

## Phenolic Constituents from the Rhizomes of *Dryopteris crassirhizoma*

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A new phenolic glycoside, dryopteroside (1), was isolated from the rhizomes of *Dryopteris crassirhizoma* (Dryopteridaceae), together with five known compounds, 4 $\beta$ -carboxymethyl-(-)-epicatechin (2), isobiflorin (3), biflorin (4), 1- $\beta$ -D-glucopyranosyloxy-3-methoxy-5-hydroxybenzene (5) and (+)-catechin-6-C- $\beta$ -D-glucopyranoside (6). The new compound was elucidated to be 1-butanoyl-3-C- $\beta$ -D-glucopyranosyl-5-methyl-phloroglucinyl-6-O- $\beta$ -D-glucopyranoside (1) by chemical and various spectroscopic analyses. The known compounds 2—6 were first reported from the genus *Dryopteris*.

**Key words** *Dryopteris crassirhizoma*; Dryopteridaceae; dryopteroside

*Dryopteris crassirhizoma* NAKAI (Dryopteridaceae) is distributed mainly in the northeast of China, which rhizomes (common name: Dong-Bei-Guan-Zhong) are widely used as a traditional Chinese medicine for the treatment of intestinal worms, fever caused by influenza and vomiting of blood.<sup>1)</sup> Phloroglucinol derivatives and flavonoid glycosides have been reported from the rhizomes of *D. crassirhizoma*,<sup>2,3)</sup> and have demonstrated antibacterial, antitumor-promoting, antioxidant and HIV-1 reverse transcriptase inhibitory activities.<sup>4—7)</sup> As a part of our ongoing investigation on medicinal plants in the northeast of China, we carried out a phytochemical investigation on the rhizomes of *D. crassirhizoma*, which resulted in the isolation of a new phenolic glycoside, dryopteroside (1), together with five known compounds. In this paper, we report the isolation and structural determination of the new compound on the basis of chemical and various spectroscopic analyses.

The air-dried rhizomes of *D. crassirhizoma* were extracted with MeOH and the extract was partitioned with EtOAc and H<sub>2</sub>O. The H<sub>2</sub>O layer was passed through a Diaion HP-20 column, and washed with H<sub>2</sub>O, 40% MeOH and MeOH. The 40% MeOH eluate fraction was separated by normal-phase and reversed-phase (RP) silica gel column chromatography (CC), and purified by repeated RP-HPLC to afford compounds 1—6. The known compounds 2—6 were identified as

4 $\beta$ -carboxymethyl-(-)-epicatechin (2),<sup>8)</sup> isobiflorin (3),<sup>9)</sup> biflorin (4),<sup>9)</sup> 1- $\beta$ -D-glucopyranosyloxy-3-methoxy-5-hydroxybenzene (5)<sup>10)</sup> and (+)-catechin-6-C- $\beta$ -D-glucopyranoside (6)<sup>11)</sup> by comparison of their spectral data with the reported values. The known compounds 2—6 were first reported from the genus *Dryopteris*.

Dryopteroside (1) was obtained as a pale yellow amorphous powder with a molecular formula of C<sub>23</sub>H<sub>34</sub>O<sub>14</sub>, as determined by the HR-FAB-MS data, implying the presence of seven degrees of unsaturation in the molecule. In the <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of 1, the signals assignable to the C- $\beta$ -glucopyranosyl and the O- $\beta$ -glucopyranosyl moieties were observed with the aromatic proton signals at  $\delta$ <sub>H</sub> 5.79 (1H, d,  $J$ =9.8 Hz, H-1') and 5.22 (1H, d,  $J$ =7.4 Hz, H-1''), correlated in the HMQC spectrum with two anomeric carbon signals at  $\delta$ <sub>C</sub> 76.8 (C-1') and 105.9 (C-1''), respectively. The presence of the butanoyl moiety was indicated by the proton signals of one methyl at  $\delta$ <sub>H</sub> 0.81 (3H, t,  $J$ =7.5 Hz, H<sub>3</sub>-10), two methylene at  $\delta$ <sub>H</sub> 1.66 (1H, m, H<sub>a</sub>-9), 1.73 (1H, m, H<sub>b</sub>-9) and 3.24 (1H, ddd,  $J$ =17.0, 8.7, 6.1 Hz, H<sub>a</sub>-8),  $\delta$ <sub>H</sub> 3.61 (1H, ddd,  $J$ =17.0, 8.7, 6.1 Hz, H<sub>b</sub>-8) in the <sup>1</sup>H-NMR spectrum, and the carbon signals at  $\delta$ <sub>C</sub> 18.1 (C-10), 14.1 (C-9), 46.4 (C-8) and 208.9 (C-7) in the <sup>13</sup>C-NMR spectrum,<sup>12)</sup> and further supported by the DQF-COSY and HMBC correlations as shown in Fig. 1. Besides the signals due to sugars and the bu-

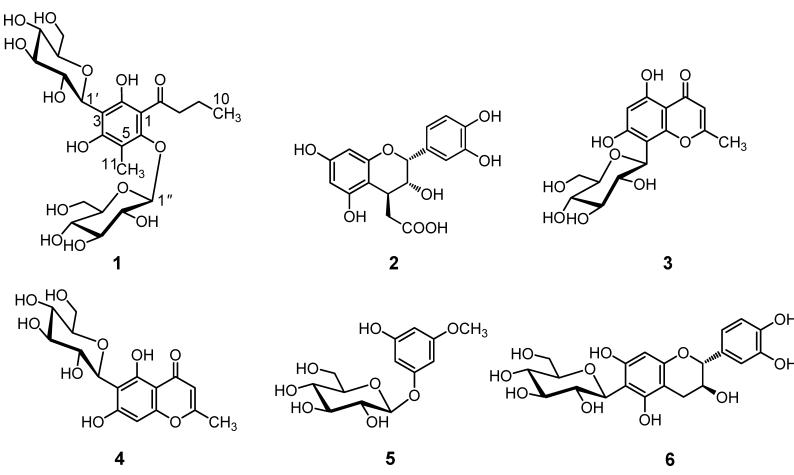
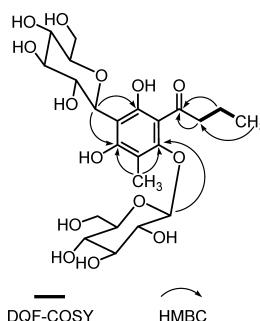


Chart 1

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Table 1.  $^1\text{H}$ - (500 MHz) and  $^{13}\text{C}$ -NMR (125 MHz) Spectral Data of **1** in  $\text{C}_5\text{D}_5\text{N}$ 

Position	<b>1</b>		Position	<b>1</b>	
	$\delta_{\text{H}}$ (mult, $J$ in Hz)	$\delta_{\text{C}}$		$\delta_{\text{H}}$ (mult, $J$ in Hz)	$\delta_{\text{C}}$
1		112.9	Glc-1'	5.79 (d, 9.8)	76.8
2		159.6	2'	4.51 (t, 9.8)	74.3
3		109.8	3'	4.36 (t, 9.2)	79.7
4		161.6	4'	4.49 (t, 9.4)	70.8
5		111.4	5'	4.06 (dt, 9.7, 2.9)	82.7
6		156.6	6'	4.45 (dd, 11.8, 2.2)	61.5
7		208.9		4.48 (dd, 11.8, 3.2)	
8	3.24 (ddd, 17.0, 8.7, 6.1) 3.61 (ddd, 17.0, 8.7, 6.1)	46.4	Glc-1"	5.22 (d, 7.4)	105.9
9	1.66 (m) 1.73 (m)	14.1	2"	4.31 (dd, 8.0, 7.4)	75.8
10	0.81 (t, 7.5)	18.5	3"	4.30 (t, 8.0)	78.3
11	2.54 (s)	9.6	4"	4.25 (t, 8.3)	71.8
			5"	3.88 (ddd, 9.3, 5.7, 2.8)	78.5
			6"	4.26 (dd, 11.4, 5.3)	62.7
				4.37 (dd, 11.4, 2.8)	

Fig. 1. Selected DQF-COSY and HMBC Correlations for **1**

tanoyl moiety, the  $^1\text{H}$ -NMR spectrum showed a methyl singlet at  $\delta_{\text{H}}$  2.54 and the  $^{13}\text{C}$ -NMR spectrum showed six additional quaternary carbons. Taking the remaining three degrees of unsaturation into consideration, the presence of one hexasubstituted aromatic ring in **1** was suggested. In the HMBC spectrum, the correlations between  $\delta_{\text{H}}$  5.79 (H-1') and  $\delta_{\text{C}}$  159.6 (C-2) and 161.6 (C-4) suggested that the C- $\beta$ -glucosyl moiety was located at C-3. The HMBC correlations between  $\delta_{\text{H}}$  2.54 (H-11) and  $\delta_{\text{C}}$  161.6 (C-4), 156.6 (C-6), and  $\delta_{\text{H}}$  5.22 (H-1") and  $\delta_{\text{C}}$  156.6 (C-6) established the attachment of the methyl group at C-5 and the O- $\beta$ -glucosyl moiety at C-6. On acid hydrolysis, **1** afforded D-glucose as a component sugar, which was identified by gas-liquid chromatography (GLC) analysis of its trimethylsilyl thiazolidine derivative.<sup>13</sup> The absolute configuration of the C- $\beta$ -glucosyl moiety was not chemically corrected. However, it was considered as D-form in keeping with those mostly encountered among plant C-glucosides.<sup>14–16</sup> Thus, the structure of dryopteroside (**1**) was determined to be 1-butanoyl-3-C- $\beta$ -D-glucopyranosyl-5-methyl-phloroglucinyl-6-O- $\beta$ -D-glucopyranoside.

## Experimental

**General Experimental Procedures** The UV spectra were obtained with a Shimadzu UV-160 spectrophotometer, whereas the IR spectra were measured with a JASCO FT/IR-300E (by a KBr disk method) spectrometer. Optical rotations were measured with a JASCO DIP-370 digital polarimeter in a 0.5-dm cell. The FAB-MS was taken on a JEOL JMS-700 MStation spectrometer. The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra were measured with a JEOL ECP-500 spectrometer or a JEOL AL-400 spectrometer in the solution with TMS as the internal reference, and chemical shifts are expressed in  $\delta$  (ppm). RP-

HPLC separation was carried out with a JASCO PU-2080 HPLC system, equipped with a Shodex RI-101 Differential Refractometer detector and a Senshu Pak RP-C<sub>18</sub> column (20×150 mm i.d.). RP CC was accomplished with RP-C<sub>18</sub> silica gel (100–200 mesh, Chromatorex DM1020T ODS, Fuji Silysia Chemical Ltd.). Silica gel CC was carried out with Kieselgel 60 (E. Merck). TLC was conducted in Kieselgel 60 F<sub>254</sub> plates (E. Merck). GLC was carried out on a PerkinElmer Clarus 500 GC-MS instrument.

**Extraction and Isolation** The rhizomes of *D. crassirhizoma* used in this study were purchased in Shenyang “Bo Kang” pharmacy, Liaoning Province, P. R. China, and identified by Professor Qishi Sun, Shenyang Pharmaceutical University. The air-dried rhizomes (1.5 kg) were extracted with MeOH by ultrasonic treatment three times for 1 h each at room temperature. Evaporation of the solvent under reduced pressure provided a MeOH extract (171.8 g). The MeOH extract was suspended in H<sub>2</sub>O and then partitioned with EtOAc. The H<sub>2</sub>O layer was subjected to a Diaion HP-20 column, and washed with H<sub>2</sub>O, 40% MeOH, and MeOH. The 40% MeOH fraction (8.4 g) was chromatographed over a silical gel column eluted with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (60:29:6) to give two fractions, A (1.1 g) and B (7.0 g). Further purification of fraction A by repeated RP-HPLC with 50% MeOH or 25% CH<sub>3</sub>CN afforded six compounds, **1** (16 mg), **2** (17 mg), **3** (27 mg), **4** (6 mg), **5** (6 mg) and **6** (55 mg).

**Dryopteroside (1):** Pale yellow amorphous powder;  $[\alpha]_D^{25} +103.4^\circ$  ( $c=1.0$ , MeOH); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 222 (4.29), 280 (4.05), 324 (3.79); IR  $\nu_{\text{max}}$  (KBr) cm<sup>-1</sup>: 3402, 2927, 1614, 1074;  $^1\text{H}$ -NMR (C<sub>5</sub>D<sub>5</sub>N, 500 MHz) and  $^{13}\text{C}$ -NMR (C<sub>5</sub>D<sub>5</sub>N, 125 MHz); see Table 1; FAB-MS (negative)  $m/z$  533 [M-H]<sup>-</sup>; HR-FAB-MS (negative)  $m/z$  533.1879 [M-H]<sup>-</sup> (Calcd for C<sub>23</sub>H<sub>33</sub>O<sub>14</sub>, 533.1870).

**Acid Hydrolysis of 1** Compound **1** (1 mg) in 1 M HCl (dioxane-H<sub>2</sub>O, 1:1 v/v, 200  $\mu$ l) was heated at 100 °C for 1 h under an Ar atmosphere. After dioxane was removed, the solution was extracted with EtOAc (1 ml×3) to remove the aglycon. The aqueous layer was neutralized by passing through an ion-exchange resin (Amberlite MB-3, Organo, Tokyo, Japan) column, concentrated under reduced pressure to dryness, to give a residue of the sugar fraction. The residue was dissolved in pyridine (1 ml), to which 0.1 M L-cysteine methyl ester hydrochloride in pyridine (2 ml) was added. The mixture was kept at 60 °C for 1.5 h. After the reaction mixture was dried *in vacuo*, the residue was trimethylsilylated with 1-trimethylsilyl imidazole (0.2 ml) for 2 h. The mixture was partitioned between hexane and H<sub>2</sub>O (0.3 ml each) and the hexane extract was analyzed by GLC under the following conditions: capillary column, EQUITY<sup>TM</sup>-1 (30 m×0.25 mm×0.25  $\mu$ m, Supelco), column temperature, 230 °C; injection temperature, 250 °C; carrier N<sub>2</sub> gas. In the acid hydrolysate of **1**, D-glucose was confirmed by comparison of the retention times of their derivatives with those of D-glucose and L-glucose derivatives prepared in a similar way, which showed retention times of 10.80 and 11.20 min, respectively.

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