## SYNTHETIC STUDIES OF 1, 2, 3, 4-TETRAHYDRO-1, 3, 4-TRIOXO- $\beta$ -CARBOLINE ALKALOIDS I<sup>1</sup>

Hideharu Suzuki, Kayoko Shinpo, Towako Yamazaki, Sachiko Niwa, Yuusaku Yokoyama, and Yasuoki Murakami\*

School of Pharmaceutical Sciences, Toho University,

2-2-1 Miyama, Funabashi, Chiba 274, Japan

Abstract - A simple and efficient total synthesis of 1, 2, 3, 4-tetrahydro-1, 3, 4-trioxo- $\beta$ -carboline (1) was accomplished *via* C<sub>3</sub>-selective acylation of indole-2-carboxylate (5). On the course of this study, we found that the cyclization of *N*-(2-indolecarbonyl)glycine (8a) with PPA gave only an *N*-cyclized 6-membered ring (10a), whereas *N*-(2-indolecarbonyl)- $\beta$ -alanine (8b) gave a C<sub>3</sub>-cyclized 7-membered ring (9b) as a main product.

Recently 1, 2, 3, 4-tetrahydro-1, 3, 4-trioxo- $\beta$ -carbolines ( $\beta$ -carboline-triones) (1, 2) were isolated from the *Simaroubaceae* plant as minor alkaloids<sup>2a</sup> along with 4-oxygenated  $\beta$ -carbolines<sup>2b</sup>(3). These have a highly oxygenated C-ring which is a novel skeleton in natural  $\beta$ -carbolines, and are expected to have some biological activity, as some of 3 have shown biological activities. However, the activities have not been examined, because of their minor productions from nature. One of these (1) has been known compound derived from natural brevicolline (4) by CrO<sub>3</sub>-oxidation during the structural determination.<sup>3</sup>

<sup>†</sup> Dedicated to the memory of late Prof. Yoshio Ban.

In this paper we report the total synthesis of 1 starting from ethyl indole-2-carboxylate (5). We have already succeeded<sup>4</sup> in the synthesis of 4-methoxy- $\beta$ -carbolines (3, R<sub>2</sub> = CH<sub>3</sub>) starting from 5 via elongation of C<sub>2</sub>-substituent to 6 and cyclization toward the C<sub>3</sub>-position to 7.

Scheme II

We planned a total synthesis of 1 via the oxidation of 9a which was synthesized via cyclization of 8a (Scheme III). Röder<sup>5</sup> had reported that the cyclization of 8a and 8b with polyphosphoric acid (PPA) gave only C<sub>3</sub>-cyclized product (9a, 9b), respectively. When we treated 8a with PPA according to Röder's report, we obtained a single product possessing melting point (242-245°) near to Röder's one (237°), and the nmr spectrum showed a similar signal pattern to the reported value observed at 60 MHz nmr.<sup>6</sup> However, we have found that the real structure of the product is N-cyclized one (10a) by the following data (measured at 500 MHz nmr in DMSO-d<sub>6</sub> at 50°, Scheme IV): the C<sub>10</sub>-H corresponding to usual indolic C<sub>3</sub>-H was observed at 7.37 ppm as a sharp singlet, which was unexchangeable with D<sub>2</sub>O, and has a NOESY to C<sub>9</sub>-H and C-H COSY to tertiary C<sub>10</sub>-carbon at 111.7 ppm. So it should be sure that they<sup>6</sup> took the C<sub>6</sub>-H signal (8.37 ppm) for N-H one on the assignment of 60 MHz nmr spectrum. The cyclization reaction of the ethyl ester of 8a with MeSO<sub>3</sub>H (neat, 70°, 2.5 h) increased the yield of N-cyclized product (10a, 92%).

Scheme III

Thus we also re-examined the cyclization of 8b, which forms a 7-membered ring, and found that the two kinds of compounds were formed from 8b, being different from Röder's result. Surprisingly the major product (77%) was C<sub>3</sub>-cyclized one (9b), while the minor product was N-cyclized one (12%) (10b), being in contrast to the cyclization of 8a. Their structures were elucidated clearly by assigning the 7.48 ppm of indolic C<sub>3</sub>-H (10b) and 12.48 ppm of N-H (9b). (Scheme IV). The reason for this interesting difference in cyclization between 6-membered ring and 7-membered ring is unknown.

Other cyclization conditions of 8a and conversion of N-cyclized compound (10a) to  $C_3$ -cyclized product (9a), failed to give the desired  $C_3$ -cyclized product (9a). Hence we synthesized 9a via a more confident route described in scheme V.

a: (CH<sub>2</sub>CO)<sub>2</sub>O, AlCl<sub>3</sub> / ClCH<sub>2</sub>CH<sub>2</sub>Cl, room temperature, 10 h b: CuBr<sub>2</sub> [1.8 mol(0.9 eq.)] / AcOEt, reflux, 1 h c: hexamethylenetetramine / CHCl<sub>3</sub>, room temperature, 1 h d: c.HCl / EtOH, 55°, 1 h e: Et<sub>3</sub>N / EtOH, room temperature, 10 h

## Scheme V

The C<sub>3</sub>-selective Friedel-Crafts acylation<sup>7</sup> of **5** gave the 3-acetyl product (**11**), and selective bromination of the  $\alpha$ -carbon of the carbonyl group with CuBr<sub>2</sub> gave the bromoacetyl compound (**12**) in good yield. The reaction of **12** with hexamethylenetetramine, followed by hydrolysis with conc. HCl gave the  $\alpha$ -aminoketone hydrochloride(**14**). The treatment of **14** with Et<sub>3</sub>N gave unexpectedly the  $\beta$ -carboline-trione (**1**) without isolation of **9a**,<sup>8</sup> as a result of cyclization followed by spontaneous air oxidation. Consequently we accomplished a total synthesis of  $\beta$ -carboline-trione (**1**). Synthesized compound (**1**) was identical with the natural product (**1**).

During the pharmacological screening examination of the synthetic sample, the compound (1) was found to possess a weak cytotoxicity against the P-388 mouse leukemia cell (IC<sub>50</sub> =  $13.6 \pm 0.5 \,\mu\text{g}$  / ml), HOC-21 human ovarian cancer cell (IC<sub>50</sub> =  $25.2 \pm 1.3 \,\mu\text{g}$  / ml) and MKN-28 human cancer cell (from human stomach) (IC<sub>50</sub> =  $16.1 \pm 1.3 \,\mu\text{g}$  / ml). Now we are interested in the activities of Picrasidine-V (2) and other analogues of 1. The syntheses of these are now in progress.

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## REFERENCES AND NOTES

- This paper is part XXXVII of "Synthetic Studies on Indoles and Related Compounds."
   Part XXXVI: Y. Yokoyama, T. Matsumoto, and Y. Murakami, J. Org. Chem., 1995, 60, 1486.
- a) K. Koike, T. Ohmoto, and K. Ikeda, *Phytochemistry*, 1990, 29, 3060.
   b) T. Ohmoto and K. Koike, *Chem. Pharm. Bull.*, 1983, 31, 3198.
- 3. P. A. Vember, I. V. Terent'eva, and G. V. Lazur'evskii, *Khim. Prir. Soedin.*, 1967, 3, 249 (*Chem. Abstr.*, 1967, 67, 108816t).
- 4. H. Suzuki, Y. Yokoyama, C. Miyagi, and Y. Murakami, Chem. Pharm. Bull., 1991, 39, 2170.
- 5. J. Pigulla and E. Röder, Liebigs Ann. Chem., 1978, 1390.
- 6. Röder assigned the signals of 8.2 8.4 ppm as N-H's of indole and imido group.
- 7. Y. Murakami, M. Tani, K. Tanaka, and Y. Yokoyama, Chem. Pharm. Bull., 1988, 36, 2023.
- 8. The early stage of this reaction gave a complex mixture on tlc (presumably, 9a was formed at first, and oxidation occurred gradually on tlc). However, the reaction mixture gave a single spot of the target compound (1) on tlc few hours later.