RACEMIC AND CHIRAL SYNTHESES OF SOME INDOLO[2,3- α]QUINOLIZIDINE ALKALOIDS THROUGH A "LACTIM ETHER ROUTE" †

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Abstract—The racemic and/or chiral syntheses of some Corynanthe-type indoloquinolizidine alkaloids and related bases, accomplished through a "lactim ether route" by the present authors' group, are reviewed with 107 reference citations. As a result of these syntheses, the structures and/or absolute configurations of the Neisosperma alkaloid ochromianine [(-)-35], the Ophiorrhiza alkaloid ophiorrhizine [(-)-46], and the Neisosperma and Ochrosia alkaloid ochropposinine [(-)-49] have been unequivocally established.

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I. Introduction

The indole alkaloids, a large and complex group of natural products bearing the indole or 2,3-dihydroindole (indoline) ring in their structures, include a number of indolo[2,3-a]quinolizidine alkaloids (e. g., 1), often referred to as indoloquinolizidine alkaloids. Although the chemical structures of most of the known indoloquinolizidine alkaloids have

[†]This review article is dedicated to Professor Koji Nakanishi (Columbia University) on the occasion of his 75th birthday.

been unequivocally established, there are some structural problems still remaining even in this new era of highly sophisticated spectroscopic studies. 1c Solutions to these problems could come with the racemic and/or chiral syntheses of compounds possessing the candidate structures, and the syntheses might be feasible via extensions of the "lactim ether route", 2,3 which has been developed in our laboratory as the best available vehicle for unified racemic and chiral syntheses of the structurally analogous benzo[a]-quinolizidine-type Alangium alkaloids (types $2-5^4$). 3,5 This synthetic route features the introduction of an adequate phenethyl skeleton into the key intermediate trans-lactim ether (7), which is readily available in any of the (\pm) -, 6 (+)-, 2c,7 and (--)-forms. 7

The aim of this article is to give an overview of the synthetic results obtained by us *via* such approach to the above problems, with coverage of the literature through the late part of 1996.

R = CH₂OH, CO₂H, or a heterocyclic ring

II. Indolo[2,3-a]quinolizidine

The tetracyclic base (1) is the simplest member of the indoloquinolizidine alkaloids: Its (-)-isomer, yet unnamed, has been isolated⁸ in partially racemized form from the leaves of a New Guinea tree, *Dracontomelum mangiferum* Bl., belonging to the family Anacardiaceae. Although more than 40 synthetic routes to this parent framework (1) had been reported,^{9,10} the feasibility of the above "lactim ether route" approach was tested by Fujii's group in the racemic synthesis of 1 (Scheme 1).¹¹

Scheme 1

Thus, alkylation of the lactim ether (11), obtained in 84% yield from 2-piperidone (10) according to the literature procedure, ^{2a,12} with 3-(chloroacetyl)indole ¹³ in DMF at 60°C in the presence of KBr for 24 h furnished the lactam ketone (12) in 71% yield. In the absence of KBr, the progress of alkylation was extremely slow, suggesting the intermediary formation of 3-(bromoacetyl)indole in the above alkylation. Reduction of 12 with NaBH₄ in aqueous EtOH at rt for 20 h gave the lactam alcohol (14) (96% yield), which was then hydrogenolyzed (10% Pd-C/H₂, 1 atm, 20°C, 50 min) in EtOH containing a small amount of 70% aqueous HClO₄ to afford the known lactam (13) in 96% yield. The hydroxy group of 14 at the indolylcarbinyl carbon was found to be considerably

reactive. On treatment with EtOH in the presence of a catalytic amount of HCl at 25°C for 1 h, 14 produced the ethoxy derivative (17) in 94% yield. Conversely, treatment of 17 with H₂O in MeCN containing a catalytic amount of HCl at 17°C for 1 h reproduced the lactam alcohol (14) in 86% yield. These transformations reflect the acid-catalyzed formation of the reactive "3-indolylcarbinyl system" (15 \in 16) from 14 and from 17. Catalytic hydrogenolysis of 17 to give 13 was smoothly effected under reaction conditions similar to those employed for the above conversion of 14 into 13, but the result was less satisfactory when the reaction was carried out in the absence of HClO₄. Bischler–Napieralski cyclization of 13 (POCl₃, boiling benzene, 75 min)⁹ and reduction of the resulting quaternary iminium salt (NaBH₄, MeOH, 1 h)⁹ provided the desired tetracycle (1) in 95% overall yield from 13. Thus, this "lactim ether route" may be of value, not only in preparing the alkaloid material (1) in an acceptable overall yield but also in the design and execution of the total syntheses of analogous alkaloids carrying substituents in ring A and/or ring D (vide infra). Interestingly, Govindachari's group has synthesized (±)-vincamine [(±)-6] via an analogous "lactim ether route". 15

III. Dihydrocorynantheol and Dihydrocorynantheine

(-)-Dihydrocorynantheol (corynan-17-ol) [(-)-25] has been isolated from the bark 16 and stem bark 17 of Aspidosperma marcgravianum Woodson (Apocynaceae); the bark of A. auriculatum; 18 the leaves of Amsonia tabernaemontana Walt. (Apocynaceae); 19 the leaves of Mitragyna parvifolia (Roxb.) Korth (Rubiaceae); 20 the bark of Hunteria zeylanica (Apocynaceae); 21 the trunk bark of Ochrosia moorei (Apocynaceae); 22 the root bark and trunk bark of Aspidosperma marcgravianum; 23 the roots of Rhazya stricta Decsne. (Apocynaceae); 24 the stem bark of Strychnos johnsonii Hutch et M. B. Mass (Loganiaceae); 25 the stem bark of Ochrosia alyxioides Guillaumin; 26 and the aerial part of Aspidosperma oblongum. 27

Although racemic dihydrocorynantheol [(\pm)-25] had been synthesized by many research groups, 28 its racemic synthesis from (\pm)-7 was tested by Fujii's group 11 in order to evaluate the applicability of the "lactim ether route". Thus, treatment of (\pm)-7 with 3-(chloroacetyl)indole 13 in DMF at 60°C in the presence of KBr for 38 h gave the lactam ketone [(\pm)-18] in 70% yield (Scheme 2). Reduction of (\pm)-18 with NaBH₄ (EtOH, 22–25°C, 3 h) furnished a diastereomeric mixture of the lactam alcohol [(\pm)-21] in 70% yield. The diastereomeric mixture [(\pm)-21] was then submitted to catalytic hydrogenolysis (10% Pd-C/H₂, EtOH, 1 atm, 21–22°C, 1 h) in the presence of a small amount of 70% aqueous HClO₄, affording the (\pm)-trans-lactam ester [(\pm)-22] in 74% yield. The lactam ester [(\pm)-22] has been converted into (\pm)-25^{28b} and (\pm)-dihydrocorynantheine [(\pm)-27]^{29,30} through the quaternary iminium salt [(\pm)-23] and the tetracyclic ester [(\pm)-26]. Consequently, the above results [(\pm)-7 \rightarrow (\pm)-18 \rightarrow (\pm)-21 \rightarrow (\pm)-22] imply that an alternative racemic synthesis of each of these Corynanthe-type alkaloids has now been completed in a formal sense.

Scheme 2

As regards the synthesis of (-)-25, semisynthetic (-)-25 has been prepared from various alkaloids such as dihydrocorynantheine (27), 10h , 31 geissoschizol, 32 quinine, 33 ajmalicine, 34 guettardine, 35 and yohimbine. 36 The total syntheses of (-)-25 by starting from chiral compounds, such as (R)-1,2-isopropylideneglyceraldehyde 37 and (4S,5R)-4-(2-benzyloxyethyl)-5-ethyl-3,4,5,6-tetrahydro-2-pyrone, 36 and by utilizing an asymmetric synthesis process 38 have also been accomplished. The latter asymmetric total synthesis 38 also allows entry into dihydrocorynantheine (27).

An alternative total synthesis of (-)-25, adaptable to its preparation on a gram-size scale, has been achieved by Fujii's group *via* a "lactim ether route" (Scheme 2).³⁹ The initial step was coupling of the lactim ether [(+)-7] with 3-(chloroacetyl)indole, which proceeded in DMF at 60°C in the presence of KBr for 72 h to give the lactam ketone [(+)-18] in 68% yield. Treatment of (+)-18 with POCl₃ in boiling toluene for 3 h afforded the oxazolium chloride (19), which was characterized as the crystalline perchlorate salt

[(+)-20]. The crude chloride salt (19) was then reduced by catalytic hydrogenation (Pt/H₂, EtOH, 1 atm, rt, 3 h) to furnish the lactam [(+)-22] in 52% overall yield [from (+)-18]. Conversion of (+)-22 into the tetracyclic ester [(-)-26] through the quaternary iminium salt (23) [characterized as the crystalline perchlorate salt [(+)-24]] was effected in 91% overall yield by means of Bischler-Napieralski cyclization (POCl₃, boiling toluene, 1 h) followed by catalytic hydrogenation (Pt/H₂, EtOH, 1 atm, rt, 1 h). On reduction with LiAlH₄ in THF at rt for 30 min, (-)-26 produced the desired alkaloid [(-)-25] in quantitative yield.

IV. 3,4,5,6-Tetradehydro-17-hydroxycorynanium

The gram-scale total synthesis of (-)-dihydrocorynantheol [(-)-25] through the "lactim ether route" described above (Section III) allowed Fujii's group to make their entry

Scheme 3

into the synthesis of 3,4,5,6-tetradehydro-17-hydroxycorynanium [(+)-32], the zwitterionic structure assigned²³ to an alkaloid from *Aspidosperma marcgravianum* Woodson (family Apocynaceae).

Thus, selective reduction of (+)-22 with LiBH₄ in boiling THF for 1 h gave the lactam alcohol [(+)-28] (84% yield), which was then converted into the acetate [(+)-30] (100% yield) by treatment with acetic anhydride in pyridine at rt for 1.5 h (Scheme 3). Bischler-Napieralski cyclization of (+)-30 with POCl₃ in boiling toluene for 1 h, followed by hydrolysis of the ester group with K₂CO₃ in aqueous MeOH at rt for 3 h, produced a tetracyclic alcohol, which was isolated in the form of the perchlorate salt [(+)-29] (86% yield). On treatment with palladium black and maleic acid in boiling H₂O for 8 h, (+)-29 afforded the dehydrogenated product [(+)-31] in 91% yield. Alternatively, dehydrogenation of synthetic (-)-25 with lead tetraacetate in AcOH at rt for 2 h or with palladium black and maleic acid in boiling H₂O for 9 h also gave (+)-31 (in 8% or 72% yield, respectively), which reverted to (-)-25 in quantitative yield when reduced with NaBH₄ in MeOH at rt for 1 h.

Finally, treatment of (+)-31 with aqueous NaOH in EtOH furnished the target anhydronium base [(+)-32], which was characterized as a hydrate [(+)-32·2/5H₂O] [mp 161–162.5°C (decomp); $[\alpha]_D^{18}$ +50.0° (c 0.53, MeOH)]. Unfortunately, however, the ¹H NMR spectral data and the sign of specific rotation for the synthetic (+)-32 were in disagreement with those reported²³ for a natural sample, leaving the chemistry of this A. marcgravianum alkaloid incomplete.

V. Ochromianine

In 1974, Koch and co-workers⁴¹ reported the isolation of (-)-ochromianine, a new Corynanthe-type alkaloid, from the bark of a New Caledonian plant named Ochrosia miana H. Bn. ex Guill. (family Apocynaceae) at that time. The name of the plant was later revised to Neisosperma miana (Baillon ex-White) Boiteau.⁴² The French research group deduced the structure and absolute configuration of (-)-ochromianine to be (-)-35 (11-methoxydihydrocorynantheol) on the basis of spectral analysis as well as chiroptical and biosynthetic rationales.⁴¹ The correctness of this structure assignment was confirmed by Fujii's group as a result of the racemic and chiral syntheses of the candidate structure (35), which followed a "lactim ether route" as delineated in Scheme 4.⁴³

The racemic synthesis of 35 started with coupling of (\pm) -7 with 3-chloroacetyl-6-methox-yindole to give the lactam ketone $[(\pm)$ -33]. Conversion of (\pm) -33 into the lactam $[(\pm)$ -37] through the oxazolium salt (34), cyclization of (\pm) -37 to form the tetracyclic ester $[(\pm)$ -36], and LiAlH₄ reduction of the ester group to give (\pm) -35 were carried out as in the case of the above synthesis of dihydrocorynantheol (25) (Section III).

For the chiral synthesis of (-)-35, a parallel sequence of conversions starting with (+)-7 was followed. The UV, IR, and ¹H NMR spectra and MS of the synthetic (-)-35 [mp 162-

169°C (decomp); $\left[\alpha\right]_{D}^{25}$ -28.0° (c 1.00, EtOH)] were virtually identical with those of natural (-)-ochromianine, and the chiral identity of (-)-35 with the alkaloid was shown by their virtually identical CD spectra.

Scheme 4

VI. Ophiorrhizine

In 1992, Arbain et al.⁴⁴ reported the isolation of a new C₁₉ pentacyclic quaternary indole alkaloid, named ophiorrhizine, from the fresh aerial parts of the medicinal plant Ophiorrhiza major Ridl. (family Rubiaceae). They established the structure and relative stereochemistry of ophiorrhizine as 46 on the basis of its spectral properties and X-ray molecular structure.⁴⁴ However, its absolute stereochemistry has only been inferred to be (-)-46 from the negative sign and magnitude of the specific rotation of this alkaloid.⁴⁴

The correctness of this inference was verified by Fujii's group as a result of the chiral synthesis of the target compound with the candidate structure [(-)-46]. The synthesis featured an adaptation of their favorite "lactim ether route" $[(+)-7\rightarrow(+)-38\rightarrow(+)-39\rightarrow(+)-42\rightarrow(-)-41\rightarrow(-)-40\rightarrow43\rightarrow(-)-44\rightarrow(-)-45\rightarrow(-)-46]$, as shown in Scheme 5, and the reaction sequence from (+)-7 to the tetracyclic alcohol [(-)-40] was similar to

those employed for the syntheses of (-)-25 (Section III), (-)-35 (Section V), and (-)-ochropposinine [(-)-49] (Section VII). The route beyond (-)-40 included treatment of (-)-40 with p-toluenesulfonyl chloride in pyridine at 4°C for 24 h and heating of the resulting O-tosyl derivative (43) in boiling DMF for 30 min, which gave the 4,17-cyclocorynanium tosylate [(-)-44] in 73% overall yield [from (-)-40]. Anion exchange of (-)-44 was effected with Amberlyst A-26 (Cl⁻) in aqueous MeOH to furnish the chloride salt [(-)-45] in 96% yield.

Scheme 5

Finally, debenzylation of (-)-45 by means of catalytic hydrogenolysis (10% Pd-C/H₂, EtOH, 1 atm, rt, 6 h) gave the target compound [(-)-46·H₂O] [mp 282-285°C (decomp); $[\alpha]_D^{30}$ -102° (c 0.209, MeOH)] in 93% yield. The UV (MeOH), IR (KBr), ¹H NMR (CD₃OD),

and ¹³C NMR (CD₃OD) spectra and TLC mobility of the synthetic (-)-46·H₂O were virtually identical with those of natural (-)-ophiorrhizine. More importantly, the chiral identity of the synthetic (-)-46·H₂O with the alkaloid was established on the basis of the same sign of their specific rotations in MeOH as well as their virtually superimposable CD spectra in MeOH.

VII. Ochropposinine

In 1972, Peube-Locou et al.⁴⁶ reported the isolation of (-)-ochropposinine, a member of the indoloquinolizidine alkaloids, from the trunk bark of Neisosperma oppositifolia (Lamarck) Fosberg et Sachet (Apocynaceae).⁴⁷ Later on, two other research groups also reported isolations of this alkaloid from the same plant.⁴⁸ Furthermore, another species of the same genus, Neisosperma glomerata, and two species of a closely related genus, Ochrosia vieillardii and Ochrosia moorei, have so far been found to contain (-)-ochropposinine, together with many other indole alkaloids.^{22,49}

Scheme 6

The structure including the absolute configurations of three stereogenic centers of (-)-ochropposinine has been proposed to be (-)-49 (10,11-dimethoxydihydrocorynantheol) on the basis of a combination of spectral analysis and a biosynthetic rationale.^{22,46b} Une-

quivocal confirmation of the correctness of this proposal came with the syntheses of the racemic and chiral target molecules [(\pm)-49 and (-)-49] by Fujii's group, which were also accomplished through their favorite "lactim ether route" ($7\rightarrow47\rightarrow48\rightarrow51\rightarrow50\rightarrow49$) as shown in Scheme 6.⁵⁰

VIII. Conclusions

The syntheses of the candidate target molecules described above enhance the usefulness of our "lactim ether route", originally designed for unified racemic and chiral syntheses of the benzo[a]quinolizidine-type Alangium alkaloids (types 2–5),3,5 for racemic and chiral syntheses of Corynanthe-type indoloquinolizidine alkaloids. Interestingly, the chiral syntheses of dihydrocorynantheol [(-)-25], ochromianine [(-)-35], ophiorrhizine [(-)-46], and ochropposinine [(-)-49] described above are equivalent to chemical correlations of these alkaloids with the Cinchona alkaloid cinchonine [(+)-8], because the lactim ether [(+)-7] employed as a starting material can be prepared from (+)-8 through its degradation product, cincholoipon ethyl ester [(+)-9].2c,51 It is hoped that this review has quoted vivid examples in support of the statement3b that chemical synthesis can still be an important and powerful tool for structure elucidation of natural products (particularly of those isolated only in minute quantity, unstable, and hard to crystallize) even in this new era of highly refined spectroscopic studies.

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