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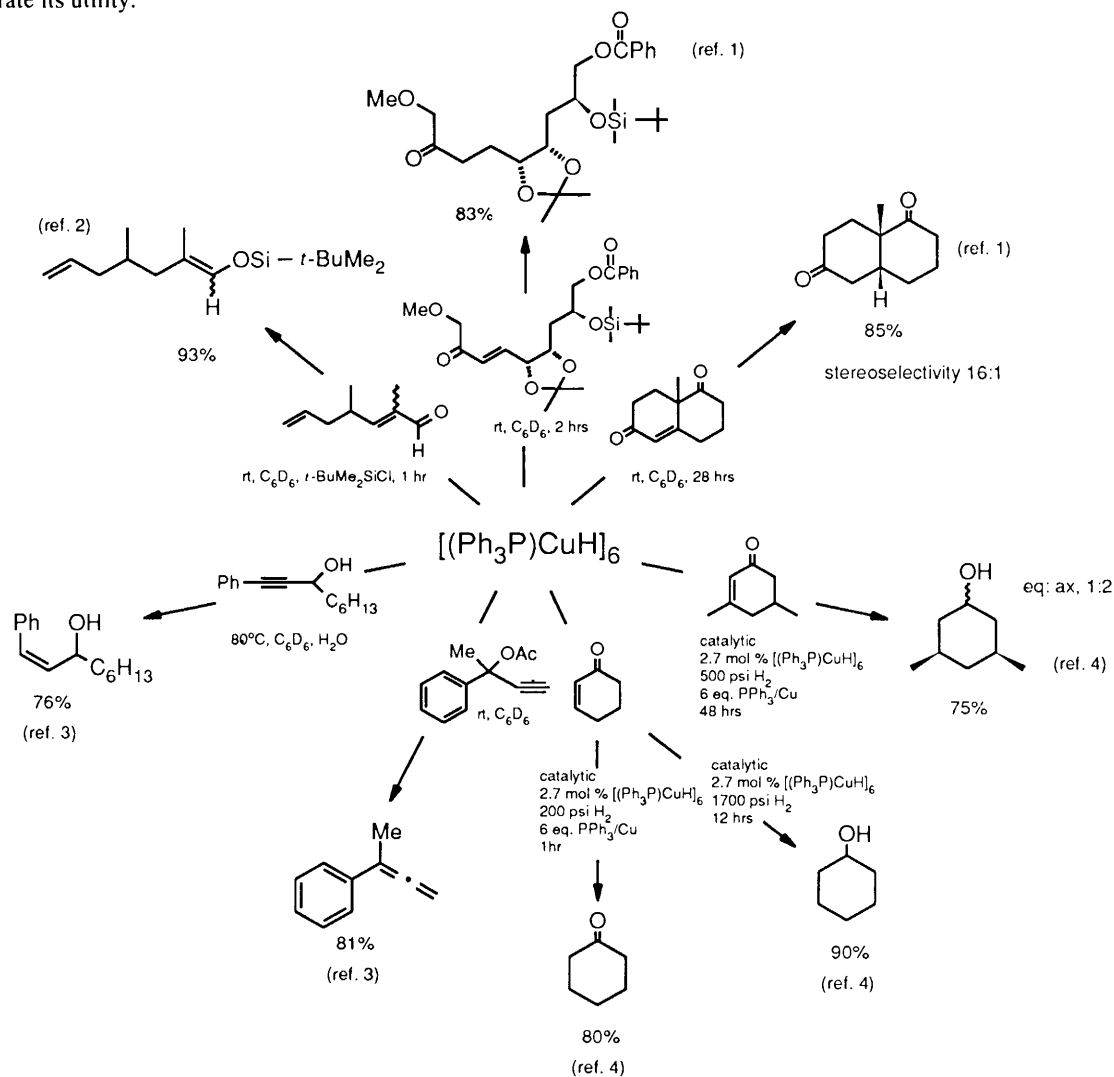
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(Triphenylphosphine)copper hydride hexamer

Dr. Jeffrey M. Stryker at Indiana University kindly suggested that we offer (triphenylphosphine)copper hydride hexamer, $[(\text{Ph}_3\text{P})\text{CuH}]_6$, a stable, conveniently handled copper(I) hydride cluster. Extensive research in Dr. Stryker's laboratory has demonstrated that this mild hydride donor is useful in a variety of synthetic applications including stoichiometric and catalytic hydrogenations, chemoselective conjugate reductions, as well as regioselective and stereoselective conjugate hydride reductions.¹⁻⁴ The versatility of the compound makes it a very useful reagent for synthetic organic chemistry.

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References:

- 1) Mahoney, W.S.; Brestensky, D.M.; Stryker, J.M. *J. Am. Chem. Soc.* **1988**, *110*, 291.
- 2) Brestensky, D.M.; Stryker, J.M. *Tetrahedron Lett.* **1989**, *30*, 5677.
- 3) Daeuble, J.F.; McGettigan, C.; Stryker, J.M. *ibid.*, in press.
- 4) Koenig, T.M.; Daeuble, J.F.; Brestensky, D.M.; Stryker, J.M. *ibid.*, in press.

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