

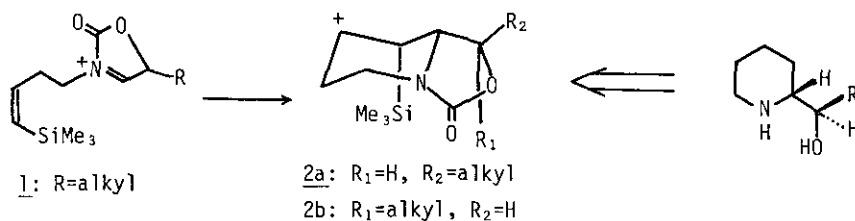
A DIASTEREOSELECTIVE SYNTHESIS OF THREO 2-(α -HYDROXYALKYL)PIPERIDINES
VIA OXACARBAMOYLIMINIUM ION-(Z)-VINYLSILANE CYCLIZATION

Shinzo Kano,* Tsutomu Yokomatsu, Yoko Yuasa, and Shiroshi Shibuya
Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan

Abstract — Reduction of 5-alkyl-N-(Z-4-silyl-3-butenyl)oxazolidine-2,4-diones (5a,b) with NaBH₄, followed by cyclization with TiCl₄ afforded the corresponding 1,8a-trans-1-alkyl-1,5,6,8a-tetrahydro-3H-oxazolo[3,4-a]pyridines (9a,b), respectively, with high diastereoselectivity. Catalytic hydrogenation of 9a, followed by alkaline hydrolysis gave threo 2-(α -hydroxyethyl)piperidine (12a). In a similar fashion, threo 2-(α -hydroxypropyl)piperidine (12b), (\pm)- β -conhydrine, was obtained from 9b.

π Cyclization of N-acyliminium ions has been documented as an important method for a synthesis of a wide variety of heterocyclic systems.¹ Such cyclizations have been found to achieve remarkable stereocontrol.^{1,2} The use of vinylsilanes as terminators for cationic cyclizations has been applied to a regiocontrolled synthesis of unsaturated azaheterocycles.³ In connection with our interest in a diastereoselective synthesis of α -(α' -hydroxyalkyl)-N-heterocycles,⁴ we have examined the cationic cyclizations of oxacarbamoyliminium ion-(Z)-vinylsilane systems (1), in the expectation that cyclization proceeds with high diastereoselectivity through the intermediate (2a) rather than 2b which have significant steric repulsion between trimethylsilyl and R₁ group. (See Scheme 1). The method would also provide a promising route to threo 2-(α -hydroxyalkyl)-piperidines. The results of our studies are described in this paper.

Scheme 1

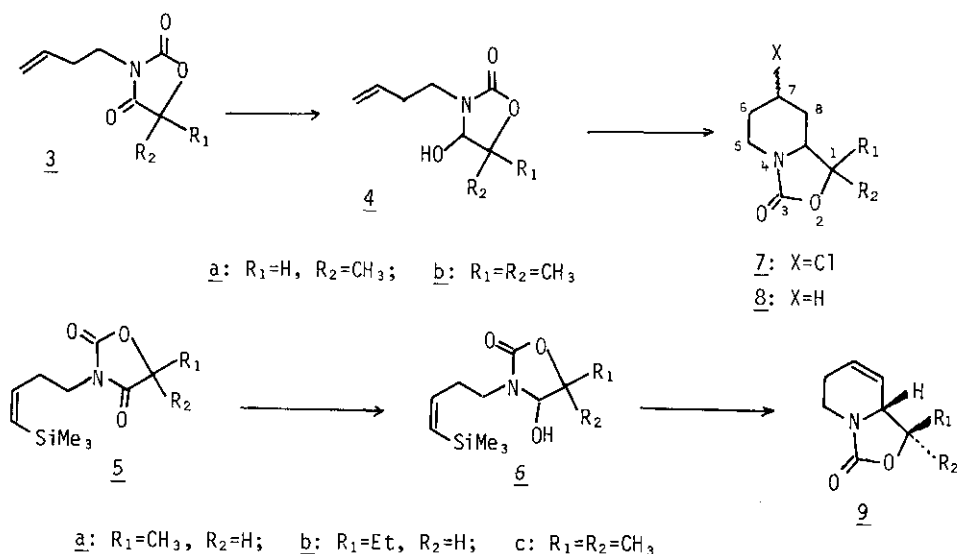


The N-butenyl- (4a,b) and N-(Z-4-trimethylsilyl-3-butenyl)-4-hydroxyoxazolidin-2-ones (6a-c) required for the formation of oxacarbamoyliminium ions were prepared as follows. Condensation of 3-buten-1-ol with 5-substituted oxazolidine-2,4-diones⁵ by an application of Mitsunobu's method⁶ afforded 3a,b. In a similar way, 5a-c were prepared by using (Z)-4-trimethylsilyl-3-buten-1-ol^{3b} and 5-substituted oxazolidine-2,4-diones.⁵ Reduction of 3a,b and 5a-c with NaBH₄ (methanol, 0°C) afforded the corresponding 4-hydroxy derivatives (4a,b) and (6a-c), respectively.

First, we examined the cyclization of 6a whether the reaction proceeded with diastereoselectivity with regard to the stereochemistry of 1-alkyl and 8a-H. Treatment of 4a with TiCl₄ (2 equiv., CH₂Cl₂, reflux, 2 h) gave 7a in 58 % yield. Dechlorination of 7a (n-Bu₃SnH, AIBN, benzene, reflux, 3 h) yielded 8a as a mixture of diastereomers in a ratio of ca 1:1.⁷ Cyclization of 4b under the same conditions yielded 7b, in 65 % yield, mp 55-59°C, dechlorination of which afforded 8b,⁸ mp 64-66°C.

Next, we explored the diastereoselective synthesis of 1,8a-trans-1-alkyl-1,5,6,8a-tetrahydro-3H-oxazolo[3,4-a]pyridines. Treatment of 6a with TiCl₄ yielded the desired cyclization product (9a)⁸ in 65 % yield, bp 122-124°C (3 torr), as a single diastereomer. Under the same conditions, 1-ethyl analogue (9b)⁸ was obtained from 6b in 70 % yield, bp 125-128°C (3 torr). In the case of 6c, the desired cyclization product (9c) was not obtained but 7b was obtained in 22 % yield. Formation of 7b from 6c can be accounted for by predominant protonation-desilylation prior to cyclization because of steric repulsion between trimethylsilyl and methyl group in the intermediate (2: R₁=R₂=CH₃).

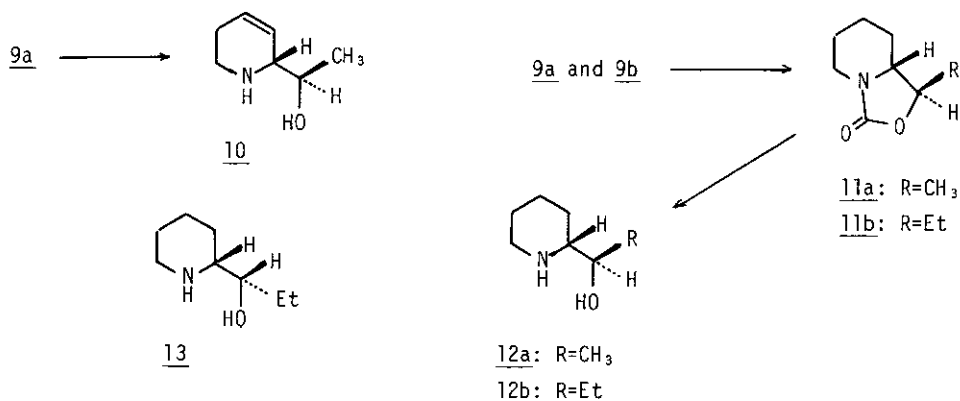
Scheme 2



Cleavage of oxazolidinone ring of 9a was carried out by heating with 10 % NaOH-EtOH (reflux, 2 h) to give 10⁸ in 88 % yield. Catalytic hydrogenation of 9a and 9b over Pt catalyst afforded 11a,⁸ mp 54-56°C, and 11b,⁸ mp 56-58°C, in nearly quantitative yield, respectively. Hydrolysis of 11a (10 % NaOH-EtOH, reflux, 2 h) afforded 88 % yield of 12a. In a similar fashion, threo 2-(α -hydroxypropyl)piperidine (12b) was obtained by hydrolysis of 11b, in 85 % yield, mp 87-88°C (lit.⁹ 87-88°C), which was identical with (\pm)- β -conhydrine [threo 2-(α -hydroxypropyl)piperidine], provided from professor Michael Vaultier. The product was not identical with (\pm)- α -conhydrine (13),⁹ erythro isomer, also donated from Professor Michael Vaultier.

As mentioned above, a facile diastereoselective synthesis of threo 2-(α -hydroxyalkyl)piperidine derivatives was achieved. The method would also be applicable to a synthesis of a variety of 2-substituted 1,2,5,6-tetrahydropyridine derivatives.

Scheme 3



ACKNOWLEDGMENT

The authors are deeply grateful to Professor Michael Vaultier, Université de Rennes, France, for the gifts of samples, ¹H NMR and ¹³C NMR spectra of (\pm)- α -conhydrine and (\pm)- β -conhydrine.

REFERENCES AND NOTES

1. (a) For a review: W. N. Speckamp, *Recl. Trav. Chim. Pays-Bas*, 1981, 100, 345. (b) J. A. M. Hamersma and W. N. Speckamp, *Tetrahedron Lett.*, 1982, 23, 3811, and references cited therein. (c) J. A. M. Hamersma and W. N. Speckamp, *Tetrahedron*, 1982, 38, 3255, and references cited therein. (d) D. J. Hart and K. Kanai, *J. Am. Chem. Soc.*, 1983, 105, 1255. (e) H. Kohn and Z.-K. Liao, *J. Org. Chem.*, 1982, 47, 2787. (f) M. S. Harley, F. D. King, and R. T. Martin,

Tetrahedron Lett., 1983, 24, 91.

2. (a) A. R. Chamberlin and Y. L. Chung, J. Am. Chem. Soc., 1983, 105, 3653. (b) B. E. Maryanoff, D. F. McComsey, and B. A. D.-Emswiler, J. Org. Chem., 1983, 48, 5062. (c) Z.-E. Liao and H. Kohn, J. Org. Chem., 1984, 49, 3812. (d) S. Kano, Y. Yuasa, T. Yokomatsu, and S. Shibuya, Heterocycles, 1984, 22, 1411. (e) S. Kano, Y. Yuasa, and S. Shibuya, Heterocycles, 1984, 22, 2327, and references cited therein.
3. (a) L. E. Overman and K. L. Bell, J. Am. Chem. Soc., 1981, 103, 1851. (b) L. E. Overman, T. C. Malone, and G. P. Meier, J. Am. Chem. Soc., 1983, 105, 6993, and references cited therein. (c) L. E. Overman, K. L. Bell, and F. Ito, J. Am. Chem. Soc., 1984, 106, 4192.
4. S. Kano, Y. Yuasa, and S. Shibuya, Heterocycles, 1985, 23, 395.
5. M. Kobayashi, M. Kitazawa, and T. Saito, Yakugaku Zasshi, 1983, 103, 1195.
6. O. Mitsunobu, M. Wada, and T. Sano, J. Am. Chem. Soc., 1972, 94, 679.
7. The ratio was determined based on the signals due to 1-CH₃ (δ 1.30 and 1.37, each d, $J=6.5$ Hz) and 1-H (δ 3.98-4.28 and 4.48-4.78, each m) in its ¹H NMR (CDCl₃) spectrum.
8. All new compounds gave satisfactory microanalysis and spectral data. Selected spectral data are as follows. Only characteristic signals are given for ¹H NMR spectra.
- 8b: ¹H NMR (CDCl₃) δ 1.32 (3H, s), 1.43 (3H, s), 2.60-2.92 (1H, m), 3.22 (1H, dd, $J=3$ and 9 Hz), 3.86 (1H, dd, $J=4$ and 12 Hz), IR (CHCl₃) 1740 cm⁻¹.
- 9a: ¹H NMR (CDCl₃) δ 1.43 (3H, d, $J=6$ Hz), 2.80-3.12 (1H, m), 3.70-4.31 (3H, m), 5.57 (1H, broad d, $J=10$ Hz), 5.82-6.00 (1H, m), IR (CHCl₃) 1738 cm⁻¹.
- 10: ¹H NMR (CDCl₃) δ 1.37 (3H, d, $J=6$ Hz), 1.92-2.19 (2H, m), 2.72-3.13 (3H, m), 3.42-3.72 (1H, m), 5.65 (1H, dd, $J=2$ and 11 Hz), 5.79-6.02 (1H, m), CI Mass m/z 128 (M⁺).
- 11a: ¹H NMR (CDCl₃) δ 1.37 (3H, d, $J=6$ Hz), 2.57-2.88 (1H, m), 3.00-3.22 (1H, m), 3.77 (1H, dd, $J=4$ and 12 Hz), 3.92-4.20 (1H, m), IR (CHCl₃) 1735 cm⁻¹, CI Mass m/z 156 (M⁺+1).
- 11b: ¹H NMR (CDCl₃) δ 1.00 (3H, t, $J=7$ Hz), 2.63-2.93 (1H, m), 3.10-3.32 (1H, m), 3.73-4.08 (2H, m), IR (CHCl₃) 1740 cm⁻¹, CI Mass m/z 170 (M⁺+1).
9. S. Pilard and M. Vaultier, Tetrahedron Lett., 1984, 25, 1555.

Received, 5th November, 1985