

CHIRAL β -LACTAMS AS SYNTHONS. STEREOSPECIFIC SYNTHESIS OF A 6-EPI-LINCOSAMINE DERIVATIVE¹ 祝古稀

Ajay K. Bose*, Chandra Mathur, Dilip R. Wagle, Raza Naqvi, and Maghar S. Manhas

Department of Chemistry and Chemical Engineering, Stevens Institute of Technology, Hoboken, NJ 07030, U.S.A.

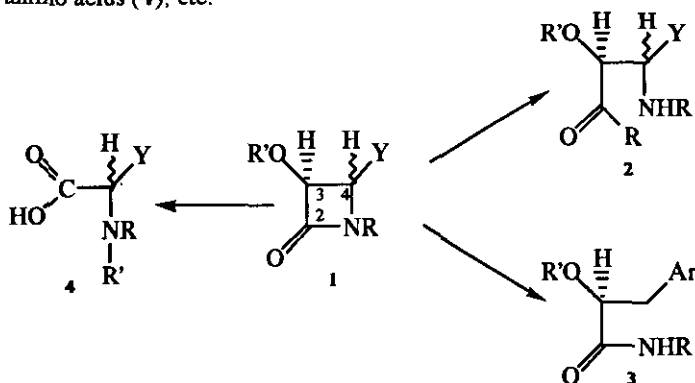
Zofia Urbanczyk-Lipkowska

Institute of Organic Chemistry, Polish Academy of Sciences, Warsaw, Poland

Abstract - A derivative of 6-epi-lincosamine has been prepared by a sequence of stereospecific steps from an optically active, *cis*- α -hydroxy- β -lactam. This β -lactam was obtained by an enantiospecific cycloaddition reaction between methoxyacetyl chloride, triethylamine and a Schiff base derived from benzylamine and an optically active aldehyde derived from D-galactopyranose.

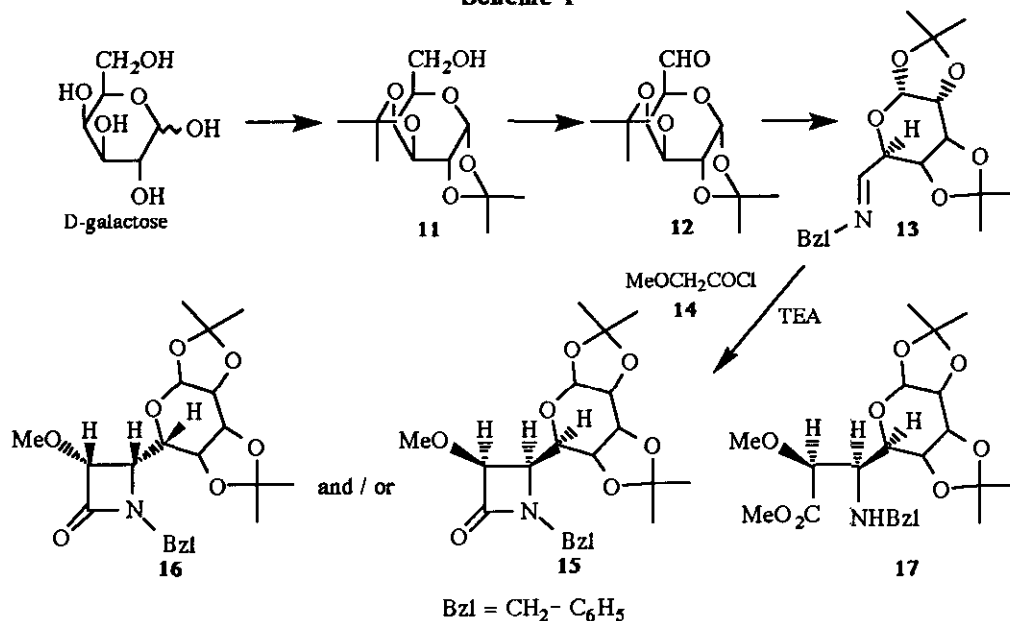
祝古稀 Dedicated to Prof. Arnold Brossi on the happy occasion of his seventieth birthday.

A substituted β -lactam (e.g. **1**) constitutes a densely functionalized molecule that can undergo a variety of stereospecific reactions including molecular rearrangements.² Additionally, the four-membered heterocyclic ring can be cleaved in several ways to produce various types of compounds such as derivatives of β -amino acids (**2**), α -hydroxyacids (**3**), α -amino acids (**4**), etc.³

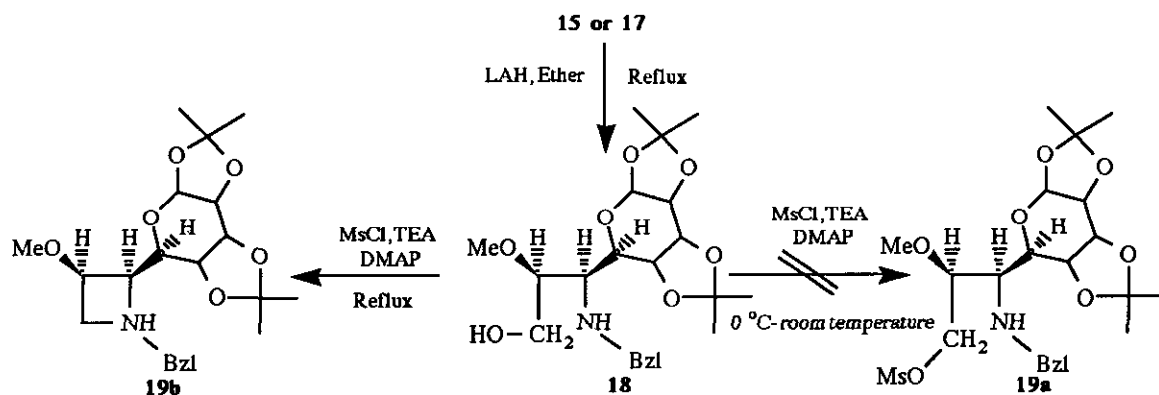


In recent years we⁴ have studied different approaches to optically pure β -lactams of predictable absolute configuration. Independent work from Hoffmann-La Roche laboratories⁵ and our research group⁶ has indicated that the reaction of an acid chloride (or equivalent) (**8**) with a Schiff base (**7**) in presence of triethylamine (or other tertiary amines) leads to a single, optically pure, *cis* stereoisomer of a β -lactam (**9**) if the Schiff base is derived

Scheme 1



Scheme 2

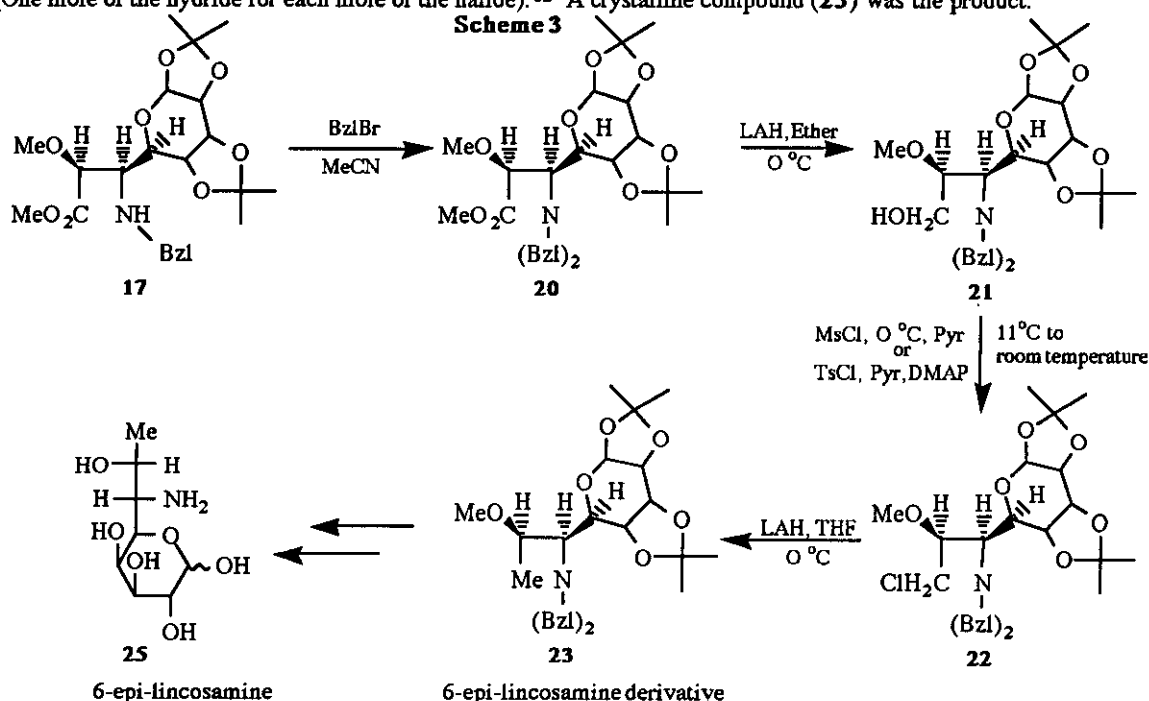


To prevent azetidine formation, it was planned to prepare an *N,N*-dibenzyl version of 18. After some experimentation suitable conditions were found for benzylating (17) with benzyl bromide and obtain a crystalline product (20). Upon reduction with lithium aluminum hydride at 0 °C, the primary alcohol (21) could be obtained in good yield.

Treatment of 21 with mesyl chloride in pyridine at 0 °C, failed to produce the desired mesylate. The product of this reaction was found to be the primary chloride (22) (Scheme 3). Reduction of the chloromethyl group to a methyl group proved unexpectedly difficult. After many trial experiments it was found that the alkyl chloride

could be reduced in very good yield if a *clear solution* of lithium aluminum hydride in THF was used *in excess* (One mole of the hydride for each mole of the halide).¹² A crystalline compound (**23**) was the product.

Scheme 3



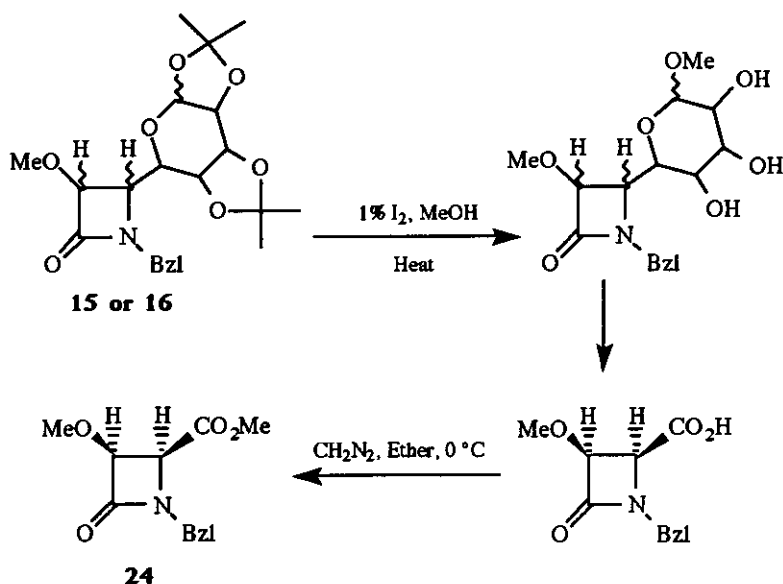
While the above chemical transformations were in progress, experiments were performed to determine the chirality induced during β -lactam formation. Following a published procedure¹³ (**15** or **16**) was heated under reflux with iodine and methanol to remove the isopropylidene protective groups. Subsequent oxidation with ruthenium tetroxide ($\text{RuCl}_3 + \text{NaIO}_4$) according to a literature method¹⁴ converted the sugar moiety to a carboxy function. Treatment with diazomethane provided a *cis*- β -lactam methyl ester (**24**) (Scheme 4).

Based on the stereospecific transformations conducted on **15** our final product should have the stereostructure (**23**) which corresponds to **25**, a 6-epi-lincosamine derivative. To remove all doubts about the structure of this compound single crystal X-ray diffraction studies were conducted on two compounds: **21** and **23**. The former proved unsuitable but the latter gave a structure (see the Pluto diagram **26**)¹⁵ that was fully in agreement with the stereostructure and absolute configuration of **23**.

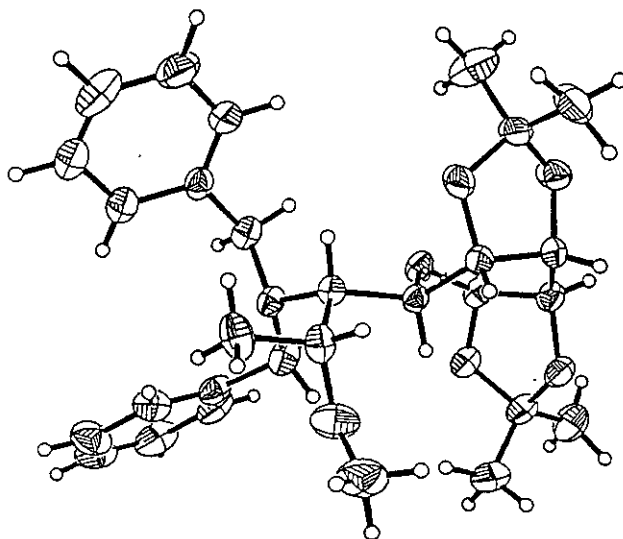
In summary, we have developed a short and stereospecific synthesis of an optically active stereoisomer of lincosamine and indicated pathways to other isomers and analogs. Comparison with the substituted β -lactam of known absolute configuration that was available in our laboratory¹⁶ established that the stereostructure (**15**) rather

than **16**) was the correct designation of the enantiopure β -lactam obtained from the Schiff base (**13**). It

Scheme 4



was noted that the correct configuration at C-4 of **15** can be predicted¹⁷ on the basis of the absolute configuration of the asymmetric carbon next to the amino group in **13**.



26

ACKNOWLEDGMENT

This research was supported by Stevens Institute of Technology and the National Science Foundation (INT-9116107).

REFERENCES

1. Studies on Lactams. Part 95; for Part 94 see A. K. Bose, B. K. Banik, S.N. Newaz, and M. S. Manhas *Synlett*, **1993**, 897.
2. For example see, (a) M. S. Manhas, S.G. Amin, and A. K. Bose, *Heterocycles*, **1976**, *5*, 669. (b) M. S. Manhas, D. R. Wagle, J. Chiang, and A. K. Bose, *Heterocycles*, **1988**, *27*, 1755.
3. B. K. Banik, M. S. Manhas, and A. K. Bose, *J. Org. Chem.*, **1993**, *58*, 307.
4. A. K. Bose, M. S. Manhas, J. M. van der Veen, S. S. Bari, and D. R. Wagle, *Tetrahedron*, **1992**, *48*, 4831.
5. C. Hubschwerlen, and G. Schmid, *Helv. Chem. Acta*, **1983**, *66*, 206.
6. A. K. Bose, M. S. Manhas, J. M. van der Veen, S. S. Bari, D. R. Wagle, V. R. Hegde, and L. Krishnan *Tetrahedron Lett.*, **1985**, *26*, 33. and subsequent papers in this series. Also see C. Palomo, F. P. Cossio, and C. Cuevas, *Tetrahedron Lett.*, **1991**, *32*, 3109 for the use of *N,O*-diprotected serinal.
7. C. Mathur Ph.D. Thesis, Stevens Institute of Technology, Hoboken, New Jersey, **1991**.
8. (a) G. B. Howorth, D. G. Lance, W. A. Szarek, and J. K. N. Jones, *Chem. Comm.*, **1969**, *47*, 75. (b) G. B. Howorth, W. A. Szarek, and J. K. N. Jones, *Chem. Comm.*, **1969**, 1339. (c) S. M. Danishefsky, E. Larson, and J. P. Springer, *J. Am. Chem. Soc.*, **1985**, *107*, 1274 and references cited therein.
9. Purchased from Pfanstiehl Laboratories.
10. D. Horton, V. Nakadate, and J. M. Tronchet, *J. Carbohydr. Res.*, **1968**, *7*, 56.
11. Apparently the sodium methoxide was contaminated with sodium ethoxide.
12. S. Krishnamurthy, and H. C. Brown, *J. Org. Chem.*, **1982**, *47*, 276.
13. W. Szarek, A. Zamojski, K. N. Tiwari, and E. R. Ison, *Tetrahedron Lett.*, **1986**, *27*, 3827.
14. E. C. Ashly, and A. B. Goel, *J. Org. Chem.*, **1981**, *46*, 3936.
15. The details of X-ray diffraction studies conducted by J. R. Ruble, T. T. Izard, and R. Naqvi will be published elsewhere.
16. D. R. Wagle, C. Garai, M. G. Monteleone, and A. K. Bose, *Tetrahedron Lett.*, **1988**, *29*, 1649.
17. A. K. Bose, J. F. Womelsdorf, L. Krishnan, Z. Urbanczyk-Lipkowska, D. C. Shelly, and M. S. Manhas, *Tetrahedron*, **1991**, *47*, 5379.

Received, 4th April, 1994