6,8-Dichloro-4-epoxyethyl-2,3-trimethylenequinoline (7).-A soln of 1 g (0.026 mole) of NaBH4 in 10 ml of H2O and 7 ml of 2 N NaOH, was added dropwise over 10 min to a stirred suspension of 4.65 g (0.013 mole) of nearly pure 6 (above) in 50 ml of MeOH. Stirring for an addl 1.5 hr, cooling for 15 min, filtering, and washing with MeOH gave 3.42 g (94.5%) of 7 (mp 134-139°); recrystd from Et₂O-hexane, mp 144-145°; ir (cm⁻¹) 2960, 2980, 3100, none for C=O; nmr (CDCl₃), 8.02 (1 H, d), 7.71 (1 H, d), 4.26 (1 H, m), 3.10 (5 H, m), 2.17 (2 H, quintuplet). Anal. (C₁₄H₁₁Cl₂NO) C, H, N.

5,7-Dichloro-2,3-dihydro-1H-cyclopenta [b] quinoline-9-(α -din-butylaminomethyl)methanol·HCl (1).—A suspension of 3.6 g of 7 in 12 ml of Bu₂NH was stirred for 4.5 hr at 105-110°, monitoring disappearance of 7 (4 hr) by tlc (silica gel G, 1:1 Et₂Ohexane). After evapn in vacuo of Bu₂NH (60°) the oil (5.1 g), dissolved in 150 ml of Et₂O, was treated with increments of Et₂O·HCl, each sufficient to give 0.2-0.4 g of 1 (each fraction being washed with $\rm Et_2O$). Fractions 1-4 contd decreasing amts of $\rm Bu_2NH\cdot HCl$; and 5-8 were largely 1 (2.65 g). Repeated recrystn from EtOH-Et₂O gave 0.5 g, light tan, mp 160-162° dec; ir (cm⁻¹) 3440, 3220 (OH), 2960, 2940, 2880 (CH), 2670, 2620, 2530 (NH). Anal. (C₂₂H₃₀Cl₂N₂O·HCl) C, H, N, Cl.

Incidental and Preliminary Experiments. Attempts to add 2-PyLi and MeLi to the 2,3-trimethylenecinchoninic acids were unsuccessful, presumably because of steric interference of the 3-CH₂ group and/or the activity of the 2-CH₂ hydrogens (cf.

2,3-Trimethylenecinchoninic acid HCl (11), pptd from Et₂O, mp 252-255° dec, was treated with PCl₅ (steam bath for 30 min, addn of C6H6, and reflux for 2 hr), giving a ppt presumed to be the acid chloride ·HCl (12) (sublimed, 8%, mp 245° dec).

2,3-Trimethylenecinchoninamide (13) was prepd from 12 by treatment with H₂O-NH₃; crystd from EtOH, mp 276-277°; ir (cm^{-1}) 3330 (s), 3140 (s) (NH_2) , 1688 (C=O). Anal. $(C_{13}$ -H₁₄N₂O) C, H.

4-Bromoacetyl-2,3-trimethylenequinoline · HBr (14).—CH₂N₂-Et₂O with 3 g of 12 (overnight) gave orange cubes of diazo ketone. Treatment with 10 ml of 48% HBr-H₂O gave 14; crystd from EtOH; 2.1 g (70%); mp 208° dec ; ir $(\text{cm}^{-1}) 1730 (C=O)$,

2500 (NH). Anal. (C₁₄H₁₈Br₂NO) N. Derivatives of 2,3-trimethylene-4-quinolones were made by the action of the appropriate aniline on ethyl cyclopentanone-2carboxylate, cyclizing at 250°, and crystn from EtOH: 3b,13 15, (a) 6,8-Cl₂, 26%, mp 305-307° (b) cyclization by refluxing Ph₂O, 8-Cl, 21%, mp 269–270° [Anal. ($C_{12}H_{10}CINO$) C, H, N]; 20, 8-F, 15%, mp 292–293° [Anal. ($C_{12}H_{10}FNO$) C, H, N].

4-Bromo-2,3-trimethylenequinolines were prepd by treating the quinolone¹³ with POBr₃ at 120°; crystd from EtOH: **21** (parent compd), 50%, mp 72-73° [Anal. (C₁₂H₁₀BrN), C, H, N]; **22**, 6,8-Me₂, from **16**, 69%, mp 124-125° [Anal. ($C_{14}H_{14}BrN$) C, H].

4,6,8-Trichloro-2,3-trimethylenequinoline (23) was prepd by refluxing POCl₃ on 15, crystd from EtOH, 80%, mp 160-162°. Anal. (C₁₂H₈Cl₃N) C, H, N.

Attempted preparation of 4-lithio-2,3-trimethylenequinolines from 21 and 22 by BuLi and addns to 2-pyridaldehyde were unsuccessful, presumably because of the activities of the 2-CH₂

$N,N^1-\alpha,\omega$ -Alkylenebis(nitroacetamides)

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Some bis(nitroacetamides) with the general structure 1 were required for screening as antispermatogenic agents. The amides were readily prepared by heating the appropriate amine with the desired nitro ester without solvent and recrystallizing the resulting solid from a suitable solvent.

The compds prepared are listed in Table I. While

TABLE I						
			R		\mathbf{R}	
O_2 NCCONH(CH ₂) $_n$ NHCOCNO $_2$						
$O_2NCCONH(CH_2)_nNHCOCNO_2$						
			R		R	
			10	1	10	
				•		
	\mathbf{R}	n	Yield, $\%$	Mp, °C	Rxt solv	${ m Formula}^a$
1	H	6	34.1	143 - 144	$\mathrm{CH_3CN}$	$C_{10}H_{18}N_4O_6$
2	H	8	50.3	147 - 148	$95\%~{ m EtOH}$	$C_{12}H_{22}N_4O_6$
3	CH_3	2	12.8	183 - 185	$\mathrm{CH_3CN}$	$C_{10}H_{18}N_4O_6$
4	CH_3	3	21.7	105 - 108	$C_6H_6-n-C_6H_{14}$	$C_{11}H_{20}N_4O_6$
5	CH_3	4	14.7	207 - 208	$\mathrm{CH_3CN}$	$C_{12}H_{22}N_4O_6$
6	$\mathrm{CH_3}$	6	30.6	168-170	$\mathrm{CH_3CN}$	$C_{14}H_{26}N_4O_6$
7	$\mathrm{CH_3}$	8	23.0	138 - 141	$\mathrm{CH_3CN}$	${ m C_{16}H_{30}N_4O_6}$
^a All compds were anal. for C, H, N.						

no antispermatogenic activity was found in this series anthelmintic activity was discovered. For example, 1 (R = H; n = 6) when administered orally to Swiss mice naturally infected with Aspicularis tetraptera (pinworm) cleared 100% of the mice (5/5 per dose level) at 100 mg/kg per day for 4 days and 1 (R = H;n=8) cleared 100% of the mice (5/5 per dose level) at 200 mg/kg per day for 4 days; also, 1 (R = CH_3 ; n =8) cleared 80% of the mice (4/5 per dose level) infected with the tapeworm Hymenolepis nana at 400 mg/kg per day for 4 days.

Experimental Section¹

 N,N^1 -Hexamethylenebis(nitroacetamide).—Ethyl tate (11.2 g, 0.0855 mole) was added to hexamethylenediamine (9.94 g, 0.855 mole). The mixt became hot and liquefied, after which a white solid pptd. The mixt was heated for 3 hr on a steam bath. It slowly turned to a thick orange liquid. The mixt was acidified with alcoholic HCl and poured into H₂O. The white solid was collected and recrystd from MeCN, mp 147-148° dec.

The other compds were prepd similarly except that in the case of the compds with no free H α to NO₂, 1 equiv of diamine was treated with 2 equiv of nitro ester and the alcoholic HCl treatment was unnecessary.

(1) Melting points were measured in open capillary tubes in a bath and are

Tricyclic Heterocycles Derived From 4-Oxo-4,5,6,7-tetrahydrothianaphthenes¹

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Recently we described the synthesis of a variety of tricyclic heterocycles from 4-oxo-4,5,6,7-tetrahydroin-

⁽¹²⁾ P. G. Campbell and P. C. Teague, J. Amer. Chem. Soc., 76, 1371 (1954).

⁽¹³⁾ D. K. Blount, W. H. Perkin, Jr., and S. G. P. Plant, J. Chem. Soc., 1975 (1929).

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