SYNTHETIC STUDY DIRECTED TOWARD NOVEL MULTI-LINKED HETEROCYCLES 1

Masanori Somei,* Yoshikazu Yamada, Keiichi Kitagawa, Katsuko Sugaya, Yayoi Tomita, Fumio Yamada, and Kyoko Nakagawa

Faculty of Pharmaceutical Sciences, Kanazawa University,

13-1 Takara-machi, Kanazawa 920, Japan

Abstract———2-Amino-4-(1-methylindol-3-yl)thiazole (11c) has a characteristic nucleophilic nature at the 5-position and add to the 4-position of acetylpyridinium acetate (13) producing 2-acetylamino-5-(1-acetyl-1,4-dihydropyridin-4-yl)-4-(1-methylindol-3-yl)-thiazole (1c). Its structure was established by X-ray single crystallographic analysis. Applying the results, simple syntheses of the related tris- (1a-b and 2-8) and tetrakis-linked heterocycles (9) were achieved.

A variety of heterocyclic compounds have biological activities.² In order to develop new lead compounds, we have designed a novel type of compounds which are consisted of plural heterocycles connected each other through single bond. We can classify these compounds as multi-linked heterocycles. In this communication, we wish to report the syntheses of tris- (1a-c and 2-8) and tetrakis-linked heterocycles (9) including indole, isoquinoline, pyridine, pyrrole, and thiazole as a component of heterocycles.

3-Chloroacetyl-1-methoxyindole (**10a**), prepared from 1-methoxyindole³ in 80% yield according to our procedure,^{3a} was converted to 4-(1-methoxyindol-3-yl)-2-aminothiazole (**11a**, 68%) by the reaction with thiourea. Similarly, 3-chloroacetylindole (**10b**) and -1-methylindole (**10c**) were converted to the corresponding **11b** (95%) and **11c** (94%). Interestingly, their 5-positions of 2-aminothiazole part were newly found to have a characteristic nucleophilic character. Thus, when **11a** was treated with a mixture of pyridine and acetic anhydride (Ac₂O) at room temperature, tris-linked heterocycle, 2-acetylamino-5-(1-acetyl-1,4-dihydropyridin-4-yl)-4-(1-methoxyindol-3-yl)thiazole (**1a**), was produced in 36% yield together with **12a** (55%). Under similar reaction conditions, **11b** and **11c** produced **1b** (65%) and **1c** (42%) in addition to **12b** (34%) and **12c** (46%), respectively.

The above results are remarkable findings because acetylpyridinium acetate (13, *in situ* formation upon mixing pyridine and Ac₂O) has not been reported to react at the pyridine part with nucreophiles except one case.⁴ Based on this character, 13 has long been utilized as acetylating reagent combining pyridine and Ac₂O. In order

to clarify the reactivity of 5-position of 2-aminothiazoles, reactions of 13 with 14a-f were examined. The results were exclusive formations of 2-acetylaminothiazoles (15a-f). Surprisingly, formations of 1 type tris-linked compounds were not detected at all in every case.

The structure of 1c was determined by X-ray single crystallographic analysis and the results are shown in Figure 1. 5-Bromo-2-bromoacetylpyrrole (16) afforded 2-aminothiazole (17a, 94%), which reacted with pyridine and Ac₂O to afford 2 (34%) and 17b (47%). DDQ oxidation of 2 successfully transformed 1,4-dihydropyridine part to pyridine and 2-acetylamino-4-(5-bromopyrrol-2-yl)-5-(pyridin-4-yl)thiazole (3) was produced in 93% yield. Similarly, 3-chloroacetyl-1-tosylpyrrole (18a) was converted to 2-aminothiazole (19a, 96%). The reaction of 19a with pyridine and Ac₂O afforded 2-acetylaminothiazole (20a, 94%) as a sole product. Under the same reaction conditions, *N*-unsubstituted pyrrole (19b), obtained in 72% yield by alkaline hydrolysis of 19a, generated a tris-linked heterocycle (4, 39%) in addition to 20b (59%).

Figure 1.
ORTEP Drawing of 1c

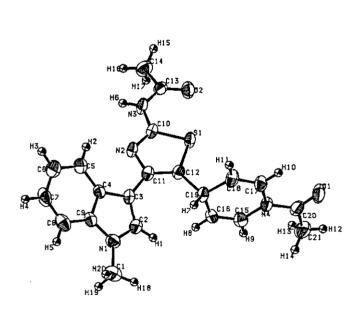


Table 1. δ-Value of 5-Position of 2-Aminothiazoles in the ¹³C-NMR Spectra

Compounds	δ-Value (ppm)				
21b	98.30				
19b	98.36				
17a	99.58				
11c	99.36				
11b	99.61				
11a	100.26				
And alternoon and alternoon					

Add	itio	n	to	13	OCCL	ITS.
-----	------	---	----	----	------	------

	_			
Compounds	δ-Value (ppm)			
14d	101.67			
19a	101.73			
14b	102.00			
14f	102.10			
21a	108.51			
Addition to 13 does not occur.				

The electron density of the 5-position seem to govern the reactivities of 2-aminothiazoles. Thus, the reaction with 13 occurred only in the cases where δ -values of the 5-position in their 13 C-NMR spectra, summarized in Table 1, are lower than 101 ppm. These results clearly suggest that 2-amino and 4-indolyl or 4-pyrrolyl groups on the thiazole nucleus cooperate to increase the electron density of the 5-position, and turn it to a soft nucleophile.

The above soft nucleophiles could also react with other iminium salts.⁵ For example, **11c** reacted with methyl nicotinate in Ac₂O to produce **5** (61%) and **12c** (38%), while the reaction with isoquinoline and Ac₂O afforded **6** (90%) and **12c** (8%). Further treatment of **5** with DDQ afforded **7** in 77% yield.

4-(Pyrrol-2-yl)-2-aminothiazole (21b), obtained from 18b through 21a in 57% overall yield by a sequential reaction with thiourea and subsequent hydrolysis, was an interesting substrate. When 21b reacted with isoquinoline and Ac_2O , the amino group and the 5-position of thiazole were completely inert and only α -position of pyrrole reacted to produce 8a (82%). The compound (8a) could further react with pyridine and Ac_2O to give 8b (45%) and the desired tetrakis-linked heterocycle, 2-acetylamino-5-(1-acetyl-1,4-dihydropyridin-4-yl)-4-[5-(2-acetyl-1,2-dihydroisoquinolin-1-yl)pyrrol-2-yl]thiazole (9, 40%).

In conclusion, we found that some of 2-aminothiazoles have an excellent nucleophilic nature at the 5-position and add even to acetylpyridinium acetate giving 1,4-dihydropyridines. Utilizing this novel reaction, simple synthesis method for various multi-linked heterocycles was developed.

REFERENCES AND NOTES

- 1. This is Part 80 of a series entitled "The Chemistry of Indoles". Part 79: M. Hasegawa, M. Tabata, K. Satoh, F. Yamada, and M. Somei, *Heterocycles*, 1996, 43, 2333. This is partly reported, Book of Abstracts, The 27th Congress of Heterocyclic Chemistry, Morioka, October, 1996, p. 184. All new compounds gave satisfactory spectral data and elemental analyses. 1a) mp 189-190°C; 1b) mp 223-224°C; 1c) mp 217-218°C; 2) mp 191.0-192.5°C; 3) mp 277-280°C; 4) mp 149-150°C; 5) mp 258-260°C (decomp.); 6) mp 249-251°C; 7) mp 245-247°C; 8a) mp 231-235°C; 8b) mp 254-256°C; 9) mp 225-230°C (decomp.); 11a) mp 107-108°C; 11b) mp 173-174°C; 11c) mp 154-155°C; 12a) mp 168-169°C; 12b) mp 243-244°C; 12c) mp 264.5-266.0°C; 15c) mp 215-216°C; 15d) mp 260-261°C; 15e) mp 196.5-197.0°C; 15f) mp 210.0-210.5°C; 16) mp 109.5-110.5°C; 17a) mp 154.5-156.0°C; 17b) mp 199-200°C; 18a) mp 124.0-124.5°C; 18b) mp 138-140°C; 19a) mp 179.5-180.5°C; 19b) mp 180-181°C; 20a) mp 181-183°C; 20b) mp 213-214°C; 21a) mp 174-176°C; 21b) mp 163-164°C.
- 2. A. R. Katritzky and C. W. Rees, "Comprehensive Heterocyclic Chemistry," Vol. 1, Pergamon Press, Oxford, 1984.
- a) M. Somei, H. Sato, N. Komura, and C. Kaneko, Heterocycles, 1985, 23, 1101; b) M. Somei, H. Ohnishi, and Y. Shoken, Chem. Pharm. Bull., 1986, 34, 677; c) M. Somei and T. Kawasaki, Heterocycles, 1989, 29, 1251; d) Review: M. Somei, J. Synth. Org. Chem., 1991, 49, 205; e) M. Somei, K. Yamada, M. Hasegawa, M. Tabata, Y. Nagahama, H. Morikawa, and F. Yamada, Heterocycles, 1996, 43, 1855 and references reported before 1996 are cited therein.
- 4. A. Treibs and A. Ohorodnik, Liebigs Ann. Chem., 1958, 611, 149.
- 5. Reactions with other various iminium salts were examined extensively. The results will be reported in due course.