

Cyano-Phosphorylation of Aldehydes Catalyzed by a Nucleophilic *N*-Heterocyclic Carbene

Yoshimasa FUKUDA, Yuka MAEDA, Kazuhiro KONDO,* and Toyohiko AOYAMA*

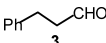
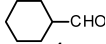
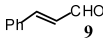
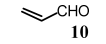
Graduate School of Pharmaceutical Sciences, Nagoya City University; 3-1 Tanabe-dori, Mizuho-ku, Nagoya 467-8603, Japan. Received October 18, 2005; accepted December 12, 2005; published online December 14, 2005

The first method for cyano-phosphorylation of aldehydes with diethyl cyanophosphonate in the presence of *N*-heterocyclic carbene prepared from 1,3-bis(2,4,6-trimethylphenyl)imidazolium chloride and KO*t*-Bu, as a nucleophilic catalyst, is described.

Key words cyano-phosphorylation; *N*-heterocyclic carbene; imidazolium salt; nucleophilic catalyst

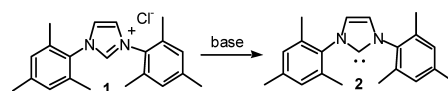
N-Heterocyclic carbenes prepared from imidazolium salts and bases, have been used mainly in transition metal-catalyzed reactions as ligands.^{1,2)} However, their suitability as nucleophilic catalysts has been underdeveloped.^{3–11)} We recently reported a catalytic method leading to the synthesis of various cyanohydrins using *N*-heterocyclic carbene **2** prepared from 1,3-bis(2,4,6-trimethylphenyl)imidazolium chloride (**1**) and KO*t*-Bu as a nucleophilic catalyst in cyanosilylation reactions.¹²⁾ Cyanohydrins and their trimethylsilyl ethers are versatile intermediates in organic synthesis,^{13,14)} but the instability of cyanohydrins and their trimethylsilyl ethers is sometimes problematic for further transformations. Therefore, the development of a one-pot cyanation-*O*-protection reaction with a stable protecting group is desirable. In 1983, the above one-pot reaction of carbonyl compounds with diethyl cyanophosphonate (DEPC) in the catalytic amount of lithium diisopropylamide has been reported by Shioiri *et al.*¹⁵⁾ Herein, we disclose the first catalytic method leading to the synthesis of various cyanohydrin-*O*-phosphonates using *N*-heterocyclic carbene **2** prepared from 1,3-bis(2,4,6-trimethylphenyl)imidazolium chloride (**1**) and KO*t*-Bu as a nucleophilic catalyst in cyano-phosphorylation reactions.

Table 1. Cyano-Phosphorylation of Various Aldehydes with **2** as a Catalyst^{a)}

$\text{R-CHO} \xrightarrow[\text{Solvent, 0}^\circ\text{C, <5 min}]{\begin{matrix} \text{1 (5 mol\%)} \\ \text{KO}t\text{-Bu (4 mol\%)} \\ \text{DEPC (1.2 mol equiv)} \end{matrix}} \text{R-CH(O}^i\text{P(OEt)}_2\text{)CN}$			
Entry	Aldehyde	Solvent	Yield (%) of cyanohydrin- <i>O</i> -phosphonate ^{b)}
1		THF	91
2		THF	94
3	<i>t</i> -Bu-CHO 5	THF	98
4	Ph-CHO 6	THF	94
5	6	DMF	91
6	4-Cl-C ₆ H ₄ -CHO 7	THF	92
7	4-MeO-C ₆ H ₄ -CHO 8	THF	90
8 ^{c)}		THF	19 ^{d)}
9 ^{e)}		THF	90

a) All reactions were performed with 5 mol% of **1**, 4 mol% of KO*t*-Bu and 1.2 moleq of DEPC in the solvent (0.3 M solution of aldehydes). b) Isolated yield. c) Reaction conditions: rt, 30 min. d) Accompanied by recovery of the starting material (30%). e) Reaction conditions: rt, 15 min.

The cyano-phosphorylation of various aldehydes **3–10** was performed as shown in Table 1. The aldehydes except for *trans*-cinnamaldehyde (**9**) were reacted with *N*-heterocyclic carbene **2** prepared from 5 mol% of the imidazolium chloride **1** and 4 mol% of KO*t*-Bu, and 1.2 moleq of DEPC in THF, affording the corresponding cyanohydrin-*O*-phosphonates in good yields.¹⁶⁾ A polar solvent such as DMF was also a suitable solvent (entry 5). The use of ketones such as acetophenone and methyl vinyl ketone afforded little or none of the desired products.



In summary, *N*-heterocyclic carbene **2** was found to function as catalyst in cyano-phosphorylation reaction of aldehydes with DEPC. Ongoing efforts are focused on developing an asymmetric version^{17,18)} of this reaction with a chiral *N*-heterocyclic carbene.

Experimental

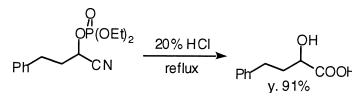
IR spectra were measured on a JASCO IR Report-100 diffraction grating IR spectrophotometer. ¹H- (270 MHz) and ¹³C-NMR (68 MHz) spectra were measured on a JEOL JNM-EX-270 NMR spectrometer. MS spectra were measured on a JEOL JMS-SX-102A instrument. Commercially available aldehydes and reagents were used without any purification. THF was distilled from Na/benzophenone ketyl under a nitrogen atmosphere. DMF was distilled from CaH₂ under reduced pressure. Silica gel column chromatography was performed on Fuji silysia PSQ 60B.

Representative Procedure for the Cyano-Phosphorylation Reaction
To a stirred solution of 1,3-bis(2,4,6-trimethylphenyl)imidazolium chloride (**1**) (5.1 mg, 15 μmol) in THF (1.0 ml) was added KO*t*-Bu (1.4 mg, 13 μmol) at rt and the mixture was stirred for 15 min at the same temperature. After dissolving the imidazolium chloride **1** completely, the mixture was cooled to 0 °C. Then, benzaldehyde (**6**) (32.1 mg, 0.302 mmol) and DEPC (59.0 mg, 0.362 mmol) were added to this mixture. The reaction mixture was stirred for 5 min at the same temperature, then quenched with water and extracted with EtOAc. The organic extracts were successively washed with water and brine, dried (Na₂SO₄) and concentrated. Purification by silica gel chromatography (hexane:EtOAc=4:1) gave phosphoric acid cyano-phenyl-methyl ester diethyl ester (76.3 mg, 94%) as a colorless oil. The physical data were comparable to those reported.¹⁹⁾

The physical data of the known cyanohydrin-*O*-phosphonates shown below were comparable to those of the corresponding literature: Phosphoric acid 1-cyano-3-phenyl-propyl ester diethyl ester.¹⁹⁾ Phosphoric acid cyano-cyclohexyl-methyl ester diethyl ester.²⁰⁾ Phosphoric acid 1-cyano-2,2-dimethyl-propyl ester diethyl ester.²⁰⁾ Phosphoric acid (4-chlorophenyl)-cyano-methyl ester diethyl ester.¹⁹⁾ Phosphoric acid cyano-(4-methoxyphenyl)-methyl ester diethyl ester.¹⁹⁾ Phosphoric acid 1-cyano-3-phenyl-allyl ester diethyl ester.¹⁹⁾ Phosphoric acid 1-cyano-allyl ester diethyl ester.²⁰⁾

References and Notes

- 1) For review: César V., Bellemin-Laponnaz S., Gade L. H., *Chem. Soc. Rev.*, **33**, 619—636 (2004).
- 2) For review: Herrmann W. A., *Angew. Chem., Int. Ed.*, **41**, 1291—1309 (2002).
- 3) For review: Nair V., Bindu S., Sreekumar V., *Angew. Chem. Int. Ed.*, **43**, 5130—5135 (2004).
- 4) For review: Grasa G. A., Singh R., Nolan S. P., *Synthesis*, **2004**, 971—985.
- 5) For example: Connor E. F., Nyce G. W., Myers M., Möck A., Hedrick J. L., *J. Am. Chem. Soc.*, **124**, 914—915 (2002).
- 6) For example: Grasa G. A., Kissling R. M., Nolan S. P., *Org. Lett.*, **4**, 3583—3586 (2002).
- 7) For example: Suzuki Y., Toyota T., Imada F., Sato M., Miyashita A., *Chem. Commun.*, **2003**, 1314—1315.
- 8) For example: Singh R., Kissling R. M., Letellier M.-A., Nolan S. P., *J. Org. Chem.*, **69**, 209—212 (2004).
- 9) For example: Sohn S. S., Rosen E. L., Bode J. W., *J. Am. Chem. Soc.*, **126**, 14370—14371 (2004).
- 10) For example: Burstein C., Glorius F., *Angew. Chem., Int. Ed.*, **43**, 6205—6208 (2004).
- 11) For example: Kano T., Sasaki K., Maruoka, K., *Org. Lett.*, **7**, 1347—1349 (2005).
- 12) Fukuda Y., Maeda Y., Ishii S., Kondo K., Aoyama Y., *Synthesis*, **2006**, 589—590.
- 13) Green T. W., Wuts P. G. M., “Protective Groups in Organic Synthesis,” 3th ed., John Wiley and Sons, Inc., New York, 1999, pp. 348—349.
- 14) Rasmussen J. K., Heilmann S. M., Krepski L. R., “Advances in Silicon Chemistry,” Vol. 1, ed. by Larson G. L., Jai Press Inc, England, 1991, pp. 75—86.
- 15) Harusawa S., Yoneda R., Kurihara Y., Hamada Y., Shioiri T., *Chem. Pharm. Bull.*, **31**, 2932—2935 (1983).
- 16) The cyanohydrin-*O*-phosphonate as shown below gave smoothly α -hydroxycarboxylic acid on treatment with 20% HCl.



- 17) For asymmetric cyano-phosphorylations of aldehydes, see: Baeza A., Casas J., Nájera C., Sansano M. J., Saá M. J., *Angew. Chem. Int. Ed.*, **42**, 3143—3146 (2003).
- 18) For asymmetric cyano-phosphorylations of aldehydes, see: Abiko Y., Yamagiwa M., Sugita M., Tian J., Matsunaga S., Shibasaki M., *Synlett*, **2004**, 2434—2436.
- 19) Baeza A., Nájera C., Sansano M. J., Saá M. J., *Chem. Eur. J.*, **11**, 3849—3862 (2005).
- 20) Micó I., Nájera C., *Tetrahedron*, **49**, 4327—4332 (1993).
- 21) Kurihara T., Miki M., Harusawa S., Yoneda R., *Chem. Pharm. Bull.*, **34**, 4620—4628 (1986).