

Adirubine: a Novel Carboxy-indole Alkaloid

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Summary Adirubine (**1a**), the first example of a new series of carboxy-alkaloids with the *Corynanthe* skeleton, has been isolated from *Adina rubescens* and its structure determined from chemical and spectral data.

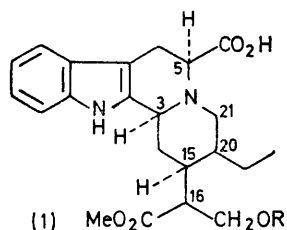
THE isolation of cordifoline¹ and related carboxy-alkaloids^{2,3} predicted the existence of a common precursor derived by condensation between tryptophan and secologanin (see preceding communication). This tetrahydrodesoxycordifoline (TDC) could be (a) a biogenetic cul-de-sac, (b) an alternative intermediate to vincoside⁴ for some indole alkaloids, or (c) the progenitor of a novel range of acidic indole alkaloids in which the carboxy-group of tryptophan was retained.⁵ The last possibility stimulated a search for members of this hypothetical tryptophan family, and we now report the discovery in *Adina rubescens* of the first example, adirubine, to which we have assigned the *Corynanthe* type structure (**1a**).

By a combination of gel permeation chromatography, gradient pH extraction, and preparative t.l.c. adirubine was isolated as the monoacetate, C₂₄H₃₀O₅N₂, m.p. 151—154°, [α]_D²⁵ - 19° (CHCl₃). The u.v. spectrum indicated an indolic chromophore and carbonyl absorption at 1735 cm⁻¹ and 3H singlets at τ 6.30 and 8.02 were consistent with

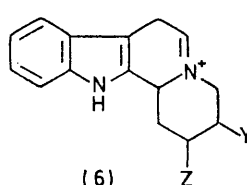
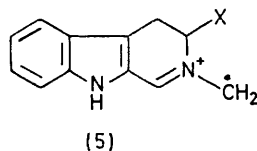
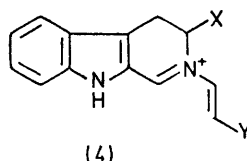
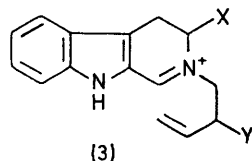
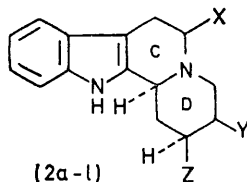
CO₂Me and AcO functions respectively. The presence of an ethyl group was shown by a 3H triplet at τ 9.17. Treatment with diazomethane gave methyladirubine acetate, C₂₅H₃₂O₆N₂, [α]_D²⁵ - 32° (MeOH), with an additional methoxy signal at τ 6.44. Both compounds had a 2H multiplet in the region τ 5.6—6.1 which was attributed to an acetylated primary alcohol since there was an upfield shift to τ 6.0—6.5 on deacetylation to methyladirubine, C₂₃H₃₀O₅N₂, [α]_D²⁵ - 34° (CHCl₃). Reduction (LiAlH₄) of methyladirubine acetate afforded a triol, C₂₁H₃₀O₃N₂, and the deuteride a corresponding [²H₄]triol, confirming the presence of three ester functions.

The mass spectra of adirubine derivatives and various model compounds (**2a—1**) showed clearly the basic tetracyclic *Corynanthe* structure and the positions of substituents (see Table).

It was apparent that there were many similarities between the fragmentation patterns of adirubine derivatives (**2e—f**) and those of dihydrositsirikine (**2i—1**).⁶ In particular both had ions (**3**) due to loss of the C-15 substituent and breakdown of ring D [(**4**) and (**5**)] in addition to the usual peaks at *m/e* 169 and 156. However, the presence in the former of major peaks at *m/e* 168 and 182 as opposed to 170 and 184 in the latter found an analogy in polynneuridine⁷ where it



a; R = H
b; R = Ac



TABLE

Structure (2)	Substituents			<i>m/e</i> for ion ^a				
	X	Y	Z	<i>M</i> ⁺	(3)	(4)	(5)	(6)
a	CO ₂ H	Et	MeO ₂ C-CH-CH ₂ OAc	442	297	269	228	397
b	CO ₂ Me	"	"	456	311	283	242	397
c	CO ₂ CH ₂ D	"	"	457	312	284	243	397
d	CO ₂ Me	"	MeO ₂ C-CH-CH ₂ OH	414	311	283	242	355
e	CH ₂ OH	"	HOCH ₂ CH-CH ₂ OH	358	283	255	? ^b	327
f	CD ₂ OH	"	HOCD ₂ CH-CH ₂ OH	362	285	257	? ^b	329
g	CO ₂ Me	H	H	284	—	255	242	225
h	CH ₂ OH	"	"	256	—	?	?	225
i	H	Et	MeO ₂ C-CH-CH ₂ OAc	398	253	225	184	—
j	"	Et	MeO ₂ C-CH-CH ₂ OH	356	253	225	184	—
k	"	"	HOCH ₂ CH-CH ₂ OH	328	253	225	184	—
l	"	CH ₂ OH	"	330	255	227	184	—

^a Assignments were confirmed by accurate mass measurement; ^b Absent, but peaks were present 2 m.u. lower at 212 and 214 respectively corresponding to fully aromatised ions.

was diagnostic of an additional C-C bond in ring c. This could be correlated with the most striking difference which with adirubine acetate (2a), for example, was the ready loss of CO₂H to give the base peak at *m/e* 397, invariably accompanied by a smaller one due to loss of two more hydrogen atoms. The obvious inference was that the carboxy-group was attached to C-5 and cleavage was generating a favourable immonium ion (6), which in turn

would readily aromatise. Identical behaviour was shown by methyladirubine, its acetate, and adirubine triol.

For confirmation that a group at C-5 would control the fragmentation in this way model compounds were synthesised. Heating of methyl L-tryptophanate and α -oxo-adipic acid in acetic acid afforded a lactam, m.p. 176°, [α]_D + 198° (CHCl₃), which was smoothly reduced by diborane to the amino-ester (2g), [α]_D + 99° (MeOH), and by LiAlH₄ to the amino-alcohol (2h). In its mass spectra the ester was remarkably similar to methyladirubine with a base peak at *M* - CO₂Me and other important ions at *m/e* 255, 223 (*M* - CO₂Me - 2H), 182, 169, 168, and 156; the alcohol was likewise analogous to adirubine triol.

Finally the ions (4) and (5) placed the ethyl group in adirubine at C-20. The unlikely alternative of C-21 was excluded firstly by the absence of any ion corresponding to (5) with retention of the ethyl group, and secondly by lack of the substantial *M* - 29 peak that might be expected if the ethyl were α to N_b like the C-5 substituent.

The c.d. spectrum of methyladirubine acetate displayed a positive Cotton effect between 300 and 250 nm and the absolute configuration at C-3 was thus readily established as α .⁸ Since 15-H is almost invariably α one would expect a *cis* relationship between 3-H and 15-H and this was confirmed by the presence of Bohlmann i.r. bands at 2800, 2760, and 2735 cm⁻¹ and the lack of any signal due to 3-H below τ 6.2 in its n.m.r. spectrum. On the assumption that adirubine is derived from L-tryptophan, it is likely that 5-H is also α , but the stereochemistry at C-16 and C-20 is unknown.

We thus deduce the structure (1b) for adirubine acetate and (1a) for the naturally occurring adirubine, surely only the first of many congeners of the *Corynanthe*-*Strychnos* type. It will be of interest to see whether carboxy-representatives of the rearranged *Iboga* and *Aspidosperma* families will also be found.

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