SIMPLE TOTAL SYNTHESES OF (-)-ERGOT ALKALOIDS AND THEIR (+)-EN-ANTIOMERS BY A COMMON SYNTHESIS METHOD UTILIZING OPTICAL RESOLUTION¹

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Abstract———The first and simple total syntheses of (–)-isochanoclavine-I ((–)-1b), (–)-agroclavine ((–)-3), (–)-agroclavine-I ((–)-4), and (–)-norchanoclavine-I ((–)-5c) and their (+)-enantiomers are achieved from indole-3-carboxaldehyde (8) by a common synthesis method utilizing optical resolution. Absolute configuration of (–)-agroclavine-I is determined to be 5R and 10S for the first time. Preparations of both enantiomers of chanoclavine-I (1c) are also included.

The first enantioselective total synthesis of (+)-paliclavine was achieved in 1984 by Kozikowski and coworkers.^{2a} Since that time, several additional enantioselective total syntheses of ergot alkaloids have been reported.^{2b-d} Although these syntheses are wonderful and artistic, there remains conspicuous need of a simple and general method for preparing optically active ergot alkaloids.

We have realized simple total syntheses of various (\pm) -ergot alkaloids establishing a common synthesis method for them.³ The method was also effective for the first syntheses of (-)- and (+)-6,7-secoagroclavine ((-)- and (+)-1a)⁴ and for determining their absolute configurations.⁴ Thus, (\pm) -4,5-trans-5-(2-methyl-1-propenyl)-4-nitro-1,3,4,5-tetrahydrobenz[colindole $((\pm)$ -7a) was prepared through 9 and 10 in three steps in 34% overall yield from indole-3-carboxaldehyde (8, Scheme 1).³ After establishing interconversion method between (\pm) -7a and (\pm) -4,5-cis-isomer $((\pm)$ -11a),^{3f} both enantiomers of 7a and 11a were obtained using preparative chiral column chromatography.⁴ Then, (-)-7a and (+)-7a were converted in one pot operation to (-)- and (+)-6,7-secoagroclavine ((-)- and (+)-1a) by the reaction with MeMgI, followed by the reduction of the resultant 4-N-methylhydroxyamine ((-)- and (+)-12) with Zn-HCI. Alternative three steps synthesis route to (-)-1a and (+)-1a was also established from (-)-7a and (+)-7a, respectively, through (-)-5a and (-)-13a, and (+)-5a and (+)-13a.⁴

Now, we wish to report that the community of the above method is further heightened by its extention to the total syntheses of both enantiomers of isochanoclavine-I (1b),^{5a} chanoclavine-I (1c),^{5a},b agroclavine-I (4),^{5d} and norchanoclavine-I (5c)^{5e} only by changing reagents without altering the synthesis route³ as shown below.

Oxidation of (Z)-methyl on the isobutenyl group at the 5-position of (-)-1a was attained by the

treatment with SeO₂ in dioxane to produce (-)-isochanoclavine-I ((-)-1b, mp 183.5–185.0°C, $[\alpha]_D^{25}$ –212.8° (c = 0.50, pyridine)) in 29% yield together with unreacted (-)-1a in 41% yield. This regioselective functionalization can be explained by the coordination of the 4-methylamino group to selenium, putting SeO₂ close to the (*Z*)-methyl group.^{3f} Similar oxidation of (+)-1a afforded (+)-1b (mp 179.5–181.0°C, $[\alpha]_D^{25}$ +210.8° (c = 0.50, pyridine)) in 31% yield together with unreacted (+)-1a in 39% yield. Subsequent cyclization of (-)-1b and (+)-1b with POCl₃ in the presence of K₂C O₃ in MeCN proceeded smoothly to give (-)-agroclavine ((-)-3, mp 209–210°C, $[\alpha]_D^{19}$ –183.0° (c = 0.30, pyridine)) and (+)-agroclavine ((+)-3, mp 210–211°C, $[\alpha]_D^{20}$ +185.2° (c = 0.31, pyridine)) in 90 and 88% yields, respectively.

On the other hand, reduction of 4,5-*cis*-4-nitro compounds ((-)-11a and (+)-11a) with amaigamated Zn in methanolic HCl afforded (-)-4,5-*cis*-4-amino-5-(2-methyl-1-propenyl)-1,3,4,5-tetrahydrobenz[*cd*]indole ((-)-6a, mp 156–157°C, $[\alpha]_D^{25}$ –258.8° (c = 0.30, 99.5% EtOH)) and (+)-6a (mp 157.0–157.5°C, $[\alpha]_D^{25}$ +259.4° (c = 0.30, 99.5% EtOH)) in 95 and 96% yields, respectively. Treatment of (-)-6a and (+)-6a with methyl chloroformate (CICO₂Me) in the presence of Et₃N afforded the corresponding carbamates, (-)-14a (mp 169–170°C, $[\alpha]_D^{22}$ –224.8° (c = 0.30, 99.5% EtOH)) and (+)-14a (mp 169–

170°C, [α]D²² +226.8° (c = 0.30, 99.5% EtOH)) in 86 and 86% yields, respectively. Further reductions of (-)-14a and (+)-14a with LiAH₄ in refluxing THF afforded (-)-2a (caramel, [α]D²⁸ -282.8° (c = 0.30, 99.5% EtOH)) and (+)-2a (caramel, [α]D²⁷ +286.7° (c = 0.30, 99.5% EtOH)) in 99 and 98% yields, respectively. Regioselective allylic oxidation of (-)-2a with 30% SeO₂ on celite in dioxane and Et₃N afforded the corresponding (*Z*)-hydroxymethyl compounds, (-)-2b (mp 178–179°C, [α]D¹⁸ -302.2° (c = 0.30, 99.5% EtOH)) in 29% yield together with unreacted (-)-2a in 32% yield. Similar oxidation of (+)-2a afforded (+)-2b (mp 178.5–180.0°C, [α]D¹⁸ +301.0° (c = 0.30, 99.5% EtOH)) in 24% yield together with unreacted (+)-2a in 45% yield. It should be mentioned that the change in sign of optical rotation was observed in the case of ring closure of (-)-2b with POCl₃ in the presence of K₂C O₃ in MeCN giving (+)-agroclavine-I ((+)-4, viscous oil, [α]D¹⁷ +157.3° (c = 0.28, pyridine)) in 86% yield. (-)-Agroclavine-I ((-)-4, viscous oil, [α]D²² -157.3° (c = 0.27, pyridine)) was also obtained by the similar cyclization of (+)-2b in 83% yield.

We have already proved unequivocally that the absolute configuration of (–)-7 \mathbf{a} is 4R and 5R.⁴ Since the treatment of (–)-11 \mathbf{a} and (+)-11 \mathbf{a} with Et₃N afforded (–)-7 \mathbf{a} and (+)-7 \mathbf{a} in 79% and 80% yields, respectively, the absolute configuration of (+)-11 \mathbf{a} is proved to be 4R and 5S. The compound ((+)-11 \mathbf{a}) is chemically correlated with (+)-2 \mathbf{b} as described above. Consequently, the absolute configuration of a natural product, (–)-agroclavine-I ((–)-4), is determined to be 5R and 10S for the first time, though it had been deduced based on ¹H-NMR comparison studies.^{5d}

Oxidation of (–)-7a with t-BuO₂H in the presence of 5%SeO₂ on SiO₂⁶ in dioxane, followed by the reduction of the resultant mixture of (–)-7c and overoxidized aldehyde compound ((–)-7d) with NaBH₄, produced (–)-(£)-hydroxymethyl compound ((–)-7c, mp 133–134°C, $[\alpha]_D^{20}$ –113.3° (c = 0.29, 99.5% EtOH)) in 53% yield together with unreacted (–)-7a in 22% yield. Under similar reaction conditions, (+)-7a was converted to the corresponding (+)-(£)-hydroxymethyl compound ((+)-7c, mp 132–133°C, $[\alpha]_D^{21}$ +112.4° (c = 0.27, 99.5% EtOH) in 43% yield together with unreacted (+)-7a in 25% yield. Reduction of (–)-7c and (+)-7c with amalgamated Zn in methanolic HCl afforded (–)-norchanoclavine-I ((–)-5c, mp 207–208°C, $[\alpha]_D^{25}$ –178.0° (c = 0.30, pyridine)) and (+)-5c (mp 205.5–207°C, $[\alpha]_D^{24}$ +179.6° (c = 0.30, pyridine)) in 98 and 100% yields, respectively. Treatment of (–)-5c and (+)-5c with CICO₂Me in the presence of Et₃N produced (–)-13c (oil, $[\alpha]_D^{21}$ –80.4° (c = 0.23, 99.5% EtOH)) and (+)-13c (oil, $[\alpha]_D^{21}$ +80.3° (c = 0.23, 99.5% EtOH)) in 89 and 88% yields, respectively. (–)-Chanoclavine-I ((–)-1c, mp 221–222°C, $[\alpha]_D^{24}$ –241.6° (c = 0.18, pyridine)) and (+)-1c (mp 222–223°C, $[\alpha]_D^{23}$ +242.7° (c = 0.18, pyridine)) were obtained by reducing (–)-13c and (+)-13c with LiAH₄ in refluxing THF in 90 and 88% yields, respectively.

The data of optical rotations of our synthetic (–)-compounds were identical with those of natural alkaloids.^{5,7} In literatures,^{5c,8} agroclavine ((–)-enantiomer) had already been led to festuclavine, costaclavine, isosetoclavine, and setoclavine. Therefore, formal total syntheses of them are also

achieved.

In conclusion, both enantiomers of isochanoclavine-I (1b), agroclavine (3), agroclavine-I (4), and nor-chanoclavine-I (5c) are now readily available in less than nine steps from indole-3-carboxaldehyde (8) utilizing our common synthesis method. In seven steps (–)- and (+)-chanoclavine-I ((–)- and (+)-1c) can be obtained. Preparations of various optically active derivatives of ergot alkaloids are now in progress.

REFERENCES AND NOTES

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- 7. Reported data⁵ for optical rotations and melting points of natural alkaloids: (-)-isochanoclavine-I ((-)-1b): mp 181°C, $[\alpha]_D^{20}$ +216° (c = 0.50, pyridine); (-)-chanoclavine-I ((-)-1c): mp 220–222°C, $[\alpha]_D^{20}$ –240° (c = 1.0, pyridine); (-)-agroclavine ((-)-3): mp 210–212°C, $[\alpha]_D^{20}$ –183° (pyridine), (-)-agroclavine-I ((-)-4): amorphous powder, $[\alpha]_D^{22}$ –155° (c = 0.15, pyridine).
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