CHIRAL β-LACTAMS AS SYNTHONS. STEREOSPECIFIC SYNTHESIS OF A 6-EPI-LINCOSAMINE DERIVATIVE 1 祝古稀

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Abstract - A derivative of 6-epi-lincosamine has been prepared by a sequence of stereospecific steps from an optically active, cis- α -hydroxy- β -lactam. This β -lactam was obtained by an enantiospecific cycloaddition reaction between methoxyacetyl chloride, triethylamine and a Schiff base derived from benzylamine and an optically active aldehyde derived from D-galactopyranose.

祝古稀 Dedicated to Prof. Arnold Brossi on the happy occasion of his seventieth birthday.

A substituted β -lactam (e.g. 1) constitutes a densely functionalized molecule that can undergo a variety of stereospecific reactions including molecular rearrangements.² Additionally, the four-membered heterocyclic ring can be cleaved in several ways to produce various types of compounds such as derivatives of β -amino acids (2), α -hydroxyacids (3), α -amino acids (4), etc.³

In recent years we⁴ have studied different approaches to optically pure β -lactams of predictable absolute configuration. Independent work from Hoffmann-La Roche laboratories⁵ and our research group ⁶ has indicated that the reaction of an acid chloride (or equivalent) (8) with a Schiff base (7) in presence of triethylamine (or other tertiary amines) leads to a single, optically pure, cis stereoisomer of a β -lactam (9) if the Schiff base is derived

from an optically active aldehyde of type (5) and an achiral amine (6).

We⁷ initiated a study on the suitability of optically active β -lactams of type (9) as synthons for higher amino sugars found in nature. We wish to report here on our synthetic approach to isomers of lincosamine⁸ (10), an eight carbon amino sugar that is a part of the antibiotic lincomycin.

Our starting point was commercially available 9 1,2,3,4-di-O-isopropylidene- α -D-galactopyranose (11) which was oxidized with pyridinium chlorochromate to the known aldehyde 10 (12). Condensation with benzylamine in presence of molecular sieves converted 12 to the Schiff base (13) which was characterized by ir and mass spectral data and then used without further purification for β -lactam formation.

Reaction of methoxyacetyl chloride (14) with 13 in presence of triethylamine (TEA) in methylene chloride at low temperature led to a single cis- β -lactam in 50% yield which should correspond to 15 or 16 (Scheme 1). The strategy (Scheme 2) at this point was to hydrolyze the amide bond in the β -lactam and reduce the carboxy group to the desired methyl group (C-8) of lincosamine without affecting the stereochemistry of the methoxy group bearing carbon. Basic hydrolysis of 15 with sodium methoxide in methanol led to the desired β -lactam cleavage and produced the expected methyl ester (17) which was contaminated with the corresponding ethyl ester. ¹¹ This impurity was no hindrance in the next step which was the lithium aluminum hydride reduction to the primary alcohol (18) which could also be abtained directly from the β -lactam (15). The next step was to be the standard conversion of primary alcohol to its mesylate (19a). The density of functional groups in 18, however, favored the formation of the azetidine (19b) which was of no value to the proposed synthesis.

Scheme 2

15 or 17

To prevent azetidine formation, it was planned to prepare an N,N-dibenzyl version of 18. After some experimentation suitable conditions were found for benzylating (17) with benzyl bromide and obtain a crystalline product (20). Upon reduction with lithium aluminum hydride at 0 $^{\circ}$ C, the primary alcohol (21) could be obtained in good yield.

Treatment of 21 with mesyl chloride in pyridine at 0 °C, failed to produce the desired mesylate. The product of this reaction was found to be the primary chloride (22) (Scheme 3). Reduction of the chloromethyl group to a methyl group proved unexpectedly difficult. After many trial experiments it was found that the alkyl chloride

could be reduced in very good yield if a *clear solution* of lithium aluminum hydride in THF was used *in excess* (One mole of the hydride for each mole of the halide). ¹² A crystalline compound (23) was the product.

While the above chemical transformations were in progress, experiments were performed to determine the chirality induced during β -lactam formation. Following a published procedure¹³ (15 or 16) was heated under reflux with iodine and methanol to remove the isopropylidene protective groups. Subsequent oxidation with ruthenium tetroxide (RuCl₃ + NaIO₄) according to a literature method¹⁴ converted the sugar moiety to a carboxy function. Treatment with diazomethane provided a cis- β -lactam methyl ester (24) (Scheme 4).

Based on the stereospecific transformations conducted on 15 our final product should have the stereostructure (23) which corresponds to 25, a 6-epi-lincosamine derivative. To remove all doubts about the structure of this compound single crystal X-ray diffraction studies were conducted on two compounds: 21 and 23. The former proved unsuitable but the latter gave a structure (see the Pluto diagram 26)¹⁵ that was fully in agreement with the stereostructure and absolute configuration of 23.

In summary, we have developed a short and stereospecific synthesis of an optically active stereoisomer of lincosamine and indicated pathways to other isomers and analogs. Comparison with the substituted β -lactam of known absolute configuration that was available in our laboratory 16 established that the stereostructure (15) rather

than 16) was the correct designation of the enantiopure β -lactam obtained from the Schiff base (13). It Scheme 4

was noted that the correct configuration at C-4 of 15 can be predicted 17 on the basis of the absolute configuration of the asymmetric carbon next to the amino group in 13.

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