SHORT STEP SYNTHESIS OF 4-SUBSTITUTED INDOLES USING PALLADIUM-CATALYZED C-C BOND FORMING REACTION

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Abstract----Enol triflate of *N*-tosyl-4-oxo-tetrahydroindole (2) was a good precursor for a synthesis of 4-substituted indoles. Various functional groups were introduced to form corresponding indoles using palladium-catalyzed carbonylation and a cross-coupling reaction followed by dehydrogenation by palladium on charcoal.

Indoles bearing substituents at the 4-position have attracted attention as important building blocks for biologically active compounds such as ergot alkaloids. Though a few methods for the preparation of the 4-substituted indoles have been reported, a simple and practical synthetic procedure is needed. Here we report an attempt to synthesize 4-substituted indoles using a palladium-catalyzed carbon-carbon bond forming reaction starting from enol triflate (3) as a substrate, which was obtained from readily available N-tosyl-4-oxotetrahydroindole (2).

Initially, we tried to synthesize 4-carbomethoxyindole using palladium-catalyzed carbonylation. When a toluene solution of enol triflate (3) (0.12 mmol), Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %), K₂CO₃ (3 eq.) and MeOH (10 eq.) was stirred at 60 °C for 2.5 h under carbon monoxide (1 atm), the desired carbonylative product (4) was obtained in 22% yield along with dimeric compound (6) (61% yield). The latter product can be obtained by the Heck reaction followed by isomerization of olefin.⁴ The reaction was carried out under various conditions

Dedicated to the memory of emeritus professor Dr. Yoshio Ban (Hokkaido University).

(Table I). The carbonylation occurred at room temperature in CH₃CN and DMF (Runs 3 and 4), and a good yield (81%) was attained in DMF. Dehydration of 4 with 10% Pd on charcoal smoothly proceeded in refluxing p-xylene for 14 h to give 4-carbomethoxyindole (5) in good yield. The result indicates that enol triflate (3) is a good precursor for the synthesis of 4-substituted indoles by a Pd-catalyzed carbon-carbon bond formation.⁵

Subsequently, a Pd-catalyzed cross-coupling reaction of 3 with 1-trimethylsilyl-2-tributylstannylacetylene (7) was carried out. A DMF solution of 3 (0.12 mmol) and 7 (1.05 eq.) was stirred in the presence of Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %) and LiCl (3 eq.) at room temperature for 16 h to afford 9a in low yield (5%) (Table II, Run 1). An improved yield (66%) was achieved in refluxing CH₃CN (Run 3) using PdCl₂(PPh₃)₂ (5 mol %) as a catalyst. When the reaction of 3 with β-tributylstannylstyrene (8) as an organostannyl reagent was carried out in DMF at 100 °C, 9b was produced in 67% yield (Run 5). The dihydroindoles (9a and 9b) were aromatized by 10% Pd on charcoal in refluxing adequate solvent to give corresponding indoles (10a and 10b).

trace

Table II Cross-coupling reaction of enol triflate

Run	RSnBu ₃	Solvent	Temp	Time	9 (%)	6 (%)	3 (%)
1 ^a 2	7	DMF DMF	room termperature	16 h 10 min	5 48	14 trace	33
3 4	8	CH ₃ CN CH ₃ CN	reflux 60 °C	20 min 8 h	66 trace	trace 22	_ 41
5		DMF	100 °C	7 min	67	trace	_

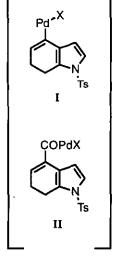
^a Pd(OAc)₂ (5 mol %) and PPh₃ (10 mol %) was used in the place of PdCl₂(PPh₃)₂

A carbonylative cross-coupling reaction is valuable for the introduction of carbonyl moiety. A DMF solution of enol triflate (3) (0.12 mmol) and 7 (1.05 eq.) was stirred at 80 °C in the presence of $PdCl_2(PPh_3)_2$ (5 mol %) and LiCl (3 eq.) under carbon monoxide giving desired 4-(trimethylsilyl)propynoyldihydroindole (12a) in only 2% yield with the coupling product (9a) in 14% yield (Table III, Run 1). In order to obtain the desired 12a in a higher yield, CH₃CN was used and the solution was stirred at a lower temperature (Runs 2-4). A maximum yield (57%) was achieved at 40 °C for this reaction. In the case of 1-trimethylsilyl-2-trisbutylstannylethylene (11) as an organostannyl compound, however, 4-(trimethylsilyl)propenoyldihydroindole (12b) was given as a single product in 68% yield (Run 5). The dehydrogenation of the carbonylative coupling products (12a and 12b) by 10% Pd on charcoal in refluxing p-xylene gave the corresponding indoles (13a and 13b) in 60% and 53% yields, respectively.

The results indicate that the formation of acylpalladium complex (II) can proceed at room temperature. The transmetalation of alkylpalladium complex (I) with an organotin compound requires higher temperature. However, the transmetalation of acylpalladium complex (II) with an organotin compound proceeds at a lower temperature than that of alkylpalladium complex (I).6

Table III Carbonylative cross-coupling reaction of enol triflate

Run	RSnBu₃	Solvent	Temp	Time	12 (%)	9 (%)
1	7	DMF	80 °C	15 min	2	14
2		CH ₃ CN	80 °C	55 mìn	23	42
3		CH ₃ CN	60 °C	40 min	38	34
4		CH ₃ CN	40 °C	3.5 h	57	16
5	11	CH₃CN	60 °C	7.5 h	68	_



If the Suzuki-coupling⁷ of enol triflate (3) with organoboron reagents is possible, it will be a strong tool to introduce aryl and vinyl groups to the 4-position of indole. When a dioxane solution of arylboronic acid (14) (1.05 eq.), which was prepared from 13 and B(OBu)₃ followed by treatment with 10% HCl, enol triflate (3) (0.12 mmol), $PdCl_2(PPh_3)_2$ (5 mol %) and K_2CO_3 (3 eq.) was refluxed for 10 min, the desired 15 was obtained in 52% yield. Carbonylative coupling of this boronic acid in refluxing dioxane for 15 min gave 17 in

43% yield. Conversion of 15 and 17 into indole derivatives (16 and 18) proceeded smoothly by 10% Pd on charcoal in refluxing p-xylene in 59% and 80% yields, respectively.

In summary, enol triflate (3) which was easily obtained from 2, was a good precursor for the synthesis of 4-substituted indoles using Pd-catalyzed carbonylation and Pd-catalyzed cross-coupling with organostannyl and organoboronic compounds. Further studies for the synthesis of 4-substituted indoles are now in progress.

REFERENCES AND NOTES

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- 3. M. Matsumoto and N. Watanabe, Heterocycles, 1984, 22, 2313.
- 4. An CH₃CN (1.2 ml) solution of 3 (0.12 mmol), Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %) and K₂CO₃ (2 eq.) was refluxed for 10 min to give dimeric compound (6) in 92% yield. The structure of 6 was determined by ¹H-nmr and COSY spectra.
- 5. N-Tosyl group could be deprotected by sodium hydroxide in alcoholic solvent if desired.
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