

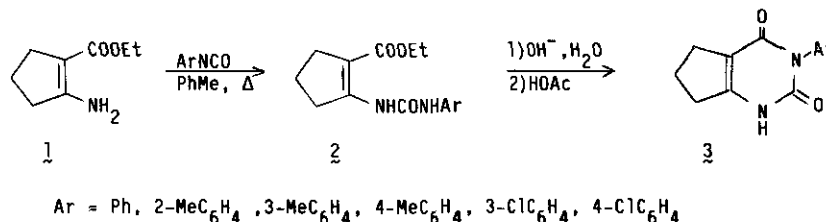
## REACTIONS OF ISOCYANATES WITH ETHYL 2-AMINO-1-CYCLOPENTENE-1-CARBOXYLATE

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**Abstract** - Aromatic isocyanates react with ethyl 2-amino-1-cyclopentene-1-carboxylate to yield the corresponding ureas, which cyclize by base to 3-substituted 6,7-dihydro-1H-cyclopentapyrimidine-2,4(3H,5H)-diones.

As part of a study of cyclization reactions of 2-aminoesters and 2-aminonitriles with isocyanates,<sup>2-4</sup> we investigated briefly such reactions of ethyl 2-amino-1-cyclopentene-1-carboxylate (**1**). Only isolated examples of reactions of **1** with isocyanates have been reported in the literature.<sup>5-7</sup> We have found that when equimolar quantities of **1** and an aromatic isocyanate are refluxed in toluene for 16-20 hours, the corresponding urea (**2**) is obtained in 45-78% yield.



In earlier work, it was established that analogous ureas derived from methyl 2-aminobenzoate cyclize to 3-substituted 2,4(1H,3H)-quinazolinediones by the action of dilute acid,<sup>8-11</sup> or base,<sup>4</sup> but to N-substituted 2-amino-4H,-3,1-benzoxazin-4-ones by treatment with concentrated sulfuric acid.<sup>4</sup> In the present case, ureas **2** have been found to resist cyclization upon refluxing with hydrochloric acid in ethanol, but to be converted into the expected 3-aryl-6,7-dihydro-1H-cyclopentapyrimidine-2,4(3H,5H)-diones (**3**) in 61-72% yield by treatment with hot, dilute alkali. Earlier preparations of compounds **3** were generally based on the acid catalyzed reactions of ethyl 2-oxocyclopentane-1-carboxylate with ureas.<sup>12,13</sup>

The room temperature treatment of ureas **2** with concentrated sulfuric acid failed to yield the anticipated<sup>4</sup> 2-(N-arylamino)-6,7-dihydrocyclopent-4H-3,1-oxazin-4(5H)-ones. Whereas relatively brief reaction periods led to the recovery of starting material, more prolonged treatment yielded only water soluble products, which were not investigated.

## EXPERIMENTAL

Melting points were determined in capillaries with a Thomas-Hoover Uni-Melt apparatus and are uncorrected. Infrared and  $^1\text{H}$ -nmr spectra were recorded on a Perkin-Elmer 337 and a Varian EM 360 spectrometer, respectively, as indicated in the Tables.

Ethyl 2-[[ (Arylamino)carbonyl]amino]-1-cyclopentene-1-carboxylates (2). After a mixture of 0.020 mole of ethyl 2-amino-1-cyclopentene-1-carboxylate (1), 20 ml of toluene, and 0.020 mole of an isocyanate had been refluxed overnight (16-20 hours), the solvent was removed by distillation under reduced pressure to yield the corresponding **2** (Table 1).

3-Aryl-6,7-dihydro-1H-cyclopentapyrimidine-2,4-(3H,5H)-diones (3). A mixture of 1.0 g of a urea **2** and 20 ml of 5% aqueous sodium hydroxide was heated on a steambath to form a solution, which was filtered from a small amount of insoluble material, cooled, and acidified with acetic acid to afford the corresponding **3** (Table 2).

Table 1. Ethyl 2-[[ (Arylamino)carbonyl]amino]-1-cyclopentene-1-carboxylates (**2**)<sup>a</sup>

Ar	Yield(%)	Mp(°C)	IR(cm <sup>-1</sup> ) <sup>b</sup>	$^1\text{H}$ -NMR(ppm) <sup>c</sup>
Ph	45	126-128 <sup>d,e</sup>	3250(N-H), 1660(C=O), 1620(C=C)	1.2(t, 3, CH <sub>3</sub> ), 1.8(m, 2, 4-CH <sub>2</sub> ), 2.4(m, 2, 5-CH <sub>2</sub> ), 3.1(m, 2, 3-CH <sub>2</sub> ), 4.2(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 6.9-7.6(m, 5, ArH), 9.8(s, 1, NH), 10.0(s, 1, NH)
2-MeC <sub>6</sub> H <sub>4</sub>	50	141-142 <sup>f</sup>	3280(N-H), 1670(C=O), 1620(C=C)	1.3(t, 3, CH <sub>2</sub> CH <sub>3</sub> ), 1.9(m, 2, 4-CH <sub>2</sub> ), 2.3(s, 3, CH <sub>3</sub> ), 2.5(m, 2, 5-CH <sub>2</sub> ), 3.2(m, 2, 3-CH <sub>2</sub> ), 4.2(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 7.6-7.9(m, 4, ArH), 9.3(s, 1, NH), 9.9(s, 1, NH)
3-MeC <sub>6</sub> H <sub>4</sub>	66	151-153 <sup>f</sup>	3250(N-H), 1670(C=O), 1630(C=C)	1.2(t, 3, CH <sub>2</sub> CH <sub>3</sub> ), 1.8(m, 2, 4-CH <sub>2</sub> ), 2.3(s, 3, CH <sub>3</sub> ), 2.4(m, 2, 5-CH <sub>2</sub> ), 3.1(m, 2, 3-CH <sub>2</sub> ), 4.2(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 6.8-7.4(m, 4, ArH), 9.8(s, 1, NH), 9.9(s, 1, NH)
4-MeC <sub>6</sub> H <sub>4</sub>	83	124.5-126 <sup>g</sup>	3460, 3280(N-H), 1680, 1650(C=O), 1620(C=C)	1.2(t, 3, CH <sub>2</sub> CH <sub>3</sub> ), 1.9(m, 2, 4-CH <sub>2</sub> ), 2.2(s, 3, CH <sub>3</sub> ), 2.4(m, 2, 5-CH <sub>2</sub> ), 3.1(m, 2, 3-CH <sub>2</sub> ), 4.1(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 7.0-7.5(m, 4, ArH), 9.8(s, 1, NH), 9.9(s, 1, NH)
3-ClC <sub>6</sub> H <sub>4</sub>	78	159.5-161 <sup>f</sup>	3250(N-H), 1670, 1660(C=O), 1620(C=C)	1.2(t, 3, CH <sub>3</sub> ), 1.9(m, 2, 4-CH <sub>2</sub> ), 2.4(m, 2, 5-CH <sub>2</sub> ), 3.1(m, 2, 3-CH <sub>2</sub> ), 4.2(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 6.9-7.3(m, 3, ArH), 7.7(m, 1, ArH), 9.8(s, 1, NH), 10.1(s, 1, NH)
4-ClC <sub>6</sub> H <sub>4</sub>	63	129.5-130.5 <sup>h</sup>	3450, 3280(N-H), 1680, 1660(C=O), 1620(C=C)	1.2(t, 3, CH <sub>3</sub> ), 1.9(m, 2, 4-CH <sub>2</sub> ), 2.4(m, 2, 5-CH <sub>2</sub> ), 3.1(m, 2, 3-CH <sub>2</sub> ), 4.2(q, 2, CH <sub>2</sub> CH <sub>3</sub> ), 7.3-7.8(m, 4, ArH), 9.9(s, 1, NH), 10.2(s, 1, NH)

<sup>a</sup>Satisfactory microanalytical data ( $\pm 0.20\%$  for C, H, N) were obtained for all compounds listed on this table. <sup>b</sup>Mineral oil mulls. <sup>c</sup>Solutions in hexadeuteriodimethylsulfoxide containing tetramethylsilane as internal standard. <sup>d</sup>Recrystallized from aqueous methanol. <sup>e</sup>Lit.<sup>14</sup> mp 127-128°. <sup>f</sup>Recrystallized from ethanol. <sup>g</sup>Recrystallized from petroleum ether (bp 63-75°).

<sup>h</sup>Recrystallized from methanol.

Table 2. 3-Aryl-6,7-dihydro-1H-cyclopentapyrimidine-2,4-(3H,5H)-diones (3)<sup>a</sup>

Ar	Yield(%)	Mp(°C)	IR(cm <sup>-1</sup> ) <sup>b</sup>	<sup>1</sup> H-NMR(ppm) <sup>c,d</sup>
Ph	61	280-281 <sup>e,f</sup>	3180,3100(N-H),1720,1630(C=O)	1.9-2.3(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.4-2.9(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),7.0-7.5(m,5,ArH)
2-MeC <sub>6</sub> H <sub>4</sub>	71	216-219 <sup>e</sup>	3200(N-H),1700,1640(C=O)	2.0(s,3,CH <sub>3</sub> ),1.8-2.2(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.3-2.8(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),6.9-7.2(m,4,ArH)
3-MeC <sub>6</sub> H <sub>4</sub>	71	274-276 <sup>g,h</sup>	3200,3100(N-H),1720,1640(C=O)	2.3(s,3,CH <sub>3</sub> ),1.8-2.2(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.4-2.8(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),6.8-7.4(m,4,ArH)
4-MeC <sub>6</sub> H <sub>4</sub>	70	270-271 <sup>g</sup>	3200,3100(N-H),1725,1640(C=O)	2.2(s,3,CH <sub>3</sub> ),1.8-2.2(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.3-2.8(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),6.8-7.3(m,4,ArH)
3-ClC <sub>6</sub> H <sub>4</sub>	72	288-290 <sup>g,i</sup>	3200,3100(N-H),1730,1650(C=O)	1.8-2.3(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.4-2.9(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),7.0-7.5(m,4,ArH)
4-ClC <sub>6</sub> H <sub>4</sub>	72	287-289 <sup>g,j</sup>	3250,3100(N-H),1710,1650(C=O)	1.8-2.2(m,2,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),2.4-2.9(m,4,CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> ),7.1-7.6(m,4,ArH)

<sup>a</sup>Satisfactory microanalytical data ( $\pm 0.30\%$  for C,H,N) were obtained for all compounds listed on this table. <sup>b</sup>Mineral oil mulls. <sup>c</sup>Solutions in hexadeuteriodimethylsulfoxide containing tetramethylsilane as internal standard. <sup>d</sup>The <sup>1</sup>H-nmr spectra showed that compounds 3 formed persistent hydrates. <sup>e</sup>Recrystallized from aqueous ethanol. <sup>f</sup>Lit. mp 319-321° (Ref. 5), 278° (Ref. 12), 288° (Ref. 14), 281.5-282° (Ref. 15). <sup>g</sup>Recrystallized from ethanol. <sup>h</sup>Lit. <sup>15</sup> mp 276-277°. <sup>i</sup>Lit. <sup>15</sup> mp 293-295°. <sup>j</sup>Lit. <sup>15</sup> mp 296-297°.

## REFERENCES AND NOTES

- (1) Undergraduate research participant.
- (2) E. P. Papadopoulos, *J. Heterocycl. Chem.*, 1980, 17, 1553; *ibid.*, 1981, 18, 515.
- (3) E. P. Papadopoulos and C. D. Torres, *Heterocycles*, 1982, 19, 1039.
- (4) E. P. Papadopoulos and C. D. Torres, *J. Heterocycl. Chem.*, 1982, 19, 269.
- (5) G. DeStevens, A. Halamandaris, P. Wenk, R. A. Mull, and E. Schlittler, *Arch. Biochem. Biophys.*, 1959, 83, 141.
- (6) J. Perronnet, A. Teche, and J. P. Demoute, French Patent 2,079,535; *Chem. Abstr.*, 1972, 77, 62019m.
- (7) Roussel-UCLAF, *French Patent* 2,335,511; *Chem. Abstr.*, 1978, 88, 136655s.
- (8) C. Paal, *Chem. Ber.*, 1894, 27, 974.
- (9) R. P. Staiger and E. C. Wagner, *J. Org. Chem.*, 1953, 18, 1427.
- (10) B. Taub and J. B. Hino, *ibid.*, 1961, 26, 5238.
- (11) D. M. O'Mant, *British Patent* 1,059,271; *Chem. Abstr.*, 1967, 67, 54161e.
- (12) S. Senda and H. Fujimura, *Japanese Patent* 4892 (1962); *Chem. Abstr.*, 1963, 59, 642h.
- (13) E. J. Soboczenski, *U.S. Patent* 3,235,360; *Chem. Abstr.*, 1966, 64, 14196f.
- (14) S. Senda, K. Hirota, and K. Maeno, *Chem. Pharm. Bull.*, 1973, 21, 1894.
- (15) Z. Eckstein, A. Kunicki, and W. Walczak-Korzeniowska, *Przem. Chem.*, 1980, 59, 541; *Chem. Abstr.*, 1981, 94, 192252u.

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