

## Crystal Structure of the Gallophosphate Framework: X-Ray Characterization of $\text{Ga}_9\text{P}_9\text{O}_{36}\text{OH}\cdot\text{HNEt}_3$

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A gallophosphate,  $\text{Ga}_9\text{P}_9\text{O}_{36}\text{OH}\cdot\text{HNEt}_3$ , has been synthesised and its framework structure has been determined by single crystal X-ray diffraction.

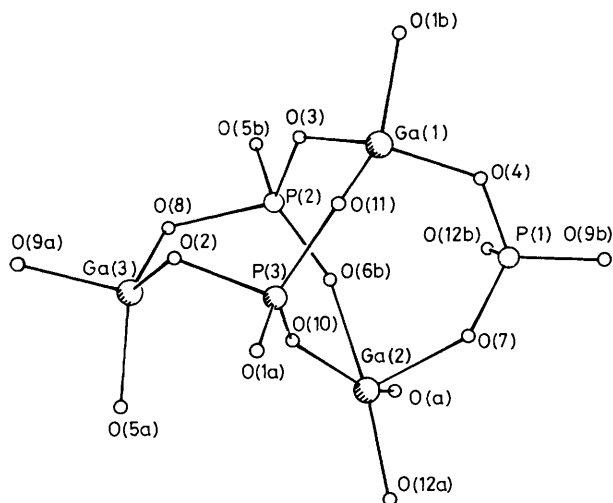
Recently, the syntheses of open framework aluminophosphates with structures and properties similar to aluminosilicate zeolites have been reported.<sup>1-3</sup> This was followed by the report<sup>4</sup> of the gallophosphates, some of which are analogues of aluminophosphate molecular sieves ( $\text{AlPO}_4\text{-n}$ ). Replacement of aluminium in a framework structure should result in materials with unique properties and structures, with a variety of cavity and pore geometries. Reported here is the synthesis and the single crystal X-ray structure of a gallophosphate,  $\text{Ga}_9\text{P}_9\text{O}_{36}\text{OH}\cdot\text{HNEt}_3$ .

The gallophosphate was synthesised by a hydrothermal procedure and associated with a specific 'template,' triethylamine. An aqueous mixture of hydrated  $\text{GaOOH}$ , phosphoric acid, and  $\text{Et}_3\text{N}$  (1.5  $\text{Et}_3\text{N}$ :1.0  $\text{Ga}_2\text{O}_3$ :1.2  $\text{P}_2\text{O}_5$ :50  $\text{H}_2\text{O}$ ) was heated at 180 °C for 4 days under autogeneous pressure. Comparison with the characteristic X-ray powder diffraction patterns of the  $\text{AlPO}_4$  frameworks reported<sup>1,4-7</sup> so far

indicated that the gallophosphate had a unique, novel structure.

The structure<sup>†</sup> was solved by direct methods and refined using block matrix-least-squares. On the basis of the elemental analysis, the asymmetric unit contents shown in Figure 1 appear to be  $\text{Ga}_3\text{P}_3\text{O}_{12}\cdot\frac{1}{3}\text{OH}\cdot\frac{1}{3}\text{HNEt}_3$ ,  $\frac{1}{3}\text{H}^+$  being added to the organic unit to balance the charge of the inorganic unit.

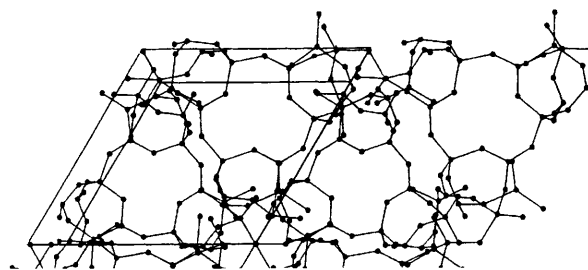
<sup>†</sup> Crystal data:  $\text{Ga}_3\text{P}_3\text{O}_{12}\cdot\frac{1}{3}\text{OH}\cdot\frac{1}{3}\text{HNEt}_3$ ,  $M_r = 524.74$ ; space group  $P6_3$ ,  $a = 12.2665(32)$ ,  $c = 16.7462(50)$  Å,  $U = 2182.16$  Å<sup>3</sup>,  $Z = 6$ ;  $D_c = 2.396$  g cm<sup>-3</sup>; number of reflections for cell refinement 25; crystal shape: hexagonal prism, 40 × 100 μm long; Mo- $K_\alpha$  0.71069 Å, graphite monochromator, Nicolet XRD R<sub>3</sub> diffractometer;  $h, \pm k, l$ ; scan mode  $\theta$ -2 $\theta$ , variable speed; 1188 unique out of 3885 measured intensities [ $I > 2.60\sigma(I)$ ];  $\mu = 61.81$  cm<sup>-1</sup>, SHELXTL; isotropic thermal parameters. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.



**Figure 1.** The asymmetric unit of the gallophosphate structure. Selected and average bond lengths: P–O 1.55, Ga(1)–O 1.791, Ga(2)–O 1.920, Ga(2)–O(a) 1.836, Ga(3)–O 1.820 Å. O(a) represents the position of a hydroxy group.

Each phosphorus atom is tetrahedrally co-ordinated and shares an oxygen atom with four adjacent gallium atoms. Of the three types of gallium atoms, two lie at tetrahedral centres, and the third is located in a distorted trigonal bipyramid. Twelve of the thirteen types of oxygen atom are bonded to one phosphorus and one gallium atom, whereas the thirteenth, the position of a hydroxy group, is symmetrically bonded to three gallium atoms (type 2). All the co-ordination polyhedra are vertex-shared. The co-ordination states of the gallium atoms are related to those in  $\text{GaPO}_4\text{-14}$  reported by Parise<sup>4</sup> in which gallium atoms are 4-, 5-, and 6-co-ordinated with oxygen atoms or hydroxy groups.

In the three-dimensional net, there are two open channels running along the *c* axis as shown in Figure 2; the smaller passes through the origin and its axis coincides with the *c* axis, while the larger coincides with the three-fold axis and accommodates the charged triethylamine template. Both the channels are formed from gallium and phosphorus equilateral triangles of different sizes, perpendicular to the *c* axis.



**Figure 2.** Packing plot projected along *c* showing the framework structure, with O, Ga, and P atoms all indicated by small filled circles.

Alternatively, they can be viewed as zigzag 6-rings packed in the direction of the *c* axis, in which gallium and phosphorus atoms are cross-linked *via* oxygen atoms. As a result, the gallium and phosphorus atoms in the channels are not aligned.

The two channels are connected in the 001 plane through 10- and 8-rings to form a hexagonal packing pattern with a rectangular cross-section.

In summary, a gallophosphate with an open framework structure has been synthesised and its crystal structure solved. It has channels in both the 001 and 010 planes which accommodate the triethylamine template.

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