



## TRITERPENOID SAPONINS FROM THE BARK OF *NOTHOPANAX DAVIDII*

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**Key Word Index**—*Nothopanax davidii*; Chinese folk medicine; Yiyeliang Wanosides IX, X and XI; Serratagenic acid; triterpenoid Saponin.

**Abstract**—Three new triterpenoid saponins were isolated from the alcoholic extract of the bark of *Nothopanax davidii*. Their structures have been determined on the basis of spectral and chemical data as 3-*O*- $\alpha$ -(4'-*O*-acetyl)-L-arabinopyranosyl-3 $\beta$ -hydroxyolean-12-ene-28,29-dioic acid-28-*O*-[ $\alpha$ -L-rhamnopyranosyl(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl(1  $\rightarrow$  6)- $\beta$ -D-glucopyranosyl]ester, named yiyeliangwanoside IX; 3-*O*- $\alpha$ -(2'-*O*-acetyl)-L-arabinopyranosyl-3 $\beta$ -hydroxy-olean-12-ene-28,29-dioic acid-28-*O*-[ $\alpha$ -L-rhamnopyranosyl(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl(1  $\rightarrow$  6)- $\beta$ -D-glucopyranosyl] ester, named yiyeliangwanoside X; and 3-*O*- $\beta$ -D-xylopyranosyl-3 $\beta$ -hydroxyolean-12-ene-28,29-dioic acid-28-*O*-[ $\alpha$ -L-rhamnopyranosyl(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl(1  $\rightarrow$  6)- $\beta$ -D-glucopyranosyl]ester, named yiyeliangwanoside XI.

### INTRODUCTION

*Nothopanax davidii* (Franch) Harms, a well-known Chinese folk medicine 'Liang Wang Cha' is a tall tree which grows in south-west China. The bark is used to remedy rheumatoid arthritis, fractures and strains. In preceding papers [1-4], we reported the isolation and structural elucidation of eight new triterpenoid saponins from the alcoholic extract of the bark of *N. davidii*. As a continuation of studies on this plant, we present here the spectral and chemical evidence for three new triterpenoid saponins, designated as yiyeliangwanoside IX (1), X (2) and XI (3).

### RESULTS AND DISCUSSION

Dried and pulverized bark of *N. davidii* was extracted with 70% ethanol. The purified extract obtained by means of column chromatography with highly porous polymer (see Experimental) was subjected to column chromatography to give 1-3, which responded to the Liebermann-Burchard test [5]. The IR spectra of 1-3 showed ester group absorptions (1: 1720, 1710  $\text{cm}^{-1}$ ; 2: 1740, 1710  $\text{cm}^{-1}$ ; 3: 1710  $\text{cm}^{-1}$ ) together with strong hydroxyl and olefinic absorptions.

Acid hydrolysis of 1-3 afforded the same aglycone 4, which was identified as 3 $\beta$ -hydroxyolean-12-ene-28,29-dioic acid(serratagenic acid) [6] by comparison with an authentic sample (mp,  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data).

On acid hydrolysis, both 1 and 2 gave L-arabinose, L-rhamnose and D-glucose; 3 gave D-xylose, L-rhamnose

and D-glucose, which were identified by HPTLC [7] and PC. The  $^{13}\text{C}$  NMR spectra indicated the presence of four sugar residues in 1-3 by their anomeric carbon signals (1:  $\delta$  95.7, 102.7, 104.8 and 107.2; 2:  $\delta$  96.0, 102.9, 104.8 and 105.0; 3:  $\delta$  95.7, 102.7, 104.9 and 107.6). The signals at 95.7, 96.0 and 95.7 suggested that 1-3 have a glycosidic ester linkage [8] at either the 28 or 29-COOH, which was supported by slight shifts of the signals of one of two carboxyl groups [9] (Table 1). The simultaneous presence of a 3-*O*-glycosidic linkage was easily seen by the attendant downfield shifts of C-3 [10]. Thus 1-3 are bidesmosidic saponins.

On alkaline hydrolysis, both 1 and 2 yielded 5, and 3 yielded 6, which were, respectively, formulated as the 3-*O*- $\alpha$ -L-arabinopyranoside and 3-*O*- $\beta$ -D-xylopyranoside of 4 based on their acid hydrolysis,  $^1\text{H}$ ,  $^{13}\text{C}$  NMR and FAB mass spectral data.

The positive FAB mass spectra of 1 and 2 yielded the same molecular ions at  $m/z$  1169 [ $\text{M} + \text{K}$ ] $^+$ , indicating their molecular weights of 1130; The FD mass spectrum of 3 revealed a molecular ion at  $m/z$  1112 [ $\text{M} + \text{Na} + \text{H}$ ] $^+$ , indicating a molecular weight of 1088. On the other hand, the EI mass spectra of the peracetates of 1 and 2 displayed the same fragment ions at  $m/z$  561 [(Glc-Rha)Ac<sub>6</sub>] $^+$ , 273 [(Rha)Ac<sub>3</sub>, terminal rhamnose] $^+$ , 259 [(Ara)Ac<sub>3</sub>, terminal arabinose] $^+$ ; and the peracetate of 3 similarly displayed fragment ions at  $m/z$  561 [(Glc-Rha)Ac<sub>6</sub>] $^+$ , 273 [(Rha)Ac<sub>3</sub>, terminal rhamnose] $^+$ , 259 [(Xyl)Ac<sub>3</sub>, terminal xylose] $^+$ . Therefore, the sequence of oligosaccharide chains of 1-3 is rhamnose-glucose-glucose-aglycone.

It was determined that one acetyl group was attached to the sugar moieties of both 1 and 2 by FAB mass

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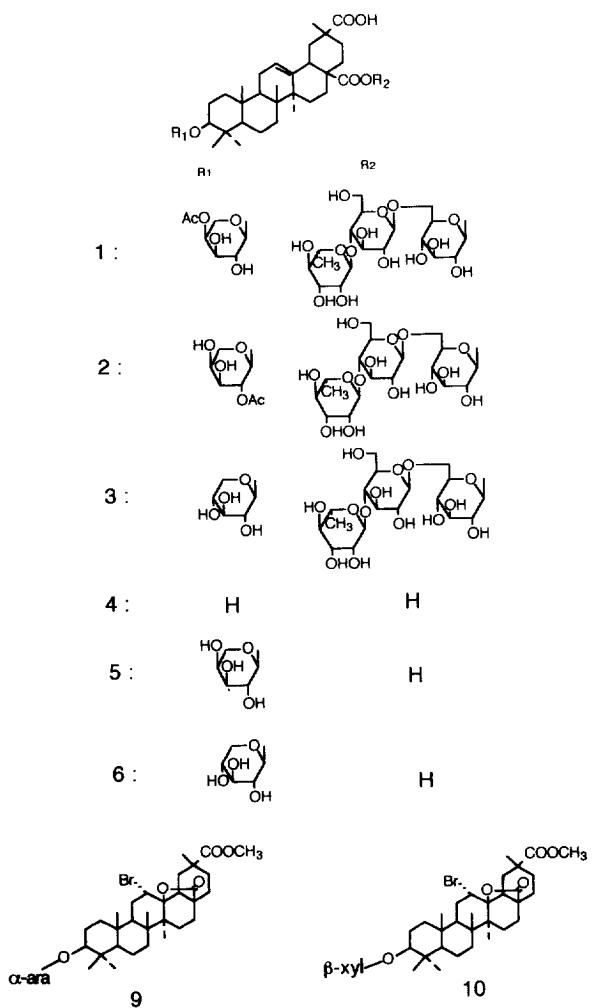


Table 1.  $^{13}\text{C}$  NMR chemical shifts of aglycone moieties (125 MHz, in  $\text{C}_5\text{D}_5\text{N}$ )

C	1	2	3	4	5	6
1	38.8	38.9	38.8	38.9	38.8	38.8
2	26.6	26.9	26.7	28.0	26.7	26.6
3	89.0	89.3	88.6	78.1	88.6	88.7
4	39.5	39.8	39.6	39.4	39.6	39.5
5	55.9	56.0	55.9	55.8	55.9	55.9
6	18.5	18.8	18.5	18.8	18.5	18.5
7	33.1	33.4	33.1	33.2	33.2	33.2
8	40.0	40.2	39.9	39.7	39.8	39.8
9	48.1	48.3	48.0	48.0	48.0	48.0
10	37.1	37.3	37.1	37.4	37.1	37.1
11	23.8	23.8	23.8	23.8	23.9	23.9
12	123.0	124.0	123.1	122.2	123.3	123.2
13	143.6	144.0	143.6	144.3	144.3	144.3
14	42.3	42.7	42.3	42.5	42.6	42.6
15	28.2	28.5	28.2	28.3	28.3	28.3
16	23.6	24.2	23.5	23.8	23.9	23.9
17	47.0	47.3	47.0	46.9	46.6	46.6
18	40.8	40.2	40.8	41.1	41.1	41.1
19	40.8	41.1	40.8	41.1	41.1	41.1
20	42.2	42.5	42.2	42.2	42.6	42.6
21	29.1	29.4	29.1	29.4	29.3	29.3
22	31.7	31.9	31.7	32.4	32.4	32.4
23	28.2	28.3	28.2	28.3	28.2	28.2
24	16.9	17.2	17.0	16.6	17.0	16.9
25	15.6	15.9	15.6	15.5	15.5	15.5
26	17.5	17.8	17.5	17.4	17.4	17.4
27	26.0	26.3	26.0	26.1	26.1	26.1
28	176.2	177.0	176.3	180.0	179.9	179.8
29	180.5	181.9	180.9	181.1	180.2	180.5
30	19.9	20.2	19.9	20.0	20.0	20.0

spectra,  $^1\text{H}$  and  $^{13}\text{C}$  NMR. On examination of  $^1\text{H}$ ,  $^{13}\text{C}$  NMR,  $^1\text{H}$ - $^1\text{H}$  COSY and  $^{13}\text{C}$ - $^1\text{H}$  COSY spectra of **1** and **2**, it was found that H-4 of arabinose shifted downfield to  $\delta$  5.55 (*m*), C-4 of arabinose shifted downfield to  $\delta$  72.2, whereas C-3 and C-5 shifted upfield for **1** (compared with these of **5**, Table 2), H-2 of arabinose shifted downfield to  $\delta$  5.70, C-2 of arabinose downfield to  $\delta$  73.9, whereas C-1 and C-3 of arabinose shifted upfield for **2** (compared with these of **5**, Table 2). Thus, the acetyl group was linked to C-4 of arabinose for **1** and C-2 of arabinose for **2**.

In the oligosaccharide chains of **1-3**, glycosylation shifts were all at C-4 and C-6 of two glucoses (shifted downfield to  $\delta$  78.7 and 69.5 for **1**;  $\delta$  78.9 and 69.0 for **2**;  $\delta$  78.7 and 69.4 for **3**; respectively) by analysis of their  $^{13}\text{C}$  NMR,  $^1\text{H}$ - $^1\text{H}$  COSY and  $^{13}\text{C}$ - $^1\text{H}$  COSY spectra. The correlation between rhamnose H-1 and outer glucose C-4 was observed in COLOC experiments of **1** and **3**. The correlations between H-1 of rhamnose and H-4 of outer glucose, H-1 of outer glucose and H-6, H-6' of inner glucose were shown in the NOESY spectrum of **2**. So, it was concluded that rhamnose was linked to C-4 of outer glucose, and outer glucose was attached to C-6 of inner glucose in the oligosaccharide chains of **1-3**.

The location of a glycosyl linkage at the 28-COOH was revealed as follows. Formation of the characteristic bromolactone has been used as chemical evidence of the presence of a free carboxyl group at C-17 of olean-12-ene triterpenes [11]. On treatment with diazomethane, **1** afforded a monomethyl ester, **7**. On selective hydrolysis of the ester linkage with alkali, **7** yielded a monodesmoside **8**. Formation of a bromolactone, **9**, from **8** was observed on treatment of **8** with bromine in the presence of sodium acetate. It follows that **1** should be formulated as the 28-glycosyl ester of 3-*O*- $\alpha$ -arabinopyranosyl-serratagenic acid. Location of the oligosaccharide chains of **2** and **3** at 28-COOH was supported by the formation of the bromolactone as in the case of **1**. When subjected to the same procedure, **2** also afforded **9**, and **3** gave **10**.

In the  $^1\text{H}$  NMR spectra, the anomeric proton signals for **1** at  $\delta$  4.73 (1H, *d*, *J* = 6.0 Hz), 4.93 (1H, *d*, *J* = 7.8 Hz), 5.79 (1H, *s*), 6.22 (1H, *d*, *J* = 8.0 Hz), for **2** at  $\delta$  4.70 (1H, *d*, *J* = 7.4 Hz), 5.00 (1H, *d*, *J* = 7.8 Hz), 5.80 (1H, *s*), 6.20 (1H, *d*, *J* = 8.0 Hz), for **3** at  $\delta$  4.79 (1H, *d*, *J* = 7.5 Hz), 4.92 (1H, *d*, *J* = 7.6 Hz), 5.80 (1H, *s*), 6.21 (1H, *d*, *J* = 8.1 Hz) led to the assignments of the anomeric configurations of glucose and xylose units as  $\beta$ , and rhamnose and arabinose units as  $\alpha$ ; these assignments were supported by their carbon signals (Table 2).

Table 2.  $^{13}\text{C}$  NMR chemical shifts of sugar moieties (125 MHz, in  $\text{C}_5\text{D}_5\text{N}$ )

	1	2	3	5	6
28-O-sugar					
Glc-1	95.7	96.0	95.7		
Glc-2	73.9	73.8	73.9		
inner					
Glc-3	78.7	77.9	78.7		
Glc-4	71.1	72.5	70.9		
Glc-5	77.9	77.0	77.1		
Glc-6	69.5	69.0	69.4		
Glc-1	104.8	104.8	104.9		
Glc-2	75.2	73.6	75.5		
outer					
Glc-3	76.5	77.6	77.9		
Glc-4	78.7	78.9	78.7		
Glc-5	77.1	76.1	77.1		
Glc-6	61.2	61.0	61.3		
Rha-1	102.7	102.9	102.7		
Rha-2	72.4	72.5	72.5		
Rha-3	72.7	72.4	72.7		
Rha-4	73.6	73.9	74.0		
Rha-5	70.3	70.1	70.3		
Rha-6	18.4	18.3	18.5		
3-O-Sugar					
Xyl-1		107.6	107.5		
Xyl-2		73.8	74.0		
Xyl-3		78.7	78.5		
Xyl-4		71.2	71.2		
Xyl-5		67.0	67.0		
Ara-1	107.2	105.0 (– 2.2)	107.2		
Ara-2	73.1	73.9 (+ 1.0)	72.9		
Ara-3	72.2 (– 2.3)	72.0 (– 2.5)	74.5		
Ara-4	72.4 (+ 2.1)	69.0	69.3		
Ara-5	65.7 (– 0.8)	66.9	66.5		
MeCO	170.7	171.0			
Me	21.0	21.8			

Based on the above results, **1–3** were established as 3-O- $\alpha$ -(4'-O-acetyl)-L-arabinopyranosyl-, 3-O- $\alpha$ -(2'-O-acetyl)-L-arabinopyranosyl- and 3-O- $\beta$ -D-xylopyranosyl-3 $\beta$ -hydroxyolean-12-ene-28,29-dioic acid-28-O-[ $\alpha$ -L-rhamnopyranosyl(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl (1-6)- $\beta$ -D-glucopyranosyl] ester, respectively.

## EXPERIMENTAL

**General procedures.** NMR spectra were taken on a Bruker AM-500 ( $^1\text{H}$  NMR at 500 MHz and  $^{13}\text{C}$  NMR at 125 MHz) spectrometer in  $\text{C}_5\text{D}_5\text{N}$  with tetramethylsilane (TMS) as int. standard.

MS were recorded on JEOL JMS-DX 300 and JMS-DX 300 mass spectrometers.

The IR were recorded on a Perkin-Elmer 683 IR spectrometer. Mps were determined on a mmp apparatus and were uncorr. Optical rotations were measured with Perkin Elmer 241 automatic digital polarimeter. The plant was identified by Prof. Jia-Lin Wu, Sichuan School of Chinese Traditional Medicine. A voucher specimen is

deposited in Institute of Materia Medica, Chinese Academy of Medical Sciences.

**Extraction and isolation of saponins.** The dried bark (11 kg) of *N. davidii* was collected in the Sichuan province of China in the autumn of 1990. The bark was pulverized and extracted with 70% EtOH (25 l  $\times$  4, 2 hr for each extraction) at 80°. The extracts were combined and concd *in vacuo* to give a brown residue (2.5 kg). The residue (0.5 kg) was subjected to CC on highly porous polymer with  $\text{H}_2\text{O}$ , 80% EtOH and EtOH, successively. The EtOH eluate (80%, 28 g) was chromatographed on silica gel with  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  (75:25:10, 70:30:10 and 65:35:10) to give 5 frs (I  $\sim$  V).

Fr. III (1.0 g) was chromatographed on silica gel ( $\text{CHCl}_3$ -MeOH-EtOAc- $\text{H}_2\text{O}$  2:2:4:1) and reversed-phase MPLC (eluted with MeOH- $\text{H}_2\text{O}$  60:40) to give **1** (100 mg).

The chromatography of fr. IV on silica gel (eluted with  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  7:3:1) and reversed-phase MPLC (eluted with MeOH- $\text{H}_2\text{O}$  57:40) gave **2** (160 mg).

Fr. V was chromatographed on silica gel ( $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  65:35:10) and reversed-phase MPLC (eluted with MeOH- $\text{H}_2\text{O}$  57:44) to afford **3** (80 mg).

**Characterization of 1.** A powder, mp 228–230° (dec.).  $[\alpha]_D^{12} = -9.43$  (MeOH;  $c 5.3 \times 10^{-2}$ ). Analyt. calcd for  $\text{C}_{55}\text{H}_{86}\text{O}_{24} \cdot 5/2\text{H}_2\text{O}$ : C 56.36, H 7.77, found C 56.18, H 7.54. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400 (OH), 2925 (C-H), 1720 (C=O, ester), 1710 (C=O, ester), 1700 (O=C-OH), 1640 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.88, 0.94, 1.11, 1.24, 1.26, 1.45 (each 3H, s, Me), 1.68 (3H, d,  $J = 6.1$  Hz, H-6 of  $\alpha$ -rha), 2.0 (3H, s, O=CMe), 4.73 (1H, d,  $J = 6.0$  Hz, H-1 of  $\alpha$ -ara), 4.93 (1H, d,  $J = 7.8$  Hz, H-1 of  $\beta$ -glc), 5.48 (1H, m, H-12), 5.79 (1H, s, H-1 of  $\alpha$ -rha), 6.20 (1H, d,  $J = 8.0$  Hz, H-1 of  $\beta$ -glc).  $^{13}\text{C}$  NMR: Tables 1 and 2. FAB-MS  $m/z$ : 1169 [M + K]<sup>+</sup>, 1123 [M - HAc - 2H + K]<sup>+</sup>, 977 [M - HAc-Rha - 2H + K]<sup>+</sup>, 653 [M - HAc - Rha - Glc - Glc - 2H + K]<sup>+</sup>. Liebermann-Burchard reaction reddish purple.

**Characterization of 2.** A powder, mp 210–212° (dec.).  $[\alpha]_D^{10} = -3.9$  (MeOH;  $c 0.1$ ). Analyt. calcd for  $\text{C}_{55}\text{H}_{86}\text{O}_{24} \cdot 5/2\text{H}_2\text{O}$ : C 56.36, H 7.77, found C 56.03, H 7.47. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400 (OH), 2925 (C-H), 1740 (C=O, ester), 1710 (C=O, ester), 1700 (O=C-OH), 1640 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.83, 0.88, 1.05, 1.08, 1.22, 1.44 (each 3H, s, Me), 1.69 (3H, d,  $J = 6.1$  Hz, H-6 of  $\alpha$ -rha), 2.10 (3H, s, O=C-Me), 4.70 (1H, d,  $J = 7.4$  Hz, H-1 of  $\alpha$ -ara), 5.00 (1H, d,  $J = 7.8$  Hz, H-1 of  $\beta$ -glc), 5.44 (1H, m, H-12), 5.84 (1H, s, H-1 of  $\alpha$ -rha), 6.25 (1H, d,  $J = 8.0$  Hz, H-1 of  $\beta$ -glc).  $^{13}\text{C}$  NMR: Tables 1 and 2. FAB-MS  $m/z$ : 1169 [M + K]<sup>+</sup>. Liebermann-Burchard reaction reddish purple.

**Characterization of 3.** A powder, mp 219–224° (dec.).  $[\alpha]_D^{12} = -21.62$  (MeOH;  $c 9.3 \times 10^{-2}$ ). Analyt. calcd for  $\text{C}_{53}\text{H}_{84}\text{O}_{23} \cdot 7\text{H}_2\text{O}$ : C 52.56, H 8.09, found C 52.56, H 7.25. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400 (OH), 1710 (C=O, ester), 1700 (O=C-OH), 1630 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.85, 1.02, 1.11, 1.22, 1.24, 1.43 (each 3H, s, Me), 1.67 (3H, d,  $J = 6.2$  Hz, H-6 of  $\alpha$ -rha), 4.79 (1H, d,  $J = 7.5$  Hz, H-1 of  $\beta$ -xyl), 4.92 (1H, d,  $J = 7.6$  Hz, H-1 of  $\beta$ -glc), 5.45 (1H, m, H-12), 5.80 (1H, s, H-1 of  $\alpha$ -rha), 6.21 (1H, d,  $J = 8.1$  Hz, H-1

of  $\beta$ -glc).  $^{13}\text{C}$  NMR: Tables 1 and 2. FD-MS  $m/z$ : 1112 [ $\text{M} + \text{H} + \text{Na}$ ] $^+$ . Liebermann–Burchard reaction reddish purple.

*Acid hydrolysis of 1–3.* A soln of sample (20 mg) and 7% HCl–EtOH (1:1) was refluxed for 4 hr. The mixt. was diluted with  $\text{H}_2\text{O}$  and extracted with  $\text{Et}_2\text{O}$ . The  $\text{Et}_2\text{O}$  layer was evapd to dryness. The residue was recrystallized in MeOH to afford **4** (6 mg), needles, mp 300.5–301 [6],  $[\alpha]_D^{20} + 23.0$  (MeOH;  $c$  0.15). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3450 (OH), 1701 (O=C–OH), 1643 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.91, 1.00, 1.03, 1.26, 1.29, 1.55 (each 3H, s, Me), 5.54 (1H, *m*, H-12).  $^{13}\text{C}$  NMR: Table 1. EI-MS  $m/z$ : 486 [ $\text{M}$ ] $^+$ , 469 [ $\text{M} - \text{H}_2\text{O} + \text{H}$ ] $^+$ , 268, 233, 189. Liebermann–Burchard reaction reddish purple.

*Identification of sugars.* The aq. layer was neutralized with 1 N NaOH, concd, and subjected to HPTLC analysis on Kieselgel 60 F254 (Merck) [using  $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  (30:12:4) 9 ml and HOAc 1 ml] and PC [using *n*-BuOH–HOAc– $\text{H}_2\text{O}$  (4:1:5); phenol– $\text{H}_2\text{O}$  (4:1); *n*-BuOH– $\text{C}_5\text{H}_5\text{N}$ –benzene– $\text{H}_2\text{O}$  (5:3:1:3)], which showed Glc, Rha and Ara in **1** and **2**; Glc, Rha and Xyl in **3**.

*Alkaline hydrolysis of 1–3.* A mixt. of sample (25 mg) and 2% KOH in 70% EtOH (7 ml) was refluxed for 6 hr. After being slowly neutralized with 0.1 N HCl, the reaction mixt. was extracted with *n*-BuOH satd with  $\text{H}_2\text{O}$ . The *n*-BuOH soln was concd *in vacuo*. The residue showing a spot on TLC ( $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  80:10:1) was recrystallized to give **5** in **1** and **2**; **6** in **3**.

Compound **5**, powder, mp 250–252° (dec.).  $[\alpha]_D^{20} - 8.9$  (MeOH;  $c$  0.10). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400 (OH), 2915 (C–H), 1700 (O=C–OH), 1645 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.86, 0.93, 0.99, 1.24, 1.29, 1.57 (each 3H, s, Me), 4.74 (1H, *d*,  $J = 6.9$  Hz, H-1 of  $\alpha$ -ara), 5.52 (1H, *m*, H-12).  $^{13}\text{C}$  NMR: Tables 1 and 2. FAB-MS  $m/z$ : 657 [ $\text{M} + \text{K}$ ] $^+$ , Liebermann–Burchard reaction reddish purple.

Compound **6**, powder, mp 265–267° (dec.).  $[\alpha]_D^{20} + 23.8$  (MeOH  $c$  0.1). Analyt. calcd for  $\text{C}_{35}\text{H}_{54}\text{O}_9\cdot 3/\text{H}_2\text{O}$ : C 65.32, H 8.89, found C 65.69, H 8.64. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3440 (OH), 2950 (C–H), 1700 (O=C–OH), 1640 (C=C).  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.84, 0.97, 0.99, 1.28, 1.29, 1.57 (each 3H, s, Me), 4.80 (1H, *d*,  $J = 7.5$  Hz, H-1 of  $\beta$ -xyl), 5.53 (1H, *m*, H-12).  $^{13}\text{C}$  NMR: Tables 1 and 2. FAB-MS  $m/z$ : 657 [ $\text{M} + \text{K}$ ] $^+$ . Liebermann–Burchard reaction reddish purple.

*Acetylation of 1–3.* A soln of sample (10 mg) in a mixt. of  $\text{Ac}_2\text{O}$  (0.4 ml) and pyridine (0.4 ml) was allowed to stand at room temp., and the mixt. was worked-up as usual to give the peracetate of **1** (6 mg, **2** gave the same peracetate as that of **1**), a powder (EtOH), 157–158° (dec.), EI-MS  $m/z$ : 561 [(glc-rha) $\text{Ac}_6$ ] $^+$ , 273 [(rha) $\text{Ac}_3$ ] $^+$ , 259 [(ara) $\text{Ac}_3$ ] $^+$ ; peracetate of **3** (7 mg), a powder (EtOH), mp 153–154° (dec.); EI-MS  $m/z$ : 561 [(glc-rha) $\text{Ac}_6$ ] $^+$ , 273 [(rha) $\text{Ac}_3$ ] $^+$ , 259 [(xyl) $\text{Ac}_3$ ] $^+$ .

*Formation of the bromolactone* [6]. Compound **1** (21 mg) was methylated with  $\text{CH}_2\text{N}_2$  in MeOH, and the product **7** was dissolved in 1 N NaOH (9 ml) and heated

at 80° for 2 hr. After cooling, the reaction mixt. was neutralized with 2 N  $\text{H}_2\text{SO}_4$  and passed through RA highly porous polymer (Seventh Chemical and Industrial Factory of Beijing) eluting with  $\text{H}_2\text{O}$  and MeOH. The MeOH soln was concd to give **8**. AcOH (6 ml) containing  $\text{Br}_2$  (0.3 ml) was added to a soln of **8** and AcONa (30 mg) in 90% AcOH (3 ml) and the mixt. was stirred at room temp. for 15 hr under a  $\text{N}_2$  stream. The reaction mixt. was poured into an aq. soln with  $\text{NaHSO}_3$  and extracted with  $\text{CHCl}_3$ . The  $\text{CHCl}_3$  layer was evapd to dryness and the residue was purified by silica gel CC ( $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$ , 80:10:1) to give **9** (6 mg) as a powder, IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3450 (OH), 1768 (five-membered ring lactone), 1724 (COOMe),  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.82, 1.00, 1.20, 1.25, 1.28, 1.48 (each 3H, s, Me), 3.64 (3H, s, H-29), 4.26 (1H, *m*, H-12). When subjected to the same procedure, **2** also afforded **9** (6.5 mg) and **3** gave **10** (7 mg), a powder, IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3450 (OH), 1765 (five-membered ring lactone), 1720 (COOMe);  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.82, 1.00, 1.21, 1.24, 1.28, 1.47 (each 3H, s, Me), 3.61 (3H, s, H-29), 4.23 (1H, *m*, H-12).

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