40	14.22	46.29	5.23	14.39	13.99
7c	275—278 (dec.) (aq. NH ₄ OH)	25	$C_{18}H_{19}N_5O_4$	58.53	5.18		18.96	58.53	5.24		18.94

in vitro and in vivo activities of 3b, c, as would be reflected in the enhancement of their bioavailability. This paper describes the synthesis and the antibacterial activity of 6 and 7.

Chemistry In the previous paper,⁵⁾ we dealt with two methods for construction of the imidazole ring fused to quinolones. For easy synthesis of the target compounds 6 and 7, we planned to develop another type of imidazole ring formation (Chart 2).

Compounds 6 and 7 were assumed to us to be favorably derived from nitro quinolones 8^{10} and 9^{11} with appropriate functional groups, respectively. Both of the C-6 and C-7 positions of 8 could react with a nucleophile, owing to the electron-withdrawing properties of the C-5 nitro and C-4 oxo groups. The electron-donating effect of the C-8 alkoxy groups of 8 might decrease the reactivity at C-7 of 8. Hence nucleophilic substitution of 8 would proceed predominantly at its C-6 position. For a similar reason, nucleophilic substitution of 9 would proceed predominantly at C-9. The reaction of 8 and 9 with ammonium carbonate in N,N-dimethylformamide (DMF) actually occurred at C-6 and C-9, and gave 6-amino-5-nitroquinolones 10 and 11, respectively, in high yields.

The nitro groups at C-5 of 10 and at C-8 of 11 were hydrogenated to amino groups, and the resulting intermediate diamines, without isolation, were subsequently treated with triethyl orthoformate and formic acid at

80 °C for 1 h to give the imidazole-fused compounds 12 and 13 in 49 and 67% yields, respectively. Compounds 12 and 13 were allowed to react with a cyclic amine such as piperazine, N-methylpiperazine, 3-aminopyrrolidine, or 3-aminomethylpyrrolidine in hot dimethyl sulfoxide (DMSO), thus giving the corresponding displacement products 6a—d and 7a—c in 10% to 43% yields after purification.

Antibacterial Activity The *in vitro* antibacterial activity of compounds **6a—d** and **7a—c** was tested against gram-positive (*Staphylococcus aureus* 209P JC-1) and gram-negative bacteria (*Escherichia coli* NIHJ JC-2 and *Pseudomonas aeruginosa* 12). The results are summarized in Table 2, in which the data for enoxacin (1), ofloxacin (5) and **3a—d** are included for comparison.

Among the 5-methoxy compounds **6a**—**d**, the activity against gram-positive bacterium (*S. aureus*) was arranged in the decreasing order of 3-aminomethylpyrrolidinyl (**6d**) > 3-aminopyrrolidinyl (**6c**) > piperazinyl (**6a**) > 4-methylpiperazinyl (**6b**) derivatives. The tendency observed in the activity of **6a**—**d** against *S. aureus* holds true for the ofloxacin type compounds **7a**—**c** and 5-fluoro compounds **3a**—**d**. The piperazinyl derivative **6a** among **6a**—**d** was the most active against gram-negative bacteria (*E. coli* and *P. aeruginosa*). The activity of the 3-aminomethylpyrrolidinyl derivative **6d** against gram-negative bacteria was noticeably weak.

Table 2. In Vitro Antibacterial Activity of the Imidazoquinolones and Related Compounds

			Minimum inhibitory conc., ^{a)} (μg/ml)				
Compound		R	S. aureus 209P JC-1	E. coli NIHJ JC-2	P. aeruginosa 12		
6a	₩H Ö	HN_N -	0.1	0.1	0.78		
6b	й Соон	MeN_N -	0.2	0.1	3.13		
6c	R → N I	H ₂ N \ \ \ \ N -	0.05	0.2	1.56		
6d		H₂N ÛN -	0.025	1.56	12.5		
7a	NH O COOH	HN N -	0.78	25	>100		
7b	RITIN	MeN_N -	0.78	0.78	3.13		
7 c	Me	H ₂ N \ \ \ \ \ \ N -	0.2	1.56	12.5		
3a	√и́н о́	HN_N -	0.2	0.05	0.39		
3 b	и соон	MeN_N -	0.2	0.05	0.2		
3c	R N I	H ₂ N \ \ \ \ \ \ N -	0.05	0.025	0.2		
3d	. Ф	$H_2N \sim N$	0.025	0.1	0.78		
1 2	Enoxacin Ofloxacin		0.39 0.2	0.1 0.05	0.78 0.78		

a) See Experimental.

In comparison of the 5-methoxy compounds $6\mathbf{a} - \mathbf{c}$ with the corresponding ofloxacin type compounds $7\mathbf{a} - \mathbf{c}$ (i.e., $6\mathbf{a}$ vs. $7\mathbf{a}$, $6\mathbf{b}$ vs. $7\mathbf{b}$, and $6\mathbf{c}$ vs. $7\mathbf{c}$), the former, $6\mathbf{a} - \mathbf{c}$, were 4 times or more active than the latter, $7\mathbf{a} - \mathbf{c}$, with the exception of $6\mathbf{b}$ vs. $7\mathbf{b}$, which were equipotent against *P. aeruginosa*. The replacement of the C-5 fluorine of $3\mathbf{a} - \mathbf{d}$ with a methoxy group (giving $6\mathbf{a} - \mathbf{d}$) caused no change in activity against gram-positive bacterium but diminished the activity against gram-negative bacteria in a range from one-half (the piperazinyl derivative $6\mathbf{a}$ vs. $3\mathbf{a}$) to one-sixteenth (the 3-aminomethylpyrrolidinyl derivative $6\mathbf{d}$ vs. $3\mathbf{d}$).

Among the compounds prepared in the present study, **6a** showed the most potent and well-balanced activity toward the gram-positive and gram-negative bacteria tested. The antibacterial activity of **6a** against *S. aureus* was superior to that of enoxacin (1) and ofloxacin (5). Against gram-negative bacteria, the activity of **6a** was almost comparable to that of the reference quinolones 1 and 5. Compound **6a**, however, was not better than **3b**, **c** in *in vitro* activity, contrary to our expectation.

The 5-methoxy compounds $6\mathbf{a}$ — \mathbf{c} were selected for testing their efficacy on systemic infection due to P. aeruginosa 12 in mice. It follows from the data listed in Table 3 that $6\mathbf{a}$ — \mathbf{c} are less efficacious than $3\mathbf{a}$, \mathbf{b} , enoxacin (1), and ofloxacin (5). A remarkable decrease in oral efficacy (ED₅₀, p.o.) of $6\mathbf{a}$ — \mathbf{c} was observed as compared to their intravenous efficacy (ED₅₀, i.v.).

In summary, we developed an easy method for the construction of the imidazo[4,5-f]quinolone ring and synthesized 4-substituted 6-cyclopropyl-6,9-dihydro-5-

Table 3. In Vivo Efficacy of Selected and Reference Compounds on Systemic Infections

			P. aeruginosa 12				
Compd.	X	R	MIC ^{a)}	$\begin{array}{c} \mathrm{ED}_{50} \\ (p.o.)^{b)} \end{array}$	ED ₅₀ (i.v.) ^{b)}		
6a	OMe	HN_N -	0.78	> 25	4.04		
6b	OMe	MeN_N -	3.13	34.7	6.25		
6c	OMe	H_2N N -	1.56	>100	3.13		
3b	F	MeN_N -	0.2	8.42	0.982		
3c	F	H ₂ N \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ -	0.2	36.0	0.443		
1 5	Enoxacin Ofloxacin	.~	0.78 0.78	8.41 6.62	2.82		

a) Minimum inhibitory concentration (μ g/ml). b) Shown in milligrams per kilogram. See Experimental.

methoxy-9-oxo-1*H*-imidazo[4,5-*f*]quinoline-8-carboxylic acids (**6a**—**d**). Compounds **6a**—**d** were found to have considerable potency in terms of antibacterial activity; in particular, the 4-piperazinyl compound **6a** was more active against *S. aureus* than enoxacin (1) and ofloxacin

990 Vol. 44, No. 5

(5). The activity of **6a—d** against gram-negative bacteria, however, was less than that of the corresponding 5-fluoro compounds **3a—d** which had been reported in the previous paper. ⁵⁾ The oral protective efficacy of **6a—c** did not reflect their *in vitro* activity.

Experimental

Chemistry All melting points were determined on a Yanagimoto micromelting point apparatus and are uncorrected. IR spectra were recorded on a Jasco A-102 or Perkin Elmer 1600 Series FTIR spectro-photometer. $^1\text{H-NMR}$ spectra were taken at 80 MHz on a Varian FT-80A spectrometer or at 200 MHz on a Varian Gemini-200 spectrometer. Chemical shifts are expressed in ppm (δ) with tetramethylsilane as an internal standard. Mass spectra were obtained on a JEOL JMS D-300 mass spectrometer for electron impact-mass spectrum (EI-MS), Hitachi M-80B mass spectrometer for secondary ion mass spectrum (SI-MS), or Hitachi M-1000 LC API mass spectrometer for atmospheric pressure chemical ionization mass spectrum (APCI-MS). The spectral data for all compounds were obtained and were consistent with the assigned structures. All solid compounds were analyzed for C, H, F, and N.

6-Amino-1-cyclopropyl-7-fluoro-1,4-dihydro-8-methoxy-5-nitro-4-oxoquinoline-3-carboxylic Acid (10) A mixture of 1-cyclopropyl-6,7-difluoro-1,4-dihydro-8-methoxy-5-nitro-4-oxoquinoline-3-carboxylic Acid **(8)**¹⁰⁾ (16.7 g, 48.8 mmol) and (NH₄)₂CO₃ (14.1 g, 147 mmol) in DMF (50 ml) was heated at 90 °C for 2 h. To the suspended mixture was added a mixture of ice and diluted HCl. The resulting solid was collected by filtration, washed successively with water, EtOH, and iso-Pr₂O, and then dried to give 16.2 g (98%) of **10**, mp 242—245 °C (dec., DMF–EtOH). IR (KBr) cm⁻¹: 3330, 3222, 1727, 1644. ¹H-NMR (200 MHz, DMSO- d_6) δ: 1.00—1.25 (4H, m, cyclopropyl CH₂CH₂), 4.04 (3H, d, J=2.0 Hz, Me), 4.10—4.25 (1H, m, cyclopropyl CH), 6.50 (2H, br s, NH₂), 8.63 (1H, s, 2-H), 14.40 (1H, br s, COOH). SI-MS m/z: 338 (M⁺ + 1), 320. *Anal*. Calcd for C₁₄H₁₂FN₃O₆: C, 49.86; H, 3.59; F, 5.63; N, 12.46. Found: C, 50.00; H, 3.53; F, 5.56; N, 12.34.

9-Amino-10-fluoro-2,3-dihydro-3-methyl-8-nitro-7-oxo-7*H*-pyrido-[1,2,3-*de*][1,4]benzoxazine-6-carboxylic Acid (11) According to the same method for the preparation of 10, 9,10-difluoro-2,3-dihydro-3-methyl-8-nitro-7-oxo-7*H*-pyrido[1,2,3-*de*][1,4]benzoxazine-6-carboxylic Acid (9)¹¹⁾ (7.50 g, 23.0 mmol) was treated and gave 6.76 g (91%) of 11, mp>300 °C (aqueous ammonia). IR (KBr)cm⁻¹: 3460, 1710, 1645, 1610. ¹H-NMR (80 MHz, NaOD-D₂O) δ: 1.50 (3H, d, J=6.5 Hz, Me), 4.3—4.9 (3H, m, OCH₂CHN), 8.30 (1H, s, 5-H). EI-MS m/z: 323 (M⁺), 279. *Anal.* Calcd for C₁₃H₁₀FN₃O₆: C, 48.31; H, 3.12; F, 5.88; N, 13.00. Found: C, 48.60; H, 3.18; F, 5.66; N, 13.18.

6-Cyclopropyl-4-fluoro-6,9-dihydro-5-methoxy-9-oxo-1*H***-imidazo[4,5-**f**]quinoline-8-carboxylic Acid (12)** A mixture of **10** (15.50 g, 46.0 mmol) and DMF (300 ml) was hydrogenated over 5% Pd–C (1.55 g) under atmospheric pressure at 50 °C for 8 h. The mixture was filtered to remove the catalyst. The filtrate was allowed to react with HC(OEt)₃ (150 ml) and HCOOH (7.5 ml) at 80 °C for 1 h. The mixture was concentrated *in vacuo* to leave a solid residue, which was triturated with EtOH. The resultant solid was collected by filtration, washed successively with EtOH and iso-Pr₂O, and then dried to give 6.91 g (49%) of **12**, mp 281—282 °C (dec., CHCl₃–EtOH). IR (KBr) cm⁻¹: 1728, 1614. ¹H-NMR (200 MHz, DMSO- d_6) δ: 1.05—1.30 (4H, m, cyclopropyl CH₂CH₂), 4.00 (3H, s, Me), 4.25—4.40 (1H, m, cyclopropyl CH), 8.42 (1H, s, 2-H), 8.82 (1H, s, 7-H), 13.38 (1H, br s, NH), 14.94 (1H, br s, COOH). SI-MS m/z: 318 (M⁺+1), 300, 273. *Anal.* Calcd for C₁₅H₁₂FN₃O₄: C, 56.78; H, 3.81; F, 5.99; N, 13.24. Found: C, 56.72; H, 3.63; F, 5.95; N, 13.21.

8-Fluoro-1,5,6,11-tetrahydro-5-methyl-1-oxo-imidazo[4,5-g]pyrido- [1,2,3-de][1,4]benzoxazine-2-carboxylic Acid (13) According to the same method for the preparation of **12, 11** (2.00 g, 6.19 mmol) was treated and gave 1.26 g (67%) of **13**, mp > 300 °C (aqueous ammonia). IR (KBr) cm⁻¹: 3362, 1712, 1645, 1622. 1 H-NMR (200 MHz, DMSO- 4 d) δ : 1.51 (3H, d, 2 H-7.0 Hz, Me), 4.40 (1H, dd, 2 H-2.0, 11.5 Hz, OCH), 4.63 (1H, dd, 2 H-1.5, 11.5 Hz, OCH), 5.07 (1H, br q, 2 H-7.0 Hz, 5-H), 8.40 (1H, s, 10-H), 9.13 (1H, s, 3-H), 13.4 (1H, br s, NH), 15.22 (1H, br s, COOH). EI-MS 2 M/z: 303 (M $^{+}$), 259. 2 Anal. Calcd for C 1 4H 1 0FN 3 O 4 C, 55.45; H, 3.32; F, 6.26; N, 13.86. Found: C, 55.10; H, 3.64; F, 6.17; N, 13.70.

4-Substituted 6-Cyclopropyl-6,9-dihydro-5-methoxy-9-oxo-1*H*-imidazo-

[4,5-f]quinoline-8-carboxylic Acids (6a—d) and 8-Substituted 1,5,6,11-Tetrahydro-5-methyl-1-oxo-imidazo[4,5-g]pyrido[1,2,3-de][1,4]benz-oxazine-2-carboxylic Acids (7a—c) A mixture of 12 (500 mg, 1.577 mmol) and piperazine (678 mg, 7.88 mmol) in DMSO (5.0 ml) was heated at 130—140 °C for 2 h. The solvent was distilled *in vacuo*. The residue was triturated with EtOH. The resultant solid was collected by filtration, washed successively with EtOH and iso-Pr₂O, and then dried to give 567 mg of a solid. Recrystallization from aqueous HCl gave 75 mg (10%) of 6a \cdot 2HCl \cdot 3/4H₂O. IR (KBr) cm⁻¹: 3360. 1702, 1617. 1 H-NMR (200 MHz, D₂O) δ : 0.95—1.1, 1.2—1.4 (both 2H, m, total cyclopropyl CH₂CH₂), 3.55—3.7, 3.9—4.1 (both 4H, m, total 2 \times HNCH₂CH₂N), 3.84 (3H, s, OMe), 4.25—4.4 (1H, m, cyclopropyl CH), 8.90 (1H, s, 7-H), 8.92 (1H, s, 2-H). APCI-MS m/z: 384 (M⁺ + 1).

In a similar manner, compounds 6b—d and 7a—c were prepared from 12 and 13, respectively (see Table 1).

In Vitro Antibacterial Activity According to the assay method recommended by the MIC Committee of the Japan Society of Chemotherapy, ¹²⁾ the MIC (in microgram per milliliter) was determined by the two-fold agar dilution method using Mueller-Hinton agar (pH 7.4, Difco); the bacterial inocula contained approximately 10⁶ colony-forming units and the bacterial growth was observed after 20 h of incubation at 37 °C.

In Vivo Efficacy on Systemic Infections An in vivo activity assay was carried out according to the method of Nakamura, et al. $^{1.3}$ Groups of 8 or more male mice (Std-ddY, $20\pm2\,\mathrm{g}$) were infected with P. aeruginosa 12 (i.p., 4×10^3 cells). For evaluation of ED $_{50}$ (p.o.), the test compounds were suspended in 0.4% carboxymethyl cellulose sodium salt and administered orally at 0 and 6 h post-infection. For the determination of ED $_{50}$ (i.v.), the test compounds were dissolved in water with equimolar NaOH and injected intravenously at 0 and 6 h post-infection. Survival rates were evaluated after 1 week, and ED $_{50}$ (p.o.) and ED $_{50}$ (i.v.) were calculated from the rates.

Acknowledgements We are grateful to Dr. M. Hashimoto for his encouragement throughout this work and to Dr. K. Chiba for his helpful discussions. Thanks are also due to Dr. S. Nakamura and his co-workers for biological testing and to members of the Department of Physicochemical Analysis of these laboratories for elemental analyses and spectral measurements.

References and Notes

- This paper is Part XVII in a series of "Pyridonecarboxylic Acids as Antibacterial Agents," Part XVI: see reference 5.
- Hooper D. C., Wolfson J. S., N. Eng. J. Med., 324, 384—394 (1991).
- Matsumoto J., Miyamoto T., Minamida A., Nishimura Y., Egawa H., Nishimura H., J. Med. Chem., 27, 292—301 (1984).
- Miyamoto T., Matsumoto J., Chiba K., Egawa H., Shibamori K., Minamida A., Nishimura Y., Okada H., Kataoka M., Fujita M., Hirose T., Nakano J., J. Med. Chem., 33, 1645—1656 (1990).
- Fujita M., Egawa H., Kataoka M., Miyamoto T., Nakano J., Matsumoto J., Chem. Pharm. Bull., 43, 2123—2132 (1995).
- Hayakawa I., Hiramitsu T., Tanaka Y., Chem. Pharm. Bull., 32, 4907—4913 (1984).
- 7) Kyorin Pharm. Co., Ltd., *Drugs of the Future*, **18**, 203—205 (1993), and the references cited therein.
- Kosuge K., Wada K., Uematsu T., Nakashima M., Ooie T., Kusajima H., Ishida R., Uchida H., Abstracts of Papers, The 32nd Interscience Conference on Antimicrobial Agents and Chemotherapy, Anaheim, California, October 1992, Abstract No. 1003.
- a) Neuman M., Clin. Pharmacokinet., 14, 96—121 (1988);
 b) Lode H., Höffken G., Borner K., Koeppe P., ibid., 16 (Suppl. 1), 1—4 (1989).
- Masuzawa K., Suzue S., Hirai K., Ishizaki T., Eur. Patent Appl. EP230295 [Chem. Abstr., 108, 75230g (1988)].
- Hayakawa I., Tanaka Y., Jpn. Patent Kokai 57-149286 [Chem. Abstr., 98, 72117q (1983)].
- MIC Committee of the Japan Society of Chemotherapy, Chemotherapy, 29, 76 (1981).
- Nakamura S., Minami A., Nakata K., Kurobe N., Kouno K., Sakaguchi Y., Kashimoto S., Yoshida H., Kojima T., Ohue T., Fujimoto K., Nakamura M., Hashimoto M., Shimizu M., Antimicrob. Agents Chemother., 33, 1167—1173 (1989).