

SOME NEW SYNTHETIC APPROACHES FOR THE PREPARATION OF PTERIDINE-3-OXIDES AND PTERIDINES

Marjan Kočevar, Branko Stanovnik, and Miha Tišler

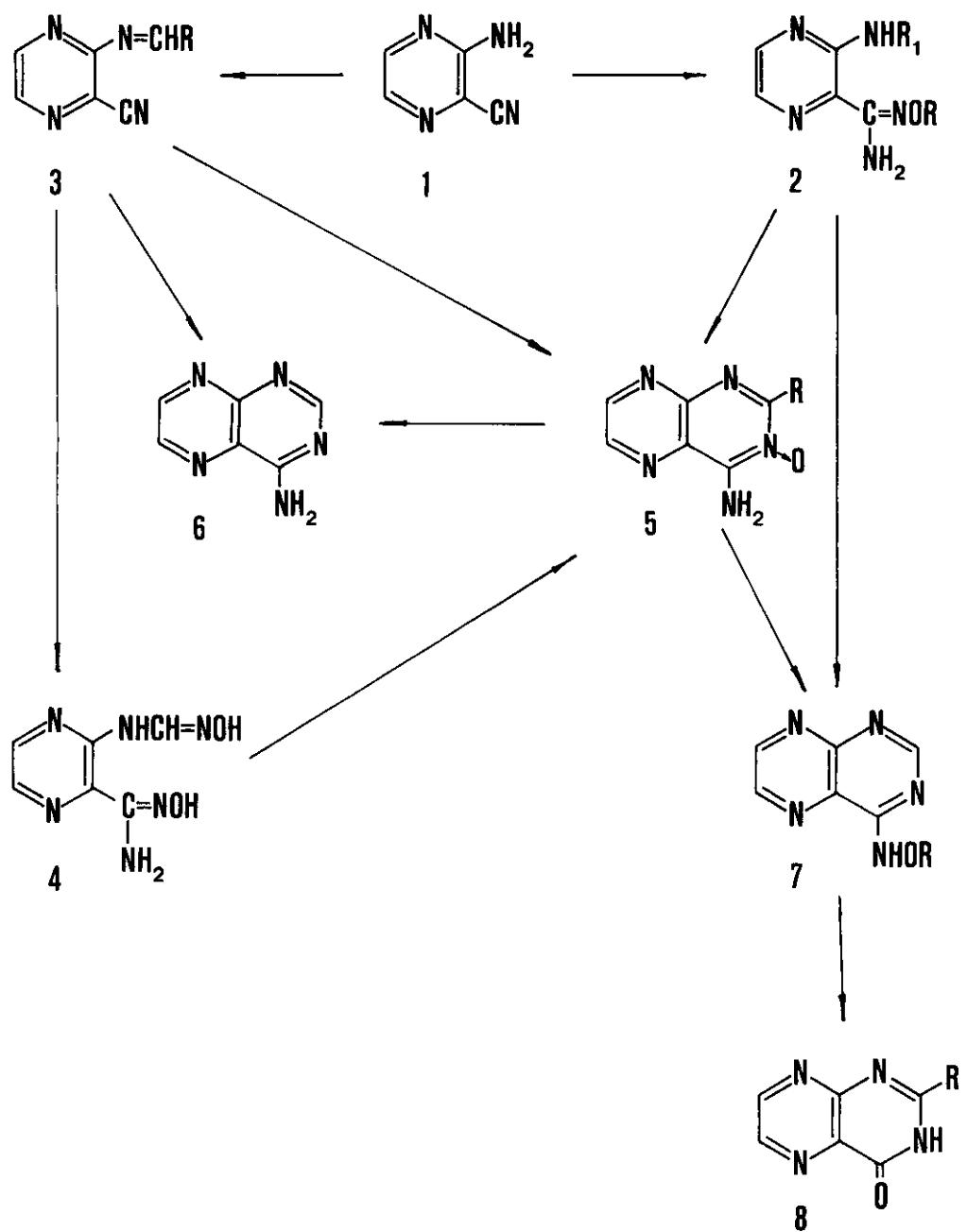
Department of Chemistry, E.Kardelj University, 61000 Ljubljana,
Yugoslavia

Abstract - Starting from 2-amino-3-cyanopyrazine some new syntheses of 4-substituted pteridines and pteridine-3-oxides have been devised.

The chemistry of o-aminonitriles has been extensively investigated¹ and several useful syntheses have been elaborated in the past for the formation of bi- and polycyclic heterocycles. As a continuation of our investigations to use substituted heterocyclic amidines as synthons for new heterocycles²⁻¹² we like to report on a new and useful approach for the preparation of various 4-substituted pteridines and their 3-oxides.

As starting compound 2-amino-3-cyanopyrazine (1) was used. It reacted at room temperature to give the amidoxime (2, R=R₁=H) in 85% yield, mp 185-187°C (H₂O); m/e 153 (M⁺) and nmr δ (DMSO-d₆) 7.40 and 7.58 (d, H₅ and H₆), 5.62 and 7.02 (broad s, NH₂), 9.64 (s, OH), J_{5,6} = 2.4 Hz. The amidoxime can be O-acylated and the acylation site is shown by further transformations into compound 7 (R = COMe). Thus, with acetic anhydride the O-acetyl compound 2 (R₁ = H, R = COMe) was obtained in almost quantitative yield, mp 170-171.5°C (EtOH), m/e 195 (M⁺), and with benzoyl chloride and in the presence of triethylamine the O-benzoyl derivative 2 (R₁ = H, R = COPh) was obtained in 97% yield, mp 216-218°C (EtOH); m/e 257 (M⁺). Treatment of the amidoxime 2 (R=R₁=H) with methyl iodide in the presence of sodium propoxide afforded the O-methyl derivative 2 (R₁ = H, R = Me) in 48% yield, mp 108-110°C (H₂O); m/e 167 (M⁺) and nmr δ (DMSO-d₆) 7.23 and 7.43 (d, H₅ and H₆), 3.52 (s, Me), 5.67 and 6.80 (broad s, NH₂), J_{5,6} = 2.4 Hz.

On the other hand, compound 1 was transformed with N,N-dimethylformamide dimethyl acetal at room temperature into 3 (R = NMe₂) in 89% yield, mp 86-88°C (CHCl₃ and petroleum ether); m/e 175 (M⁺) and nmr δ (DMSO-d₆) 8.31 (s, CH), 7.83 and 8.07 (d, H₅ and H₆), 3.06 (s, Me) 2.97 (s, Me), J_{5,6} = 2.4 Hz.¹³ Alternatively, from 1 and triethyl orthoformate after 15 h under reflux the ethoxymethylene derivative 3 (R = OEt) was obtained in 50% yield, mp 119°C (from cyclohexane); m/e 176 (M⁺) and nmr δ (CDCl₃) 8.16 (s, CH), 8.04 (s, H₅ and H₆), 4.31 (q, OCH₂Me), 1.36 (t, OCH₂CH₃), J_{Et} = 7.0 Hz. The amidine 3 (R = NMe₂) reacted readily with free hydroxylamine at 0° to give compound 4 in 89% yield, mp 230-233°C (conversion into 5, R = H) (EtOH); m/e 196 (M⁺) and nmr δ (DMSO-d₆) 7.77 and 7.86 (d, H₅ and H₆), 7.60 (d, CH), 7.77 (broad s, NH₂), 10.05 (s, OH), 10.59 (d, NH), J_{5,6} = 2.4, J_{CHNH} = 9.5 Hz.



Several attempts to obtain 4-aminopteridine-3-oxide from the above precursors should be mentioned. The reaction between 2 ($R = R_1 = H$) and triethyl orthoformate afforded the N-oxide 5 ($R = H$) in 80% yield, mp 275-278°C (H_2O); m/e 163 (M^+) and nmr δ (D_2O , 95°) 8.52 and 8.62 (d, H_6 and H_7), 8.49 (s, H_2), $J_{6,7} = 2.0$ Hz. Alternatively, the amidine 3 ($R = NMe_2$) when heated in an alcoholic solution of hydroxylamine hydrochloride for 40 min afforded a mixture of several products. The main product, which was filtered from the mixture, obtained after evaporation of the solvent and treatment of the residue with water, was the N-oxide 5 ($R = H$) (80% yield) and from the filtrate compound 2 ($R = R_1 = H$) could be isolated in 11% yield. By tlc compound 1 and 2-amino-3-pyrazinecarboxamide could be detected in the filtrate.

4-Aminopteridine-3-oxide (5, $R = H$) could be reduced with aqueous solution of titanium trichloride at room temperature into 4-aminopteridine (6) in low yield (11%). The later compound is obtainable in high yield (87%) from the ethoxymethylene derivative 3 ($R = OEt$) and methanolic ammonia at room temperature after 24 h, mp 309-312°C (EtOH) (Lit. ¹⁵ gives mp 305°); m/e 147 (M^+) and nmr δ ($DMSO-d_6$) 8.14 (s, H_2), 8.36 and 8.63 (d, H_6 and H_7), 7.50 (broad s, NH_2), $J_{6,7} = 1.8$ Hz. In the same manner the amidine 3 ($R = NMe_2$) afforded at room temperature after 1 week 4-aminopteridine 6 in 58% yield.

4-Aminopteridine-3-oxide could be easily transformed in boling water after 7 h into the 4-hydroxylamino compound 7 ($R = H$) in 73% yield, mp over 310°C (from H_2O); m/e 163 (M^+) and nmr δ ($DMSO-d_6$, 130°) 7.33 (s, H_2), 7.96 and 8.10 (d, H_6 and H_7), $J_{6,7} = 2.3$ Hz. This transformation proceeds by ring opening at the pyrimidine ring followed by ring closure involving the former 4-amino function. This is substantiated by the presence of a small amount of compounds 1 and 2 ($R = R_1 = H$) in the filtrate. The hydroxylamino compound 7 ($R = H$) is acetylated with acetic anhydride to give the O-acetyl derivative (7, $R = COMe$) which is in turn obtained in 29% yield from the reaction between 2 ($R_1 = H$, $R = COMe$) and triethyl orthoformate after 6 h under reflux, mp 231-234°C (from H_2O); m/e 205 (M^+) and nmr δ ($DMSO-d_6$, 50°) 7.52 (s, H_2), 8.15 and 8.30 (d, H_6 and H_7), 2.08 (s, Me), 3.35 (broad s, NH), $J_{6,7} = 2.2$ Hz. The O-methyl derivative can be prepared in a similar way from 2 ($R_1 = H$, $R = Me$) in 54% yield, mp 263-266°C (from EtOH); m/e 177 (M^+).

The above described reaction paths are suitable also for the synthesis of 2-substituted derivatives as exemplified in the following cases. Acetylation of 1 afforded a mixture of the monoacetyl and diacetyl derivatives in a ratio of about 2:1 and recrystallization from ethyl acetate afforded the pure monoacetyl derivative, mp 147°C, m/e 162 (M^+). The diacetyl derivative had mp 70-71°C (from H_2O). Each of these compounds reacted with free hydroxylamine at room temperature to give after 80 min the corresponding monoacetylated amidoxime (2, $R_1 = COMe$, $R = H$) in about 80% yield, mp 159-161°C (from EtOH), m/e 195 (M^+). The later compound, when heated either in the presence of glacial acetic acid for 1 h under reflux or in the presence of polyphosphoric acid at 70-80°C for 1.5 h, afforded the pteridine 5 ($R = Me$) in 79% and 40% yield, respectively, mp 278-280°C; m/e 177 (M^+).

and nmr δ (D_2O , 78°) 8.19 and 8.32 (d, H_6 and H_7), 2.60 (s, Me), $J_{6,7} = 1.8$ Hz. However, if compound 1 was acetylated with a mixture of acetic anhydride and glacial acetic acid (1:1) for 3 h, 2-acetylaminopyrazine-3-carboxamide was obtained in 50% yield, mp 219 - 221° C (from EtOH); m/e 181 (M^+) (Lit. ¹⁶ gives mp 219° C). At its mp or when heated at 230 - 235° C the later compound is cyclized into the pteridinone 8 ($R = Me$) in 52% yield, mp over 200° C (from EtOH and N,N-dimethyl-formamide) (Lit. ¹⁷ gives mp over 200° C); nmr δ (DMSO- d_6) 8.66 and 8.75 (d, H_6 and H_7), 2.39 (s, Me), 3.3 (broad s, NH), $J_{6,7} = 2.1$ Hz. The analogous desmethyl analog 8 ($R = H$) could be prepared from the hydroxylamino compound 7 ($R = H$) by acid hydrolysis (hydrochloric acid, 1:1, 10 min at boiling temperature) in 28% yield, mp over 350° C (Lit. ¹⁵ gives mp over 350° C); nmr δ (DMSO- d_6 , 115°) 8.25 (s, H_2), 8.78 and 8.95 (d, H_6 and H_7), $J_{6,7} = 2.2$ Hz.

The described reactions represent a new and versatile approach for the preparation of some functionalized pteridines. Satisfactory analyses (C,H,N) were obtained for all compounds.

Acknowledgment: This work was partially supported by the Research Council of Slovenia.

REFERENCES

- 1 E.C. Taylor and A.McKillop, "The Chemistry of Cyclic Enaminonitriles and o-Aminonitriles", Interscience Publ., New York, 1970.
- 2 S.Polanc, B.Verček, B.Šek, B.Stanovnik, and M.Tišler, J.Org.Chem., 1974, 39, 2143.
- 3 B.Stanovnik and M.Tišler, Synthesis, 1974, 120.
- 4 J.Faganeli, S.Polanc, B.Stanovnik, and M.Tišler, Croat.Chem.Acta, 1976, 48, 161.
- 5 K.Babič, S.Molan, S.Polanc, B.Stanovnik, J.Stres-Bratoš, M.Tišler, and B.Verček, J.Heterocycl.Chem., 1976, 13, 487.
- 6 B.Jenko, B.Stanovnik, and M.Tišler, Synthesis, 1976, 833.
- 7 M.Debeljak-Šuštar, B.Stanovnik, M.Tišler, and Z.Zrimšek, J.Org.Chem., 1978, 43, 393.
- 8 B.Verček, B.Stanovnik, M.Tišler, and Z.Zrimšek, Org.Prep.Proced.Intern., 1978, 10, 293.
- 9 B.Verček, B.Stanovnik, and M.Tišler, Heterocycles, 1978, 11, 313.
- 10 B.Verček, I.Leban, B.Stanovnik, and M.Tišler, J.Org.Chem., 1979, 44, 1695.
- 11 M.Tišler, B.Stanovnik, and B.Verček, Vestn.Slov.Kem.Društva, 1980, 27, 65
- 12 M.Tišler and B.Stanovnik, J.Chem.Soc., Chem.Commun., 1980, 313.
- 13 This compound is mentioned in the literature ¹⁴, yet only as an intermediate and it was not isolated in pure form.
- 14 A.Albert and K.Ohta, J.Chem.Soc.C, 1970, 1540.
- 15 A.Albert, D.J. Brown, and G.Cheeseman, J.Chem.Soc., 1951, 474.
- 16 R.C.Ellingson, R.L.Henry, and F.G.McDonald, J.Amer.Chem.Soc., 1945, 67, 1711.
- 17 A.Albert and C.Howell, J.Chem.Soc., 1962, 1591.

Received, 15th July, 1980