SYNTHESES OF NOVEL GROUP 15 AND 16 THIENO[2,3-b]-, THIENO[3,4-b]-, AND THIENO [3,2-b]-HETEROLES

Shuji Yasuike, Jyoji Kurita, and Takashi Tsuchiya*

Faculty of Pharmaceutical Sciences, Hokuriku University, Kanagawa-machi, Kanazawa 920-11, Japan

Abstract - Group 15 (P, As and Sb) and group 16 (S, Se, and Te) thieno[2,3-b]-(1), thieno[3,4-b]-(2), and thieno[3,2-b]heteroles (3) were prepared from the corresponding 1,4-dilithium intermediates, derived from (Z)-(β -bromo- β -trimethylsilylvinyl)thiophenes (6), (10), and (13) by treatment with tert-butyllithium, respectively. These thienoheteroles obtained are novel fused heterocyclic ring systems, except for thienothiophenes and thienoselenophenes.

Although a variety of new five-five fused aromatic heterocyclic ring systems containing one heteroatom in each ring have been prepared, most of these systems so far reported consist of the well-known simple heteroles such as furan, pyrrole, and thiophene rings. With regard to thienoheteroles represented by the structures (1)-(3), thienopyrroles, thienothiophenes, and thienoselenophenes 3,5 have been prepared. However, their synthetic routes are not always versatile and thienoheteroles containing an element other than the above elements (N, S, and Se) are little known. Therefore, we are interested in the synthesis of these fused systems containing Main Group heavier elements and report here versatile synthetic routes to group 15 (P, As, Sb) and group 16 (S, Se, Te) thieno[2,3-b]- (1), thieno[3,4-b]- (2), and thieno[3,2-b]- heteroles (3), each of which can be obtained from a corresponding common starting material.

Scheme 1

The thienoheteroles (1)-(3) obtained are novel heterocyclic ring systems except for thienothiophenes and thienoselenophenes.

We have previously reported that the 1,4-dilithium intermediate, derived from (Z)-\(\textit{B}\)-bromo-\(\textit{B}\)-trimethylsilylstyrene, reacted with dihalogeno metal reagents to give the corresponding 1-benzoheteroles \(^6\) and the 1-benzoheteroepines could be prepared from 1-bromo-1-trimethylsilyl-4-(o-bromophenyl)buta-1,3-diene \(\via \) the 1,6-dilithium intermediate.\(^7\) These results led us to examine the present synthetic routes.

3-lodothiophene (4A) was coupled with trimethylsilyl(TMS)acetylene in the presence of a catalytic amount of a mixture of bis(triphenylphosphine)palladium dichloride and copper (I) iodide\(^8\) to give the 3-(trimethylsilylethynyl)thiophene (5A) in \(\cap a\). 80\(^8\) yield. The ethynyl compound (5A) was hydraluminated with dissobutylaluminium hydride (DIBALH) followed by bromination with \(N\)-bromosuccinimide (NBS) to give the (Z)-3-(\(\text{B}\)-bromo-\(\text{B}\)-trimethylsilylvinyl)thiophene (6A) regio- and stereo-selectively in 75\(^9\) yield. The vinyl compound (6A) was treated with \(\text{tert}\)-butyllithium in dry ether at -80 \(^8\)C, then with a reagent [(PhSO_2)_2Se, (PhSO_2)_2Se, TeCl_4, PhPCl_2, PhAsCl_2 or PhSbCl_2], resulting in ring closure forming the 2-trimethylsilylthieno[2,3-b]heteroles (8a-1)\(^9\) in 45-65\(^9\) yields, presumably \(\nu ia\) the key 1,4-dilithium intermediate (7). The trimethylsilyl group in 8a-e was removed by treatment with tetrabutylammonium fluoride (TBAF) in tetrahydrofuran containing water to give the desired C-unsubstituted thieno[2,3-b]-

*) Reagent; (PhSO₂)₂S, (PhSO₂)₂Se, TeCl₄, PhPCl₂, PhAsCl₂ or PhSbCl₂

heteroles (1a-e)¹⁰ in 50-60% yields, however, the stibole (8f) was gradually decomposed under the reaction condition to afford no detrimethylsilylation product (1f).

The 2-bromothiophene derivative (6 B), prepared similarly from 2-bromo-3-iodothiophene (4 B)¹¹ through 5 B, also afforded 8 *via* the same dilithium intermediate (7), but in lower yields (20-40%).

Similarly, thieno[3,4-b]heteroles (2a-f) and thieno[3,2-b]heteroles (3a-f) could be synthesized. The starting dibromo compounds (10) and (13) were obtained in ca. 90% yields from 3-bromo-4-(trimethylsilyl)ethynyl-thiophene (12),which were prepared from 3-bromo-4-iodothiophene ¹² and 3-bromo-2-iodothiophene, ¹¹ respectively. Treatment of the dibromo compounds (10) and (13) with the reagents resulted in the formation of 2-trimethylsilylthieno[3,4-b]heteroles (11a-f) and 2-trimethylsilylthieno[3,2-b]heteroles (14a-f), which afforded the corresponding C-unsubstituted thienoheteroles (2a-f) ¹³ and (3a-f), ¹⁴ respectively, by treatment with TBAF.

Scheme 3

In all cases of the ring-closure reactions *via* dilithium intermediates, the formation of other possible thienoheteroles, such as thieno[3,4-b]heteroles (11) from 6A and 3-bromothieno[2,3-b]heteroles from 10, was not observed. These results show that the lithiation of the α -position of the thiophene ring proceeds smoothly as well as bromo-lithium exchange reactions, but the lithiation of the β -position dose not occur, and the exchange reaction occurs predominantly over the α -lithiation. The trimethylsilyl group is also essential for the present synthetic routes, by analogy with the syntheses of 1-benzoheteroles starting from phenylacetylene and of 1-benzoheteroepines from 1-phenylbut-1-en-3-ynes.

As stated in the introductory paragraphs, although the thienothiophenes (1a, 2a, and 3a) and thienoselenophenes (1b, 2b, and 3b) are known, the other thienoheteroles reported are new ring systems, whose structures were elucidated mainly by ¹H-NMR spectral comparison with known thienoheteroles. Further studies on the reactivities of these novel ring systems are now under investigation.

This work was supported by Grant-in-Aid for Scientific Research from the Ministry of Education, Science, Sports and Culture, Japan.

REFERENCES AND NOTES

- M. P. Cava and M. V. Lakshmikantham, "Comprehensive Heterocyclic Chemistry", Vol. 4, ed by A. R. Katritzky and C. W. Rees, Pergamon Press, Oxford, 1984, pp. 1037-1083.
- 2. J. Elguero and R. M. Claramunt, Adv. Heterocyclic Chem., 1978, 22, 183.
- 3. V. P. Litvinov and Ya. L. Goldfarb, Adv. Heterocyclic Chem., 1976, 19, 123.
- 4. R. L. P. de Jong and L. Brandsma, J. Chem. Soc., Chem. Commun., 1983, 1056; L. Brandsma and H. D. Verkruijsse, Syn. Communn., 1990, 20, 2275.
- A. Bugge, Acta Chem. Scand., 1969, 23, 1823; V. P. litvinov, Ja. L. Goldfarb, V. S. Bogdanov,
 I. P. Konjaeva, and A. N. Sukiasjan, J. Prakt. Chem., 1973, 315, 850.
- J. Kurita, M. Ishii, S. Yasuike, and T. Tsuchiya, J. Chem. Soc., Chem. Commun., 1993, 1309;
 Chem. Pharm. Bull., 1994, 42, 1437.
- 7. S. Yasuike, H. Ohta, S. Shiratori, J. Kurita, and T. Tsuchiya, J. Chem. Soc., Chem. Commun., 1993, 1817.
- 8. S. Takahashi, Y. Kuroyama, K. Sonogashira, and N. Hagihara, Synthesis, 1980, 627.
- Satisfactory elemental analyses and spectroscopic data were obtained for all new compounds reported.
 8a-8c: oil; 8d: mp 73-74 °C; 8e: mp 57-58 °C; 8f: mp 40-41 °C.
- 10. 1a: Colorless oil; 1b: pale yellow oil; 1c: pale yellow oil; 1d: mp 55.5-57 °C; 1e: mp 45.5-46.5 °C.
- 11. S. Gronowitz and B. Holm, Acta Chem. Scand., B, 1976, 30, 423.
- 12. P. Spagnolo, P. Zanirato, and S. Gronowitz, J. Org. Chem., 1982, 47, 3177.
- 13. 2a: Colorless oil; 2b: pale yellow oil; 2c: pale yellow oil; 2d: mp 69-71 °C; 2e: mp 53-54 °C; 2f: mp 51-53 °C.
- 14. **3a**: mp 52-53.5 °C; **3b**: mp 89-90 °C; **3c**: mp 95-96 °C; **3d**: mp 50-52 °C; **3e**: mp 36-38 °C; **3f**: colorless oit.