1,3-DIPHENYLTHIENO [2,3-c] FURAN. A NEW HETEROCYCLIC SYSTEM Willy Friedrichsen + and Axel Schöning

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Abstract - The synthesis of 1, 3-diphenylthieno[2, 3-c] furan (7) is described.

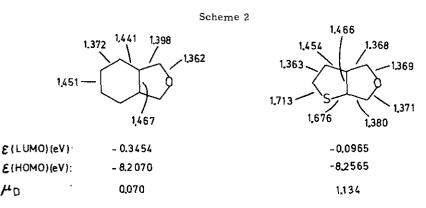
Isobenzofurans (benzo[c]furans, 1)¹ are of considerable importance both as trapping reagents for unstable 2π components (olefins¹, sulfenes^{2a}, nitroso compounds^{2b, c}) and as building blocks in inter-^{1,3} and intra-molecular⁴ Diels-Alder reactions. Despite of these interesting properties, the area of heteroanellated derivatives^{1,5}(2) has remained almost unexplored. In this paper we describe the synthesis and some reactions of a member (7) of the hitherto unknown thieno[2, 3-c] furan system 3. 4,4-Dimethyl-2-(thien-2-yl)-2-oxazoline (4, Scheme 1)⁶ was allowed to react with n-BuLi and benzaldehyde yielding 5^6 (mp 62° C), which on treatment with methyl iodide gave an oxazolinium salt (6, mp $154-156^{\circ}$ C)⁷. The reaction of 6 with phenylmagnesium bromide proceeded without difficulty⁸; after acidic work-up, 1,3-diphenylthieno[2,3-c]furan (7) was obtained as tiny yellow needles with mp 136° C (85%; UV(acetonitrile): $\lambda(\lg \epsilon) = 252$ (4.42), 298 (4.15), 325 (4.10), 343 (4.09), 370 nm (4.15)). Compound 7 reacts with simple olefins (e.g., dimethyl

Scheme 1

Het

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 $\frac{1}{2}$
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acetylenedicarboxylate 9) in rates comparable with 1,3-diphenylisobenzofuran. The reaction of $\overline{2}$ with tetrachloro-o-benzoquinone yielded a dioxole 10 (9, mp 150°C; IR(KBr): 1445, 1660 cm -1). The results of MNDO 11 calculations with complete geometry optimization 12 are shown in Scheme 2. As can be seen from these figures, both benzo[c]furan and thieno[2, 3-c]furan can be considered as cyclic polyenes. A small decrease of $\mathcal{E}(HOMO)$ and a small increase of $\mathcal{E}(LUMO)$ is in accord with the observed hypsochromic shift of the first UV maximum.



MNDO - CGO results (bond lengths in A)

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REFERENCES

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- 1. 1a. W. Friedrichsen, Adv. Heterocycl. Chem., 26, 135 (1980). 1b. U. E. Wiersum, Aldrichim. Acta, 14, 53 (1981).
- 2. 2a. E. Block and M. Aslam, Tetrahedron Lett., 1982, 4203. 2b. P. Horsewood, G. W. Kirby, R. P. Sharma and J. G. Sweeny, J. Chem. Soc., Perkin Trans. I, 1981, 1802. 2c. G. W. Kirby and J.G. Sweeny, ibid., 1981, 3250.
- 3. B.A. Keay and R. Rodrigo, Canad. J. Chem., 61, 637 (1983).
- 4. B.A. Keay and R. Rodrigo, J. Am. Chem. Soc., 104, 4725 (1982).
- 5. M.G. Saulnier and G.W. Gribble, Tetrahedron Lett., 1983, 5435.
- 6. L.D. Vecchia and I. Vlattas, J. Org. Chem., 42, 2649 (1977).
- 7. If the preparation of $\underline{6}$ is conducted in the presence of a small amount of copper powder a very clean product is obtained.
- 8. A.I. Meyers and E. M. Smith, J. Org. Chem., 37, 4289 (1972).
- 9. 8: mp 123-124°C; IR(KBr): 1720, 1730 cm⁻¹ (sh).
- 10. W. M. Horspool, J. M. Tedder and Z. U. Din, J. Chem. Soc. C, 1969, 1694.
- 11. M. J. S. Dewar and W. Thiel, J. Am. Chem. Soc., 99, 4899 (1977).
- 12. MOPAC package; J. P. Stewart, QCPE 455.

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