

CLERODANE DITERPENOIDS FROM *SALVIA UROLEPIS*

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**Key Word Index**—*Salvia urolepis*; Labiatae; clerodane derivatives; diterpenoids.

**Abstract**—From the aerial parts of *Salvia urolepis*, three new neoclerodane diterpenoids were isolated. Their structures were determined by spectroscopic means. The known languidulane diterpenoid 2 $\alpha$ -hydroxy-7-*epi*-8 $\beta$ ,17-dihydro-languiduline was also obtained.

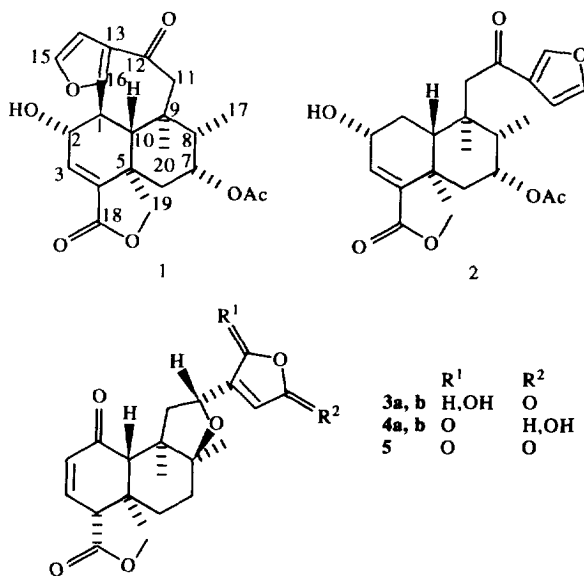
## INTRODUCTION

In a continuation of our systematic studies of Mexican *Salvia* species we have analysed the diterpenoid content of *Salvia urolepis*, Fern. This species has been classified [1] in section Angulatae, subsection Glumacea (*Salvia*, subgenus Calosphace), which also includes *S. languidula* [2]. A phytochemical study of *S. languidula* led to the isolation of several diterpenoids with rearranged clerodane skeletons named languidulane [2] and salvilanguidulane [3]. Diterpenoids with rearranged skeletons of clerodanic origin were also isolated from *S. tiliaefolia* [4] and *S. rhyacophila* [5], two *Salvia* species included [1] in the section Angulatae, subsection Tiliaefolia. In this paper, we describe the isolation and structure determination of the diterpenoids found in *S. urolepis*.

## RESULTS AND DISCUSSION

The aerial parts of *S. urolepis* afforded a mixture of oleanolic and ursolic acids, eupatorine [6] (6,7,4'-trimethoxy-5,5'-dihydroxy flavone); 5,6,3'-trihydroxy-7,4'-dimethoxy flavone [7], the languidulane diterpene 2 $\alpha$ -hydroxy-7-*epi*-8 $\beta$ ,17-dihydrolanguiduline (1) previously isolated [8] from *S. soussae* (*Salvia*, sect. Polystachyae) and named salvissousolide, and three new neoclerodane diterpenoids whose structures (2–4) were determined by spectroscopic means.

The languidulane 1 was the most abundant diterpenoid. It was obtained as an amorphous powder and identified by comparison of its IR and <sup>1</sup>H NMR spectra with those of the previously obtained diterpenoid [8]. The use of high-field 2D NMR experiments allowed the unambiguous assignment of all the proton resonances (Table 1). The COSY experiments showed a *W* coupling between H-10 and the Me-20 protons which is only possible in a *trans* A/B ring fusion of the decalin with the Me-20  $\alpha$ -axial. The <sup>13</sup>C NMR spectrum of 1, not previously described, was consistent with the languidulane



structure shown. The carbon resonances (Table 2) of the protonated carbon 5 were assigned by DEPT and HETCOR experiments. The three sp<sup>3</sup> doublets of carbons not bound to oxygen, were unambiguously assigned to C-1, C-8 and C-10. The quaternary and carbonyl carbon signals were established by comparison with the spectra of similar structures [2, 9].

The neoclerodane diterpene 2 was obtained as an unstable oil which showed in its IR spectrum bands due to a hydroxy group (3603 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone (1772 cm<sup>-1</sup>), an ester function (1738 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated ketone (1673 cm<sup>-1</sup>) and a furan ring (1607, 1560 and 872 cm<sup>-1</sup>). The mass spectrum was consistent with the molecular formula C<sub>22</sub>H<sub>26</sub>O<sub>7</sub> (see experimental). A strong peak at *m/z* 95 (93%) suggested that the  $\alpha,\beta$ -unsaturated ketone group was at C-12 [10]. The

Table 1.  $^1\text{H}$  NMR spectral data for 1–4 (200 MHz,  $\text{CDCl}_3$ , TMS)

H	1*	2	3a	3b	4a	4b
1	3.11 <i>t</i> (9.3)		—	—	—	—
2	4.72 <i>dd</i> (9.3, 1.5)	4.52 <i>br dd</i> (8, 4)	6.21 <i>dd</i> (9.9, 2.7)	6.22 <i>dd</i> (10, 2.4)	6.21 <i>dd</i> (10.2, 3.0)	6.21 <i>dd</i> (10.2, 3)
3	6.68 <i>d</i> (1.5)	6.6 <i>br s</i> $W_{1/2} = 4$	6.60 <i>dd</i> (9.9, 3.3)	6.59 <i>dd</i> (10, 3.3)	6.57 <i>dd</i> (10.2, 3.3)	6.57 <i>dd</i> (10.2, 3.3)
4			3.25 <i>dd</i> (3.3, 2.7)	3.23 <i>dd</i> (3.3, 2.4)	3.23 <i>dd</i> (3.3, 3.0)	3.26 <i>dd</i> (3.3, 3.0)
6 $\alpha$	2.38 <i>dd</i> (15, 2.2)					
6 $\beta$	1.55 <i>ddd</i> (15, 4.0, 2.0)					
7	5.3 <i>dt</i> (4, 2.2)	5.26 <i>dt</i> (4, 2)				
8	1.68 <i>dq</i> (7, 4.0)					
10	2.28 <i>d</i> (9.3)		2.85 <i>s</i>	2.83 <i>s</i>	2.83 <i>s</i>	2.84 <i>s</i>
11A	3.04 <i>d</i> (15.1)	2.98 <i>d</i> (16)	2.92 <i>br t</i> (8.1)	2.97 <i>br t</i> (8.1)	3.01 <i>m</i>	3.0 <i>m</i>
11B	2.54 <i>d</i> (15.1)	2.72 <i>d</i> (16)				
12			4.89 <i>ddd</i> (9.6, 8.1, 1.8)	4.80 <i>ddd</i> (9.6, 8.1, 1.8)	4.76 <i>br t</i> (7.8)	4.77 <i>br t</i> (7.8)
14	6.79 <i>d</i> (2)	6.75 <i>d</i> (1.6)	6.04 (1)	5.99 <i>d</i> (1)	7.05 <i>br s</i>	7.05 <i>br s</i>
15	7.47 <i>d</i> (2)	7.47 <i>t</i> (1.6)	—	—	6.13 <i>br s</i>	6.13 <i>br s</i>
16		8.1 <i>br s</i>	5.97 <i>s</i>	6.13 <i>s</i>		
H <sub>3</sub> -17	1.04 <i>d</i> (7)	0.94 <i>d</i> (7.2)	1.12 <i>s</i>	1.12 <i>s</i>	1.15 <i>s</i>	1.12 <i>s</i>
19 <i>pro</i> -R	4.96 <i>d</i> (8.3)	4.82 <i>d</i> (8)	4.37 <i>d</i> (9)	4.37 <i>d</i> (9)	4.37 <i>d</i> (9)	4.38 <i>d</i> (9)
19 <i>pro</i> -S	4.11 <i>dd</i> (8.3, 2.0)	4.0 <i>dd</i> (8, 2)	4.06 <i>d</i> (9)	4.06 <i>d</i> (9)	4.06 <i>d</i> (9)	4.07 <i>d</i> (9)
H <sub>3</sub> -20	0.90 <i>s</i>	0.91 <i>s</i>	1.08 <i>s</i>	1.07 <i>s</i>	1.06 <i>s</i>	1.05 <i>s</i>
MeCO <sub>2</sub>	2.13 <i>s</i>	2.1 <i>s</i>	—	—	—	—

\* Run at 300 MHz.

*J* in Hertz in parentheses.

$^1\text{H}$  NMR spectrum (Table 1) showed the typical resonances of a  $\beta$ -substituted furan ring conjugated to the ketone group ( $\delta$ 8.1, *br s*, H-16). The signal at  $\delta$ 193.65 in the  $^{13}\text{C}$  NMR spectrum of **2** (Table 2) was attributed to this carbon. An AB system at  $\delta$ 2.98 and 2.72 (two doublets,  $J = 16$  Hz) was ascribed to the C-11 methylene. The rest of the  $^1\text{H}$  NMR spectrum of **2** was very similar to that of **1** (Table 1). These data suggested the same substitution pattern in the decalin ring system of both diterpenoids, with the exception of the H-2 resonance observed as a doublet ( $J = 8$  and 4 Hz) at  $\delta$ 4.52 in **2**. The multiplicity and coupling constants of this signal confirmed that **2** had no substituent at C-1. The carbon resonances and multiplicities (Table 2) were established by comparison with those found for the spectra of similar structures [9] and DEPT experiments. A triplet at  $\delta$ 30.25 assigned to C-1 confirmed the absence of a substituent at this carbon. The

neoclerodane **2** can be considered as a biogenetic precursor of the languidulane **1** [11].

The hydroxy lactone **3** was obtained as an oily C-16 epimeric mixture (**3a** and **b**) whose mass spectrum was in accordance with the molecular formula  $\text{C}_{20}\text{H}_{22}\text{O}_7$ , although the parent peak was not observed. The IR spectrum contained bands due to a hydroxy group (3582 and 3360  $\text{cm}^{-1}$ ), a  $\gamma$ -lactone (1774  $\text{cm}^{-1}$ ), an  $\alpha,\beta$ -unsaturated ketone (1681  $\text{cm}^{-1}$ ), an ether function (1026  $\text{cm}^{-1}$ ) and double bonds (1632 and 830  $\text{cm}^{-1}$ ). The  $^1\text{H}$  NMR spectrum showed duplicate signals for most protons (Table 1) which suggested that the sample was a mixture of C-16 epimers. The vinylic protons, H-14, of the hydroxy lactone moiety, were observed as doublets ( $J = 1$  Hz) at  $\delta$ 5.99 and 6.04. Two singlets  $\delta$ 6.13 and 5.97 were ascribed to the protons geminal to the hydroxyls of this function. For simplicity, we will analyse the  $^1\text{H}$  NMR of the

Table 2.  $^{13}\text{C}$ NMR spectral data for compounds **1** and **2** (50 MHz,  $\text{CDCl}_3$ , TMS)

C	1	2
1	45.1 <i>d</i>	30.3 <i>t</i>
2	75.1 <i>d</i>	69.5 <i>d</i>
3	139.5 <i>d</i>	140.1 <i>d</i>
4	143.3 <i>s</i>	137.3 <i>s</i>
5	46.3 <i>s</i>	44.5 <i>s</i>
6	36.9 <i>t</i>	37.5 <i>t</i>
7	71.6 <i>d</i>	73.2 <i>d</i>
8	44.6 <i>d</i>	39.6 <i>d</i>
9	37.3 <i>s</i>	39.9 <i>s</i>
10	51.8 <i>d</i>	42.3 <i>d</i>
11	56.5 <i>t</i>	45.9 <i>t</i>
12	192.4 <i>s</i>	193.7 <i>s</i>
13	124.5 <i>s</i>	128.9 <i>s</i>
14	109.3 <i>d</i>	108.5 <i>d</i>
15	143.3 <i>d</i>	144.6 <i>d</i>
16	157.4 <i>s</i>	147.0 <i>d</i>
17	12.1 <i>q</i>	11.9 <i>q</i>
18	167.4 <i>s</i>	168.7 <i>s</i>
19	70.7 <i>t</i>	72.1 <i>t</i>
20	13.9 <i>q</i>	19.2 <i>q</i>
21	169.8 <i>s</i>	169.8 <i>s</i>
22	21.1 <i>q</i>	21.2 <i>q</i>

hydroxy lactone **3b** in which two double doublets at  $\delta$ 6.22 ( $J=10$  and 2.4 Hz) and 6.59 ( $J=10$  and 3.3 Hz) were assigned to the vinyl protons H-2 and H-3, respectively. They were shown to be coupled to H-4 which appeared at  $\delta$ 3.23 (*dd*,  $J=3.3$  and 2.4 Hz). The chemical shifts and multiplicities of H-2 and H-3 suggested that the ketone group must be located at C-1. An AB system observed at  $\delta$ 4.37 and 4.06 (*d*,  $J=9$  Hz) was attributed to the C-19 methylene. A complex signal at  $\delta$ 4.80 was ascribed to H-12, geminal to an ethereal oxygen. Two methyl singlets at  $\delta$ 1.07 and 1.12 were assigned to Me-20 and Me-17, respectively. Therefore, the ethereal function observed in the IR ( $1026\text{ cm}^{-1}$ ) must be bound to C-8 and C-12 as found in kerlin (from *S. keerlii*) [12] and a related neoclerodane diterpenoid isolated from *S. rhyacophila* [5]. A singlet at  $\delta$ 2.83 was assigned to H-10. COSY experiments proved that it was *W* coupled to the Me-20 protons, thus showing a *trans* steroidal decalin ring system in **3** [9].

The mass spectrum of the C-15 epimeric mixture of the hydroxy lactones **4a, b** was determined by FAB-mass spectrometry and indicated the molecular formula  $\text{C}_{20}\text{H}_{22}\text{O}_7$ . It showed the same main bands in the IR spectrum as **3a, b**. The  $^1\text{H}$ NMR was very similar to that of **3a, b** from which it differed in the chemical shifts and multiplicities of the hydroxy lactone protons H-14 ( $\delta$ 7.05, *br s*) and H-15 ( $\delta$ 6.13, *br s*) as expected for an  $\alpha$  substituted hydroxylated butenolide. The H-12 resonance was observed as a broad triplet ( $J=7.8$  Hz) at  $\delta$ 4.76.

Attempted oxidation of the hydroxy lactones **3a, b** and **4a, b** to the anhydride **5**, were unsuccessful.

The presence of the languidulane diterpenoid **1** in *S. urolepis* has chemotaxonomic interest. It supports the botanical classification [1] of this species in sect. Angulatae, subsect Glumacea in which *S. languidula* [2] is also included. It is interesting to note that *S. soussae*, from which salvisousolide (**1**) was first isolated, has been included in sect. Polystachyae, a section botanically related to sect. Angulatae, subsect. Glumacea.

## EXPERIMENTAL

Mps.: uncorr; MS: 70 eV, direct inlet;  $^1\text{H}$ NMR: 200 and 300 MHz,  $\text{CDCl}_3$ , TMS as int. standard;  $^{13}\text{C}$ NMR: 50 MHz. Plant material was collected in Zimapán, State of Hidalgo (México) and a voucher specimen (TPR4852) is deposited in the Herbarium of Instituto de Biología UNAM.

*Isolation of the constituents.* Dried aerial parts of *Salvia urolepis* (3 kg) were extracted twice ( $2 \times 20\text{ l}$ ) with  $\text{Me}_2\text{CO}$  at room temp. for five days. The solvent was removed under red. pres. to yield 116.1 g of a gummy residue. This extract (78 g) was subjected to dry CC over silica gel (1 kg, 35–70 mesh, deactivated with 10%  $\text{H}_2\text{O}$ ). Mixtures of petrol–EtOAc and EtOAc–MeOH of increasing polarity were used as eluents.

Elution with petrol–EtOAc (4:1) afforded  $\beta$ -sitosterol and a mixture of oleanolic and ursolic acids, which were identified by comparison with authentic samples. Elution with petrol–EtOAc (7:3) gave the flavone eupatorine (40 mg). The 5,6,3'-trihydroxy-7,4'-dimethoxy flavone (20 mg) was obtained from the fractions eluted with petrol–EtOAc (1:1). Both flavonoids were identified by comparison with literature data.

The fractions eluted with petrol–EtOAc (1:4) were subjected to extensive chromatographic purifications to afford 100 mg of **1**, 40 mg of **2**, 50 mg of the epimeric mixture **3a** and **b** and 15 mg of the mixture of hydroxylactones **4a** and **b**.

*Salvisousolide (1).* Amorphous powder, mp 116–117°;  $[\alpha]_{\text{D}} -136.5$  ( $\text{CHCl}_3$ ;  $c$  0.2); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 202 (4.5), 265 (3.4); IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3580, 1770, 1740, 1660, 1590, 920, 870;  $^1\text{H}$ NMR: Table 1;  $^{13}\text{C}$ NMR: Table 2; MS  $m/z$  (rel. int.): 401 (5), 400 (10), 179 (20), 163 (5), 162 (40), 161 (100), 159 (10), 115 (20), 95 (10), 91 (10), 79 (5), 43 (60).  $\text{C}_{22}\text{H}_{24}\text{O}_7$  requires  $[\text{M}]^+$  at  $m/z$  400.

*2 $\alpha$ -Hydroxy-7 $\alpha$ -acetoxy-12-oxo-15 : 16-epoxy-neoclerodan-3,13(16),14-trien-18:19-olide (2).* Oily compound; IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3603, 1772, 1738, 1673, 1607, 1560, 872;  $^1\text{H}$ NMR: Table 1,  $^{13}\text{C}$ NMR: Table 2; MS  $m/z$  (rel. int.): 384 (0.2), 342 (0.2), 233 (10), 215 (10), 171 (15), 159 (15), 149 (20), 145 (20), 110 (22), 95 (92.7), 43 (100).  $\text{C}_{22}\text{H}_{26}\text{O}_7$  requires  $[\text{M}]^+$  at  $m/z$  402, not observed.

*Mixture of hydroxylactones 3a and b.* Oily mixture; IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3582, 3360, 1774, 1681, 1632, 1026, 830;  $^1\text{H}$ NMR: Table 1; MS  $m/z$  (rel. int.): 359 (4.9), 341 (1.8), 231 (2.5), 115 (20), 91 (50), 76 (20), 55 (40).  $\text{C}_{20}\text{H}_{22}\text{O}_7$  requires  $[\text{M}]^+$  at  $m/z$  374, not observed.

*Mixture of hydroxylactones 4a and b.* Oily mixture; IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3400, 1771, 1678, 1630, 1020, 832, 753;  $^1\text{H}$ NMR: Table 1; FABMS  $m/z$  (rel. int.): 375.2 (10),

277.2 (8), 231.2 (6), 185.2 (85), 93.1 (100), 75.1 (25), 57.0 (18).  $C_{20}H_{22}O_7$  requires  $[M]^+$  at  $m/z$  374.

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