PREPARATION OF 1,2-DIHYDRO-1,2-DIPHENYL-1 $^{\lambda}$,2 $^{\lambda}$ -DIPHOSPHORIN 1,2-DISULFIDE

Takayuki Kawashima, Michio Shimamura, and Naoki Inamoto*

Department of Chemistry, Faculty of Science, The University of Tokyo, Hongo, Tokyo 113, Japan

<u>Abstract</u>: 1,2-Dihydro-1,2-diphenyl-1 $^{\lambda}$,2 $^{\lambda}$ -diphosphorin 1,2-disulfide was synthesized from the corresponding 1,2,3,6-tetrahydro derivative by lithiation, bromination with bromotrichloromethane and dehydrobromination successively. Direct bromination of 1,2,3,6-tetrahydro derivative resulted in the P-P bond cleavage.

There is no report on 1,2-dihydrophosphorin or $1\underline{H}$ -phosphorin derivative containing a hetero atom at α -position except 2-mesity1-1,2 $^{\lambda}$ -oxaphosphorin 2-oxide¹ which was employed for study on generation and reactivity of mesity1metaphosphonate. In this paper, we would like to report a novel synthesis of the titled compound which could be derived to a precursor of phenylphosphinothioylidene dimer postulated as a reaction intermediate in our laboratory.
1,2-Dipheny1-1,2-dipotassiodiphosphane (1) was allowed to react with cis-1,4-di-chloro-2-butene in tetrahydrofuran (THF) under nitrogen atmosphere, followed by sulfurization to afford 1,2,3,6-tetrahydro-1,2-dipheny1-1 $^{\lambda}$,2 $^{\lambda}$ -diphosphorin 1,2-disulfide (2) in 53.2% yield.

<u>Trans</u> and <u>cis</u> isomers were separated by dry column chromatography on silica gel in a ratio of 8:1. Their structure determinations were done by their behavior on silica gel and spectral analyses, ⁵ especially ¹H-NMR.

 $\begin{array}{l} \frac{\text{Trans-2}}{\text{Cm}^{-1}} \text{ (less polar isomer): mp } 198.2\text{-}199.5^{\circ}\text{C (acetone); IR (KBr disc): } \nu_{\text{C=C}} 1650 \\ \text{cm}^{-1}; \\ \stackrel{\text{i}}{\text{H-NMR}} \text{ (CD}_2\text{Cl}_2\text{): } \delta \text{ 2.69-3.45 (m,2H), } 3.70\text{-}4.50 \text{ (m,2H), } 5.95\text{-}6.34 \text{ (m,2H), } \\ 7.16\text{-}7.73 \text{ (m,6H), and } 7.73\text{-}8.23 \text{ (m,4H); } \\ ^{31}\text{P-NMR} \text{ (using } 85\% \text{ H}_3\text{PO}_4 \text{ as an external standard) (CDCl}_3\text{): } \delta_{\text{p}} \text{ 25.31 ppm; High resolution MS (70 eV) Found: m/e } 334.0171. \\ \text{Calcd for C}_{16}\text{H}_{16}\text{P}_2\text{S}_2\text{: } 334.0169. \text{ Anal. Found: C, 57.76; H, 5.05; S, 18.77\%. Calcd } \\ \text{for C}_{16}\text{H}_{16}\text{P}_2\text{S}_2\text{: C, 57.47; H, 4.82; S, 19.18\%.} \end{array}$

Cis-2 (more polar one): mp 183-185°C (decomp)(acetone); IR (KBr disc): $v_{C=C}$ 1650 cm⁻¹; $^{1}\text{H-NMR}$ (CD₂Cl₂): δ 3.02-3.71 (m,4H), 5.80-6.40 (m,2H), and 7.00-8.79 (m,10H); $^{31}\text{P-NMR}$ (CDCl₃): δ_{p} 28.41 ppm; High resolution MS (70 eV) Found: m/e 334.0170. Calcd: 334.0169. Anal. Found: C, 57.42; H, 4.92; S, 19.07%.

Reaction of <u>trans-2</u> with bromine of pyridinium tribromide in benzene gave no corresponding 4,5-dibromide in contrast to $1,2^{\lambda}$ -oxaphosphorin system, 1 but afforded bisphosphinothioy1 dibromide (3), which was recyclized with magnesium to give a 1:1 mixture of <u>trans-</u> and <u>cis-2</u> in 31% yield.

$$\underbrace{\frac{\text{S S}}{\text{Ph-P-P-Ph}}}_{\text{Ph-P-P-Ph}} \underbrace{\frac{\text{Br}_2 \text{ or } \bigcirc \text{N}^+ \text{H Br}_3^-}{\text{benzene}}}_{\text{benzene}} \underbrace{\frac{\text{S}}{\text{Ph-Br}}}_{\text{Ph-Br}} \underbrace{\frac{\text{S}}{\text{Br}_{PPh}}}_{\text{THF}} \underbrace{\frac{\text{trans-}}{\text{cis-2}}}_{\text{THF}}$$

<u>Trans-2</u> was treated with <u>n</u>-butyllithium or <u>sec</u>-butyllithium at -100° C in THF and the resulting lithio derivative was allowed to react with bromotrichloromethane and then the reaction mixture was quenched by conc HCl at -100° C. After rapid silica-gel chromatography with benzene, recrystallization from accetone gave the corresponding 3-bromide (4) in 24.4% yield, along with unchanged 2 in 11% recovery.

 $\begin{array}{l} \underline{\text{Trans-4:}} \text{ mp 165-166}^{\circ}\text{C (acetone); IR (KBr disc): } \nu_{\text{C=C}} \text{ 1640 cm}^{-1}; \quad ^{1}\text{H-NMR (CDCl}_{3}): \delta \\ 2.72-3.31 \text{ (m,2H), } 3.92-4.15 \text{ (m,1H), } 5.40-5.72 \text{ (m,1H), } 5.80-7.30 \text{ (m,2H), } \text{and } 7.35-8.10 \text{ (m,10H); } ^{31}\text{P-NMR (CDCl}_{3}): \delta_{\text{p}} \text{ 16.39 (d,P}_{1}) \text{ and } 17.56 \text{ (d,P}_{2}), J_{\text{P_1},\text{P_2}} = 36.62 \text{ Hz; } \\ \text{High resolution MS Found: m/e 411.9275. Calcd for C_{16}^{\text{H}}_{15}$^{\text{BrP}}_{2}$^{\text{S}}_{2}: 411.9275. Anal. } \\ \text{Found: C, 46.75; H, 3.65; Br, 19.78$. Calcd for C_{16}^{\text{H}}_{15}$^{\text{BrP}}_{2}$^{\text{S}}_{2}: C, 46.50; H, 3.67; } \\ \text{Br, 19.34$$.} \end{array}$

From $\underline{\text{cis-2}}$ was obtained $\underline{\text{cis-4}}$ similarly in 4.8% yield; High resolution MS Found: m/e 411.9259. Cis-4 was more polar than trans-4.

Dehydrobromination of <u>trans-4</u> by triethylamine in refluxing acetonitrile afforded 1,2-dihydro-1,2-diphenyl- 1^{λ} , 2^{λ} -diphosphorin 1,2-disulfide (5) in 46.8% yield as yellow crystals. Reduction reaction also occurred competitively to give <u>trans-2</u> in 14% yield. This undesirable reaction was predominant when the bulky 1,8-di-azabicyclo[5.4.0] undec-7-ene instead of triethylamine was used.

Trans-5: mp 168-170°C (acetone); IR (KBr disc): $\nu_{C=C}$ 1639 cm⁻¹; ¹H-NMR (CDC1₃): δ 6.38-7.37 (m,4H), 7.37-7.72 (m,6H), and 7.72-8.11 (m,4H); ³¹P-NMR (CDC1₃): δ_p 17.16 ppm; High resolution MS Found: m/e 331.9975. Calcd for $C_{16}H_{14}P_2S_2$: 332.0010. Analytically pure sample has not been obtained yet.

Reaction of <u>trans-5</u> with tetracyanoethylene afforded elemental sulfur in 50% yield. Reactions of 5 with other dienophiles are being investigated now, and the results will be reported in a future.

Acknowledgments: This work was supported by Grant-in-Aid (No. 474224) from the Ministry of Education, Science and Culture, which is gratefully acknowledged.

REFERENCES AND NOTES

- 1. I. Sigal and L. Loew, J. Am. Chem. Soc., 1978, 100, 6394.
- 2. S. Nakayama, M. Yoshifuji, R. Okazaki, and N. Inamoto, <u>Bull. Chem. Soc. Jpn.</u>, 1975, <u>48</u>, 546.
- K. Issleib and K. Krech, Chem. Ber., 1966, 99, 1310. Tetraphenylcyclotetraphosphane was prepared as described in the following paper: M. Eppstein and F. S. Seichter, J. Am. Chem. Soc., 1963, 85, 2462.
- 4. J. M. Bobbitt, L. H. Amundsen, and R. I. Steiner, <u>J. Org. Chem.</u>, 1959, <u>25</u>, 2230.
- 5. 1 H-NMR and 1 H noise-decoupled 31 P-NMR spectra were measured with Varian EM 390 and JEOL FX 900 spectrometers, respectively. High resolution mass spectra

were obtained with a JEOL D 300 mass spectrometer.

The half-chair conformation is more favored than the boat one for this tetrahydrodiphosphorin system, because all substituents on phosphorus atoms in the latter must be eclipsed about the P-P bond. In the case of $\underline{\text{trans-2}}$, the half-chair conformation with phenyl groups at the equatorial positions and sulfur atoms at the axial positions seems more stable than the other one, since phenyl group is bulkier than sulfur atom. On the other hand, if phenyl group on P_1 atom would occupy the equatorial position in the $\underline{\text{cis-2}}$, phenyl group on P_2 atom must take the axial position, so that two half-chair conformations are energetically equivalent each other. Therefore, the ring inversion results in some average for signals assignable to unequivalent protons of methylene groups adjacent to phosphorus atoms for $\underline{\text{cis-2}}$, but does not for $\underline{\text{trans-2}}$.

6. The ³¹P-NMR spectrum of 4 shows to be a single compound. Similarity of signals of methylene and methine protons attached to phosphorus atoms to those of trans-2 and the result that undesirable reduction reaction of 4 gave only trans-2 indicate that stereochemistry of diphosphane moiety is the same as that of trans-2. Since lithiation and bromination are expected to occur from the less hindered side, the bromo atom in 4 seems to be in trans configuration to phenyl group, but the actual stereochemistry of bromo atom could not be determined from the spectral data for lack of the data in analogous compounds.

Received, 4th September, 1981