

Regio- and Stereoselective Ring-Opening Reactions of Chiral Substituted Spiro[2.4]hepta-4,6-dienes: A New, Simple, and Versatile Approach to the Synthesis of Optically Active Bidentate Cyclopentadienyl–Phosphine Ligands. X-ray Crystal Structure of (S)-[Rh(η^2 -C₂H₄)(η^5 -C₅H₄CH₂CHPhPPh₂- κ P)]

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Summary: The ring-opening reaction of (S)-1-phenylspiro[2.4]hepta-4,6-diene with LiPPh₂ proceeds with full regio- and stereoselectivity, taking place at phenyl-substituted carbon and with complete inversion of the configuration at the stereogenic center, yielding (S)-[Li-(C₅H₄CH₂CHPhPPh₂)]. Optically active complexes of rhenium, rhodium, zirconium, and iron with this new chiral bidentate ligand have been prepared; additionally, the X-ray crystal structure of (S)-[Rh(η^2 -C₂H₄)(η^5 -C₅H₄-CH₂CHPhPPh₂- κ P)] is reported.

Ligands possessing both a cyclopentadienyl and a heteroatom connected by an appropriate spacer are receiving considerable interest in synthesis and catalysis, as they lead to increased rigidity in a metal complex through additional intramolecular coordination of the heteroatom, avoiding free rotation of the cyclopentadienyl group.¹ For example, half-sandwich complexes of group 4 metals containing linked cyclopentadienyl–amido ligands (the so-called “constrained geometry catalysts”, CGC) have already found technological applications.² Among the possible heteroatoms, phosphorus has been receiving growing interest in view of the general importance of phosphine ligands in organometallic chemistry and homogeneous catalysis. Available data clearly suggest that the cyclopentadienyl–phosphino ligand, as a (5 + 2)-electron ligand, forms a fairly stable chelate ring with early and late transition metals.³ Additionally, the introduction of chirality into these

systems, in which the phosphorus is coordinated to the metal, may assist in controlling the stereochemistry of reactions taking place at the metal center and therefore could eventually increase the stereoselection in catalytic reactions, particularly if the stereogenic center is in close proximity to the metal. Although recent reports based on such compounds have appeared in the literature,⁴ general and simple synthetic procedures for obtaining these hybrid ligands and their metal complexes in optically active form remain relatively undeveloped.

In this communication we describe a new simple general pathway to obtain chiral cyclopentadienyl–phosphine bidentate ligands containing stereogenic centers on the side chain. We have adapted the method first described by Kauffmann, based on opening reactions of spiro[2.4]hepta-4,6-dienes.⁵ Traditionally, such substances have been synthesized by bis-alkylation of freshly cracked cyclopentadiene with the corresponding dibromoalkane in the presence of sodium amide or sodium hydride.⁶ However, a more convenient approach to introduce chirality into these spiroannulated precursors is the use of the corresponding bis-mesylates or -tosylates as alkylating reagents. Our investigations started with the utilization of (R)-1-phenyl-1,2-ethanediol ((R)-1), which is easily converted into the bis(methane-

(1) For reviews see: (a) Okuda, J. *Comments Inorg. Chem.* **1994**, *16*, 185. (b) Jutzi, P.; Redeker, T. *Eur. J. Inorg. Chem.* **1998**, 663.

(2) (a) Thayer, A. M. *Chem. Eng. News* **1995**, 73(38), 15. (b) McKnight, A. L.; Waymouth, R. M. *Chem. Rev.* **1998**, *98*, 2587.

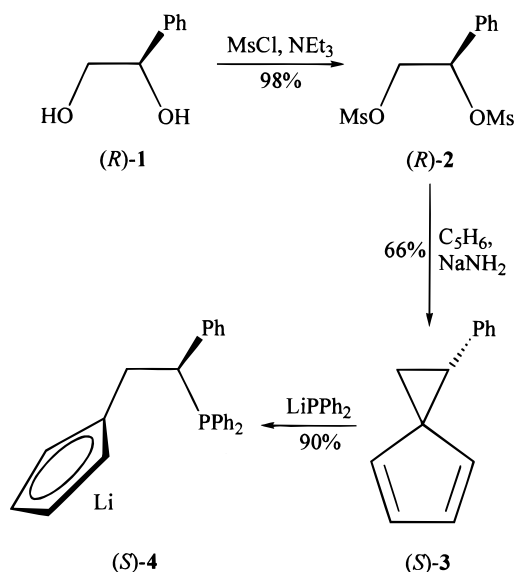
(3) (a) Barthel-Rosa, L. P.; Catalano, V. J.; Maitra, K.; Nelson, J. H. *Organometallics* **1996**, *15*, 3924. (b) Lee, I.; Dahan, F.; Maisonnat, A.; Poilblanc, R. *J. Organomet. Chem.* **1997**, *532*, 159. (c) Wang, T.-F.; Lai, C.-Y. *J. Organomet. Chem.* **1997**, *545–546*, 179. (d) Bosch, B. E.; Erker, G.; Fröhlich, R.; Meyer, O. *Organometallics* **1997**, *16*, 5449. (e) Foerstner, J.; Kakoschke, A.; Stellfeldt, D.; Butenschön, H.; Wartchow, R. *Organometallics* **1998**, *17*, 893. (f) Lefort, L.; Crane, T. W.; Farwell, M. D.; Baruch, D. M.; Kaeuper, J. A.; Lachicotte, R. J.; Jones, W. D. *Organometallics* **1998**, *17*, 3889. (g) Karsch, H. H.; Graf, V.; Reisky, M.; Witt, E. *Eur. J. Inorg. Chem.* **1998**, 1403. (h) Krut'ko, D. P.; Borzov, M. V.; Veksler, E. N.; Kirsanov, R. S.; Churakov, A. V. *Eur. J. Inorg. Chem.* **1999**, 1973. (i) Urtel, K.; Frick, A.; Huttner, G.; Zsolnai, L.; Kircher, P.; Rutsch, P.; Kaifer, E.; Jacobi, A. *Eur. J. Inorg. Chem.* **2000**, 33.

(4) (a) Kataoka, Y.; Saito, Y.; Nagata, K.; Kitamura, K.; Shibahara, A.; Tani, K. *Chem. Lett.* **1995**, 833. (b) Nishibayashi, Y.; Takei, I.; Hidai, M. *Organometallics* **1997**, *16*, 3091. (c) Kataoka, Y.; Saito, Y.; Shibahara, A.; Tani, K. *Chem. Lett.* **1997**, 621. (d) Antelmann, B.; Winterhalter, U.; Huttner, G.; Janssen, B. C.; Vogelgesang, J. *J. Organomet. Chem.* **1997**, *545–546*, 407. (e) Bosch, B.; Erker, G.; Fröhlich, R. *Inorg. Chim. Acta* **1998**, *270*, 446. (f) Antelmann, B.; Huttner, G.; Winterhalter, U. *J. Organomet. Chem.* **1998**, *553*, 433. (g) Mobley, T. A.; Bergman, R. G. *J. Am. Chem. Soc.* **1998**, *120*, 3253. (h) Kataoka, Y.; Shibahara, A.; Saito, Y.; Yamagata, T.; Tani, K. *Organometallics* **1998**, *17*, 4338. (i) Trost, B. M.; Vidal, B.; Thommen, M. *Chem. Eur. J.* **1999**, *5*, 1055. (j) van der Zeijden, A. A. H.; Jimenez, J.; Mattheis, C.; Wagner, C.; Merzweiler, K. *Eur. J. Inorg. Chem.* **1999**, 1919. (k) Ganter, C.; Kaulen, C.; Englert, U. *Organometallics* **1999**, *18*, 5444. (l) Kataoka, Y.; Iwato, Y.; Yamagata, T.; Tani, K. *Organometallics* **1999**, *18*, 5424. (m) Dodo, N.; Matsushima, Y.; Uno, M.; Onitsuka, K.; Takahashi, S. *J. Chem. Soc., Dalton Trans.* **2000**, 35.

(5) Kauffmann, T.; Ennen, J.; Lhotak, H.; Rensing, A.; Steinseifer, F.; Woltermann, A. *Angew. Chem., Int. Ed. Engl.* **1980**, *92*, 328.

(6) (a) Wilcox, C. F.; Craig, R. R. *J. Am. Chem. Soc.* **1961**, *83*, 3866. (b) D'yachenko, A. I.; Menchikov, L. G.; Nefedov, O. M. *Bull. Acad. Sci. USSR, Div. Chem. Sci. (Engl. Transl.)* **1985**, 649.

Scheme 1



sulfonate) ester (*R*)-**2** in quantitative yield by reaction with methanesulfonyl chloride (Scheme 1; for experimental details and analytical data see the Supporting Information).

Displacement of the methanesulfonate groups in the intermediate (*R*)-**2** by cyclopentadiene/sodium amide formed the spiroannulated diene (*S*)-**3**.⁷ The formation of the spiro compound involves first a mesylate displacement by Na(C₅H₅), followed by a fast deprotonation of the substituted cyclopentadienyl ring, and finally a second mesylate displacement, producing intramolecular cyclization. As a result, both carbons in (*R*)-**2** undergo an S_N2 pathway with complete inversion of configuration at the stereogenic center, as has been shown by Halterman in many examples.⁸

The ring-opening reaction of (*S*)-**3** with LiPPh₂ proceeds with complete regioselectivity by attack of the nucleophile at the substituted carbon atom of the cyclopropane ring, bringing the stereogenic center close to the phosphorus atom. Interestingly, this regioselectivity is the opposite to that found for the nucleophilic opening of phenyl epoxide and phenyl episulfide with lithium phosphides, yielding respectively β-hydroxyphosphines and β-mercaptophosphines.⁹ Substituted cyclopropanes with electron-withdrawing groups on the other hand are ring-opened at the most substituted bond of the ring.¹⁰ As far as we know, the only precedent for a similar opening reaction of a chiral (but racemic) substituted spiro[2.4]hepta-4,6-diene was reported by Rieger in order to synthesize unsymmetrical *ansa*-indenyl and -fluorenyl ligands with chiral ethylene

bridges. The regioselectivity found there in the reaction between spiro[(2-phenylcyclopropane)-1,1'-indene] and NaCp was the same as that reported here, with the phenyl group located in a position α to the incoming Cp group.¹¹ The ring-opening reaction of (*S*)-**3** with LiPPh₂ proceeds with complete inversion of the configuration at the stereogenic center, as is demonstrated by characterization of the metal complexes synthesized from the lithium salt (*S*)-**4** (vide infra).

This synthetic pathway has the following advantages.

(i) The chiral precursors are easily available in multigram quantities, particularly after Jacobsen's recent spectacular report on an inexpensive, hydrolytic, kinetic resolution of terminal epoxides which produces optically active monosubstituted 1,2-ethanediols.¹²

(ii) The lithium salt (*S*)-**4** obtained from the opening reaction either can be used directly without further purification to synthesize organometallic complexes.

(iii) The method has been extended by us to chiral, C₂-symmetric 1,2-disubstituted spiro[2.4]hepta-4,6-dienes leading to ligands with stereogenic centers in both positions of the side chain.¹³

(iv) The method can be combined with other ways to introduce chirality into the ligands. For example, the simple opening reaction of the spiroannulated indene derivative reported by Rieger¹¹ should lead to diastereomeric ligands combining both planar-indenyl-based chirality as well as central chirality on the side chain. Alternatively, opening reactions with chiral nucleophiles might produce ligands combining chirality on the spacer and also chiral information directly attached to the heteroatom.

(v) According to Kauffmann's protocol,⁵ the length of the spacer between the cyclopentadienyl ring and the phosphorus could be also modified just by changing the length of the diol precursor. The scope of nucleophiles that can be used in the opening reaction also shows high variability.

The lithium salt (*S*)-**4** was reacted with some metal halide precursors to produce optically active transition-metal complexes. The reaction of 1 equiv of (*S*)-**4** with [Re(CO)₅Br] led to the 18-electron tricarbonylrhenium complex (*S*)-**5** (Scheme 2). The ³¹P NMR spectrum of the crude product showed only a single resonance at −0.6 ppm, proving the regioselectivity of the opening reaction of (*S*)-**3**. To determine the enantiomeric purity of complex (*S*)-**5**, it was first treated with an excess of BH₃·THF, to protect the phosphorus atom against eventual oxidation and also to increase the polarity of the resulting adduct, (*S*)-**5**·BH₃. The compound (*S*)-**5**·BH₃ was then found to be 97% enantiomerically pure by HPLC.¹⁴ As the diol precursor also had an ee of 97%, the pathway represented in Scheme 1 is fully stereospecific.

(7) To the best of our knowledge, compound **3** in optically active form has not been reported before in the literature. For syntheses of racemic **3**, see for example: (a) Johnson, C. R.; Lockard, J. P.; Kennedy, E. R. *J. Org. Chem.* **1980**, *45*, 264. (b) Reference 6b.

(8) (a) Chen, Z.; Eriks, K.; Halterman, R. L. *Organometallics* **1991**, *10*, 3449. (b) Halterman, R. L. *Chem. Rev.* **1992**, *92*, 965 and references therein. (c) Chen, Z.; Halterman, R. L. *Organometallics* **1994**, *13*, 3932.

(9) (a) Tsvetkov, E. N.; Bondarenko, N. A.; Malakhova, I. G.; Kabachnik, M. I. *Synthesis* **1986**, 198. (b) Kagan, H. B.; Tahar, M.; Fiaud, J.-C. *Tetrahedron Lett.* **1991**, *32*, 5959. (c) Pellon, P. *Tetrahedron Lett.* **1992**, *33*, 4451. (d) Hauptmann, E.; Fagan, P. J.; Marshall, W. *Organometallics* **1999**, *18*, 2061 and references therein.

(10) Houben-Weyl; de Meijere, A.; von Angerer, S., Eds.; Thieme: Stuttgart, Germany, 1997; Vol. E17c, p 2082.

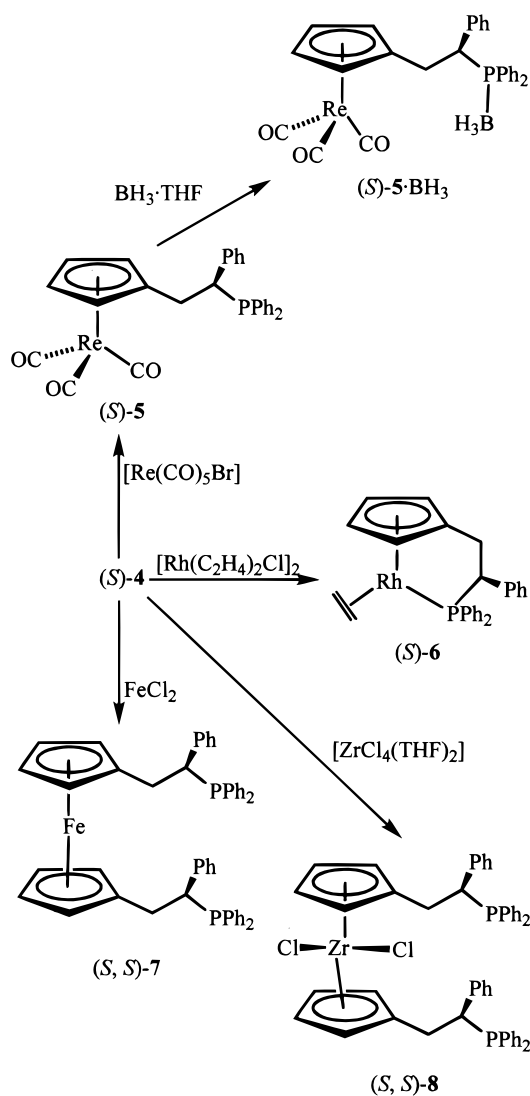
(11) (a) Rieger, B.; Jany, G.; Fawzy, R.; Steimann, M. *Organometallics* **1994**, *13*, 647. (b) Rieger, B.; Jany, G.; Steimann, M.; Fawzy, R. *Z. Naturforsch.* **1994**, *49*, 451.

(12) Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. *Science* **1997**, *277*, 936.

(13) Ciruelos, S.; Salzer, A. Publication in preparation.

(14) Racemic **5**·BH₃ prepared from racemic 1-phenyl-1,2-ethanediol (**1**) was resolved using the column (S,S)-Whelk-01 (4 × 250 mm), 99/1 *n*-heptane/2-propanol as eluent, flow 1 mL/min: *t*₁ = 10.7 min (*S* enantiomer), *t*₂ = 11.9 min (*R* enantiomer). The optically active compound (*S*)-**5**·BH₃ was run under the same conditions and was found to be 98.7% *S* enantiomer, 1.3% *R* enantiomer; ee = 97%.

Scheme 2



The reaction of 1 equiv of (S)-4 with $[\text{Rh}(\eta^2\text{-C}_2\text{H}_4)_2\text{-Cl}]_2$ led to (S)-6. The intramolecular coordination of the phosphorus atom in (S)-6 was confirmed by the large coupling constant between phosphorus and rhodium observed in the ^{31}P NMR spectrum ($^1J_{\text{P-Rh}} = 217$ Hz). The absolute configuration of (S)-6 could be determined by an X-ray crystal structure analysis and was found to be *S* (Figure 1).¹⁵ This agrees with our expectation, as the opening reaction of (S)-3 normally should proceed

