A SYNTHESIS OF 3-HYDROXYAPORPHINE AND HOMOAPORPHINE

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<u>Abstract</u> — 3-Hydroxy-2,9,10-trimethoxyaporphine ($\underline{8}$) and 3-hydroxy-2,10,11,12-tetramethoxyhomoaporphine ($\underline{9}$) were synthesized via the corresponding o-quinol acetates ($\underline{13}$ and $\underline{14}$).

A facile synthesis of 1-hydroxyaporphine $(\underline{1})$ and homoaporphine $(\underline{2})$ via the corresponding p-quinol acetates $(\underline{3} \text{ and } \underline{4})$ have been already established. Recently, we reported that lead tetraacetate oxidation in $\operatorname{CH_2Cl_2}$ of 5-hydroxy-6-methoxy-2-methy1-1,2,3,4-tetrahydroisoquinoline $(\underline{5})$ has given the o-quinol acetate $(\underline{6})$, which has been treated with AcOH to afford the 8-acetoxy derivative $(\underline{7})$, showing that the 8-position is susceptible to the nucleophilic attack and the reaction should be applicable to ring closure. Here we wish to report a novel synthesis of 3-hydroxyaporphine $(\underline{9})$.

Three starting phenols ($\underline{10}$, $\underline{11}$, and $\underline{12}$) were prepared by a conventional method. 1,3 Lead tetraacetate oxidation of $\underline{10}$ in $\mathrm{CH_2Cl_2}$ at 0°C for 2 min and careful work-up 2 (<30°C) gave the o-quinol acetate ($\underline{13}$) [IR: 1730 cm $^{-1}$ (OAc) and 1670 cm $^{-1}$ (dienone)], quantitatively. Without purification, $\underline{13}$ was treated with $\mathrm{CF_3CO_2H}$ in $\mathrm{CH_2Cl_2}$ at room temperature for 2 h to give an oil, which was purified on preparative TLC affording 3-hydroxy-2,9,10-trimethoxyaporphine ($\underline{8}$), mp 213-214°C (lit. 4 214-215°C), in 73% yield (from $\underline{10}$). Three aromatic protons at δ 6.82, 7.07, and 7.16 on NMR spectrum reasonably proved the aporphine structure for $\underline{8}$.

Similarly, oxidation of $\underline{11}$ gave the o-quinol acetate ($\underline{14}$), acid treatment of which furnished 3-hydroxy-2,10,11,12-tetramethoxyhomoaporphine ($\underline{9}$), mp 240-245°C (methiodide of the acetate: mp 204-205°C), in 90% yield (from $\underline{11}$). NMR spectrum of $\underline{9}$ showed signals of four methoxyl groups [δ 3.92, 4.03 (6H), 4.09] and two aromatic protons (δ 6.85, 7.13). Evidently, one aromatic proton at δ 7.13 assignable to the

 $\frac{1}{n}$: n=1, R=H

2: n=2, R=OMe

3 : n=1, R=H

 $\frac{4}{\cdot}$: n=2, R=OMe

8 : n=1, R=H

9 : n=2, R=OMe

18 : n=0, R=H

10 : n=1, R=H

11: n=2, R=OMe

12 : n=0, R=H

13: n=1, R=H

14 : n=2, R=OMe

<u>17</u> : n=0, R=H

$$\begin{array}{c} \text{MeO} \\ \text{HO} \\ \text{NMe} \\ \text{OMe} \\ \underline{16} \\ \end{array}$$

C-1 position was deshielded by the C-12 methoxyl group.

Analoguously, oxidation of the 1-aryltetrahydro-5-isoquinolinol ($\underline{12}$) gave the oquinol acetate ($\underline{17}$), cyclization of which however was unsuccessful. Namely, $\underline{17}$ was treated with $\mathrm{CF_3CO_2H}$ in $\mathrm{CH_2Cl_2}$ to give rise to no cyclized product ($\underline{18}$) but the p-quinone ($\underline{19}$) in 66% yield (from $\underline{12}$). Structure of $\underline{19}$ was determined both spectroscopically and chemically. One olefinic proton [δ 5.65 (NMR)] and two absorption bands [1645, $1600\,\mathrm{cm}^{-1}$ (IR)] pointed out clearly the p-quinone structure. Reduction of $\underline{19}$ with Zn in $\mathrm{Ac_2O^2}$ gave the diacetate ($\underline{20}$) (methiodide: mp 238-240°C), spectral data of which showed aromatic acetoxyl function ($1750\,\mathrm{cm}^{-1}$) on IR and four aromatic protons and two singlets of acetoxyl groups on NMR. Signal due to the C-8 acetoxyl protons was shifted to higher field owing to the anisotropic effect caused by the C-1 benzene ring.

Mechanism of the aporphine cyclization and the formation of p-quinone could be illustrated as shown in the Scheme. Especially, in the case of $\underline{17}$, steric strain imposed by the five membered ring in $\underline{18}$ probably prohibited the intramolecular arylation, allowing the nucleophilic attack by CF₃COO $^-$. Hydrolysis of trifluoroacetate and subsequent oxidation gave the quinone $\underline{19}$.

Scheme

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- All new compounds gave reasonable spectroscopic data and satisfactory combusion analytical values.
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- 5. The sole formation of $\underline{9}$ from $\underline{14}$ was interesting when compared with the result that the p-quinol acetate $(\underline{4})^{1c}$ gave three products, the 1-hydroxyhomoaporphine $[(\pm)$ -kreysigine] $(\underline{2})$, the homomorphinandienone $[(\pm)$ -O-methylandrocymbine) $(\underline{15})$, and the homoproaporphine $(\underline{16})$ under similar conditions.

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