

## SYNTHESIS AND ELABORATION OF 3-SUBSTITUTED 4-NITROISOXAZOLES

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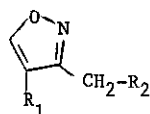
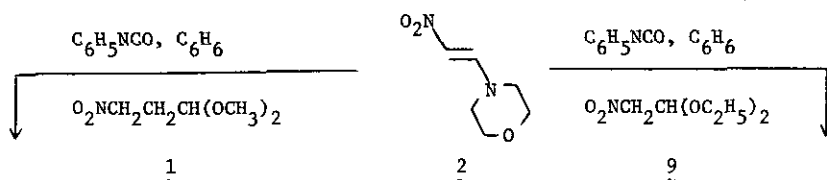
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**Abstract** — A regioselective Mukaiyama reaction involving nitroacetals 1 and 9, phenyl isocyanate and 1-morpholino-2-nitroethene 2 in benzene provided, respectively, the 4-nitroisoxazole acetals 3 and 10 in excellent yield. These acetals in turn served as convenient synthetic entries into the series of 3,4-difunctionalized isoxazoles of structures 4-8 and 11-13, respectively.

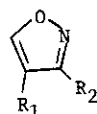
Isoxazoles<sup>1</sup> are versatile synthetic intermediates<sup>2</sup> which incorporate latent functionality<sup>3</sup> corresponding to  $\gamma$ -amino alcohols,  $\alpha,\beta$ -unsaturated ketones,  $\beta$ -hydroxy ketones, cyano and imino ketones, as well as other combinations of juxtaposed functional groups. We required a series of 3-substituted 4-nitroisoxazoles which could be further elaborated while maintaining the isoxazole ring intact. Direct nitration<sup>4</sup> of 3-methylisoxazole, for example, requires vigorous conditions not compatible with sensitive substituents. We therefore chose to adapt methodology<sup>5</sup> which has led to 4-nitroisoxazoles by the Mukaiyama reaction<sup>6</sup> of nitrile oxides with nitro enamines.<sup>7</sup>

Reaction of nitro acetal 1<sup>8</sup> with a ten-fold excess of phenyl isocyanate in benzene containing 1-morpholino-2-nitroethene 2<sup>9</sup> gave 4-nitroisoxazole 3 regiospecifically and in excellent yield. This compound in turn could be converted under standard conditions into the functionalized derivatives 4-8 while keeping the isoxazole ring intact. Noteworthy is the selective reduction of the nitro group to give amine 6 without fission or reduction of the isoxazole ring. Also, the conversion of acetal 3 into thioacetal 5 allows for the possibility of further elaboration at the terminal carbon atom through umpolung<sup>10</sup> alkylation chemistry.

The generality of this approach is illustrated by the synthesis of the next lower homolog isoxazoles 10-13 starting with nitro acetal 9.<sup>11</sup>



	R <sub>1</sub>	R <sub>2</sub>
3	NO <sub>2</sub>	CH(OCH <sub>3</sub> ) <sub>2</sub>
4	NO <sub>2</sub>	CHO
5	NO <sub>2</sub>	CH
6	NH <sub>2</sub>	CH(OCH <sub>3</sub> ) <sub>2</sub>
7	NHCOCH <sub>3</sub>	CH(OCH <sub>3</sub> ) <sub>2</sub>
8	NHCOCH <sub>3</sub>	CH



	R <sub>1</sub>	R <sub>2</sub>
10	NO <sub>2</sub>	CH(OC <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>
11	NH <sub>2</sub>	CH(OC <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>
12	NHCOC <sub>6</sub> H <sub>5</sub>	CH(OC <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>
13	NHCOC <sub>6</sub> H <sub>5</sub>	CHO

#### EXPERIMENTAL SECTION

3-(2,2-Dimethoxyethyl)-4-nitroisoxazole (3). To dry benzene (50 ml) containing phenyl isocyanate (23.8 g, 200 mmol) and 2<sup>9</sup> (3.08 g, 20.0 mmol) was added with stirring a solution of 1<sup>8</sup> (6.72 g, 45.4 mmol) and Et<sub>3</sub>N (350 mg, 3.48 mmol) in benzene (20 ml). After a 2 h reflux period, the mixture was filtered and the filtrate was concentrated in vacuo. Flash chromatography over silica gel (ether-hexane, 1:9) gave 3 (3.63 g, 90%) as an oil: NMR (CDCl<sub>3</sub>) δ 3.20 (s, 6), 3.20 (d, 2), 4.86 (t, 1), 9.25 (s, 1). Anal. Calcd for C<sub>7</sub>H<sub>10</sub>N<sub>2</sub>O<sub>5</sub>: C, 41.59; H, 4.99; N, 13.86. Found: C, 41.64; H, 4.65; N, 13.51.

3-(2-Oxoethyl)-4-nitroisoxazole (4). Acetal 3 (32 mg, 0.16 mmol) was dissolved in HOAc (4 ml) containing CF<sub>3</sub>CO<sub>2</sub>H (10 drops) and water (10 drops) and then heated at 95°C for 30 min. Evaporation of the solvent in vacuo gave the sensitive aldehyde 4 (25 mg, 100%; >90% pure by NMR) as an oil which tended to undergo decomposition during attempted purification. NMR (CDCl<sub>3</sub>) δ 3.42 (s, 2), 9.32 (s, 1), 9.86 (s, 1); MS m/e 156.016 (M<sup>+</sup>, calcd for C<sub>5</sub>H<sub>4</sub>N<sub>2</sub>O<sub>4</sub>, 156.017) (9), 128 (49), 91 (67), 86 (82), 53 (74), 43 (100).

3-(1,3-Dithiacyclohex-2-ylmethyl)-4-nitroisoxazole (5). To a refluxing solution of BF<sub>3</sub>·Et<sub>2</sub>O (123 mg, 0.86 mmol) and propane dithiol (39 mg, 0.36 mmol) in dry CHCl<sub>3</sub> (10 ml) was added over 1 h acetal 3 (59 mg, 0.29 mmol) in CHCl<sub>3</sub> (10 ml). After 5 h at reflux, the usual workup gave an oil which was filtered through silica gel (CHCl<sub>3</sub>) giving dithiane 5 (52 mg, 72%) as an oil: NMR

(CDCl<sub>3</sub>)  $\delta$  1.90-2.20 (m, 2), 2.78-3.02 (m, 4), 3.60 (d, 2), 4.46 (t, 1), 9.29 (s, 1); MS m/e 246.013 (M<sup>+</sup>, calcd for C<sub>8</sub>H<sub>10</sub>N<sub>2</sub>O<sub>3</sub>S<sub>2</sub>, 246.013) (50), 229 (13), 165 (26), 149 (14), 132 (13), 119 (100).

3-(2,2-Dimethoxyethyl)-4-aminoisoxazole (6). Isoxazole 3 (320 mg, 1.58 mmol) and NH<sub>4</sub>Cl (2.0 g, 37 mmol) were dissolved in water (8 ml) at 0°C. Then zinc dust (3.2 g, 49 mg-atom) was added in portions over 15 min with stirring. After 30 min at 0°C, the mixture was filtered and the cake was washed with MeOH (20 ml). The combined filtrate was evaporated in vacuo to give amine 6 (224 mg, 82%) as an oil of suitable purity for the next reaction: NMR (CDCl<sub>3</sub>)  $\delta$  2.98 (d, 2), 3.42 (s, 6), 3.42 (s, 2), 4.58 (t, 1), 7.95 (s, 1).

3-(2,2-Dimethoxyethyl)-4-acetamidoisoxazole (7). To a stirred solution of 6 (224 mg, 1.30 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (12 ml) was added pyridine (364 mg, 4.38 mmol) and Ac<sub>2</sub>O (248 mg, 2.43 mmol). After 2 h the solution was diluted with water and extracted with ether. The extract was dried (K<sub>2</sub>CO<sub>3</sub>) and concentrated in vacuo to give amide 7 (275 mg, 98%) as an oil of suitable purity for the next reaction: NMR (CDCl<sub>3</sub>)  $\delta$  2.16 (s, 3), 3.10 (d, 2), 3.51 (s, 6), 4.57 (t, 1), 8.37-8.60 (s, 1), 9.04 (s, 1).

3-(1,3-Dithiacyclohex-2-ylmethyl)-4-acetamidoisoxazole (8). To a refluxing solution of BF<sub>3</sub>·Et<sub>2</sub>O (156 mg, 1.1 mmol) and propane dithiol (119 mg, 1.1 mmol) in CHCl<sub>3</sub> (10 ml) was added dropwise 7 (235 mg, 1.1 mmol) in CHCl<sub>3</sub> (10 ml). After 2 h the usual workup followed by crystallization from CH<sub>2</sub>Cl<sub>2</sub>-hexane gave 8 (245 mg, 87%) as colorless needles: mp 111-112°C; NMR (CDCl<sub>3</sub>)  $\delta$  1.72-2.30 (m, 2), 2.20 (s, 3), 2.78-3.00 (m, 4), 3.15 (d, 2), 4.36 (t, 1), 8.48 (s, 1), 9.00 (s, 1). Anal. Calcd for C<sub>10</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C, 46.49; H, 5.46; N, 10.84. Found: C, 46.41; H, 5.57; N, 10.61.

3-(Diethoxymethyl)-4-nitroisoxazole (10). Following the procedure used to prepare 3, crude 10 was obtained (from 9) as an oil of suitable purity for the next reaction: NMR (CDCl<sub>3</sub>)  $\delta$  1.28 (t, 6), 3.65-3.96 (m, 4), 6.10 (s, 1), 9.24 (s, 1).

3-(Diethoxymethyl)-4-aminoisoxazole (11). A solution of 10 (150 mg) in MeOH (7 ml) containing 10% Pd/C (150 mg) was stirred under H<sub>2</sub> (1 atm.) until 3 equivalents of H<sub>2</sub> were absorbed. Filtration followed by evaporation gave amine 11 (128 mg, 100%) as an oil suitably pure for the next reaction: NMR (CDCl<sub>3</sub>)  $\delta$  1.26 (t, 6), 3.44-3.84 (m, 4), 3.64 (s, 2), 5.59 (s, 1), 7.92 (s, 1).

3-(Diethoxymethyl)-4-benzamidoisoxazole (12). Benzamide 12 was obtained as an oil by benzoylation of 11 with benzoyl chloride and pyridine under standard conditions: NMR (CDCl<sub>3</sub>)  $\delta$  2.32 (t, 6), 3.60-3.94 (m, 4), 5.80 (s, 1), 7.40-7.64 (m, 3), 7.80-7.96 (m, 2), 8.96 (s, 1), 9.28 (s, 1); MS m/e 290.127 (M<sup>+</sup>, calcd for C<sub>15</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>, 290.127) (2), 122 (47), 105 (100), 103 (27), 91 (24), 77 (60).

3-Formyl-4-benzamidoisoxazole (13). Acetal 12 (10 mg) was stirred in THF (4 ml) containing 3 N HCl (4 ml) for 22 h at 25°C. After the usual workup 8 mg (100%) of crude 13 was isolated which could be recrystallized from CH<sub>2</sub>Cl<sub>2</sub>-hexane to give aldehyde 13 as colorless, fluffy microcrystals, mp 125° (dec). NMR (CDCl<sub>3</sub>) δ 7.50-7.64 (m, 3H), 7.84-7.98 (m, 2H), 9.36 (s, 1H), 9.45 (s, 1H), 10.35 (s, 1H); MS m/e 216.054 (M<sup>+</sup>, calcd for C<sub>11</sub>H<sub>8</sub>N<sub>2</sub>O<sub>3</sub>, 216.053) (2), 105 (100), 91 (2), 77 (78), 51 (38).

#### ACKNOWLEDGMENT

This research was supported by PHS research grants GM-24951 and 27137 from the National Institute of General Medical Sciences.

#### REFERENCES

1. A. I. Meyers in "Heterocycles in Organic Synthesis," Wiley-Interscience, New York, 1974.
2. For some recent applications see, A. Barco, S. Benetti, G. P. Pollini, P. G. Baraldi, M. Guarneri, D. Simoni, and C. Gandolfi, *J. Org. Chem.*, 1981, 46, 4518.
3. For a review see, A. A. Akhrem, F. A. Lakhvich, and V. A. Khripach, *Chem. Het. Comp.*, 1982, 17, 853. See also, D. H. Hoskin and R. A. Olofson, *J. Org. Chem.*, 1982, 47, 5222.
4. A. Quilico and C. Musanti, *Gazz. Chim. Ital.*, 1941 71, 327.
5. S. Rajappa, B. G. Advani, and R. Sreenivasan, *Synthesis*, 1974, 656.
6. T. Mukaiyama and T. Hoshimo, *J. Am. Chem. Soc.*, 1960, 82, 5339.
7. D. R. Britelli and G. A. Boswell, Jr., *J. Org. Chem.*, 1981, 46, 316.
8. E. J. Corey, I. Vlittas, N. H. Andersen, K. Harding, *J. Am. Chem. Soc.*, 1968, 90, 3247.
9. C. D. Hurd and L. T. Sherwood, Jr., *J. Org. Chem.*, 1948, 13, 471.
10. D. Seebach, *Angew. Chem. Internat. Ed. Engl.*, 1979, 18, 239.
11. S. Kabusz and W. Tritschler, *Synthesis*, 1971, 312.

Received, 16th February, 1983