A PHOTOCHEMISTRY OF 1-(1-PHENYLVINYL)-3,4-DIHYDROISOQUINOLINE

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Abstract —— Irradiation of the 1-(1-phenylviny1)-3,4-dihydroisoquinoline (3) in benzene in the presence of xanthone gave the spirobenzylisoquinoline (4) in low yield. On the other hand irradiation of 3-hydrochloride in alcohol solvent (methanol, ethanol, and iso-propanol) afforded the hydroxyalkylated product (10), (13), and (14) in moderate yield, respectively.

In a previous paper, we reported preparation of the spirobenzylisoquinoline (2) from the 1-o-toluoyl-3,4-dihydroisoquinoline (1) by photochemical reaction. However the isolated yield of 2 was low. The Norrish type II split of the photoproduct (2) may occur as a secondary reaction. And the photochemical reaction of the 1-(1-phenylviny1)-3,4-dihydroisoquinoline (3) might yield the spirobenzylisoquinoline (4) without the Norrish type II split. Then we examined the photochemical reaction of 3. We wish to describe the conversion of 3 to 4 and the preparation of the hydroxyalkylated products which were encountered during the course of our studies.

Scheme 1

The 1-(1-phenylvinyl)-3,4-dihydroisoquinoline (3) was prepared from the corresponding phenylacetic acid (5). (Scheme 2) Treatment of 5 with 2 equiv. of lithium diisopropylamide (LDA) in tetrahydrofuran (THF) containing hexamethyl-phosphoric triamide (HMPA) followed by bubbling of formaldehyde gas gave the 2-(hydroxymethyl)phenylacetic acid (6) in 73% yield. Dehydration of 6 was effected by treatment with methanesulfonyl chloride in 1,8-diazabicyclo[5,4,0]undec-7-ene (DBU) to give the 2-(methylene)phenylacetic acid (7) in 52% yield. The acid (7) was converted to the corresponding amide (8) in 56% yield on treatment with oxalyl chloride followed by β-(3,4-dimethoxyphenyl)ethylamine. Cyclisation of 8 was carried out by heating with phosphorus oxychloride (2 equiv.) in benzene for 2 h to yield 3 as its hydrochloride in 86% yield.

Scheme 2

Irradiation of 3 with 200 watt high pressure mercury lamp using a vycol filter in benzene in the presence of xanthone at 10°C for 25 h gave the spirobenzylisoquino-line (4) in 6% yield und unknown reaction mixture. The structure of 4 was established by comparison with the spectral data of analogous compounds. Although the photochemical reaction of 3 was examined under many conditions, we could not get over 6% yield of 4. Next, the photochemical reaction of 3-hydrochloride in alcohol was carried out. Irradiation of 3-hydrochloride with the same mercury lamp in methanol at 10°C for 25 h gave the hydroxymethylated product (10) and the reduced product (2) in 29% and 6% yield, respectively. The starting material was recoverd in 9% yield. The structure of 10 was identified by an alternative synthesis starting from the 1-benzylisoquinoline (11).

benzindolizine (12) as a stereoisomeric mixture in 72% yield by treatment of methanesulfonyl chloride in pyridine followed by sodium borohydride reduction. Furthermore, we undertook the photochemical reaction of 3-hydrochloride in ethanol and iso-propanol. Irradiation of 3-hydrochloride in ethanol for 25 h gave 13 and 9 in 20% and 8% yield, respectively. In iso-propanol, irradiation of 3-hydrochloride afforded 14 in 61% yield.

We are continuing to explore mechanistic details of this photoreaction and application to other simpler 1-aza-1,3-diene compounds.

REFERENCE AND NOTES

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- All new compounds gave satisfactory elemental analyses and mass spectroscopic data. Some selected characteristics are listed below.
 ir (Nujol) cm⁻¹ 3400, 1655, 1605; ¹H-nmr (CDCl₃) δ 2.11 (3H, s, ArCH₃),

5.42 (1H, d, J=3Hz, olefinic proton), 6.46 (1H, d, J=3Hz, olefinic proton);

3: ir (CDCl₃) cm⁻¹ 1605; uv (MeOH) nm 315, 283; lH-nmr (CDCl₃) & 2.21 (3H, s, ArCH₃), 2.65 (2H, t, J=7Hz, C₄-H), 3.78 (2H, t, J=7Hz, C₃-H), 5.61 (1H, d, J=2Hz, olefinic proton); 5.87 (lH, d, J=2Hz, olefinic proton); 4: lH-nmr (CDCl₃) & 3.20 (2H, br s, ArCH₂C), 4.80 (lH, s, olefinic proton), and 5.13 (lH, s, olefinic proton); 9: ir (CHCl₃) cm⁻¹ 1621; lH-nmr (CDCl₃) & 1.46 (3H, d, J=7Hz, CH₃CHAr), 2.51 (3H, s, ArCH₃), 4.42 (lH, q, J=7Hz, MeCHAr); l0: ir (CHCl₃) cm⁻¹ 3200, 1622; lH-nmr (CDCl₃) & 2.20 (2H, q, J=5Hz, OCH₂CH₂-), 2.54 (3H, s, ArCH₃), 3.43-3.66 (2H, m, HOCH₂-), 4.68 (lH, t, J=5Hz, methine proton); l3: ir (CHCl₃) cm⁻¹ 3200, 1620; lH-nmr (CDCl₃) & 1.15 (3H, d, J=5Hz, CH₃CH(OH)-), 2.56 (3H, s, ArCH₃), 3.68-3.80 (lH, m, MeCHOH), 4.80 (lH, t, J=4Hz, CH₂CHAr); l4: ir (CHCl₃) cm⁻¹ 3200, 1620; lH-nmr (CDCl₃) & 1.27 and 1.30 (6H, each s, -CCCH₃), 2.53 (3H, s, CH₃Ar), 4.50 and 4.67 (lH, dd, J=3 and 7Hz, methine proton).

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