Intramolecular Cyclizations of Allyl- and Prop-2-ynyl-silanes with the Terminal Group governing the Regiochemical Outcome of the Cyclization

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The intramolecular additions of allyl- and prop-2-ynyl-silanes to enones and dienones are described; cyclizations carried out using 1.1 equiv. of either $EtAlCl_2$ in toluene or $TiCl_4$ in methylene chloride at low temperature have shown that the terminal group governs the regiochemical outcome of the cyclization, with allylsilane leading to 1,6 addition and prop-2-ynylsilane to 1,4 addition.

As part of a general study of Lewis-acid promoted intramolecular reactions of allyl- and prop-2-ynyl-silanes we have recently introduced EtAlCl₂ as a useful catalyst for intramolecular Sakurai reactions that lead stereoselectively to spiro-[4.5]decanones^{1,2} and to fused [4.3.0]nonanones.³ The products obtained are potentially useful precursors for a number of natural products.³ We now describe further syntheses and

additions of allyl- and prop-2-ynyl-silanes to enones and conjugated dienones. Related work by Majetich and co-workers⁴ has recently been published.

Following Stork's⁵ early studies, 3-ethoxycyclohex-2-enone was used to construct the desired starting materials. Alkylation of 3-ethoxycyclohex-2-enone with 5-trimethylsilyl-pent-3-enyl or -pent-3-ynyl iodide⁶ yielded the alkylation products.

O SiMe₃

(1)
$$R = H, Me$$

(2)

Reagents and conditions: i, $EtAlCl_2$, toluene, -78 °C; ii, $TiCl_4$, CH_2Cl_2 , -78 °C.

These were then used in reactions with e.g., disobutylaluminium hydride (DIBAH), MeLi, vinyl Grignard reagents, and 6-trimethylsilylhex-4-ynylmagnesium bromide³ to afford the starting materials (1), (3), (4), (8), and (10). The cyclizations were carried out in toluene at -78 °C with 1.1 equiv. of EtAlCl₂ or in CH₂Cl₂ with 1.1 equiv. of TiCl₄ as Lewis acid catalysts. From (1) we obtained single isomers of the expected cis fused products (2) in 50% (R = H) and 87% (R = Me) yields.†

The intramolecular addition of allyl- and prop-2-ynyl-silanes (3) and (4) to conjugated dienones showed an interesting behaviour. Compound (3) cyclized only in 1,6 fashion leading to the bicyclo[4.5.0]undecenones (5). The reaction proceeded diastereoselectively at -78 °C forming a single diastereoisomer (60% yield).†‡ In contrast with (3), the prop-2-ynylsilanes (4) reversed the regioselectivity to afford the 1,4 addition products, bicyclo[4.3.0]nonanones (6) with an angular vinyl group as the major products, with only a minor amount of the 1,6 addition products (7) [ratio (6): (7) = 10:1; yields R = H: (6) 80%, (7) 8%; R = Me: (6) 68%, (7) 6%].†

The preference for 1,4 addition is probably influenced by the low nucleophilicity of the prop-2-ynylsilane terminal group. Therefore, the formation of the five-membered ring reflects a reaction at the centre with the highest charge density. The 1,6 addition of the allylsilane could be a result of higher nucleophilicity; steric interactions become more important and the molecule reacts at the less hindered site forming a seven-membered ring.

Treatment of compounds (8) and (10) with EtAlCl₂ as the Lewis acid catalyst did not afford the cyclization products. However, the reactions proceeded smoothly with TiCl₄ at -78 °C. The spirocyclic product (9) is diastereoisomerically pure and was obtained in 46% yield.†§ Finally we have examined the cyclization of the sterically hindered compound (10). Again, reaction proceeded with TiCl₄ at -78 °C forming a highly functionalized hydrindanone (11) with two quaternary centres in 62% yield.

These results demonstrate an interesting change in ring-size selectivity between allyl- and prop-2-ynyl-silanes, governed by the terminal group, and provide a route to controlled annulation.

This work was supported by the Deutsche Forschungsgemeinschaft, the Alexander von Humboldt-Foundation (Humboldt fellowship to S. S.), and the Fonds der Chemischen Industrie (Liebig fellowship to D. S.). We thank Dr. V. Wray, Gesellschaft für Biotechnologische Forschung (GBF), Braunschweig–Stöckheim, for carrying out the n.O.e. experiments.

Received, 3rd February 1986; Com. 154

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[†] All new compounds have been fully characterized.

[‡] Compound (**5**) shows only one set of lines in the 13 C n.m.r. spectrum, (100.61 MHz CDCl₃): δ 199(s), 171(s), 143.4(d), 127.6(d), 112.6(t), 44.9(d), 39.5(d), 37.3(t), 35.5(t), 34.7(t), 33.0(t), 30.56(t), 30.0(t).

[§] Assignment of the configurations of the product followed from nuclear Overhauser effect (n.O.e.) experiments.