

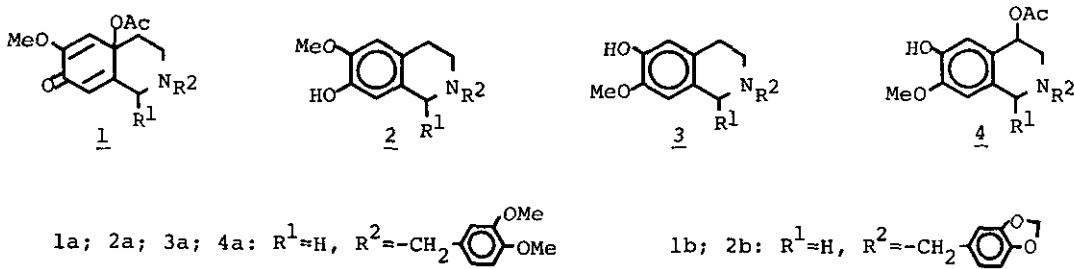
A NOVEL SYNTHESIS OF DIBENZO[*c,f*]-1-AZABICYCLO[3.3.1]NONANES<sup>†</sup>

Hiroshi Hara, Osamu Hoshino, and Bunsuke Umezawa\*

Faculty of Pharmaceutical Sciences, Science University of Tokyo,  
Shinjuku-ku, Tokyo, 162, Japan

Abstract—Treatment of the *N*-benzylated *p*-quinol acetates (1a and 1b) with trifluoroacetic acid gave ( $\pm$ )-3-hydroxydibenzoazabicyclononanes (5a and 5c) in good yields. On the other hand, lead tetraacetate oxidation of 2-benzyl-6-hydroxy-7-methoxy-1,2,3,4-tetrahydroisoquinoline (3a) gave the *o*-quinol acetate (6), which rearranged into the 4-acetoxy-6-hydroxy derivative (4a) at room temperature. Acid treatment of the 4-acetate (4a) afforded a cyclization product (5e) having the same skeleton as that of 5a.

The *p*-quinol acetate (1), easily prepared from 7-hydroxy-6-methoxy-1,2,3,4-tetrahydroisoquinoline (2) by lead tetraacetate (LTA) oxidation, is a key compound for the synthesis of isoquinoline alkaloids<sup>1)</sup>, aporphine<sup>2)</sup>, homoaporphine, homoproaporphine, and homomorphinandienone.<sup>3)</sup> On the other hand, LTA oxidation of 6-hydroxy-7-methoxy congener (3) gives the corresponding 4-acetoxy derivative (4)<sup>4)</sup>, which undergoes acid-catalysed cyclization to isopavine alkaloids.<sup>5)</sup> Now we wish to report the synthesis of dibenzo[*c,f*]-1-azabicyclo[3.3.1]nonanes, through two routes via the *N*-benzylated *p*-quinol acetate (1) and 4-acetoxy derivative (4).



<sup>†</sup> Dedicated to Prof. T. Kametani on the occasion of his retirement.

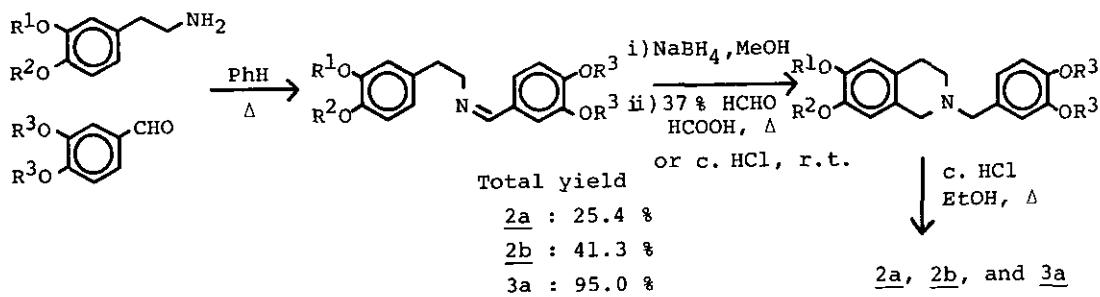
The starting phenols (2a,b and 3a) were prepared from benzaldehyde and  $\beta$ -phenethylamines according to Kametani's method<sup>6)</sup> (condensation, reduction, Mannich's reaction, and debenzylation) as shown in Scheme I.

LTA (1.2 eq.) oxidation of 2a (100 mg) in acetic acid (AcOH) (1 ml) gave the  $\text{p}$ -quinol acetate (1a) [IR ( $\text{cm}^{-1}$ ): 1735 (OAc), 1670, 1650, 1625 (dienone)] quantitatively, which was treated with trifluoroacetic acid ( $\text{CF}_3\text{COOH}$ ) (1 ml) in methylene chloride ( $\text{CH}_2\text{Cl}_2$ ) (10 ml) at room temperature for 1 hr to give ( $\pm$ )-3-hydroxy-2,9,10-trimethoxydibenzo[*c,f*]-1-azabicyclo[3.3.1]nonane (5a)<sup>7)</sup>, m.p. 213-215°, in 38% yield, which showed four singlets due to aromatic protons ( $\delta$  6.39, 6.41, 6.56, 6.58) on its nuclear magnetic resonance (NMR) spectrum and was methylated with diazomethane to give a tetramethyl ether (5b). NMR spectra of both tetramethyl ether (5b) and the authentic sample<sup>8)</sup> were completely superimposable.

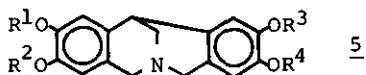
Similarly, oxidation and the subsequent acid treatment of 2b afforded ( $\pm$ )-3-hydroxy-2-methoxy-9,10-methylenedioxydibenzoazabicyclononane (5c), m.p. 203.5-205.5° (dec.), in 50% yield, the structure of which was confirmed by its conversion to the known methyl ether (5d).<sup>9)</sup>

Oxidation [LTA (1.2 eq.)] of 3a in  $\text{CH}_2\text{Cl}_2$  and careful work-up<sup>10)</sup> gave the oily  $\text{p}$ -quinol acetate (6) [IR ( $\text{cm}^{-1}$ ): 1740 (OAc), 1685 (C=O); NMR ( $\delta$ ): 2.03 (OCOMe), 3.36 (aliph. OMe), 3.77 (2 x arom. OMe), 5.73, 5.79 (each 1H, olefin. H)], which was allowed to stand overnight to give a diastereomeric mixture of the 4-acetoxy derivatives (4a)<sup>10)</sup> [IR ( $\text{cm}^{-1}$ ): 3550 (OH), 1720 (OAc); NMR ( $\delta$ ): 1.95, 2.02 (3H, each s, OCOMe (1 : 1.3))] as an oil. Without purification, the 4-acetoxy derivatives (4a) were treated with  $\text{CF}_3\text{COOH}$  at room temperature for 1 hr to afford an amorphous ( $\pm$ )-2-hydroxy-3,9,10-trimethoxydibenzoazabicyclononane (5e) (HCl salt: m.p. 238-240°) in 80% yield from 3a. The structure of 5e was verified by comparison of its methyl ether with the authentic sample (5b)<sup>8)</sup> in all respects.

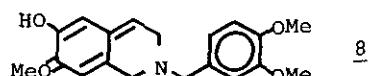
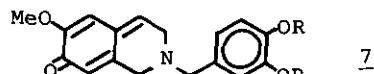
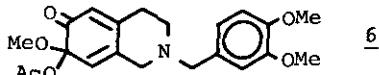
Thus a novel synthesis of ( $\pm$ )-dibenzo[*c,f*]-1-azabicyclo[3.3.1]nonanes (5) was accomplished from either 7- or 6-hydroxy-N-benzyltetrahydroisoquinoline (2a,b or 3a) via the intermediacy of either the  $\text{p}$ -quinol acetate (1a,b) or the 4-acetoxy derivatives (4a) presumably by the following reaction pathway; deacetoxylation of the former (1a,b) or the latter (4a) with acid would generate Michael-type acceptor, a  $\text{p}$ -quinone methide (7) or a cation (8), which would then immediately react together in a manner of intramolecular conjugate addition to form the products.



Scheme I



	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>
<u>5a</u>	Me	H	Me	Me
<u>5b</u>	Me	Me	Me	Me
<u>5c</u>	Me	H	-CH <sub>2</sub> -	
<u>5d</u>	Me	Me	-CH <sub>2</sub> -	
<u>5e</u>	H	Me	Me	Me



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7. All new compounds gave satisfactory analytical data. NMR and IR spectra were taken in  $CDCl_3$  and  $CHCl_3$  solution, respectively. Preparative t.l.c. was run on silica gel HF<sub>254</sub> (Merck).
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