

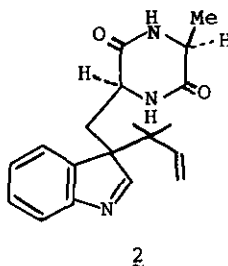
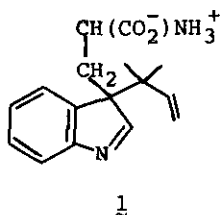
THE PREPARATION AND REARRANGEMENT OF 1-PRENYLINDOLES AND  
3-PRENYLINDOLENINES

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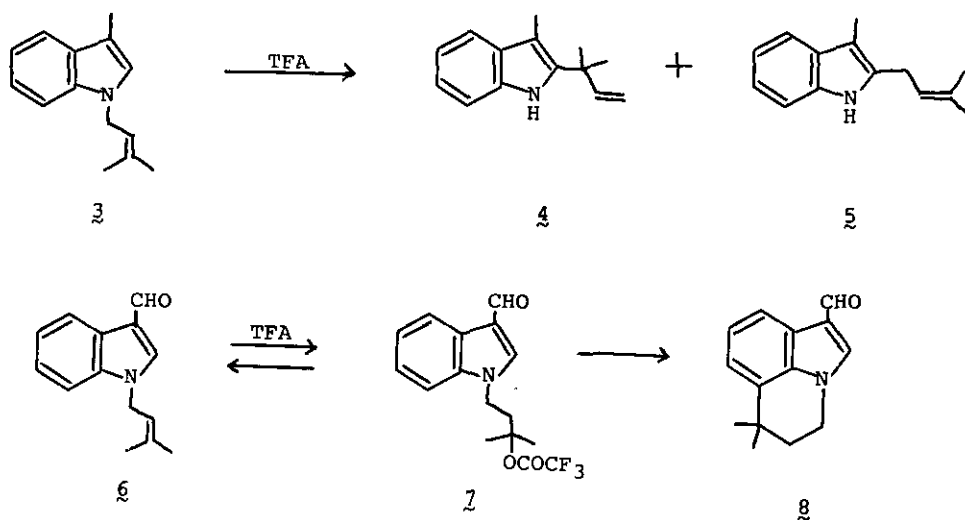
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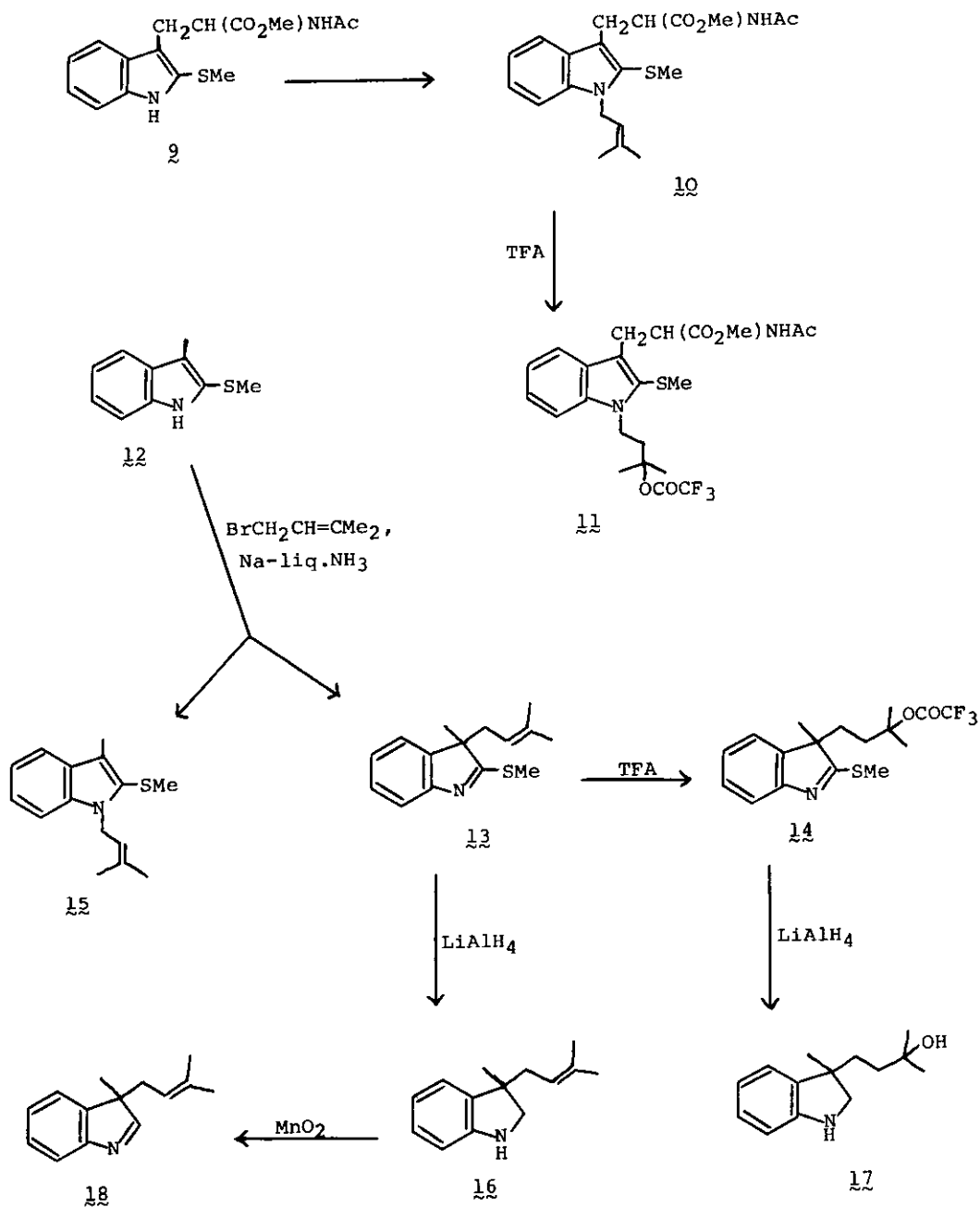
**Abstract** - The preparation of 3-alkyl-3-prenylindolenines was explored by rearrangement of 1-prenylindoles and by allylation of 2-(methylthio)indoles. Reaction of the 3-formylindole (5) with trifluoroacetic acid gave first the addition product (7) and then the pyridinoindole (8); the esters (11) and (14) were formed similarly from the tryptophan derivative (10) and the 2-(methylthio)indolenine (13), respectively. Successive reaction of the indolenine (13) with lithium aluminium hydride and manganese dioxide gave 3-(3,3-dimethylallyl)-3-methylindolenine (18) in good yield.

The reversed prenyl group present in the 2-position of the indole ring of echinulin and other fungal diketopiperazines may be introduced *in vivo* by direct alkylation at C-2 of an indole precursor.<sup>1</sup> An alternative pathway involves the 3-alkyl-3-(1,1-dimethylallyl)indolenines (1) and/or (2). In connection with our biosynthetic studies<sup>2</sup> we sought a simple synthesis of these indolenines and of indoles substituted at the 2-position by a dimethylallyl group.



One approach to the synthetic problem stemmed from the work of Casnati and co-workers<sup>3</sup> who studied acid-catalysed reactions of 1-allylindoles. We have confirmed that the rearrangement products (96% yield) obtained from 1-(3,3-dimethylallyl)-3-methylindole (**3**) in trifluoroacetic acid (TFA) consist of the (1,1-dimethylallyl)-indole (**4**) and the (3,3-dimethylallyl)-indole (**5**) in an approximate ratio of 1:1 (NMR); the compounds were separated by preparative t.l.c. on silica gel. In exploring the scope of this reaction, it was found that 1-(3,3-dimethylallyl)-3-formylindole (**6**) in TFA at 33°C gave the addition product (**7**) [IR (neat)  $\nu_{\text{max}}$ . 1650 (CHO) and 1775  $\text{cm}^{-1}$  ( $\text{CO}_2\text{CF}_3$ ); NMR ( $\text{CDCl}_3$ )  $\delta$  9.95 (1H, s,  $\text{CHO}$ ), 8.3 (1H, m, C-4H), 7.66 (1H, s, C-2H), ~ 7.3 (3H, m, arom.), 4.28 (2H) and 2.38 (2H), AA'BB' pattern,  $-\text{CH}_2\text{CH}_2-$  and 1.67 (6H, s,  $\text{CMe}_2$ ); MS  $m/e$  327 ( $\text{M}^+$ , 56%), 213 ( $\text{M}-\text{CF}_3\text{CO}_2\text{H}$ , 20%), 158 ( $\text{M}-\text{CH}_2\text{C}(\text{OCOCF}_3)\text{Me}_2$ , 100%) and 130 ( $\text{M}-\text{CH}_2\text{C}(\text{OCOCF}_3)\text{Me}_2$ ,  $-\text{CHO}$ , 11%], but in refluxing TFA the product (**8**) of an intramolecular Friedel-Crafts reaction was formed in quantitative yield from an equilibrium mixture of the prenyl derivative (**6**) and the ester (**7**); the structure of the pyridinoindole (**8**) was indicated by IR absorption at 1650  $\text{cm}^{-1}$  (CHO) by the mass spectrum [ $m/e$  213 ( $\text{M}^+$ , 87%) and 198 ( $\text{M}-\text{Me}$ , 100%) and by the NMR spectrum in  $\text{CDCl}_3$  [ $\delta$  9.89 (1H, s,  $\text{CHO}$ ), 8.0 (1H, dd,  $J$  3.5 and 10 Hz, C-4H), 7.56 (1H, s, C-2H), 7.30-7.07 (2H, m, C-5H and C-6H), 4.16 (2H) and 1.96 (2H), AA'BB' pattern,  $-\text{CH}_2\text{CH}_2-$  and 1.35 (6H, s,  $\text{CMe}_2$ )]. Similar cyclisation reactions were reported recently by Sammes and co-workers.<sup>4</sup>





Scheme

We have also studied the synthesis of 3-prenylindolenines lacking a 2-substituent. These compounds are apparently readily converted into 2-prenylindoles<sup>5</sup> and this acid-induced shift is presumably involved in the conversion of the 1-prenylindole (3) into 2-prenylindoles (4) and (5).<sup>6</sup> In the preparation of 1-(3,3-dimethylallyl)-3-methylindole (3) from 3-methylindole, 3,3-dimethylallyl bromide and NaH in DMF, the indolenine (18) was detected as a minor product in the crude reaction mixture [NMR (CDCl<sub>3</sub>)  $\delta$  7.97 (1H, s,  $-\text{CH}=\text{N}-$ ), 7.7-6.9 (4H, m, arom.), 4.93 (1H, t, J 7.5 Hz,  $-\text{CH}=\text{C}$ ), 2.38 (2H, d, J 7.5 Hz,  $-\text{CH}_2\text{CH}=\text{C}$ ), 1.60 and 1.53 (6H, 2 x s,  $\text{Me}_2\text{C}=\text{}$ ) and 1.30 (3H, s, Me)]. Attempts to isolate this indolenine by chromatography on silica gel led to its quantitative isomerisation to the 2-(3,3-dimethylallyl) indole (5).

The work of Bycroft and Landon<sup>7</sup> on the reaction of 3-methyl-2-(methylthio)-indole with 3,3-dimethylallyl bromide seemed to offer the attractive possibility of employing a methylthio substituent as a blocking group to stabilise 3-prenylindolenines.

We decided, therefore, to prepare the (2-methylthio)indoles (9) and (12) and investigate their reactions (Scheme).

Treatment of L- N-acetyl tryptophan methyl ester with methane sulphenyl chloride in acetic acid gave the 2-methylthio derivative (9) in 58% yield, which was converted with 3,3-dimethylallyl bromide in Na-liquid ammonia into the 1-(3,3-dimethylallyl)indole (10) (32% yield), m.p. 124°C. Under the same conditions, 3-methyl-2-(methylthio)indole (12) afforded two products, the 1-(3,3-dimethylallyl)indole (15) (19% yield) [NMR (CDCl<sub>3</sub>)  $\delta$  7.6-7.0 (4H, m, arom.), 5.20 (1H) and 4.85 (2H), AB<sub>2</sub> pattern ( $J_{\text{AB}}$  ~ 7Hz),  $\text{CH}_2\text{CH}=\text{CMe}_2$ , 2.40 (3H) and 2.15 (3H), 2 x s,  $\text{SCH}_3$  and indolyl  $\text{CH}_3$  (or v.v.), 1.85 (3H) and 1.65 (3H), 2 x s,  $\text{C}=\text{C}(\text{CH}_3)_2$ ], and the 3-(3,3-dimethylallyl)indolenine<sup>7</sup> (13) (62% yield) [NMR (CDCl<sub>3</sub>)  $\delta$  7.5-7.0 (4H, m, arom.), 4.68 (1H, t, J 7.5Hz,  $\text{CH}_2\text{CH}$ ), 2.57 (3H, s, SMe), 2.4 (2H, m,  $\text{CH}_2\text{CH}$ ), 1.47 (3H) and 1.41 (3H), broad singlets,  $\text{C}=\text{CMe}_2$ , 1.28 (3H, s, indolyl Me)]. The tryptophan derivatives (10) with trifluoroacetic acid at 50°C gave in low yield the addition product (11) as an oil [NMR (CDCl<sub>3</sub>)  $\delta$  7.37-7.0 (4H, m, arom.), 6.13 (1H, d,  $\text{NHCOMe}$ ), 4.99-4.77 (1H, m,  $\text{CH}_2\text{CH}$ ), 4.25-4.13 (2H, m,  $\text{>NCH}_2\text{CH}_2$ ), 3.70 (3H, s,  $\text{CO}_2\text{Me}$ ), 3.41 (2H, d,  $\text{CH}_2\text{CH}$ ), 2.27 (3H, s, SMe), 2.10-1.92 (5H, m,  $\text{NHCOCH}_3$  and  $\text{>NCH}_2\text{CH}_2$  and 1.39 (6H, s,  $\text{CMe}_2$ )]. In TFA at 30°C the 3-(3,3-dimethylallyl) indolenine (13) also gave an addition product (14) [NMR

(CDCl<sub>3</sub>)  $\delta$  7.4-7.1 (4H, m, arom.), 2.60 (3H, s, SMe), 2.1-1.6 (2H, m) and 1.3-0.8 (2H, m) (2 X CH<sub>2</sub>) and 1.39 (3H), 1.35 (3H), 1.32 (3H), 3 X s, 3 X Me; MS m/e 359 (M<sup>+</sup>, 98%), 245 (M-CF<sub>3</sub>CO<sub>2</sub>H, 33%) and 177 (M-C<sub>7</sub>H<sub>9</sub>O<sub>2</sub>F<sub>3</sub>, 100%)], which was converted by lithium aluminium hydride into the indoline (17).

The 2-(methylthio)indolenine (13) was reduced quantitatively with lithium aluminium hydride in boiling ether to the indoline (16) [NMR (CDCl<sub>3</sub>)  $\delta$  7.10-6.45 (4H, m, arom.), 5.15 (1H, t, J 7.5Hz, -CH=), 3.13 and 3.31 (2H, m, J 9Hz, -CH<sub>2</sub>NH-) 2.25 (2H, d, J 7.5Hz, -CH<sub>2</sub>CH=), 1.69, 1.53 (6H, 2 X s, Me<sub>2</sub>C=) and 1.25 (3H, s, Me); MS m/e 201 (M<sup>+</sup>, 12.5%), 132 (M-C<sub>5</sub>H<sub>9</sub>, 100%), 131 (10%) and 130 (15%)]. Reoxidation with MnO<sub>2</sub> in boiling benzene or toluene then furnished 3-(3,3-dimethylallyl)-3-methylindolenine (18) (60-80% yield). This sequence which avoids acidic conditions may prove to be useful in the preparation of highly functionalised and labile 3,3-dialkyl-indolenines since MnO<sub>2</sub> oxidation may be performed as the terminal step in the synthesis.

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#### REFERENCES

1. G. Casnati, A. Dossena, and A. Pochini, Tetrahedron Letters, 1972, 5277.
2. M. F. Grundon, M. R. Hamblin, D. M. Harrison and (in part) J. N. D. Logue, (Mrs.) M. Maguire, and J. A. McGrath, J.C.S. Perkin I, 1980, 1294.
3. G. Casnati and A. Pochini, Chem. Comm., 1970, 1328; G. Casnati, R. Marchelli, and A. Pochini, J.C.S. Perkin I, 1974, 754.
4. P. G. Sammes and A. C. Weedon, J.C.S. Perkin I, 1979, 3053.
5. A. H. Jackson and A. E. Smith, Tetrahedron, 1965, 21, 989.
6. M. Schmid, H. J. Hansen and H. Schmid, Helv. Chem. Acta, 1973, 56, 105.
7. B. W. Bycroft and W. Landon, Chem. Comm., 1970, 967.

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