## A SHORT STEP SYNTHESIS OF LESPEDAMINE

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<u>Abstract</u>——— A convenient synthetic method for l-hydroxy-, l-methoxy-, and l-acetoxy-2-oxindole was disclosed starting from methyl 2-nitrophenylacetate. A five-step synthesis of lespedamine was achieved utilizing this method.

In this report, we describe a practical synthetic method for 1-hydroxy-2-oxindole derivatives and a short step synthesis of lespedamine (1),  $^2$ ,  $^3$  one of eight naturally occurring 1-methoxy indole derivatives.  $^4$ 

## I. Syntheses of 1-Hydroxy-2-oxindole Derivatives

Various synthetic methods so far reported for 1-hydroxy- (2) and 1-methoxy-2-oxindole (3) are known to give unsatisfactory results. However, we found that the readily available methyl 2-nitrophenylacetate (4) simply upon treatment with zinc (20 mol eq.) and ammonium chloride (3.8 mol eq.) in methanol for 3h afforded 2 in 48% yield. When an excess amount of reducing agents was used or a longer reaction time was adopted, the yield of 2 was decreased mainly due to its sensitivity toward reductive decomposition, resulting in the formation of 2-oxindole.

We have also found that a significant amount of 2 was lost by the formation of a complex with zinc iron, which was rather insoluble in organic solvents. The structure of the complex was tentatively assigned to be 5 based mainly on its mass spectrum which showed the ratio of 2 and zinc iron to be 2 to 1.

Direct treatments of the reaction mixture, obtained by the reaction of 4 with zinc and ammonium chloride, with diazomethane and acetic anhydride and pyridine were found to give 1-methoxy- (3)<sup>8</sup> and 1-acetoxy-2-oxindole (6)<sup>9</sup> in 77% and 70% overall yields, respectively. Furthermore, hydrolysis of 6 with sodium carbonate afforded 2 in 94% yield.

## II. Synthesis of Lespedamine

The reaction of 3 with ethylene dibromide in the presence of sodium hydride afforded spiro compound (7) $^{10}$  in 90% yield. Subsequent treatment of 7 with aq. dimethylamine (50 mol eq.) and its hydrochloride (9 mol eq.) in N,N-dimethylformamide produced the desired 3-(2-N,N-dimethylaminoethyl)-1-methoxy-2-oxindole (8) 11 in 54% yield, together with 10% yield of a phenylhydroxylamine derivative (9). 12 The reduction of 8 with lithium aluminum hydride (LiAlH4) in ether was found to produce 3-(2-N,N-dimethylaminoethyl)-2-hydroxy-1-methoxy-2,3-dihydroindole (10) in 62% yield as a mixture of diastereoisomers. The compound (10) was unstable and instantaneously changed by treatment with aq. hydrochloric acid to lespedamine 14 (1) in 95% yield. On the basis of the above results, the final step was improved as follows. Thus, after reduction of 8 with LiAlH4, the reaction mixture was treated briefly with aq. hydrochloric acid. By this modification, lespedamine (1) was prepared in 64% yield directly from 8. Thus, the total synthesis  $\frac{3}{2}$  of  $\frac{1}{2}$  was achieved in five steps with 24% overall yield from 4. The spectral data of synthetic material and melting point of its picrate were identical with those of lespedamine. $^{2}$ In conclusion, building blocks such as 2, 3, and 6 for 1-hydroxyindole derivatives have now become readily available from 4 in excellent yields. Investigation of their reactions and preparation of other naturally occurring 1-methoxyindole derivatives are currently in progress.

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## REFERENCES AND NOTES

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- Pharm. Bull., in press.
- 2. H. Morimoto and H. Oshio, Liebigs Ann. Chem., 682, 212 (1965).
- 3. Total synthesis of lespedamine was reported with 2.6% overall yield. R.M. Acheson, P.G. Hunt, D.M. Littlewood, B.A. Murrer, and H.E. Rosenberg, J. Chem. Soc., perkin I, 1978, 1117.
- E. Wenkert, J.C. Orr, S. Garratt, J.H. Hansen, B. Wickberg, and C.L. Leicht, J. Org. Chem., 27, 4123 (1963); S.R. Johns, J.A. Lamberton, and J.L. Occolowitz, Aust. J. Chem., 20, 1737 (1967); M. Nomoto and S. Tamura, Agr. Biol. Chem., 34, 1590 (1970); D.W. Nagel, K.G.R. Pachler, P.S. Steyn, R. Vleggaar, and P.L. Wessels, Tetrahedron, 32, 2625 (1976); H. Wagner and T. Nestler, Tetrahedron Letters, 1978, 2777; Y. Konda, M. Onda, A. Hirano, and S. Omura, Chem. Pharm. Bull., 28, 2987 (1980).
- 5. A. Reissert, Ber., 41, 3921 (1908); W.B. Wright, Jr. and K.H. Collins, J. Am. Chem. Soc., 78, 221 (1956); T. Andre, M. Georges, and R. Gilbert, C. R. Acad. Sci., Ser. C, 273, 1378 (1971)[C.A., 76, 59359w (1972)]; C.w. Muth and R.N. Beers, Proc. W. Va. Acad. Sci., 41, 235 (1969) [C.A., 74, 42104v (1971)].
  6. mp 200.5-202.0°C (lit. 5 mp 199-200°C). IR (KBr): 1675, 1617 cm<sup>-1</sup>. H-NMR (10%)
- CD<sub>3</sub>OD in CDCl<sub>3</sub>)  $\delta$ : 3.35 (1H, br s), 3.43 (2H, s), 6.65-7.41 (4H, m). 7. mp>300°C. IR (KBr): 1630, 1605 cm<sup>-1</sup>. <sup>1</sup>H-NMR (pyridine-d<sub>5</sub>)  $\delta$ : 3.28 (4H, s),
- 6.62-7.38 (8H, m). High MS m/z: Calcd for  $C_{16}H_{12}N_{2}O_{4}Zn$ : 360.0087 and 362.0057. Found: 360.0109 and 361.9963.
- 8. mp 84.5-86.0°C (lit. mp 84-86°C). IR (KBr): 1712, 1617 cm<sup>-1</sup>.  $^{1}$ H-NMR (CDCl<sub>2</sub>)  $\mathcal{S}$ :
- 3.42 (2H, s), 3.95 (3H, s), 6.65-7.42 (4H, m). MS m/z: 163 (M<sup>+</sup>), 132. 9. mp 97-99°C. IR (KBr): 1807, 1727 cm<sup>-1</sup>.  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\mathcal{S}$ : 2.33 (3H, s), 3.55 (2H, s), 6.50-7.35 (4H, m). Anal. Calcd for C<sub>10</sub>H<sub>q</sub>NO<sub>3</sub>: C, 62.82; H, 4.75; N, 7.33. Found: C, 63.00; H, 4.72; N, 7.04.
- 10. Oil. IR (film): 1723, 1619 cm<sup>-1</sup>.  ${}^{1}$ H-NMR (CCl<sub>4</sub>)  $\delta$ : 1.17-1.54 (2H, m), 1.54-1.87 (2H, m), 3.92 (3H, s), 6.41-7.21 (4H, m). High MS m/z: Calcd for C<sub>11</sub>H<sub>11</sub>NO<sub>2</sub>: 189.0789. Found: 189.0795
- 11. Oil. IR (film): 1727, 1616 cm<sup>-1</sup>.  ${}^{1}H$ -NMR (CCl<sub>4</sub>)  $\delta$ : 1.69-2.50 (4H, A<sub>2</sub>B<sub>2</sub>, m), 2.06 (6H, s), 3.32 (1H, t, J=5.6 Hz), 3.86 (3H, s), 6.57-7.29 (4H, m). High MS m/z: Calcd for  $C_{13}H_{18}N_2O_2$ : 234.1367. Found: 234.1375.
- 12. Oil. IR (film): 3480, 1647 cm<sup>-1</sup>.  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\mathcal{E}$ : 1.67-2.57 (4H, m), 2.32 (6H, s), 2.73 (3H, s), 2.86 (3H, s), 3.66 (3H, s), 3.90 (1H, dd, J=8.8 and 5.2)Hz), 6.50-7.30 (4H, m), 6.93 (1H, br s). High MS m/z: Calcd for  $C_{15}H_{25}N_3O_2$ : 279.1944. Found: 279.1937.
- 13. Oil. IR (film): 3340, 1612, 1596, 1475, 1463 cm<sup>-1</sup>.  ${}^{1}$ H-NMR (CCl<sub>4</sub>)  $\mathcal{S}$ : 1.84-2.75 (5H, m), 2.25 (6H, s), 3.82 (3H, s), 4.60 and 4.92 (total 1H, each d, J=8 Hz), 5.83 (1H, br s), 6.44-7.15 (4H, m). High MS m/z: Calcd for  $C_{13}H_{20}N_2O_2$ : 236.1523. Found: 236.1539.
- 14. Spectra of IR and <sup>1</sup>H-NMR were identical with those of lespedamine. Charts of IR and H-NMR spectra of lespedamine are reported in the ref. 2. Oil. IR.  $(CHCl_3): 1459 \text{ cm}^{-1}. \ ^{1}\text{H-NMR} \ (CCl_4) \ \mathcal{S}: 2.19 \ (6\text{H}, \text{s}), 2.32-2.96 \ (4\text{H}, \text{m}), 3.92$ (3H, s), 6.62-7.45 (5H, m). MS m/z: 218  $(M^+)$ , 187  $(M^+$ -OMe). Picrate: mp 161-163°C (lit. 2 mp 160-162°C).

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